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## 2-(4-Bromophenyl)acetohydrazide

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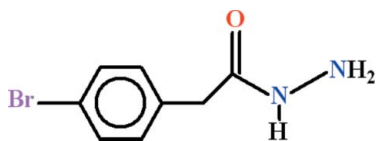
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Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.023;  $wR$  factor = 0.059; data-to-parameter ratio = 15.8.

In the title compound,  $\text{C}_8\text{H}_9\text{BrN}_2\text{O}$ , the 1-bromo-4-methylbenzene group and the formic hydrazide moiety [r.m.s. deviations of 0.0129 and 0.0038 Å] are oriented at a dihedral angle of 80.66 (11)°. In the crystal, molecules are linked *via* strong  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds, leading to the formation of chains in the [010] direction. These chains are linked *via* weaker  $\text{N}-\text{H}\cdots\text{N}$  and  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds, with  $R_2^2(7)$  and  $R_3^2(7)$  ring motifs, forming a two-dimensional network parallel to (001).

## Related literature

For background literature and the crystal structure of 2-chlorobenzohydrazide, see: Ahmad *et al.* (2012). For graph-set notation, see: Bernstein *et al.* (1995).



## Experimental

## Crystal data

$\text{C}_8\text{H}_9\text{BrN}_2\text{O}$	$V = 441.88$ (2) Å <sup>3</sup>
$M_r = 229.07$	$Z = 2$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation
$a = 6.0798$ (2) Å	$\mu = 4.60$ mm <sup>-1</sup>
$b = 4.8565$ (1) Å	$T = 296$ K
$c = 15.1126$ (5) Å	$0.36 \times 0.23 \times 0.22$ mm
$\beta = 98.003$ (2)°	

## Data collection

Bruker Kappa APEXII CCD diffractometer	4331 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2009)	1814 independent reflections
$T_{\min} = 0.298$ , $T_{\max} = 0.366$	1677 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.023$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.023$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.059$	$\Delta\rho_{\text{max}} = 0.44$ e Å <sup>-3</sup>
$S = 1.06$	$\Delta\rho_{\text{min}} = -0.47$ e Å <sup>-3</sup>
1814 reflections	Absolute structure: Flack (1983), 583 Friedel pairs
115 parameters	Flack parameter: 0.007 (11)
1 restraint	

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^{\text{i}}$	0.86	2.02	2.863 (3)	165
$\text{N2}-\text{H2A}\cdots\text{N2}^{\text{ii}}$	0.84 (4)	2.37 (4)	3.192 (4)	167 (3)
$\text{N2}-\text{H2B}\cdots\text{O1}^{\text{iii}}$	0.73 (3)	2.59 (4)	3.230 (3)	147 (4)

 Symmetry codes: (i)  $x, y + 1, z$ ; (ii)  $-x - 1, y - \frac{1}{2}, -z + 1$ ; (iii)  $-x, y + \frac{1}{2}, -z + 1$ .

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2009); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2429).

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## supporting information

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## 2-(4-Bromophenyl)acetohydrazide

Shakeel Ahmad, Abdul Jabbar, Muhammad Tahir Hussain and M. Nawaz Tahir

### S1. Comment

Recently, we have reported the crystal structure of 2-chlorobenzohydrazide (Ahmad *et al.*, 2012). In continuation of this work we have synthesized the title compound, a hydrazide derivative, and report herein on its crystal structure.

In the title molecule, Fig. 1, the 1-bromo-4-methylbenzene group A (C1–C7/Br1) and the formic hydrazide moiety B (O1/C8/N1/N2) are planar with r. m. s. deviations of 0.0129 Å and 0.0038 Å, respectively. The dihedral angle between these mean planes, A/B, is 80.66 (11)°.

In the crystal, molecules are linked via N—H···O hydrogen bonds to form one-dimensional polymeric chains along [010]. These chains are linked via N—H···N and N—H···O hydrogen bonds to form a two-dimensional polymeric network in (001). The hydrogen bonds give rise to  $R_2^2(7)$  and  $R_3^3(7)$  ring motifs (Bernstein *et al.*, 1995; Table 1 and Fig. 2).

### S2. Experimental

2-(4-Bromophenyl)acetic acid (4.42 g, 0.022 mol) was converted to methyl 2-(4-bromophenyl)acetate by refluxing in methanol (25 ml) in the presence of catalytic amount of sulfuric acid. This ester was then converted into the title compound by refluxing with hydrazine hydrate (80%, 10 ml) in dry methanol. The title compound was purified by recrystallization from dry methanol, giving colourless rod-like crystals [M.p. 438–439 K].

### S3. Refinement

The coordinates of the H-atoms of the NH<sub>2</sub> group were refined with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$ . The remainder of the H-atoms were included in calculated positions and treated as riding atoms: N—H = 0.86 Å, C—H = 0.93–0.97 Å, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N}, \text{C})$ .

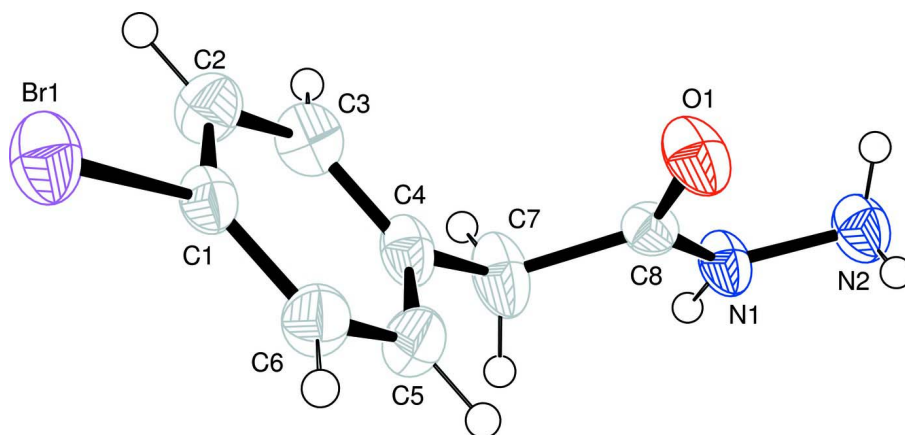


Figure 1

View of the title molecule with the atom numbering. The displacement ellipsoids are drawn at the 50% probability level.

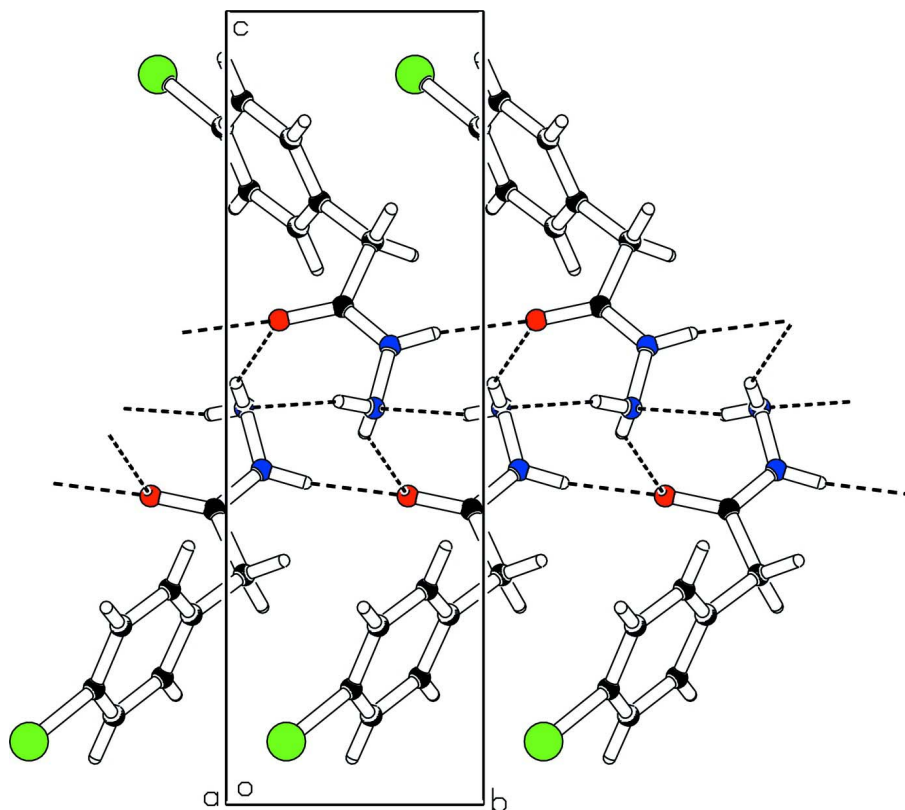


Figure 2

A view along the *a* axis of the crystal packing of the title compound. The two-dimensional hydrogen bonded network extends in the plane (001). Hydrogen bonds are shown as dashed lines - see Table 1 for details.

## 2-(4-Bromophenyl)acetohydrazide

### Crystal data

$C_8H_9BrN_2O$   
 $M_r = 229.07$   
 Monoclinic,  $P2_1$   
 Hall symbol:  $P\ 2_1yb$   
 $a = 6.0798\ (2)\ \text{\AA}$   
 $b = 4.8565\ (1)\ \text{\AA}$   
 $c = 15.1126\ (5)\ \text{\AA}$   
 $\beta = 98.003\ (2)^\circ$   
 $V = 441.88\ (2)\ \text{\AA}^3$   
 $Z = 2$

$F(000) = 228$   
 $D_x = 1.722\ \text{Mg m}^{-3}$   
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$   
 Cell parameters from 1677 reflections  
 $\theta = 2.7\text{--}28.3^\circ$   
 $\mu = 4.60\ \text{mm}^{-1}$   
 $T = 296\ \text{K}$   
 Rod, colourless  
 $0.36 \times 0.23 \times 0.22\ \text{mm}$

### Data collection

Bruker Kappa APEXII CCD  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 Detector resolution:  $7.50\ \text{pixels mm}^{-1}$   
 $\omega$  scans

Absorption correction: multi-scan  
 (*SADABS*; Bruker, 2009)  
 $T_{\min} = 0.298$ ,  $T_{\max} = 0.366$   
 4331 measured reflections  
 1814 independent reflections  
 1677 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$   
 $\theta_{\text{max}} = 28.3^\circ$ ,  $\theta_{\text{min}} = 2.7^\circ$   
 $h = -8 \rightarrow 8$

$k = -6 \rightarrow 4$   
 $l = -20 \rightarrow 20$

### Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.023$   
 $wR(F^2) = 0.059$   
 $S = 1.06$   
 1814 reflections  
 115 parameters  
 1 restraint  
 Primary atom site location: structure-invariant  
 direct methods  
 Secondary atom site location: difference Fourier  
 map

Hydrogen site location: inferred from  
 neighbouring sites  
 H atoms treated by a mixture of independent  
 and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.024P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.44 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.47 \text{ e } \text{\AA}^{-3}$   
 Absolute structure: Flack (1983), 583 Friedel  
 pairs  
 Absolute structure parameter: 0.007 (11)

### Special details

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
Br1	0.42290 (4)	0.23371 (7)	0.07962 (1)	0.0435 (1)
O1	-0.1520 (3)	0.7065 (5)	0.38909 (13)	0.0457 (6)
N1	-0.2370 (4)	1.1385 (4)	0.42196 (15)	0.0364 (7)
N2	-0.3296 (5)	1.0741 (6)	0.49999 (17)	0.0420 (8)
C1	0.2739 (4)	0.4860 (5)	0.14603 (16)	0.0315 (8)
C2	0.0587 (5)	0.5632 (6)	0.11262 (19)	0.0395 (9)
C3	-0.0499 (4)	0.7519 (8)	0.16041 (16)	0.0407 (8)
C4	0.0518 (5)	0.8638 (5)	0.23979 (17)	0.0362 (8)
C5	0.2669 (5)	0.7798 (7)	0.27159 (17)	0.0416 (12)
C6	0.3788 (5)	0.5900 (6)	0.22578 (17)	0.0381 (8)
C7	-0.0693 (6)	1.0694 (6)	0.2896 (2)	0.0494 (10)
C8	-0.1528 (4)	0.9529 (5)	0.37211 (16)	0.0289 (7)
H1	-0.23523	1.30826	0.40587	0.0437*
H2	-0.01156	0.49007	0.05918	0.0474*
H2A	-0.416 (5)	0.940 (8)	0.491 (2)	0.0503*
H2B	-0.230 (6)	1.040 (8)	0.531 (2)	0.0503*
H3	-0.19442	0.80410	0.13848	0.0488*
H5	0.33752	0.85271	0.32499	0.0500*
H6	0.52191	0.53407	0.24842	0.0457*
H7A	-0.19487	1.14063	0.24952	0.0591*
H7B	0.02941	1.22250	0.30749	0.0591*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Br1	0.0503 (2)	0.0403 (2)	0.0433 (1)	0.0101 (2)	0.0184 (1)	-0.0001 (2)
O1	0.0663 (11)	0.0214 (11)	0.0547 (10)	0.0035 (12)	0.0269 (9)	0.0049 (11)
N1	0.0533 (14)	0.0226 (11)	0.0367 (11)	0.0007 (9)	0.0184 (10)	0.0019 (8)
N2	0.0511 (16)	0.0378 (14)	0.0407 (14)	0.0000 (12)	0.0193 (12)	-0.0020 (11)
C1	0.0367 (13)	0.0284 (14)	0.0315 (12)	0.0012 (10)	0.0123 (10)	0.0007 (10)
C2	0.0393 (14)	0.0426 (17)	0.0356 (13)	0.0034 (12)	0.0021 (12)	-0.0016 (12)
C3	0.0373 (11)	0.0407 (16)	0.0442 (12)	0.0084 (16)	0.0063 (10)	0.0062 (17)
C4	0.0523 (16)	0.0247 (13)	0.0352 (13)	0.0029 (11)	0.0185 (12)	0.0034 (10)
C5	0.0488 (14)	0.044 (3)	0.0321 (11)	-0.0038 (14)	0.0061 (11)	-0.0051 (12)
C6	0.0344 (14)	0.0440 (16)	0.0355 (13)	0.0028 (12)	0.0040 (12)	0.0009 (12)
C7	0.077 (2)	0.0298 (16)	0.0479 (17)	0.0083 (15)	0.0319 (16)	0.0057 (13)
C8	0.0312 (12)	0.0221 (13)	0.0338 (12)	0.0001 (10)	0.0064 (10)	-0.0008 (9)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

Br1—C1	1.893 (2)	C4—C5	1.390 (4)
O1—C8	1.224 (3)	C4—C7	1.503 (4)
N1—N2	1.411 (4)	C5—C6	1.387 (4)
N1—C8	1.323 (3)	C7—C8	1.520 (4)
N1—H1	0.8600	C2—H2	0.9300
N2—H2B	0.73 (3)	C3—H3	0.9300
N2—H2A	0.84 (4)	C5—H5	0.9300
C1—C2	1.387 (4)	C6—H6	0.9300
C1—C6	1.379 (4)	C7—H7A	0.9700
C2—C3	1.389 (4)	C7—H7B	0.9700
C3—C4	1.382 (4)		
N2—N1—C8	123.8 (2)	O1—C8—C7	123.0 (2)
C8—N1—H1	118.00	N1—C8—C7	114.4 (2)
N2—N1—H1	118.00	O1—C8—N1	122.5 (2)
N1—N2—H2B	101 (3)	C1—C2—H2	121.00
H2A—N2—H2B	112 (4)	C3—C2—H2	121.00
N1—N2—H2A	111 (2)	C2—C3—H3	119.00
Br1—C1—C2	118.61 (19)	C4—C3—H3	119.00
Br1—C1—C6	120.2 (2)	C4—C5—H5	119.00
C2—C1—C6	121.2 (2)	C6—C5—H5	119.00
C1—C2—C3	118.8 (2)	C1—C6—H6	121.00
C2—C3—C4	121.5 (3)	C5—C6—H6	121.00
C3—C4—C7	120.3 (3)	C4—C7—H7A	109.00
C3—C4—C5	118.1 (3)	C4—C7—H7B	109.00
C5—C4—C7	121.6 (3)	C8—C7—H7A	109.00
C4—C5—C6	121.7 (3)	C8—C7—H7B	109.00
C1—C6—C5	118.7 (3)	H7A—C7—H7B	108.00
C4—C7—C8	114.0 (2)		

N2—N1—C8—O1	1.3 (4)	C2—C3—C4—C7	179.5 (3)
N2—N1—C8—C7	178.1 (3)	C3—C4—C5—C6	0.1 (4)
Br1—C1—C2—C3	-178.9 (2)	C7—C4—C5—C6	179.9 (3)
C6—C1—C2—C3	1.0 (4)	C3—C4—C7—C8	104.7 (3)
Br1—C1—C6—C5	178.4 (2)	C5—C4—C7—C8	-75.1 (4)
C2—C1—C6—C5	-1.5 (4)	C4—C5—C6—C1	1.0 (4)
C1—C2—C3—C4	0.2 (5)	C4—C7—C8—O1	-10.6 (4)
C2—C3—C4—C5	-0.7 (4)	C4—C7—C8—N1	172.7 (2)

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...O1 <sup>i</sup>	0.86	2.02	2.863 (3)	165
N2—H2A...N2 <sup>ii</sup>	0.84 (4)	2.37 (4)	3.192 (4)	167 (3)
N2—H2B...O1 <sup>iii</sup>	0.73 (3)	2.59 (4)	3.230 (3)	147 (4)

Symmetry codes: (i)  $x, y+1, z$ ; (ii)  $-x-1, y-1/2, -z+1$ ; (iii)  $-x, y+1/2, -z+1$ .