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2-(Methoxyimino)-2-[2-[(2-methylphenoxy)methyl]phenyl]acetohydrazide

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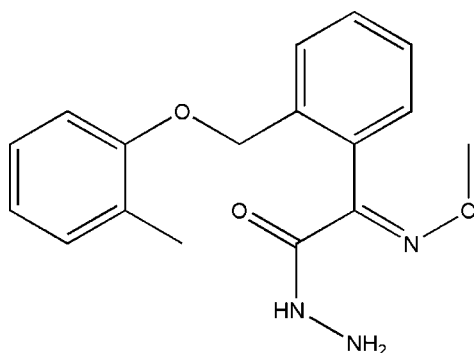
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.105; data-to-parameter ratio = 14.3.

In the title molecule, $\text{C}_{17}\text{H}_{19}\text{N}_3\text{O}_3$, the dihedral angle between the two benzene rings is $57.17(5)^\circ$. In the crystal, $\text{N}-\text{H}\cdots\text{O}$ and $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds connect molecules to form chains along $[001]$. In addition, a weak $\text{C}-\text{H}\cdots\pi$ interaction is observed.

Related literature

For the biological activities of kresoxim-methyl {methyl 2(*E*)-methoxyimino-2-[2-(2-tolyloxymethyl)phenyl] acetate}, which is a starting material in the synthesis of the title compound, see: Anke *et al.* (1977); Balba (2007); Ichinari *et al.* (1999); Grossmann & Retzlaff (1997); Ypema (1998). For the crystal structure of kresoxim-methyl, see: Chopra *et al.* (2004).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{19}\text{N}_3\text{O}_3$
 $M_r = 313.35$
 Monoclinic, $C2/c$
 $a = 21.4015(6)$ Å
 $b = 20.7277(4)$ Å

$c = 7.6975(2)$ Å
 $\beta = 109.103(3)^\circ$
 $V = 3226.60(14)$ Å³
 $Z = 8$
 Mo $K\alpha$ radiation

$\mu = 0.09$ mm⁻¹
 $T = 293$ K

 $0.3 \times 0.2 \times 0.2$ mm

Data collection

Oxford Diffraction Xcalibur
 Sapphire3 diffractometer
 Absorption correction: multi-scan
 (*CrysAlis PRO*; Oxford
 Diffraction, 2010)
 $T_{\min} = 0.906$, $T_{\max} = 1.000$

55120 measured reflections
 3174 independent reflections
 2532 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.105$
 $S = 1.02$
 3174 reflections
 222 parameters
 3 restraints

H atoms treated by a mixture of
 independent and constrained
 refinement
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.17$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg refers to the centroid of the C8–C13 phenyl ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N7}-\text{H72}\cdots\text{O1}^i$	0.86 (2)	2.40 (2)	3.220 (2)	158 (2)
$\text{N6}-\text{H61}\cdots\text{N7}^{ii}$	0.87 (2)	2.34 (1)	3.191 (2)	166 (1)
$\text{C5}-\text{H5B}\cdots\text{Cg}^{iii}$	0.96	2.69	3.461 (2)	138

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $x, -y + 1, z + \frac{1}{2}$; (iii) $-x + \frac{1}{2}, -y + \frac{1}{2}, -z$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5499).

References

- Anke, T., Oberwinkler, F., Steglich, W. & Schramm, G. (1977). *J. Antibiot.* **30**, 806–810.
 Balba, H. (2007). *J. Environ. Sci. Health Part B*, **42**, 441–451.
 Chopra, D., Mohan, T. P., Rao, K. S. & Guru Row, T. N. (2004). *Acta Cryst.* **E60**, o2421–o2423.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Grossmann, N. & Retzlaff, G. (1997). *Pestic. Sci.* **50**, 11–20.
 Ichinari, M., Masuko, M., Takenaka, H., Hasegawa, R., Ichiba, T., Hayase, Y. & Takeda, R. (1999). *Pestic. Sci.* **55**, 347–349.
 Oxford Diffraction (2010). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, Oxfordshire, England.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
 Ypema, H. L. (1998). *Plant Dis.* **83**, 4–17.

supporting information

Acta Cryst. (2012). E68, o2426 [https://doi.org/10.1107/S160053681203070X]

2-(Methoxyimino)-2-{2-[(2-methylphenoxy)methyl]phenyl}acetohydrazide**Rajni Kant, Vivek K. Gupta, Kamini Kapoor, Chetan S. Shripanavar and Kaushik Banerjee****S1. Comment**

Kresoxim-methyl is an active agrochemical (Chopra *et al.*, 2004), strobilurin fungicide (Anke *et al.*, 1977), with broad spectrum biological activity (Ypema, 1998; Ichinari *et al.*, 1999; Grossmann & Retzlaff 1997). This type of compound is easily metabolized in nature as well as in living systems, and that is the reason studies on their fate in soil, plants and animal systems (Balba, 2007) are very important. Herein, we present the crystal structure of the title compound (I) which was synthesized from kresoxim-methyl.

In (I) (Fig. 1), all bond lengths and angles are normal and correspond to those observed in the related structure (Chopra *et al.*, 2004). The dihedral angles formed by the two benzene rings (C8-C13/C16-C21) is 57.17 (5)°. In the crystal, molecules are connected by N—H···O and N—H···N hydrogen bonds into chains along [001] (Fig. 2). In addition a weak C—H··· π interaction is observed.

S2. Experimental

Kresoxim-methyl (0.313 g, 0.001 mol) was dissolved in 5 ml methanol and to it hydrazine hydrate (0.1 g, 0.002 mol) solution was added and refluxed at 343K for 1 h. The reaction mixture was then cooled and solvent was removed under reduced pressure to give a solid product. The compound was dissolved in methanol, and by the process of slow evaporation this crystalline compound was separated out, m.p. 397K.

S3. Refinement

H atoms bonded to N atoms were located in a difference map and refined independently with the constraint of N—H = 0.86 (1) Å. Other H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C—H distances of 0.93–0.97 Å and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

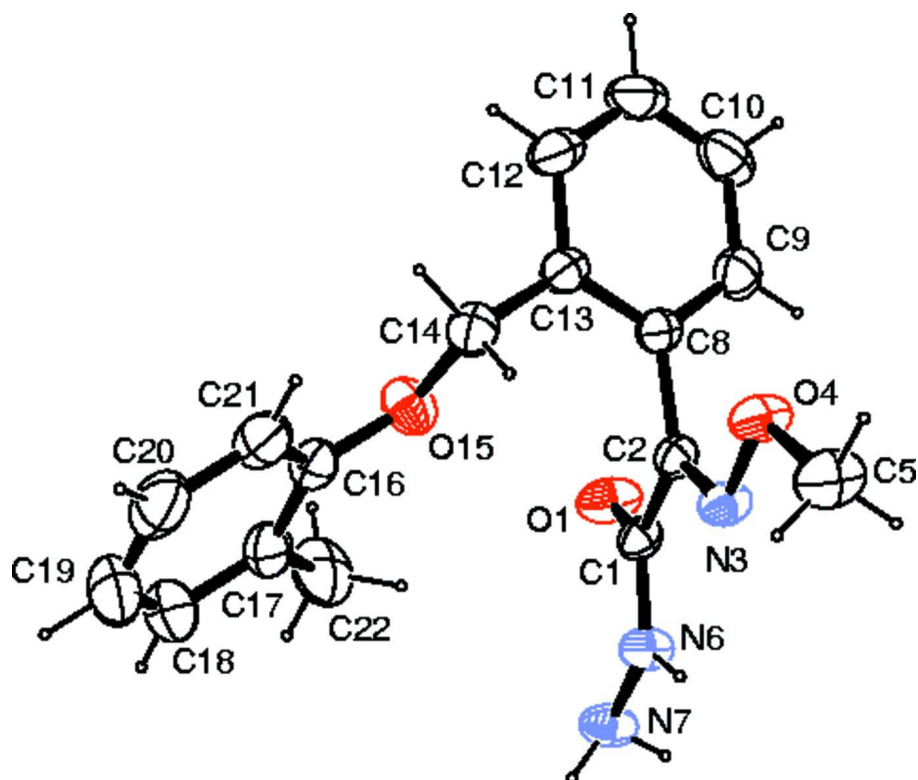


Figure 1

The molecular structure of (I) with ellipsoids drawn at the 40% probability level. H atoms are shown as small spheres of arbitrary radii.

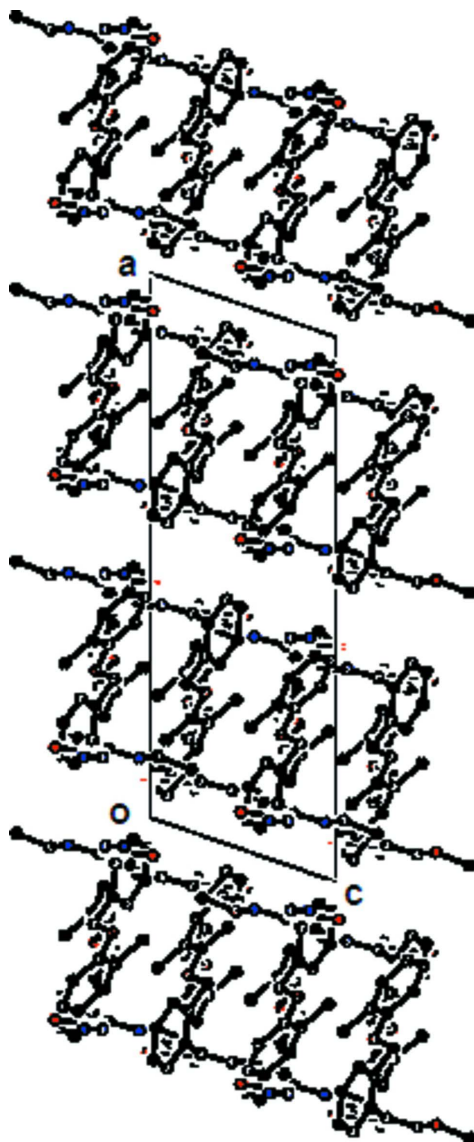


Figure 2

Part of the crystal structure viewed along the *b* axis.

2-(Methoxyimino)-2-{2-[(2-methylphenoxy)methyl]phenyl}acetohydrazide

Crystal data

$C_{17}H_{19}N_3O_3$

$M_r = 313.35$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 21.4015 (6) \text{ \AA}$

$b = 20.7277 (4) \text{ \AA}$

$c = 7.6975 (2) \text{ \AA}$

$\beta = 109.103 (3)^\circ$

$V = 3226.60 (14) \text{ \AA}^3$

$Z = 8$

$F(000) = 1328$

$D_x = 1.290 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 21236 reflections

$\theta = 3.5\text{--}29.1^\circ$

$\mu = 0.09 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Block, white

$0.3 \times 0.2 \times 0.2 \text{ mm}$

Data collection

Oxford Diffraction Xcalibur Sapphire3
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 16.1049 pixels mm⁻¹
 ω scan
Absorption correction: multi-scan
(*CrysAlis PRO*; Oxford Diffraction, 2010)
 $T_{\min} = 0.906$, $T_{\max} = 1.000$

55120 measured reflections
3174 independent reflections
2532 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$
 $\theta_{\max} = 26.0^\circ$, $\theta_{\min} = 3.5^\circ$
 $h = -26 \rightarrow 26$
 $k = -25 \rightarrow 25$
 $l = -9 \rightarrow 9$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.105$
 $S = 1.02$
3174 reflections
222 parameters
3 restraints
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H atoms treated by a mixture of independent
and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0448P)^2 + 1.9045P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. *CrysAlis PRO*, Oxford Diffraction Ltd., Version 1.171.34.40 (release 27-08-2010 *CrysAlis171.NET*) (compiled Aug 27 2010, 11:50:40) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N3	0.10242 (6)	0.35418 (5)	0.94034 (15)	0.0383 (3)
N6	0.08078 (6)	0.46145 (6)	0.74149 (16)	0.0391 (3)
N7	0.06911 (8)	0.51996 (6)	0.64119 (18)	0.0453 (3)
O15	0.23561 (6)	0.35236 (5)	0.70574 (15)	0.0510 (3)
O4	0.11743 (6)	0.29666 (5)	1.03992 (13)	0.0466 (3)
O1	0.06825 (6)	0.40146 (5)	0.48718 (13)	0.0521 (3)
C17	0.28955 (8)	0.44911 (8)	0.6901 (2)	0.0519 (4)
C18	0.34451 (11)	0.48795 (10)	0.7505 (3)	0.0694 (6)
H18	0.3437	0.5282	0.6965	0.083*
C19	0.40054 (11)	0.46909 (12)	0.8882 (3)	0.0808 (7)
H19	0.4372	0.4961	0.9252	0.097*
C20	0.40213 (10)	0.41013 (11)	0.9710 (3)	0.0706 (6)

H20	0.4400	0.3972	1.0645	0.085*
C21	0.34756 (8)	0.36974 (9)	0.9158 (2)	0.0546 (4)
H21	0.3484	0.3300	0.9728	0.066*
C16	0.29186 (8)	0.38918 (8)	0.7750 (2)	0.0444 (4)
C22	0.22894 (10)	0.47007 (10)	0.5382 (3)	0.0728 (6)
H22A	0.2339	0.5142	0.5069	0.109*
H22B	0.1910	0.4662	0.5780	0.109*
H22C	0.2231	0.4432	0.4324	0.109*
C14	0.23124 (8)	0.29575 (7)	0.8055 (2)	0.0453 (4)
H14A	0.2306	0.3074	0.9270	0.054*
H14B	0.2695	0.2685	0.8198	0.054*
C13	0.16963 (7)	0.26008 (7)	0.70408 (19)	0.0382 (3)
C12	0.17407 (9)	0.20035 (8)	0.6249 (2)	0.0523 (4)
H12	0.2155	0.1838	0.6353	0.063*
C11	0.11869 (11)	0.16554 (8)	0.5321 (3)	0.0603 (5)
H11	0.1228	0.1257	0.4814	0.072*
C10	0.05738 (10)	0.18966 (9)	0.5143 (2)	0.0593 (5)
H10	0.0198	0.1663	0.4507	0.071*
C9	0.05121 (8)	0.24886 (8)	0.5911 (2)	0.0473 (4)
H9	0.0094	0.2650	0.5783	0.057*
C8	0.10689 (7)	0.28413 (6)	0.68670 (18)	0.0347 (3)
C2	0.09848 (7)	0.34665 (6)	0.77269 (18)	0.0320 (3)
C5	0.11414 (11)	0.30815 (9)	1.2204 (2)	0.0608 (5)
H5A	0.1481	0.3382	1.2840	0.091*
H5B	0.1205	0.2683	1.2873	0.091*
H5C	0.0716	0.3257	1.2106	0.091*
C1	0.08095 (7)	0.40622 (7)	0.65305 (18)	0.0334 (3)
H61	0.0840 (8)	0.4629 (7)	0.8568 (13)	0.041 (4)*
H71	0.0964 (9)	0.5477 (8)	0.707 (3)	0.081 (7)*
H72	0.0295 (6)	0.5333 (11)	0.627 (3)	0.090 (8)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N3	0.0523 (7)	0.0327 (6)	0.0321 (6)	0.0099 (5)	0.0169 (5)	0.0061 (5)
N6	0.0587 (8)	0.0320 (6)	0.0273 (6)	0.0039 (5)	0.0149 (6)	0.0027 (5)
N7	0.0656 (10)	0.0306 (7)	0.0381 (7)	0.0039 (7)	0.0147 (7)	0.0031 (5)
O15	0.0515 (7)	0.0461 (6)	0.0510 (7)	-0.0027 (5)	0.0109 (5)	0.0143 (5)
O4	0.0752 (8)	0.0367 (6)	0.0327 (5)	0.0146 (5)	0.0244 (5)	0.0094 (4)
O1	0.0866 (9)	0.0427 (6)	0.0291 (5)	0.0151 (6)	0.0215 (5)	0.0035 (4)
C17	0.0549 (10)	0.0522 (10)	0.0582 (10)	-0.0013 (8)	0.0317 (8)	0.0044 (8)
C18	0.0698 (14)	0.0621 (12)	0.0891 (15)	-0.0130 (10)	0.0435 (12)	-0.0013 (11)
C19	0.0598 (13)	0.0881 (16)	0.1030 (18)	-0.0239 (12)	0.0382 (13)	-0.0213 (14)
C20	0.0448 (11)	0.0892 (16)	0.0760 (13)	0.0057 (10)	0.0172 (9)	-0.0183 (12)
C21	0.0504 (10)	0.0566 (10)	0.0577 (10)	0.0075 (8)	0.0188 (8)	-0.0050 (8)
C16	0.0443 (9)	0.0451 (9)	0.0491 (9)	0.0019 (7)	0.0226 (7)	-0.0006 (7)
C22	0.0763 (14)	0.0638 (12)	0.0795 (14)	-0.0005 (10)	0.0271 (11)	0.0313 (10)
C14	0.0445 (9)	0.0439 (9)	0.0489 (9)	0.0101 (7)	0.0171 (7)	0.0141 (7)

C13	0.0481 (9)	0.0341 (7)	0.0373 (7)	0.0075 (6)	0.0209 (6)	0.0088 (6)
C12	0.0712 (12)	0.0402 (9)	0.0566 (10)	0.0157 (8)	0.0360 (9)	0.0073 (7)
C11	0.0990 (16)	0.0358 (9)	0.0598 (11)	-0.0025 (9)	0.0446 (11)	-0.0069 (8)
C10	0.0804 (13)	0.0514 (10)	0.0523 (10)	-0.0246 (9)	0.0303 (9)	-0.0129 (8)
C9	0.0503 (9)	0.0516 (9)	0.0451 (9)	-0.0055 (7)	0.0225 (7)	-0.0034 (7)
C8	0.0457 (8)	0.0328 (7)	0.0299 (7)	0.0028 (6)	0.0184 (6)	0.0041 (5)
C2	0.0339 (7)	0.0331 (7)	0.0307 (7)	0.0043 (5)	0.0130 (6)	0.0027 (5)
C5	0.0999 (15)	0.0554 (10)	0.0335 (8)	0.0123 (10)	0.0307 (9)	0.0089 (7)
C1	0.0367 (7)	0.0359 (7)	0.0297 (7)	0.0058 (6)	0.0136 (6)	0.0028 (5)

Geometric parameters (Å, °)

N3—C2	1.2750 (17)	C22—H22A	0.9600
N3—O4	1.3968 (14)	C22—H22B	0.9600
N6—C1	1.3326 (17)	C22—H22C	0.9600
N6—N7	1.4153 (17)	C14—C13	1.490 (2)
N6—H61	0.868 (9)	C14—H14A	0.9700
N7—H71	0.858 (10)	C14—H14B	0.9700
N7—H72	0.864 (10)	C13—C12	1.397 (2)
O15—C16	1.3767 (19)	C13—C8	1.397 (2)
O15—C14	1.4222 (17)	C12—C11	1.372 (3)
O4—C5	1.4335 (18)	C12—H12	0.9300
O1—C1	1.2192 (16)	C11—C10	1.369 (3)
C17—C18	1.375 (3)	C11—H11	0.9300
C17—C16	1.397 (2)	C10—C9	1.387 (2)
C17—C22	1.498 (3)	C10—H10	0.9300
C18—C19	1.372 (3)	C9—C8	1.387 (2)
C18—H18	0.9300	C9—H9	0.9300
C19—C20	1.374 (3)	C8—C2	1.4926 (18)
C19—H19	0.9300	C2—C1	1.5123 (18)
C20—C21	1.386 (3)	C5—H5A	0.9600
C20—H20	0.9300	C5—H5B	0.9600
C21—C16	1.383 (2)	C5—H5C	0.9600
C21—H21	0.9300		
C2—N3—O4	112.21 (11)	C13—C14—H14A	109.8
C1—N6—N7	119.14 (11)	O15—C14—H14B	109.8
C1—N6—H61	122.7 (10)	C13—C14—H14B	109.8
N7—N6—H61	117.9 (10)	H14A—C14—H14B	108.3
N6—N7—H71	106.7 (15)	C12—C13—C8	118.36 (15)
N6—N7—H72	109.5 (16)	C12—C13—C14	119.56 (14)
H71—N7—H72	108 (2)	C8—C13—C14	122.07 (13)
C16—O15—C14	116.85 (12)	C11—C12—C13	121.52 (16)
N3—O4—C5	108.33 (11)	C11—C12—H12	119.2
C18—C17—C16	117.83 (17)	C13—C12—H12	119.2
C18—C17—C22	121.37 (17)	C10—C11—C12	119.78 (15)
C16—C17—C22	120.80 (16)	C10—C11—H11	120.1
C19—C18—C17	122.0 (2)	C12—C11—H11	120.1

C19—C18—H18	119.0	C11—C10—C9	120.17 (17)
C17—C18—H18	119.0	C11—C10—H10	119.9
C18—C19—C20	119.6 (2)	C9—C10—H10	119.9
C18—C19—H19	120.2	C10—C9—C8	120.50 (16)
C20—C19—H19	120.2	C10—C9—H9	119.7
C19—C20—C21	120.3 (2)	C8—C9—H9	119.7
C19—C20—H20	119.8	C9—C8—C13	119.66 (13)
C21—C20—H20	119.8	C9—C8—C2	119.06 (13)
C16—C21—C20	119.22 (18)	C13—C8—C2	121.27 (13)
C16—C21—H21	120.4	N3—C2—C8	125.28 (12)
C20—C21—H21	120.4	N3—C2—C1	115.94 (11)
O15—C16—C21	124.22 (15)	C8—C2—C1	118.73 (11)
O15—C16—C17	114.76 (14)	O4—C5—H5A	109.5
C21—C16—C17	121.01 (16)	O4—C5—H5B	109.5
C17—C22—H22A	109.5	H5A—C5—H5B	109.5
C17—C22—H22B	109.5	O4—C5—H5C	109.5
H22A—C22—H22B	109.5	H5A—C5—H5C	109.5
C17—C22—H22C	109.5	H5B—C5—H5C	109.5
H22A—C22—H22C	109.5	O1—C1—N6	124.57 (13)
H22B—C22—H22C	109.5	O1—C1—C2	119.84 (12)
O15—C14—C13	109.25 (12)	N6—C1—C2	115.59 (11)
O15—C14—H14A	109.8		
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C2—N3—O4—C5	-174.13 (14)	C12—C11—C10—C9	-0.5 (3)
C16—C17—C18—C19	0.7 (3)	C11—C10—C9—C8	-0.1 (2)
C22—C17—C18—C19	-179.34 (19)	C10—C9—C8—C13	0.7 (2)
C17—C18—C19—C20	-0.9 (3)	C10—C9—C8—C2	-178.03 (13)
C18—C19—C20—C21	0.1 (3)	C12—C13—C8—C9	-0.6 (2)
C19—C20—C21—C16	0.8 (3)	C14—C13—C8—C9	-179.88 (13)
C14—O15—C16—C21	10.5 (2)	C12—C13—C8—C2	178.12 (12)
C14—O15—C16—C17	-170.36 (13)	C14—C13—C8—C2	-1.2 (2)
C20—C21—C16—O15	178.15 (15)	O4—N3—C2—C8	1.5 (2)
C20—C21—C16—C17	-1.0 (2)	O4—N3—C2—C1	178.76 (11)
C18—C17—C16—O15	-178.97 (15)	C9—C8—C2—N3	96.60 (18)
C22—C17—C16—O15	1.1 (2)	C13—C8—C2—N3	-82.08 (18)
C18—C17—C16—C21	0.2 (2)	C9—C8—C2—C1	-80.61 (16)
C22—C17—C16—C21	-179.71 (17)	C13—C8—C2—C1	100.71 (15)
C16—O15—C14—C13	-176.27 (12)	N7—N6—C1—O1	-4.0 (2)
O15—C14—C13—C12	111.95 (15)	N7—N6—C1—C2	175.93 (13)
O15—C14—C13—C8	-68.73 (17)	N3—C2—C1—O1	-170.98 (14)
C8—C13—C12—C11	-0.1 (2)	C8—C2—C1—O1	6.5 (2)
C14—C13—C12—C11	179.25 (14)	N3—C2—C1—N6	9.09 (19)
C13—C12—C11—C10	0.6 (3)	C8—C2—C1—N6	-173.44 (13)

Hydrogen-bond geometry (Å, °)

Please define Cg

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N7—H72 \cdots O1 ⁱ	0.86 (2)	2.40 (2)	3.220 (2)	158 (2)
N6—H61 \cdots N7 ⁱⁱ	0.87 (2)	2.34 (1)	3.191 (2)	166 (1)
C5—H5B \cdots Cg ⁱⁱⁱ	0.96	2.69	3.461 (2)	138

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $x, -y+1, z+1/2$; (iii) $-x+1/2, -y+1/2, -z$.