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# 2-(4-Fluorophenyl)-1-phenyl-1*H*-benzimidazole

**K. Jayamoorthy,[a] S. Rosepriya,[b] A. Thiruvalluvar,[b]* J. Jayabharathi[a] and R. J. Butcher[c]**

[a]Department of Chemistry, Annamalai University, Annamalai Nagar 608 002, Tamilnadu, India, [b]PG Research Department of Physics, Rajah Serfoji Government College (Autonomous), Thanjavur 613 005, Tamilnadu, India, and [c]Department of Chemistry, Howard University, 525 College Street NW, Washington, DC 20059, USA
Correspondence e-mail: thiruvalluvar.a@gmail.com

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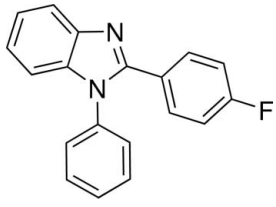
Key indicators: single-crystal X-ray study; $T$ = 123 K; mean $\sigma$(C–C) = 0.002 Å; $R$ factor = 0.063; $wR$ factor = 0.160; data-to-parameter ratio = 36.9.

In the title molecule, $C_{19}H_{13}FN_2$, the benzimidazole unit is close to planar [maximum deviation = 0.0342 (9) Å] and forms dihedral angles of 58.94 (3) and 51.43 (3)° with the phenyl and fluorobenzene rings, respectively; the dihedral angle between the phenyl and fluorobenzene rings is 60.17 (6)°. In the crystal, three C—H···F hydrogen bonds and two weak C—H···$\pi$ interactions involving the fused benzene ring lead to a three-dimensional architecture.

## Related literature

For linear and non-linear optical properties of benzimidazole compounds, see: Cross *et al.* (1995); Bu *et al.* (1996); Dirk *et al.* (1990). For a related structure, see: Rosepriya *et al.* (2011).




## Experimental

### Crystal data

$C_{19}H_{13}FN_2$
$M_r$ = 288.31
Monoclinic, $P2_1/n$
$a$ = 8.7527 (4) Å
$b$ = 10.1342 (4) Å
$c$ = 17.0211 (6) Å
$\beta$ = 104.187 (4)°
$V$ = 1463.75 (11) Å$^3$
$Z$ = 4
Mo $K\alpha$ radiation
$\mu$ = 0.09 mm$^{-1}$
$T$ = 123 K
0.47 × 0.42 × 0.15 mm

### Data collection

Agilent Xcalibur Ruby Gemini diffractometer
Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2012)
$T_{min}$ = 0.961, $T_{max}$ = 1.000
13721 measured reflections
7347 independent reflections
5352 reflections with $I > 2\sigma(I)$
$R_{int}$ = 0.031

### Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.063
$wR(F^2)$ = 0.160
$S$ = 1.04
7347 reflections
199 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.52 e Å$^{-3}$
$\Delta\rho_{min}$ = −0.22 e Å$^{-3}$

**Table 1**
Hydrogen-bond geometry (Å, °).

*Cg*2 is the centroid of the fused benzene ring (C4–C9).

| $D$—H···$A$ | $D$—H | H···$A$ | $D$···$A$ | $D$—H···$A$ |
|---|---|---|---|---|
| C4—H4···F4$^{i}$ | 0.93 | 2.46 | 3.3640 (14) | 164 |
| C7—H7···F4$^{ii}$ | 0.93 | 2.43 | 3.3058 (13) | 157 |
| C26—H26···F4$^{iii}$ | 0.93 | 2.52 | 3.4348 (14) | 166 |
| C16—H16···*Cg*2$^{iv}$ | 0.93 | 2.75 | 3.5443 (12) | 144 |
| C22—H22···*Cg*2$^{v}$ | 0.93 | 2.80 | 3.5245 (13) | 136 |

Symmetry codes: (i) $x, y+1, z$; (ii) $x-\frac{1}{2}, -y+\frac{1}{2}, z-\frac{1}{2}$; (iii) $-x+\frac{1}{2}, y+\frac{1}{2}, -z+\frac{1}{2}$; (iv) $-x, -y+1, -z$; (v) $-x+1, -y+1, -z$.

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SIR2004* (Burla *et al.*, 2005); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK5141).

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# supporting information

*Acta Cryst.* (2012). E**68**, o2708 [doi:10.1107/S1600536812035155]

## 2-(4-Fluorophenyl)-1-phenyl-1*H*-benzimidazole

**K. Jayamoorthy, S. Rosepriya, A. Thiruvalluvar, J. Jayabharathi and R. J. Butcher**

### S1. Comment

Benzimidazole based chromophores have received increasing attention due to their distinctive linear, non-linear optical properties and also due to their excellent thermal stability in guest-host systems (Cross *et al.*, 1995). The imidazole ring can be easily tailored to accommodate functional groups, which allows the covalent incorporation of the NLO chromophores into polyamides leading to NLO side chain polymers (Bu *et al.*, 1996). Most $\pi$-conjugated systems play a major role in determining second-order NLO response (Dirk *et al.*, 1990). Since our group is doing research in organic light emitting devices (OLEDs), we are interested in using the title compound as a ligand in the preparation of Ir(III) complexes and in studying the photophysical properties of these complexes. Rosepriya *et al.* (2011) have reported a related crystal structure, namely 1-(4-Methylbenzyl)-2-(4-methylphenyl)-1*H*-benzimidazole.

In the title molecule, $C_{19}H_{13}FN_2$ (Fig. 1), the benzimidazole unit is almost planar [maximum deviation = 0.0342 (9) Å for C6]. The dihedral angles between the planes of the benzimidazole and the phenyl and the fluorobenzene groups are 58.94 (3) and 51.43 (3)°, respectively. The dihedral angle between the planes of the phenyl and the fluorobenzene rings is 60.17 (6)°. Intermolecular C4—H4···F4, C7—H7···F4 and C26—H26···F4 hydrogen bonds and weak C16—H16···$\pi$ and C22—H22···$\pi$ interactions involving the fused benzene ring are found in the crystal structure (Fig. 2, Table 1).

### S2. Experimental

To *N*-phenyl-*o*-phenylenediamine (3.128 g, 17 mmol) in ethanol (10 ml) was added 4-fluorobenzaldehyde (1.9 ml, 17 mmol) and ammonium acetate (3 g) over about 1 h while maintaining the temperature at 353 K. The reaction mixture was refluxed until the completion of reaction, as monitored by TLC. The reaction mixture was extracted with dichloromethane. The solid that separated out was purified by column chromatography using petroleum ether: ethyl acetate as the eluent. Yield: 2.47 g (50%). The title compound was dissolved in acetonitrile and allowed to slowly evaporate for two days to obtain crystals suitable for X-ray diffraction studies.

### S3. Refinement

The H atoms were positioned geometrically and allowed to ride on their parent atoms, with C—H = 0.93 Å and $U_{iso}$(H) = 1.2$U_{eq}$(C).

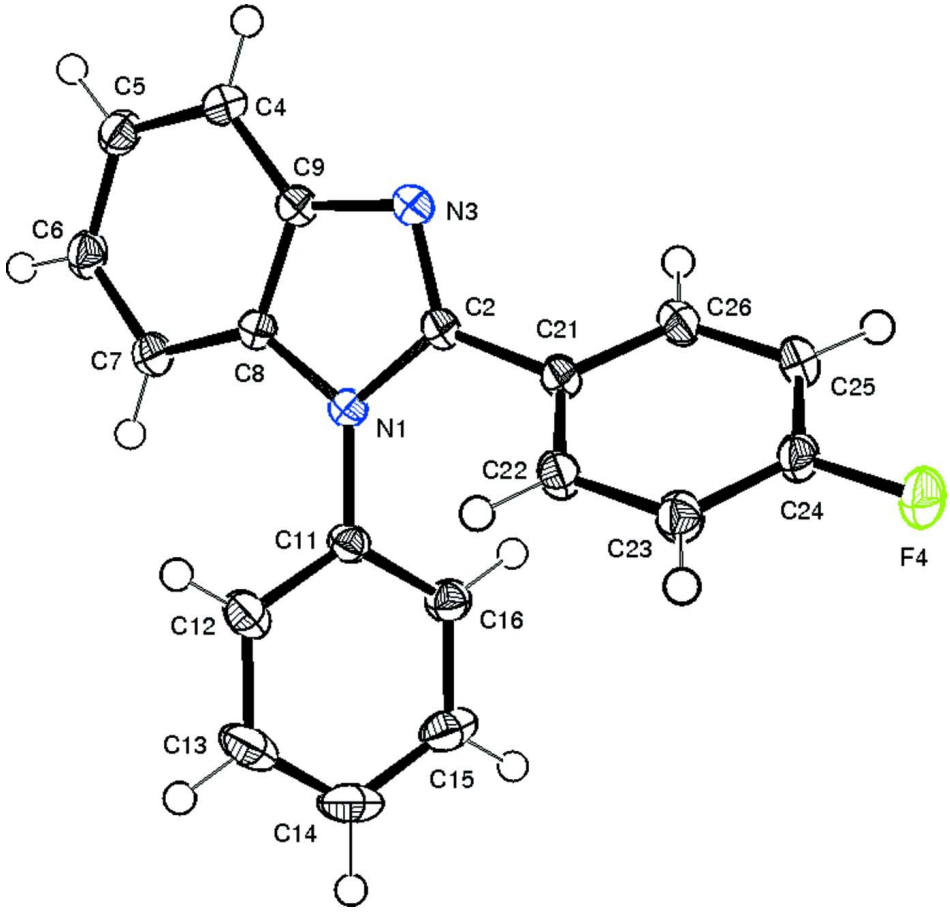


**Figure 1**

The molecular structure of the title compound, with displacement ellipsoids drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radius.

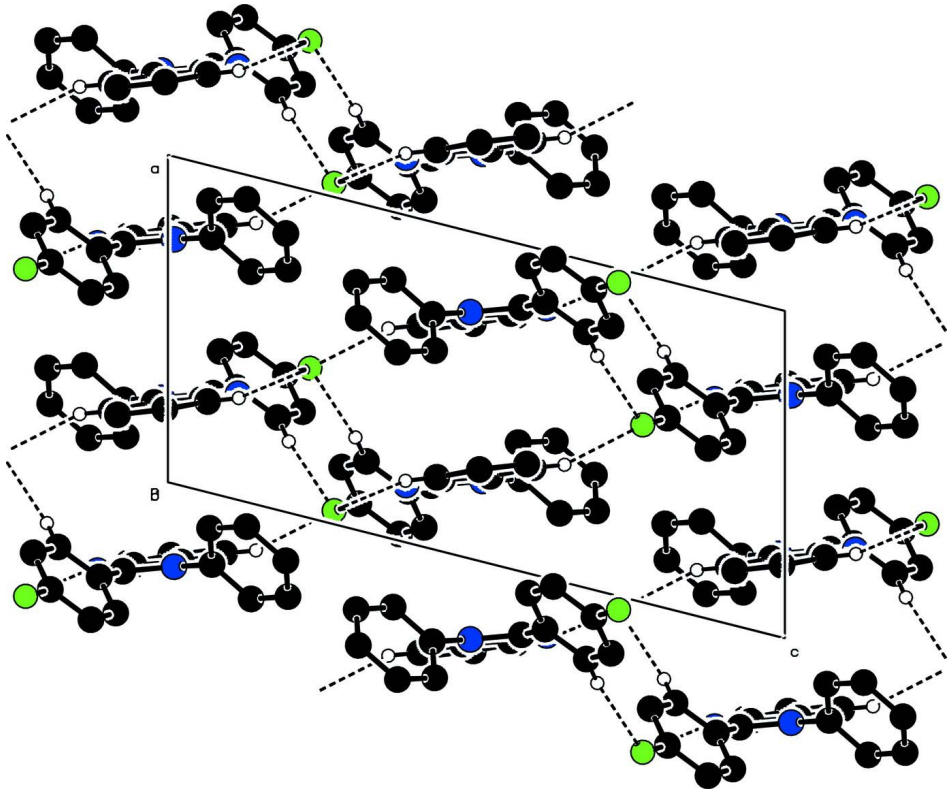


**Figure 2**

The packing of the title compound, viewed down the *b* axis. Dashed lines indicate hydrogen bonds. H atoms not involved in hydrogen bonding have been omitted.

### 2-(4-Fluorophenyl)-1-phenyl-1*H*-benzimidazole

*Crystal data*

$C_{19}H_{13}FN_2$
$M_r = 288.31$
Monoclinic, $P2_1/n$
Hall symbol: -P 2yn
$a = 8.7527$ (4) Å
$b = 10.1342$ (4) Å
$c = 17.0211$ (6) Å
$\beta = 104.187$ (4)°
$V = 1463.75$ (11) Å$^3$
$Z = 4$
$F(000) = 600$
$D_x = 1.308$ Mg m$^{-3}$
Melting point: 369 K
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 3202 reflections
$\theta = 3.1$–$37.6°$
$\mu = 0.09$ mm$^{-1}$
$T = 123$ K
Plate, colourless
0.47 × 0.42 × 0.15 mm

*Data collection*

Agilent Xcalibur Ruby Gemini diffractometer
Radiation source: Enhance (Mo) X-ray Source
Graphite monochromator
Detector resolution: 10.5081 pixels mm$^{-1}$
$\omega$ scans
Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2012)
$T_{min} = 0.961$, $T_{max} = 1.000$
13721 measured reflections
7347 independent reflections
5352 reflections with $I > 2\sigma(I)$
$R_{int} = 0.031$
$\theta_{max} = 37.7°$, $\theta_{min} = 3.1°$
$h = -12 \rightarrow 15$
$k = -12 \rightarrow 17$
$l = -28 \rightarrow 24$

*Refinement*

Refinement on $F^2$
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.063$
$wR(F^2) = 0.160$
$S = 1.04$
7347 reflections
199 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0685P)^2 + 0.2726P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} = 0.001$
$\Delta\rho_{max} = 0.52$ e Å$^{-3}$
$\Delta\rho_{min} = -0.22$ e Å$^{-3}$

*Special details*

**Geometry**. Bond distances, angles *etc*. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement**. Refinement of $F^2$ against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on $F^2$, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative $F^2$. The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on $F^2$ are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)*

| | *x* | *y* | *z* | $U_{iso}$*/$U_{eq}$ |
|---|---|---|---|---|
| F4 | 0.45822 (10) | −0.05502 (7) | 0.23042 (4) | 0.0296 (2) |
| N1 | 0.25329 (11) | 0.44506 (9) | −0.01000 (5) | 0.0181 (2) |
| N3 | 0.32447 (11) | 0.53932 (9) | 0.11392 (5) | 0.0204 (2) |
| C2 | 0.30699 (12) | 0.42744 (10) | 0.07314 (6) | 0.0181 (2) |
| C4 | 0.27719 (14) | 0.77382 (11) | 0.06432 (7) | 0.0224 (3) |
| C5 | 0.22041 (14) | 0.84918 (11) | −0.00490 (7) | 0.0241 (3) |
| C6 | 0.16727 (14) | 0.79006 (11) | −0.08140 (7) | 0.0242 (3) |
| C7 | 0.17379 (14) | 0.65470 (11) | −0.09192 (6) | 0.0227 (3) |
| C8 | 0.23331 (12) | 0.57999 (10) | −0.02222 (6) | 0.0184 (2) |
| C9 | 0.28005 (12) | 0.63684 (10) | 0.05527 (6) | 0.0192 (2) |
| C11 | 0.21724 (12) | 0.34608 (10) | −0.07164 (6) | 0.0181 (2) |
| C12 | 0.29236 (14) | 0.34944 (12) | −0.13497 (6) | 0.0250 (3) |
| C13 | 0.25650 (17) | 0.25270 (14) | −0.19472 (7) | 0.0326 (4) |
| C14 | 0.14865 (18) | 0.15428 (13) | −0.19126 (7) | 0.0342 (4) |
| C15 | 0.07463 (16) | 0.15197 (12) | −0.12781 (8) | 0.0296 (3) |
| C16 | 0.10831 (13) | 0.24826 (11) | −0.06771 (6) | 0.0222 (3) |
| C21 | 0.34251 (12) | 0.29683 (10) | 0.11128 (6) | 0.0184 (2) |
| C22 | 0.43985 (13) | 0.20730 (11) | 0.08453 (6) | 0.0207 (3) |
| C23 | 0.47933 (13) | 0.08832 (11) | 0.12460 (6) | 0.0226 (3) |
| C24 | 0.41834 (13) | 0.06137 (11) | 0.19050 (6) | 0.0216 (3) |
| C25 | 0.31950 (14) | 0.14570 (11) | 0.21819 (6) | 0.0240 (3) |
| C26 | 0.28287 (14) | 0.26534 (11) | 0.17816 (6) | 0.0224 (3) |
| H4 | 0.31207 | 0.81317 | 0.11491 | 0.0269* |
| H5 | 0.21753 | 0.94056 | −0.00049 | 0.0289* |
| H6 | 0.12661 | 0.84318 | −0.12621 | 0.0290* |
| H7 | 0.14030 | 0.61573 | −0.14271 | 0.0273* |

| | | | | |
|---|---|---|---|---|
| H12 | 0.36522 | 0.41522 | −0.13727 | 0.0299* |
| H13 | 0.30549 | 0.25419 | −0.23741 | 0.0391* |
| H14 | 0.12581 | 0.08983 | −0.23133 | 0.0410* |
| H15 | 0.00225 | 0.08581 | −0.12547 | 0.0355* |
| H16 | 0.05837 | 0.24707 | −0.02538 | 0.0266* |
| H22 | 0.47838 | 0.22748 | 0.03966 | 0.0248* |
| H23 | 0.54500 | 0.02834 | 0.10762 | 0.0271* |
| H25 | 0.27886 | 0.12332 | 0.26204 | 0.0288* |
| H26 | 0.21808 | 0.32514 | 0.19596 | 0.0269* |

*Atomic displacement parameters (Å²)*

| | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
|---|---|---|---|---|---|---|
| F4 | 0.0397 (4) | 0.0227 (3) | 0.0241 (3) | 0.0008 (3) | 0.0033 (3) | 0.0082 (3) |
| N1 | 0.0235 (4) | 0.0161 (3) | 0.0140 (3) | −0.0019 (3) | 0.0035 (3) | −0.0001 (3) |
| N3 | 0.0259 (4) | 0.0195 (4) | 0.0158 (3) | −0.0011 (3) | 0.0051 (3) | −0.0005 (3) |
| C2 | 0.0208 (4) | 0.0193 (4) | 0.0144 (3) | −0.0009 (3) | 0.0047 (3) | 0.0010 (3) |
| C4 | 0.0266 (5) | 0.0188 (4) | 0.0221 (4) | −0.0028 (4) | 0.0065 (4) | −0.0029 (4) |
| C5 | 0.0275 (5) | 0.0183 (4) | 0.0280 (5) | −0.0008 (4) | 0.0096 (4) | 0.0009 (4) |
| C6 | 0.0279 (5) | 0.0209 (5) | 0.0235 (4) | 0.0007 (4) | 0.0059 (4) | 0.0055 (4) |
| C7 | 0.0279 (5) | 0.0217 (5) | 0.0174 (4) | −0.0007 (4) | 0.0032 (3) | 0.0028 (3) |
| C8 | 0.0214 (4) | 0.0170 (4) | 0.0167 (4) | −0.0021 (3) | 0.0045 (3) | 0.0002 (3) |
| C9 | 0.0214 (4) | 0.0197 (4) | 0.0167 (4) | −0.0021 (3) | 0.0048 (3) | −0.0003 (3) |
| C11 | 0.0203 (4) | 0.0186 (4) | 0.0148 (3) | 0.0007 (3) | 0.0030 (3) | −0.0010 (3) |
| C12 | 0.0285 (5) | 0.0289 (5) | 0.0192 (4) | 0.0028 (4) | 0.0091 (4) | 0.0010 (4) |
| C13 | 0.0437 (7) | 0.0362 (7) | 0.0192 (4) | 0.0128 (6) | 0.0102 (5) | −0.0029 (4) |
| C14 | 0.0464 (8) | 0.0263 (6) | 0.0244 (5) | 0.0114 (5) | −0.0017 (5) | −0.0089 (4) |
| C15 | 0.0329 (6) | 0.0203 (5) | 0.0307 (5) | 0.0001 (4) | −0.0017 (4) | −0.0051 (4) |
| C16 | 0.0236 (5) | 0.0207 (4) | 0.0211 (4) | −0.0011 (4) | 0.0034 (3) | −0.0015 (4) |
| C21 | 0.0211 (4) | 0.0194 (4) | 0.0144 (3) | −0.0013 (3) | 0.0039 (3) | 0.0009 (3) |
| C22 | 0.0230 (4) | 0.0222 (5) | 0.0180 (4) | 0.0003 (4) | 0.0073 (3) | 0.0027 (3) |
| C23 | 0.0244 (5) | 0.0219 (5) | 0.0218 (4) | 0.0019 (4) | 0.0062 (4) | 0.0024 (4) |
| C24 | 0.0266 (5) | 0.0189 (4) | 0.0170 (4) | −0.0022 (4) | 0.0010 (3) | 0.0042 (3) |
| C25 | 0.0322 (5) | 0.0244 (5) | 0.0167 (4) | −0.0023 (4) | 0.0086 (4) | 0.0027 (4) |
| C26 | 0.0278 (5) | 0.0233 (5) | 0.0178 (4) | 0.0009 (4) | 0.0090 (4) | 0.0008 (3) |

*Geometric parameters (Å, º)*

| | | | |
|---|---|---|---|
| F4—C24 | 1.3633 (13) | C21—C22 | 1.3952 (15) |
| N1—C2 | 1.3890 (13) | C21—C26 | 1.4005 (15) |
| N1—C8 | 1.3877 (14) | C22—C23 | 1.3867 (15) |
| N1—C11 | 1.4298 (13) | C23—C24 | 1.3825 (15) |
| N3—C2 | 1.3186 (13) | C24—C25 | 1.3787 (16) |
| N3—C9 | 1.3908 (13) | C25—C26 | 1.3893 (15) |
| C2—C21 | 1.4735 (14) | C4—H4 | 0.9300 |
| C4—C5 | 1.3897 (16) | C5—H5 | 0.9300 |
| C4—C9 | 1.3976 (15) | C6—H6 | 0.9300 |
| C5—C6 | 1.4050 (16) | C7—H7 | 0.9300 |

| | | | |
|---|---|---|---|
| C6—C7 | 1.3864 (16) | C12—H12 | 0.9300 |
| C7—C8 | 1.3961 (14) | C13—H13 | 0.9300 |
| C8—C9 | 1.4052 (14) | C14—H14 | 0.9300 |
| C11—C12 | 1.3939 (15) | C15—H15 | 0.9300 |
| C11—C16 | 1.3883 (15) | C16—H16 | 0.9300 |
| C12—C13 | 1.3919 (17) | C22—H22 | 0.9300 |
| C13—C14 | 1.384 (2) | C23—H23 | 0.9300 |
| C14—C15 | 1.3891 (19) | C25—H25 | 0.9300 |
| C15—C16 | 1.3920 (17) | C26—H26 | 0.9300 |

| | | | |
|---|---|---|---|
| C2—N1—C8 | 106.15 (8) | F4—C24—C25 | 118.13 (9) |
| C2—N1—C11 | 128.03 (9) | C23—C24—C25 | 123.67 (10) |
| C8—N1—C11 | 125.75 (8) | C24—C25—C26 | 117.52 (10) |
| C2—N3—C9 | 104.91 (8) | C21—C26—C25 | 120.69 (10) |
| N1—C2—N3 | 113.10 (9) | C5—C4—H4 | 121.00 |
| N1—C2—C21 | 123.13 (9) | C9—C4—H4 | 121.00 |
| N3—C2—C21 | 123.76 (9) | C4—C5—H5 | 119.00 |
| C5—C4—C9 | 117.69 (10) | C6—C5—H5 | 119.00 |
| C4—C5—C6 | 121.32 (10) | C5—C6—H6 | 119.00 |
| C5—C6—C7 | 121.80 (10) | C7—C6—H6 | 119.00 |
| C6—C7—C8 | 116.45 (10) | C6—C7—H7 | 122.00 |
| N1—C8—C7 | 131.97 (9) | C8—C7—H7 | 122.00 |
| N1—C8—C9 | 105.45 (8) | C11—C12—H12 | 121.00 |
| C7—C8—C9 | 122.51 (9) | C13—C12—H12 | 121.00 |
| N3—C9—C4 | 129.50 (9) | C12—C13—H13 | 120.00 |
| N3—C9—C8 | 110.38 (9) | C14—C13—H13 | 120.00 |
| C4—C9—C8 | 120.12 (9) | C13—C14—H14 | 120.00 |
| N1—C11—C12 | 119.29 (10) | C15—C14—H14 | 120.00 |
| N1—C11—C16 | 119.75 (9) | C14—C15—H15 | 120.00 |
| C12—C11—C16 | 120.96 (10) | C16—C15—H15 | 120.00 |
| C11—C12—C13 | 118.90 (11) | C11—C16—H16 | 120.00 |
| C12—C13—C14 | 120.69 (12) | C15—C16—H16 | 120.00 |
| C13—C14—C15 | 119.87 (12) | C21—C22—H22 | 120.00 |
| C14—C15—C16 | 120.29 (12) | C23—C22—H22 | 120.00 |
| C11—C16—C15 | 119.30 (10) | C22—C23—H23 | 121.00 |
| C2—C21—C22 | 121.34 (9) | C24—C23—H23 | 121.00 |
| C2—C21—C26 | 118.84 (9) | C24—C25—H25 | 121.00 |
| C22—C21—C26 | 119.75 (10) | C26—C25—H25 | 121.00 |
| C21—C22—C23 | 120.23 (10) | C21—C26—H26 | 120.00 |
| C22—C23—C24 | 118.14 (10) | C25—C26—H26 | 120.00 |
| F4—C24—C23 | 118.21 (10) | | |

| | | | |
|---|---|---|---|
| C8—N1—C2—N3 | −0.88 (13) | C6—C7—C8—N1 | 178.40 (12) |
| C8—N1—C2—C21 | 179.88 (10) | C6—C7—C8—C9 | 1.94 (17) |
| C11—N1—C2—N3 | −178.02 (10) | N1—C8—C9—N3 | −1.25 (12) |
| C11—N1—C2—C21 | 2.74 (17) | N1—C8—C9—C4 | 178.72 (10) |
| C2—N1—C8—C7 | −175.66 (12) | C7—C8—C9—N3 | 176.02 (10) |
| C2—N1—C8—C9 | 1.24 (12) | C7—C8—C9—C4 | −4.00 (17) |

| | | | |
|---|---|---|---|
| C11—N1—C8—C7 | 1.56 (19) | N1—C11—C12—C13 | −179.97 (12) |
| C11—N1—C8—C9 | 178.46 (10) | C16—C11—C12—C13 | 0.00 (17) |
| C2—N1—C11—C12 | −123.95 (12) | N1—C11—C16—C15 | −179.68 (10) |
| C2—N1—C11—C16 | 56.07 (16) | C12—C11—C16—C15 | 0.33 (17) |
| C8—N1—C11—C12 | 59.44 (15) | C11—C12—C13—C14 | −0.32 (19) |
| C8—N1—C11—C16 | −120.54 (12) | C12—C13—C14—C15 | 0.3 (2) |
| C9—N3—C2—N1 | 0.11 (13) | C13—C14—C15—C16 | 0.1 (2) |
| C9—N3—C2—C21 | 179.34 (10) | C14—C15—C16—C11 | −0.36 (18) |
| C2—N3—C9—C4 | −179.25 (12) | C2—C21—C22—C23 | 176.01 (10) |
| C2—N3—C9—C8 | 0.72 (12) | C26—C21—C22—C23 | −0.90 (16) |
| N1—C2—C21—C22 | 50.81 (15) | C2—C21—C26—C25 | −177.04 (10) |
| N1—C2—C21—C26 | −132.26 (11) | C22—C21—C26—C25 | −0.06 (17) |
| N3—C2—C21—C22 | −128.35 (12) | C21—C22—C23—C24 | 0.67 (16) |
| N3—C2—C21—C26 | 48.58 (16) | C22—C23—C24—F4 | −179.40 (10) |
| C9—C4—C5—C6 | 0.15 (18) | C22—C23—C24—C25 | 0.53 (17) |
| C5—C4—C9—N3 | −177.21 (11) | F4—C24—C25—C26 | 178.48 (10) |
| C5—C4—C9—C8 | 2.83 (17) | C23—C24—C25—C26 | −1.45 (17) |
| C4—C5—C6—C7 | −2.2 (2) | C24—C25—C26—C21 | 1.18 (17) |
| C5—C6—C7—C8 | 1.13 (18) | | |

*Hydrogen-bond geometry (Å, º)*

Cg2 is the centroid of the fused benzene ring (C4–C9).

| *D*—H···*A* | *D*—H | H···*A* | *D*···*A* | *D*—H···*A* |
|---|---|---|---|---|
| C4—H4···F4$^{i}$ | 0.93 | 2.46 | 3.3640 (14) | 164 |
| C7—H7···F4$^{ii}$ | 0.93 | 2.43 | 3.3058 (13) | 157 |
| C26—H26···F4$^{iii}$ | 0.93 | 2.52 | 3.4348 (14) | 166 |
| C16—H16···*Cg*2$^{iv}$ | 0.93 | 2.75 | 3.5443 (12) | 144 |
| C22—H22···*Cg*2$^{v}$ | 0.93 | 2.80 | 3.5245 (13) | 136 |

Symmetry codes: (i) *x*, *y*+1, *z*; (ii) *x*−1/2, −*y*+1/2, *z*−1/2; (iii) −*x*+1/2, *y*+1/2, −*z*+1/2; (iv) −*x*, −*y*+1, −*z*; (v) −*x*+1, −*y*+1, −*z*.