

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

# 1-[(3*RS*,4*RS*)-1-Benzyl-4-methylpiperidin-3-yl]-1,6-dihydroimidazo[4,5-*d*]-pyrrolo[2,3-*b*]pyridine hemihydrate

Ellen Pfaffenrot,<sup>a</sup> Dieter Schollmeyer<sup>b</sup> and Stefan Laufer<sup>a\*</sup><sup>a</sup>Eberhard-Karls-University Tübingen, Auf der Morgenstelle 8, 72076 Tübingen, Germany, and <sup>b</sup>Institute of Organic Chemistry, University Mainz, Duesbergweg 10-14, 55099 Mainz, Germany

Correspondence e-mail: stefan.laufer@uni-tuebingen.de

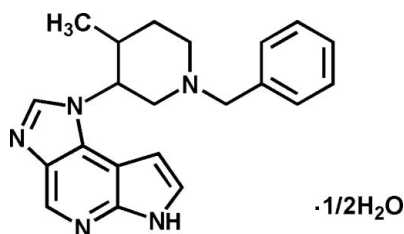
Received 17 September 2012; accepted 20 September 2012

Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.045;  $wR$  factor = 0.109; data-to-parameter ratio = 19.0.

The benzyl residue in the title compound,  $\text{C}_{21}\text{H}_{23}\text{N}_5 \cdot 0.5\text{H}_2\text{O}$ , is oriented at a dihedral angle of  $83.8(3)^\circ$  towards the 1,6-dihydroimidazo[4,5-*d*]pyrrolo[2,3-*b*]pyridine system. The piperidine ring adopts a chair conformation with the *cis* substituents displaying a torsion angle of  $-45.91(16)^\circ$ . In the crystal, molecules are accumulated as racemic dimers by two intermolecular hydrogen bonds between the pyrrolopyridine systems. Another hydrogen bond is formed between the imidazole ring and the cocrystallized water molecule, which is located on a twofold rotation axis.

## Related literature

For biological details on Janus protein tyrosine kinases, see: Kulagowski *et al.* (2012). For synthetic details, see: Bajwa *et al.* (2006).



## Experimental

## Crystal data

$\text{C}_{21}\text{H}_{23}\text{N}_5 \cdot 0.5\text{H}_2\text{O}$   
 $M_r = 354.45$   
 Monoclinic,  $C2/c$   
 $a = 17.3606(19)$  Å  
 $b = 10.0422(10)$  Å  
 $c = 22.995(2)$  Å  
 $\beta = 100.965(3)^\circ$

$V = 3935.8(7)$  Å<sup>3</sup>  
 $Z = 8$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.08$  mm<sup>-1</sup>  
 $T = 173$  K  
 $0.50 \times 0.27 \times 0.20$  mm

## Data collection

Bruker APEXII diffractometer  
 22115 measured reflections  
 4571 independent reflections

3504 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.040$   
 Standard reflections:  $\frac{1}{2}$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.109$   
 $S = 1.03$   
 4571 reflections

240 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.37$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.30$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{N8}-\text{H8} \cdots \text{N6}^i$	0.98	2.02	2.9757 (18)	165
$\text{O1W}-\text{H1W} \cdots \text{N3}$	0.93	2.13	3.0148 (16)	161

Symmetry code: (i)  $-x + \frac{1}{2}, -y + \frac{1}{2}, -z$ .

Data collection: APEX2 (Bruker, 2006); cell refinement: APEX2; data reduction: APEX2; program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: PLATON.

The authors thank Maria Leticia Barbosa, Matthias Gehringer and Peter Keck for suggestions and comments.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT6840).

## References

- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
- Bajwa, J. S., Chen, G. P., Prasad, K., Repi, O. & Blacklock, T. J. (2006). *Tetrahedron Lett.* **47**, 6425–6427.
- Bruker (2006). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.
- Kulagowski, J. J. *et al.* (2012). *J. Med. Chem.* **55**, 5901–5921.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

## supporting information

*Acta Cryst.* (2012). E68, o3052 [https://doi.org/10.1107/S1600536812039980]

## 1-[(3*RS*,4*RS*)-1-Benzyl-4-methylpiperidin-3-yl]-1,6-dihydroimidazo[4,5-*d*]pyrrolo[2,3-*b*]pyridine hemihydrate

Ellen Pfaffenrot, Dieter Schollmeyer and Stefan Laufer

### S1. Comment

Imidazopyrrolopyridine derivatives were identified as novel tricyclic JAK inhibitors. The Janus protein tyrosine kinases (JAK1, JAK2, JAK3 and TYK2) regulate the signal transduction of numerous cytokines, presenting a key role in immune and inflammatory processes (Kulagowski *et al.*, 2012).

The planar imidazopyrrolopyridine system is oriented at a dihedral angle of 83.8 (3)° and shows a distance of 6.89 (8) Å with respect to the benzyl group (Fig. 1). The equatorial methyl and the axial tricyclic substituent of the piperidine ring show a torsion angle of 45.9 (1)°. The crystal structure is characterized by two types of intermolecular hydrogen bonds. The water molecule connects two by 2-fold axis related pyrrolopyridine systems (O1W–H1W···N3 2.13 Å) while the N8—H8···N6 hydrogen bond build centrosymmetric dimers (Tabl. 1).

### S2. Experimental

The title compound was prepared by deprotection of a *N*-tosylated precursor (Bajwa *et al.*, 2006). 1-(1-benzyl-4-methylpiperidin-3-yl)-6-tosyl-1,6-dihydro-imidazo[4,5-*d*]pyrrolo[2,3-*b*]pyridine (0.164 g, 0.329 mmol) and caesium carbonate (0.986 g, 0.321 mmol) were dissolved in a mixture of dry THF (2 ml) and dry MeOH (1 ml) under argon atmosphere. The mixture was stirred for 3.5 h at room temperature and the reaction monitored by HPLC. When the reaction was complete, saturated NaHCO<sub>3</sub> solution was added and the mixture was extracted with ethylacetate. The organic phase was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated *in vacuo*. The precipitated product was obtained as crystalline solid by filtration (0.091 g, 80%). Crystals of the title compound were obtained by slow evaporation of methanol at room temperature.

### S3. Refinement

Hydrogen atoms attached to carbon and nitrogen atoms were placed at calculated positions with C—H = 0.95 Å (aromatic) or 0.99–1.00 Å (*sp*<sup>3</sup> C-atom). The position of the water H atom was taken from a difference map. All H atoms were refined using a riding model with isotropic displacement parameters set at 1.2–1.5 times of the *U*<sub>eq</sub> of the parent atom.

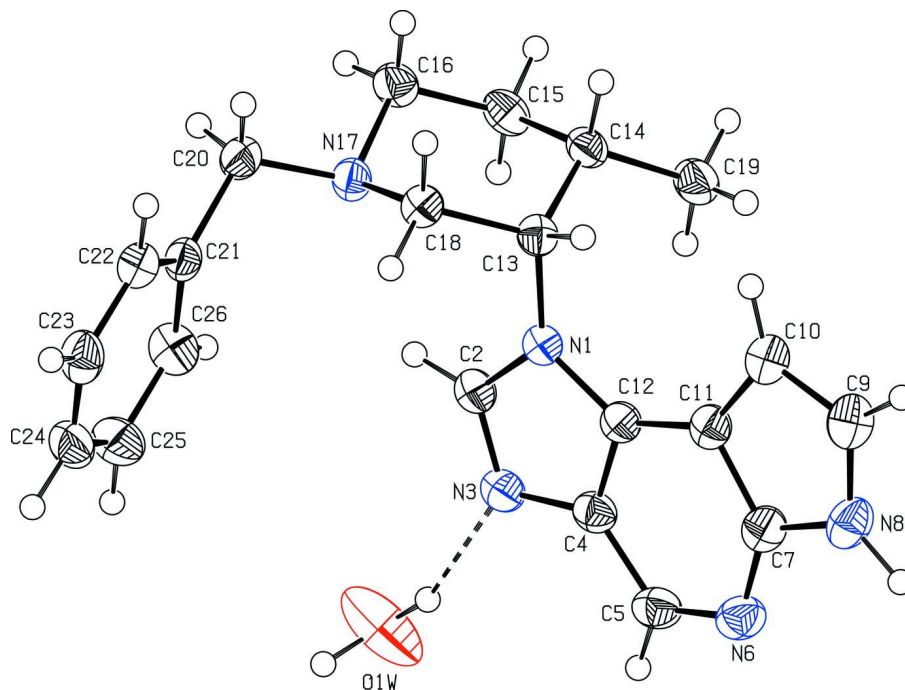


Figure 1

Crystal structure of the title compound with labelling and displacement ellipsoids drawn at the 50% probability level. Hydrogen bond is represented as dashed lines.

### 1-[(3*RS*,4*RS*)-1-Benzyl-4-methylpiperidin-3-yl]-1,6-dihydroimidazo[4,5-*d*]pyrrolo[2,3-*b*]pyridine hemihydrate

#### Crystal data

$C_{21}H_{23}N_5 \cdot 0.5H_2O$

$M_r = 354.45$

Monoclinic,  $C2/c$

Hall symbol:  $-C 2yc$

$a = 17.3606 (19) \text{ \AA}$

$b = 10.0422 (10) \text{ \AA}$

$c = 22.995 (2) \text{ \AA}$

$\beta = 100.965 (3)^\circ$

$V = 3935.8 (7) \text{ \AA}^3$

$Z = 8$

$F(000) = 1512$

$D_x = 1.196 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4892 reflections

$\theta = 2.4\text{--}26.7^\circ$

$\mu = 0.08 \text{ mm}^{-1}$

$T = 173 \text{ K}$

Block, colourless

$0.50 \times 0.27 \times 0.20 \text{ mm}$

#### Data collection

Bruker APEXII

diffractometer

Radiation source: sealed Tube

Graphite monochromator

CCD scan

22115 measured reflections

4571 independent reflections

3504 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.040$

$\theta_{\text{max}} = 27.7^\circ$ ,  $\theta_{\text{min}} = 1.8^\circ$

$h = -21 \rightarrow 22$

$k = -13 \rightarrow 12$

$l = -30 \rightarrow 30$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.109$   
 $S = 1.03$   
 4571 reflections  
 240 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0401P)^2 + 2.9543P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.37 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.30 \text{ e } \text{Å}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{Å}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.23581 (6)	0.09700 (11)	0.24008 (5)	0.0258 (3)
C2	0.30258 (8)	0.02507 (14)	0.26224 (7)	0.0293 (3)
H2	0.3148	-0.0059	0.3020	0.035*
N3	0.34774 (7)	0.00328 (13)	0.22344 (5)	0.0310 (3)
C4	0.30842 (8)	0.06430 (14)	0.17183 (6)	0.0288 (3)
C5	0.33093 (9)	0.07650 (15)	0.11641 (7)	0.0338 (3)
H5	0.3783	0.0356	0.1108	0.041*
N6	0.28808 (8)	0.14347 (13)	0.07164 (6)	0.0362 (3)
C7	0.22083 (9)	0.19640 (14)	0.08238 (7)	0.0319 (3)
N8	0.16809 (8)	0.26560 (13)	0.04086 (6)	0.0373 (3)
H8	0.1745	0.2858	0.0004	0.056*
C9	0.10645 (10)	0.30373 (15)	0.06626 (7)	0.0375 (4)
H9	0.0625	0.3533	0.0466	0.045*
C10	0.11658 (9)	0.26095 (15)	0.12371 (7)	0.0328 (3)
H10	0.0820	0.2746	0.1506	0.039*
C11	0.18991 (8)	0.19139 (14)	0.13518 (6)	0.0285 (3)
C12	0.23855 (8)	0.12243 (13)	0.18146 (6)	0.0261 (3)
C13	0.17177 (8)	0.13850 (14)	0.27023 (6)	0.0257 (3)
H13	0.1425	0.2113	0.2458	0.031*
C14	0.11248 (8)	0.02574 (14)	0.27341 (6)	0.0284 (3)
H14	0.0641	0.0683	0.2826	0.034*
C15	0.14310 (8)	-0.06998 (14)	0.32409 (7)	0.0323 (3)
H15A	0.1007	-0.1323	0.3291	0.039*
H15B	0.1867	-0.1230	0.3139	0.039*
C16	0.17175 (8)	0.00276 (15)	0.38202 (7)	0.0320 (3)

H16A	0.1278	0.0524	0.3936	0.038*
H16B	0.1915	-0.0623	0.4137	0.038*
N17	0.23465 (7)	0.09550 (12)	0.37510 (5)	0.0275 (3)
C18	0.20358 (8)	0.19706 (14)	0.33118 (6)	0.0273 (3)
H18A	0.2457	0.2614	0.3279	0.033*
H18B	0.1610	0.2463	0.3450	0.033*
C19	0.08819 (9)	-0.04706 (17)	0.21443 (7)	0.0379 (4)
H19A	0.0685	0.0175	0.1832	0.057*
H19B	0.0468	-0.1116	0.2176	0.057*
H19C	0.1337	-0.0938	0.2047	0.057*
C20	0.26827 (9)	0.15779 (16)	0.43189 (6)	0.0331 (3)
H20A	0.2786	0.0881	0.4629	0.040*
H20B	0.2296	0.2207	0.4429	0.040*
C21	0.34369 (8)	0.23169 (14)	0.43017 (6)	0.0278 (3)
C22	0.35253 (9)	0.36471 (15)	0.44601 (6)	0.0312 (3)
H22	0.3098	0.4116	0.4567	0.037*
C23	0.42287 (9)	0.43037 (16)	0.44643 (7)	0.0373 (4)
H23	0.4283	0.5213	0.4578	0.045*
C24	0.48489 (9)	0.36375 (18)	0.43038 (7)	0.0412 (4)
H24	0.5333	0.4084	0.4312	0.049*
C25	0.47669 (10)	0.23215 (19)	0.41312 (8)	0.0445 (4)
H25	0.5191	0.1866	0.4013	0.053*
C26	0.40644 (9)	0.16630 (16)	0.41310 (7)	0.0381 (4)
H26	0.4011	0.0756	0.4013	0.046*
O1W	0.5000	-0.1491 (2)	0.2500	0.0825 (8)
H1W	0.4569	-0.0982	0.2336	0.124*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0244 (6)	0.0261 (6)	0.0264 (6)	-0.0000 (5)	0.0040 (5)	-0.0001 (5)
C2	0.0261 (7)	0.0294 (7)	0.0314 (7)	0.0011 (6)	0.0028 (6)	-0.0000 (6)
N3	0.0260 (6)	0.0326 (7)	0.0344 (7)	-0.0007 (5)	0.0059 (5)	-0.0011 (5)
C4	0.0275 (7)	0.0262 (7)	0.0330 (8)	-0.0046 (6)	0.0066 (6)	-0.0010 (6)
C5	0.0321 (8)	0.0326 (8)	0.0389 (8)	-0.0052 (6)	0.0122 (6)	-0.0010 (7)
N6	0.0409 (7)	0.0353 (7)	0.0346 (7)	-0.0064 (6)	0.0126 (6)	0.0017 (6)
C7	0.0394 (8)	0.0260 (7)	0.0304 (8)	-0.0073 (6)	0.0068 (6)	0.0007 (6)
N8	0.0489 (8)	0.0342 (7)	0.0283 (7)	-0.0014 (6)	0.0063 (6)	0.0048 (6)
C9	0.0451 (9)	0.0298 (8)	0.0353 (8)	0.0034 (7)	0.0020 (7)	0.0020 (7)
C10	0.0387 (8)	0.0282 (7)	0.0304 (8)	0.0016 (6)	0.0040 (6)	-0.0004 (6)
C11	0.0330 (7)	0.0233 (7)	0.0288 (7)	-0.0052 (6)	0.0046 (6)	-0.0014 (6)
C12	0.0290 (7)	0.0220 (6)	0.0274 (7)	-0.0062 (5)	0.0056 (5)	-0.0018 (6)
C13	0.0252 (7)	0.0242 (7)	0.0274 (7)	0.0035 (5)	0.0042 (5)	-0.0005 (6)
C14	0.0215 (7)	0.0316 (7)	0.0320 (8)	0.0001 (5)	0.0048 (6)	-0.0044 (6)
C15	0.0279 (7)	0.0256 (7)	0.0424 (9)	-0.0045 (6)	0.0044 (6)	0.0001 (6)
C16	0.0308 (7)	0.0300 (7)	0.0346 (8)	-0.0056 (6)	0.0048 (6)	0.0059 (6)
N17	0.0298 (6)	0.0267 (6)	0.0251 (6)	-0.0058 (5)	0.0027 (5)	0.0018 (5)
C18	0.0308 (7)	0.0233 (7)	0.0281 (7)	-0.0009 (6)	0.0060 (6)	-0.0008 (6)

C19	0.0304 (8)	0.0443 (9)	0.0393 (9)	-0.0065 (7)	0.0069 (6)	-0.0123 (7)
C20	0.0352 (8)	0.0391 (8)	0.0251 (7)	-0.0084 (6)	0.0062 (6)	0.0003 (6)
C21	0.0301 (7)	0.0305 (7)	0.0214 (7)	-0.0027 (6)	0.0013 (5)	0.0016 (6)
C22	0.0343 (8)	0.0326 (8)	0.0262 (7)	-0.0000 (6)	0.0045 (6)	-0.0008 (6)
C23	0.0451 (9)	0.0312 (8)	0.0327 (8)	-0.0095 (7)	0.0003 (7)	0.0007 (7)
C24	0.0325 (8)	0.0513 (10)	0.0379 (9)	-0.0133 (7)	0.0018 (7)	0.0072 (8)
C25	0.0306 (8)	0.0537 (11)	0.0496 (10)	0.0062 (7)	0.0087 (7)	0.0011 (8)
C26	0.0371 (8)	0.0315 (8)	0.0447 (9)	0.0015 (6)	0.0052 (7)	-0.0030 (7)
O1W	0.0391 (11)	0.0429 (11)	0.158 (2)	0.000	0.0001 (12)	0.000

*Geometric parameters (Å, °)*

N1—C2	1.3786 (18)	C15—H15A	0.9900
N1—C12	1.3815 (18)	C15—H15B	0.9900
N1—C13	1.4777 (17)	C16—N17	1.4667 (17)
C2—N3	1.3132 (18)	C16—H16A	0.9900
C2—H2	0.9500	C16—H16B	0.9900
N3—C4	1.3936 (19)	N17—C18	1.4642 (18)
C4—C12	1.401 (2)	N17—C20	1.4654 (18)
C4—C5	1.407 (2)	C18—H18A	0.9900
C5—N6	1.332 (2)	C18—H18B	0.9900
C5—H5	0.9500	C19—H19A	0.9800
N6—C7	1.348 (2)	C19—H19B	0.9800
C7—N8	1.378 (2)	C19—H19C	0.9800
C7—C11	1.419 (2)	C20—C21	1.512 (2)
N8—C9	1.368 (2)	C20—H20A	0.9900
N8—H8	0.9793	C20—H20B	0.9900
C9—C10	1.368 (2)	C21—C22	1.385 (2)
C9—H9	0.9500	C21—C26	1.391 (2)
C10—C11	1.432 (2)	C22—C23	1.386 (2)
C10—H10	0.9500	C22—H22	0.9500
C11—C12	1.408 (2)	C23—C24	1.376 (2)
C13—C18	1.5234 (19)	C23—H23	0.9500
C13—C14	1.5414 (19)	C24—C25	1.379 (3)
C13—H13	1.0000	C24—H24	0.9500
C14—C15	1.526 (2)	C25—C26	1.387 (2)
C14—C19	1.528 (2)	C25—H25	0.9500
C14—H14	1.0000	C26—H26	0.9500
C15—C16	1.518 (2)	O1W—H1W	0.9253
C2—N1—C12	105.96 (11)	C14—C15—H15B	109.2
C2—N1—C13	128.98 (12)	H15A—C15—H15B	107.9
C12—N1—C13	125.05 (11)	N17—C16—C15	109.69 (12)
N3—C2—N1	113.88 (13)	N17—C16—H16A	109.7
N3—C2—H2	123.1	C15—C16—H16A	109.7
N1—C2—H2	123.1	N17—C16—H16B	109.7
C2—N3—C4	104.26 (12)	C15—C16—H16B	109.7
N3—C4—C12	110.30 (12)	H16A—C16—H16B	108.2

N3—C4—C5	129.35 (13)	C18—N17—C20	110.45 (11)
C12—C4—C5	120.31 (13)	C18—N17—C16	109.44 (11)
N6—C5—C4	122.23 (14)	C20—N17—C16	110.67 (11)
N6—C5—H5	118.9	N17—C18—C13	112.78 (11)
C4—C5—H5	118.9	N17—C18—H18A	109.0
C5—N6—C7	115.68 (13)	C13—C18—H18A	109.0
N6—C7—N8	123.80 (13)	N17—C18—H18B	109.0
N6—C7—C11	128.67 (14)	C13—C18—H18B	109.0
N8—C7—C11	107.52 (13)	H18A—C18—H18B	107.8
C9—N8—C7	108.42 (13)	C14—C19—H19A	109.5
C9—N8—H8	126.0	C14—C19—H19B	109.5
C7—N8—H8	125.6	H19A—C19—H19B	109.5
N8—C9—C10	110.92 (14)	C14—C19—H19C	109.5
N8—C9—H9	124.5	H19A—C19—H19C	109.5
C10—C9—H9	124.5	H19B—C19—H19C	109.5
C9—C10—C11	106.11 (14)	N17—C20—C21	112.72 (11)
C9—C10—H10	126.9	N17—C20—H20A	109.0
C11—C10—H10	126.9	C21—C20—H20A	109.0
C12—C11—C7	113.22 (13)	N17—C20—H20B	109.0
C12—C11—C10	139.75 (14)	C21—C20—H20B	109.0
C7—C11—C10	107.03 (13)	H20A—C20—H20B	107.8
N1—C12—C4	105.59 (12)	C22—C21—C26	118.41 (14)
N1—C12—C11	134.53 (13)	C22—C21—C20	121.30 (13)
C4—C12—C11	119.86 (13)	C26—C21—C20	120.28 (13)
N1—C13—C18	111.55 (11)	C21—C22—C23	120.92 (14)
N1—C13—C14	112.57 (11)	C21—C22—H22	119.5
C18—C13—C14	111.61 (11)	C23—C22—H22	119.5
N1—C13—H13	106.9	C24—C23—C22	120.00 (15)
C18—C13—H13	106.9	C24—C23—H23	120.0
C14—C13—H13	106.9	C22—C23—H23	120.0
C15—C14—C19	111.95 (12)	C23—C24—C25	119.98 (15)
C15—C14—C13	111.12 (11)	C23—C24—H24	120.0
C19—C14—C13	112.55 (12)	C25—C24—H24	120.0
C15—C14—H14	106.9	C24—C25—C26	119.96 (16)
C19—C14—H14	106.9	C24—C25—H25	120.0
C13—C14—H14	106.9	C26—C25—H25	120.0
C16—C15—C14	112.05 (12)	C25—C26—C21	120.70 (15)
C16—C15—H15A	109.2	C25—C26—H26	119.6
C14—C15—H15A	109.2	C21—C26—H26	119.6
C16—C15—H15B	109.2		
C12—N1—C2—N3	0.50 (16)	C10—C11—C12—C4	178.39 (16)
C13—N1—C2—N3	179.23 (12)	C2—N1—C13—C18	46.14 (18)
N1—C2—N3—C4	-0.21 (16)	C12—N1—C13—C18	-135.35 (13)
C2—N3—C4—C12	-0.15 (15)	C2—N1—C13—C14	-80.23 (17)
C2—N3—C4—C5	177.61 (15)	C12—N1—C13—C14	98.28 (15)
N3—C4—C5—N6	-176.99 (14)	N1—C13—C14—C15	80.55 (14)
C12—C4—C5—N6	0.6 (2)	C18—C13—C14—C15	-45.79 (15)

C4—C5—N6—C7	-1.4 (2)	N1—C13—C14—C19	-45.91 (16)
C5—N6—C7—N8	-178.11 (14)	C18—C13—C14—C19	-172.24 (12)
C5—N6—C7—C11	0.6 (2)	C19—C14—C15—C16	176.63 (12)
N6—C7—N8—C9	179.33 (14)	C13—C14—C15—C16	49.84 (15)
C11—C7—N8—C9	0.35 (16)	C14—C15—C16—N17	-58.88 (15)
C7—N8—C9—C10	-0.34 (18)	C15—C16—N17—C18	63.33 (15)
N8—C9—C10—C11	0.18 (18)	C15—C16—N17—C20	-174.72 (12)
N6—C7—C11—C12	0.9 (2)	C20—N17—C18—C13	176.90 (11)
N8—C7—C11—C12	179.84 (12)	C16—N17—C18—C13	-61.02 (14)
N6—C7—C11—C10	-179.16 (14)	N1—C13—C18—N17	-74.69 (14)
N8—C7—C11—C10	-0.24 (16)	C14—C13—C18—N17	52.20 (15)
C9—C10—C11—C12	179.93 (17)	C18—N17—C20—C21	-70.56 (15)
C9—C10—C11—C7	0.04 (16)	C16—N17—C20—C21	168.08 (12)
C2—N1—C12—C4	-0.54 (14)	N17—C20—C21—C22	124.87 (14)
C13—N1—C12—C4	-179.34 (12)	N17—C20—C21—C26	-55.85 (18)
C2—N1—C12—C11	-178.93 (15)	C26—C21—C22—C23	-1.7 (2)
C13—N1—C12—C11	2.3 (2)	C20—C21—C22—C23	177.56 (13)
N3—C4—C12—N1	0.44 (15)	C21—C22—C23—C24	0.7 (2)
C5—C4—C12—N1	-177.55 (12)	C22—C23—C24—C25	0.9 (2)
N3—C4—C12—C11	179.12 (12)	C23—C24—C25—C26	-1.3 (3)
C5—C4—C12—C11	1.1 (2)	C24—C25—C26—C21	0.2 (3)
C7—C11—C12—N1	176.48 (14)	C22—C21—C26—C25	1.3 (2)
C10—C11—C12—N1	-3.4 (3)	C20—C21—C26—C25	-178.01 (14)
C7—C11—C12—C4	-1.73 (18)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N8—H8...N6 <sup>i</sup>	0.98	2.02	2.9757 (18)	165
O1 <i>W</i> —H1 <i>W</i> ...N3	0.93	2.13	3.0148 (16)	161

Symmetry code: (i)  $-x+1/2, -y+1/2, -z$ .