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(4-Methoxyphenyl)(4-methylcyclohexyl)-methanone

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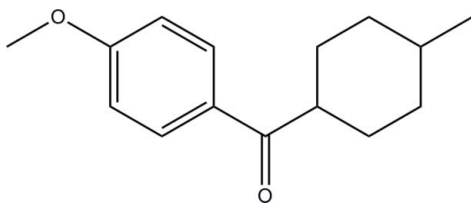
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Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.043; wR factor = 0.120; data-to-parameter ratio = 19.0.

The title compound, $\text{C}_{15}\text{H}_{20}\text{O}_2$, crystallizes with two independent molecules of similar geometry in the asymmetric unit. The cyclohexyl ring adopts a chair conformation in each molecule. In the crystal, molecules related by translation are linked into chains along the a axis via weak $\text{C}-\text{H}\cdots\text{O}$ interactions.

Related literature

For the antihyperglycemic activity of SGLT2 inhibitors, see: Shao *et al.* (2011); Zhao *et al.* (2011). For related structures, see: Meng *et al.* (2012); Wang *et al.* (2011).



Experimental

Crystal data

 $\text{C}_{15}\text{H}_{20}\text{O}_2$ $M_r = 232.31$

Triclinic, $P\bar{1}$
 $a = 5.7003$ (18) Å
 $b = 7.279$ (2) Å
 $c = 30.76$ (1) Å
 $\alpha = 96.700$ (9)°
 $\beta = 94.380$ (16)°
 $\gamma = 93.332$ (12)°

$V = 1261.0$ (7) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 113$ K
 $0.20 \times 0.18 \times 0.12$ mm

Data collection

Rigaku Saturn CCD area-detector diffractometer
 Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2009)
 $T_{\min} = 0.984$, $T_{\max} = 0.991$

12908 measured reflections
 5924 independent reflections
 3739 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.038$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.120$
 $S = 1.03$
 5924 reflections

311 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.35$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---|-------|-------------|-------------|---------------|
| $\text{C9}-\text{H9}\cdots\text{O2}^i$ | 1.00 | 2.57 | 3.4694 (18) | 150 |
| $\text{C24}-\text{H24}\cdots\text{O4}^{ii}$ | 1.00 | 2.60 | 3.5244 (18) | 154 |

Symmetry codes: (i) $x - 1, y, z$; (ii) $x + 1, y, z$.

Data collection: *CrystalClear* (Rigaku/MS, 2009); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5337).

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supporting information

Acta Cryst. (2012). E68, o2850 [https://doi.org/10.1107/S1600536812037373]

(4-Methoxyphenyl)(4-methylcyclohexyl)methanone**Wei Liu and Lida Tang****S1. Comment**

SGLT2 inhibitors represent an emerging class of promising antihyperglycemic agents, and a lot of drug candidates are now filed or in phase III clinical trials. During the study of saturated ring-bearing SGLT2 inhibitors (Shao *et al.*, 2011; Zhao *et al.*, 2011) the title compound was prepared, which is a key intermediate for the synthetic procedure.

In title compound, C₁₅H₂₀O₂, crystallizes with two independent molecules of similar geometry in the asymmetric unit. All bond lengths and angles are normal and in a good agreement with those reported previously for related compounds (Meng *et al.*, 2012; Wang *et al.*, 2011). The cyclohexyl rings (C9—C14 and C24—C29) adopt chair confirmation. In the crystal, the molecules related by translation along the *a* axis are linked into chains *via* weak C—H⋯O interactions.

S2. Experimental

A mixture of 2.46 g (10 mmol) of ethyl 3-(4-methoxyphenyl)-1*H*-pyrazole-5-carboxylate, 3.76 g (20 mmol) of 1,2-dibromoethane, 5.53 g (40 mmol) of K₂CO₃ in 30 ml of dried acetonitrile was refluxed overnight. On cooling, the reaction mixture was filtered and poured into 200 ml of brine. The resulting mixture was extracted with three 50-ml of dichloromethane, and combined extracts were washed with saturated brine, dried over Na₂SO₄ and evaporated on a rotary evaporator to afford the crude product as brown solid, which was purified by column chromatography to yield the pure product as colorless crystals (yield 88%). The single crystals suitable for single-crystal X-ray diffraction was obtained by slow evaporation at room temperature of a solution of the title compound in dichloromethane/hexane (1/5) (m.p. 88–90 Celsius degree).

S3. Refinement

All H atoms bonded on carbon were found on difference maps, with C—H = 0.95–1.00, and included in the final cycles of refinement using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and $1.5U_{\text{eq}}(\text{C})$ for the methyl H atoms.

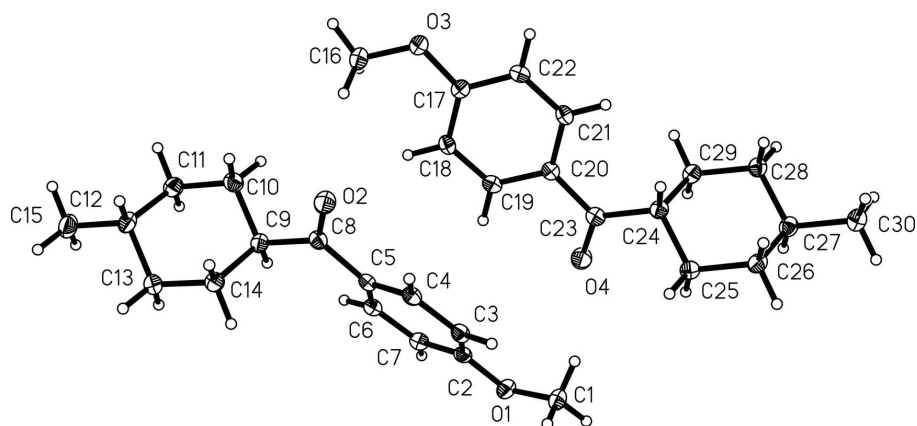


Figure 1

Two independent molecules of the title compound, with displacement ellipsoids drawn at the 40% probability level.

(4-Methoxyphenyl)(4-methylcyclohexyl)methanone

Crystal data

$C_{15}H_{20}O_2$

$M_r = 232.31$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 5.7003\ (18)\ \text{\AA}$

$b = 7.279\ (2)\ \text{\AA}$

$c = 30.76\ (1)\ \text{\AA}$

$\alpha = 96.700\ (9)^\circ$

$\beta = 94.380\ (16)^\circ$

$\gamma = 93.332\ (12)^\circ$

$V = 1261.0\ (7)\ \text{\AA}^3$

$Z = 4$

$F(000) = 504$

$D_x = 1.224\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 4144 reflections

$\theta = 2.0\text{--}27.9^\circ$

$\mu = 0.08\ \text{mm}^{-1}$

$T = 113\ \text{K}$

Prism, colourless

$0.20 \times 0.18 \times 0.12\ \text{mm}$

Data collection

Rigaku Saturn CCD area-detector
diffractometer

Radiation source: rotating anode

Multilayer monochromator

Detector resolution: $14.63\ \text{pixels mm}^{-1}$

ω and φ scans

Absorption correction: multi-scan

(*CrystalClear*; Rigaku/MS, 2009)

$T_{\min} = 0.984$, $T_{\max} = 0.991$

12908 measured reflections

5924 independent reflections

3739 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.038$

$\theta_{\max} = 27.9^\circ$, $\theta_{\min} = 2.0^\circ$

$h = -7 \rightarrow 7$

$k = -9 \rightarrow 9$

$l = -39 \rightarrow 40$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.120$

$S = 1.03$

5924 reflections

311 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0506P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.004$

$\Delta\rho_{\max} = 0.35\ \text{e \AA}^{-3}$

$\Delta\rho_{\min} = -0.18\ \text{e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|--------------|-------------|----------------------------------|
| O1 | 0.19664 (16) | 1.11167 (12) | 0.33482 (3) | 0.0263 (2) |
| O2 | 0.46796 (15) | 0.82156 (13) | 0.14498 (3) | 0.0305 (2) |
| O3 | 0.77830 (16) | 0.33602 (12) | 0.17072 (3) | 0.0275 (2) |
| O4 | 0.52284 (16) | 0.73059 (14) | 0.35409 (3) | 0.0334 (3) |
| C1 | 0.4001 (2) | 1.19400 (18) | 0.36164 (4) | 0.0295 (3) |
| H1A | 0.5278 | 1.1096 | 0.3598 | 0.044* |
| H1B | 0.3614 | 1.2175 | 0.3922 | 0.044* |
| H1C | 0.4506 | 1.3113 | 0.3513 | 0.044* |
| C2 | 0.2251 (2) | 1.05419 (16) | 0.29179 (4) | 0.0209 (3) |
| C3 | 0.4298 (2) | 1.08989 (16) | 0.27130 (4) | 0.0218 (3) |
| H3 | 0.5619 | 1.1598 | 0.2870 | 0.026* |
| C4 | 0.4368 (2) | 1.02131 (16) | 0.22748 (4) | 0.0212 (3) |
| H4 | 0.5772 | 1.0428 | 0.2135 | 0.025* |
| C5 | 0.2447 (2) | 0.92202 (16) | 0.20324 (4) | 0.0192 (3) |
| C6 | 0.0383 (2) | 0.89343 (16) | 0.22415 (4) | 0.0205 (3) |
| H6 | -0.0971 | 0.8298 | 0.2080 | 0.025* |
| C7 | 0.0302 (2) | 0.95706 (16) | 0.26805 (4) | 0.0213 (3) |
| H7 | -0.1094 | 0.9343 | 0.2821 | 0.026* |
| C8 | 0.2704 (2) | 0.84618 (17) | 0.15690 (4) | 0.0219 (3) |
| C9 | 0.0566 (2) | 0.80827 (17) | 0.12401 (4) | 0.0220 (3) |
| H9 | -0.0875 | 0.7917 | 0.1401 | 0.026* |
| C10 | 0.0752 (2) | 0.63558 (17) | 0.09157 (4) | 0.0272 (3) |
| H10A | 0.0743 | 0.5251 | 0.1075 | 0.033* |
| H10B | 0.2267 | 0.6453 | 0.0780 | 0.033* |
| C11 | -0.1284 (2) | 0.61157 (18) | 0.05559 (4) | 0.0283 (3) |
| H11A | -0.1074 | 0.5011 | 0.0346 | 0.034* |
| H11B | -0.2787 | 0.5904 | 0.0690 | 0.034* |
| C12 | -0.1415 (2) | 0.78128 (18) | 0.03088 (4) | 0.0258 (3) |
| H12 | 0.0095 | 0.7978 | 0.0168 | 0.031* |
| C13 | -0.1635 (2) | 0.95325 (18) | 0.06335 (4) | 0.0267 (3) |
| H13A | -0.3154 | 0.9416 | 0.0767 | 0.032* |
| H13B | -0.1645 | 1.0637 | 0.0474 | 0.032* |
| C14 | 0.0378 (2) | 0.98096 (17) | 0.09963 (4) | 0.0259 (3) |
| H14A | 0.0112 | 1.0894 | 0.1208 | 0.031* |
| H14B | 0.1881 | 1.0071 | 0.0867 | 0.031* |

| | | | | |
|------|-------------|--------------|--------------|------------|
| C15 | -0.3433 (3) | 0.7554 (2) | -0.00528 (4) | 0.0344 (3) |
| H15A | -0.4931 | 0.7358 | 0.0077 | 0.052* |
| H15B | -0.3195 | 0.6475 | -0.0263 | 0.052* |
| H15C | -0.3470 | 0.8664 | -0.0205 | 0.052* |
| C16 | 0.5868 (3) | 0.3403 (2) | 0.13812 (4) | 0.0324 (3) |
| H16A | 0.4446 | 0.2788 | 0.1474 | 0.049* |
| H16B | 0.6262 | 0.2756 | 0.1101 | 0.049* |
| H16C | 0.5582 | 0.4693 | 0.1345 | 0.049* |
| C17 | 0.7511 (2) | 0.41843 (16) | 0.21210 (4) | 0.0213 (3) |
| C18 | 0.5547 (2) | 0.51267 (16) | 0.22355 (4) | 0.0215 (3) |
| H18 | 0.4286 | 0.5234 | 0.2021 | 0.026* |
| C19 | 0.5460 (2) | 0.59047 (16) | 0.26667 (4) | 0.0214 (3) |
| H19 | 0.4111 | 0.6529 | 0.2747 | 0.026* |
| C20 | 0.7298 (2) | 0.57948 (16) | 0.29851 (4) | 0.0195 (3) |
| C21 | 0.9266 (2) | 0.48535 (16) | 0.28615 (4) | 0.0214 (3) |
| H21 | 1.0549 | 0.4774 | 0.3073 | 0.026* |
| C22 | 0.9355 (2) | 0.40431 (16) | 0.24356 (4) | 0.0219 (3) |
| H22 | 1.0681 | 0.3386 | 0.2357 | 0.026* |
| C23 | 0.7117 (2) | 0.67370 (17) | 0.34374 (4) | 0.0223 (3) |
| C24 | 0.9260 (2) | 0.69855 (17) | 0.37672 (4) | 0.0216 (3) |
| H24 | 1.0706 | 0.7109 | 0.3605 | 0.026* |
| C25 | 0.9183 (2) | 0.87238 (17) | 0.40970 (4) | 0.0242 (3) |
| H25A | 0.9196 | 0.9833 | 0.3939 | 0.029* |
| H25B | 0.7700 | 0.8658 | 0.4244 | 0.029* |
| C26 | 1.1281 (2) | 0.89120 (17) | 0.44422 (4) | 0.0251 (3) |
| H26A | 1.1156 | 1.0027 | 0.4654 | 0.030* |
| H26B | 1.2757 | 0.9082 | 0.4297 | 0.030* |
| C27 | 1.1396 (2) | 0.72105 (18) | 0.46893 (4) | 0.0252 (3) |
| H27 | 0.9918 | 0.7093 | 0.4842 | 0.030* |
| C28 | 1.1465 (2) | 0.54731 (17) | 0.43608 (4) | 0.0250 (3) |
| H28A | 1.2954 | 0.5536 | 0.4216 | 0.030* |
| H28B | 1.1451 | 0.4369 | 0.4521 | 0.030* |
| C29 | 0.9384 (2) | 0.52528 (17) | 0.40108 (4) | 0.0244 (3) |
| H29A | 0.7900 | 0.5053 | 0.4151 | 0.029* |
| H29B | 0.9552 | 0.4151 | 0.3797 | 0.029* |
| C30 | 1.3480 (2) | 0.74087 (19) | 0.50368 (4) | 0.0329 (3) |
| H30A | 1.3497 | 0.6296 | 0.5188 | 0.049* |
| H30B | 1.3330 | 0.8497 | 0.5250 | 0.049* |
| H30C | 1.4951 | 0.7561 | 0.4896 | 0.049* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|------------|------------|------------|------------|-------------|-------------|
| O1 | 0.0278 (5) | 0.0285 (5) | 0.0217 (5) | 0.0011 (4) | 0.0032 (4) | -0.0006 (4) |
| O2 | 0.0202 (5) | 0.0442 (6) | 0.0272 (5) | 0.0045 (4) | 0.0047 (4) | 0.0015 (4) |
| O3 | 0.0290 (5) | 0.0311 (5) | 0.0217 (5) | 0.0059 (4) | 0.0004 (4) | -0.0002 (4) |
| O4 | 0.0217 (5) | 0.0480 (6) | 0.0301 (6) | 0.0073 (5) | 0.0046 (4) | -0.0009 (5) |
| C1 | 0.0333 (8) | 0.0294 (7) | 0.0232 (7) | 0.0005 (6) | -0.0047 (6) | -0.0016 (6) |

| | | | | | | |
|-----|------------|------------|------------|-------------|-------------|-------------|
| C2 | 0.0240 (7) | 0.0176 (6) | 0.0220 (7) | 0.0046 (5) | 0.0028 (6) | 0.0040 (5) |
| C3 | 0.0187 (7) | 0.0193 (6) | 0.0270 (7) | -0.0002 (5) | -0.0017 (5) | 0.0042 (5) |
| C4 | 0.0172 (7) | 0.0214 (6) | 0.0259 (7) | 0.0016 (5) | 0.0025 (5) | 0.0058 (5) |
| C5 | 0.0175 (7) | 0.0183 (6) | 0.0227 (7) | 0.0032 (5) | 0.0009 (5) | 0.0058 (5) |
| C6 | 0.0173 (7) | 0.0178 (6) | 0.0263 (7) | 0.0000 (5) | 0.0012 (5) | 0.0035 (5) |
| C7 | 0.0197 (7) | 0.0195 (6) | 0.0256 (7) | 0.0000 (5) | 0.0050 (5) | 0.0055 (5) |
| C8 | 0.0217 (7) | 0.0223 (6) | 0.0224 (7) | 0.0027 (5) | 0.0018 (6) | 0.0053 (5) |
| C9 | 0.0182 (7) | 0.0279 (7) | 0.0197 (7) | 0.0013 (5) | 0.0024 (5) | 0.0022 (5) |
| C10 | 0.0285 (8) | 0.0238 (6) | 0.0288 (8) | 0.0031 (6) | -0.0009 (6) | 0.0025 (6) |
| C11 | 0.0295 (8) | 0.0265 (7) | 0.0269 (8) | 0.0009 (6) | -0.0016 (6) | -0.0018 (6) |
| C12 | 0.0243 (7) | 0.0334 (7) | 0.0189 (7) | 0.0012 (6) | 0.0026 (6) | 0.0001 (6) |
| C13 | 0.0281 (8) | 0.0282 (7) | 0.0241 (7) | 0.0049 (6) | 0.0000 (6) | 0.0043 (6) |
| C14 | 0.0267 (7) | 0.0245 (7) | 0.0253 (7) | 0.0021 (6) | -0.0017 (6) | 0.0000 (6) |
| C15 | 0.0362 (9) | 0.0414 (8) | 0.0234 (8) | 0.0020 (7) | -0.0038 (6) | -0.0001 (6) |
| C16 | 0.0377 (9) | 0.0345 (8) | 0.0233 (8) | 0.0044 (7) | -0.0038 (6) | -0.0001 (6) |
| C17 | 0.0237 (7) | 0.0187 (6) | 0.0214 (7) | -0.0014 (5) | 0.0034 (6) | 0.0032 (5) |
| C18 | 0.0194 (7) | 0.0209 (6) | 0.0240 (7) | -0.0002 (5) | -0.0025 (5) | 0.0057 (5) |
| C19 | 0.0177 (7) | 0.0194 (6) | 0.0279 (7) | 0.0008 (5) | 0.0035 (6) | 0.0050 (5) |
| C20 | 0.0174 (6) | 0.0188 (6) | 0.0224 (7) | -0.0012 (5) | 0.0022 (5) | 0.0047 (5) |
| C21 | 0.0184 (7) | 0.0227 (6) | 0.0236 (7) | 0.0008 (5) | 0.0004 (5) | 0.0064 (5) |
| C22 | 0.0201 (7) | 0.0195 (6) | 0.0273 (7) | 0.0030 (5) | 0.0050 (6) | 0.0052 (5) |
| C23 | 0.0205 (7) | 0.0226 (6) | 0.0247 (7) | 0.0009 (5) | 0.0046 (6) | 0.0055 (5) |
| C24 | 0.0196 (7) | 0.0244 (6) | 0.0209 (7) | 0.0009 (5) | 0.0041 (5) | 0.0022 (5) |
| C25 | 0.0244 (7) | 0.0235 (6) | 0.0247 (7) | 0.0026 (5) | 0.0034 (6) | 0.0019 (5) |
| C26 | 0.0255 (7) | 0.0250 (6) | 0.0237 (7) | 0.0001 (5) | 0.0023 (6) | -0.0015 (5) |
| C27 | 0.0245 (7) | 0.0298 (7) | 0.0208 (7) | 0.0002 (6) | 0.0033 (6) | 0.0012 (6) |
| C28 | 0.0275 (8) | 0.0251 (6) | 0.0224 (7) | 0.0028 (6) | 0.0005 (6) | 0.0033 (5) |
| C29 | 0.0260 (7) | 0.0239 (6) | 0.0226 (7) | -0.0002 (5) | 0.0013 (6) | 0.0012 (5) |
| C30 | 0.0342 (9) | 0.0367 (8) | 0.0262 (8) | 0.0037 (7) | -0.0020 (6) | -0.0006 (6) |

Geometric parameters (Å, °)

| | | | |
|--------|-------------|----------|-------------|
| O1—C2 | 1.3658 (15) | C15—H15A | 0.9800 |
| O1—C1 | 1.4289 (16) | C15—H15B | 0.9800 |
| O2—C8 | 1.2273 (14) | C15—H15C | 0.9800 |
| O3—C17 | 1.3663 (15) | C16—H16A | 0.9800 |
| O3—C16 | 1.4290 (16) | C16—H16B | 0.9800 |
| O4—C23 | 1.2269 (14) | C16—H16C | 0.9800 |
| C1—H1A | 0.9800 | C17—C22 | 1.3898 (18) |
| C1—H1B | 0.9800 | C17—C18 | 1.3936 (17) |
| C1—H1C | 0.9800 | C18—C19 | 1.3860 (17) |
| C2—C7 | 1.3897 (18) | C18—H18 | 0.9500 |
| C2—C3 | 1.3935 (17) | C19—C20 | 1.3908 (17) |
| C3—C4 | 1.3863 (17) | C19—H19 | 0.9500 |
| C3—H3 | 0.9500 | C20—C21 | 1.4022 (17) |
| C4—C5 | 1.3910 (17) | C20—C23 | 1.4915 (17) |
| C4—H4 | 0.9500 | C21—C22 | 1.3785 (17) |
| C5—C6 | 1.4000 (16) | C21—H21 | 0.9500 |

| | | | |
|------------|-------------|---------------|-------------|
| C5—C8 | 1.4886 (17) | C22—H22 | 0.9500 |
| C6—C7 | 1.3806 (17) | C23—C24 | 1.5152 (18) |
| C6—H6 | 0.9500 | C24—C25 | 1.5317 (16) |
| C7—H7 | 0.9500 | C24—C29 | 1.5427 (17) |
| C8—C9 | 1.5142 (18) | C24—H24 | 1.0000 |
| C9—C10 | 1.5251 (17) | C25—C26 | 1.5264 (18) |
| C9—C14 | 1.5418 (17) | C25—H25A | 0.9900 |
| C9—H9 | 1.0000 | C25—H25B | 0.9900 |
| C10—C11 | 1.5290 (18) | C26—C27 | 1.5289 (17) |
| C10—H10A | 0.9900 | C26—H26A | 0.9900 |
| C10—H10B | 0.9900 | C26—H26B | 0.9900 |
| C11—C12 | 1.5266 (18) | C27—C30 | 1.5244 (18) |
| C11—H11A | 0.9900 | C27—C28 | 1.5283 (17) |
| C11—H11B | 0.9900 | C27—H27 | 1.0000 |
| C12—C15 | 1.5252 (18) | C28—C29 | 1.5277 (17) |
| C12—C13 | 1.5251 (18) | C28—H28A | 0.9900 |
| C12—H12 | 1.0000 | C28—H28B | 0.9900 |
| C13—C14 | 1.5256 (18) | C29—H29A | 0.9900 |
| C13—H13A | 0.9900 | C29—H29B | 0.9900 |
| C13—H13B | 0.9900 | C30—H30A | 0.9800 |
| C14—H14A | 0.9900 | C30—H30B | 0.9800 |
| C14—H14B | 0.9900 | C30—H30C | 0.9800 |
| | | | |
| C2—O1—C1 | 117.19 (10) | H15B—C15—H15C | 109.5 |
| C17—O3—C16 | 117.13 (10) | O3—C16—H16A | 109.5 |
| O1—C1—H1A | 109.5 | O3—C16—H16B | 109.5 |
| O1—C1—H1B | 109.5 | H16A—C16—H16B | 109.5 |
| H1A—C1—H1B | 109.5 | O3—C16—H16C | 109.5 |
| O1—C1—H1C | 109.5 | H16A—C16—H16C | 109.5 |
| H1A—C1—H1C | 109.5 | H16B—C16—H16C | 109.5 |
| H1B—C1—H1C | 109.5 | O3—C17—C22 | 115.81 (11) |
| O1—C2—C7 | 115.38 (11) | O3—C17—C18 | 124.01 (12) |
| O1—C2—C3 | 124.43 (12) | C22—C17—C18 | 120.19 (12) |
| C7—C2—C3 | 120.18 (12) | C19—C18—C17 | 118.95 (12) |
| C4—C3—C2 | 118.62 (12) | C19—C18—H18 | 120.5 |
| C4—C3—H3 | 120.7 | C17—C18—H18 | 120.5 |
| C2—C3—H3 | 120.7 | C18—C19—C20 | 121.69 (11) |
| C3—C4—C5 | 122.11 (11) | C18—C19—H19 | 119.2 |
| C3—C4—H4 | 118.9 | C20—C19—H19 | 119.2 |
| C5—C4—H4 | 118.9 | C19—C20—C21 | 118.35 (12) |
| C4—C5—C6 | 118.18 (12) | C19—C20—C23 | 118.46 (11) |
| C4—C5—C8 | 118.67 (11) | C21—C20—C23 | 123.16 (12) |
| C6—C5—C8 | 123.13 (12) | C22—C21—C20 | 120.54 (12) |
| C7—C6—C5 | 120.43 (12) | C22—C21—H21 | 119.7 |
| C7—C6—H6 | 119.8 | C20—C21—H21 | 119.7 |
| C5—C6—H6 | 119.8 | C21—C22—C17 | 120.26 (11) |
| C6—C7—C2 | 120.42 (11) | C21—C22—H22 | 119.9 |
| C6—C7—H7 | 119.8 | C17—C22—H22 | 119.9 |

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| C2—C7—H7 | 119.8 | O4—C23—C20 | 119.73 (12) |
| O2—C8—C5 | 119.42 (12) | O4—C23—C24 | 120.28 (12) |
| O2—C8—C9 | 119.92 (11) | C20—C23—C24 | 119.99 (10) |
| C5—C8—C9 | 120.59 (10) | C23—C24—C25 | 111.55 (10) |
| C8—C9—C10 | 112.24 (10) | C23—C24—C29 | 109.11 (11) |
| C8—C9—C14 | 106.83 (11) | C25—C24—C29 | 110.09 (10) |
| C10—C9—C14 | 110.54 (11) | C23—C24—H24 | 108.7 |
| C8—C9—H9 | 109.1 | C25—C24—H24 | 108.7 |
| C10—C9—H9 | 109.1 | C29—C24—H24 | 108.7 |
| C14—C9—H9 | 109.1 | C26—C25—C24 | 111.27 (10) |
| C9—C10—C11 | 111.49 (10) | C26—C25—H25A | 109.4 |
| C9—C10—H10A | 109.3 | C24—C25—H25A | 109.4 |
| C11—C10—H10A | 109.3 | C26—C25—H25B | 109.4 |
| C9—C10—H10B | 109.3 | C24—C25—H25B | 109.4 |
| C11—C10—H10B | 109.3 | H25A—C25—H25B | 108.0 |
| H10A—C10—H10B | 108.0 | C25—C26—C27 | 111.83 (11) |
| C12—C11—C10 | 111.75 (11) | C25—C26—H26A | 109.2 |
| C12—C11—H11A | 109.3 | C27—C26—H26A | 109.3 |
| C10—C11—H11A | 109.3 | C25—C26—H26B | 109.3 |
| C12—C11—H11B | 109.3 | C27—C26—H26B | 109.3 |
| C10—C11—H11B | 109.3 | H26A—C26—H26B | 107.9 |
| H11A—C11—H11B | 107.9 | C30—C27—C28 | 111.83 (10) |
| C15—C12—C13 | 111.66 (11) | C30—C27—C26 | 111.76 (11) |
| C15—C12—C11 | 111.45 (12) | C28—C27—C26 | 109.44 (11) |
| C13—C12—C11 | 109.48 (11) | C30—C27—H27 | 107.9 |
| C15—C12—H12 | 108.0 | C28—C27—H27 | 107.9 |
| C13—C12—H12 | 108.0 | C26—C27—H27 | 107.9 |
| C11—C12—H12 | 108.0 | C29—C28—C27 | 112.45 (10) |
| C12—C13—C14 | 112.18 (10) | C29—C28—H28A | 109.1 |
| C12—C13—H13A | 109.2 | C27—C28—H28A | 109.1 |
| C14—C13—H13A | 109.2 | C29—C28—H28B | 109.1 |
| C12—C13—H13B | 109.2 | C27—C28—H28B | 109.1 |
| C14—C13—H13B | 109.2 | H28A—C28—H28B | 107.8 |
| H13A—C13—H13B | 107.9 | C28—C29—C24 | 111.02 (11) |
| C13—C14—C9 | 111.44 (11) | C28—C29—H29A | 109.4 |
| C13—C14—H14A | 109.3 | C24—C29—H29A | 109.4 |
| C9—C14—H14A | 109.3 | C28—C29—H29B | 109.4 |
| C13—C14—H14B | 109.3 | C24—C29—H29B | 109.4 |
| C9—C14—H14B | 109.3 | H29A—C29—H29B | 108.0 |
| H14A—C14—H14B | 108.0 | C27—C30—H30A | 109.5 |
| C12—C15—H15A | 109.5 | C27—C30—H30B | 109.5 |
| C12—C15—H15B | 109.5 | H30A—C30—H30B | 109.5 |
| H15A—C15—H15B | 109.5 | C27—C30—H30C | 109.5 |
| C12—C15—H15C | 109.5 | H30A—C30—H30C | 109.5 |
| H15A—C15—H15C | 109.5 | H30B—C30—H30C | 109.5 |
| C1—O1—C2—C7 | 173.31 (10) | C16—O3—C17—C22 | 177.02 (11) |
| C1—O1—C2—C3 | -7.90 (17) | C16—O3—C17—C18 | -3.19 (17) |

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| O1—C2—C3—C4 | 179.14 (11) | O3—C17—C18—C19 | 179.81 (11) |
| C7—C2—C3—C4 | -2.13 (17) | C22—C17—C18—C19 | -0.41 (18) |
| C2—C3—C4—C5 | 1.44 (17) | C17—C18—C19—C20 | 1.11 (18) |
| C3—C4—C5—C6 | 0.71 (17) | C18—C19—C20—C21 | -0.56 (18) |
| C3—C4—C5—C8 | -177.36 (10) | C18—C19—C20—C23 | 177.42 (11) |
| C4—C5—C6—C7 | -2.21 (17) | C19—C20—C21—C22 | -0.68 (18) |
| C8—C5—C6—C7 | 175.77 (11) | C23—C20—C21—C22 | -178.56 (11) |
| C5—C6—C7—C2 | 1.54 (17) | C20—C21—C22—C17 | 1.37 (18) |
| O1—C2—C7—C6 | 179.51 (10) | O3—C17—C22—C21 | 178.99 (11) |
| C3—C2—C7—C6 | 0.67 (17) | C18—C17—C22—C21 | -0.81 (18) |
| C4—C5—C8—O2 | 21.98 (17) | C19—C20—C23—O4 | 12.43 (18) |
| C6—C5—C8—O2 | -155.99 (12) | C21—C20—C23—O4 | -169.69 (12) |
| C4—C5—C8—C9 | -155.03 (11) | C19—C20—C23—C24 | -167.66 (11) |
| C6—C5—C8—C9 | 27.00 (17) | C21—C20—C23—C24 | 10.22 (18) |
| O2—C8—C9—C10 | 39.30 (16) | O4—C23—C24—C25 | -29.70 (16) |
| C5—C8—C9—C10 | -143.71 (11) | C20—C23—C24—C25 | 150.39 (11) |
| O2—C8—C9—C14 | -82.01 (14) | O4—C23—C24—C29 | 92.15 (14) |
| C5—C8—C9—C14 | 94.98 (13) | C20—C23—C24—C29 | -87.76 (13) |
| C8—C9—C10—C11 | -173.66 (11) | C23—C24—C25—C26 | 177.09 (10) |
| C14—C9—C10—C11 | -54.51 (14) | C29—C24—C25—C26 | 55.81 (14) |
| C9—C10—C11—C12 | 56.90 (15) | C24—C25—C26—C27 | -57.41 (14) |
| C10—C11—C12—C15 | 179.37 (10) | C25—C26—C27—C30 | -179.46 (10) |
| C10—C11—C12—C13 | -56.60 (14) | C25—C26—C27—C28 | 56.11 (14) |
| C15—C12—C13—C14 | -179.84 (11) | C30—C27—C28—C29 | 179.88 (11) |
| C11—C12—C13—C14 | 56.25 (14) | C26—C27—C28—C29 | -55.74 (14) |
| C12—C13—C14—C9 | -55.65 (14) | C27—C28—C29—C24 | 56.02 (14) |
| C8—C9—C14—C13 | 176.27 (10) | C23—C24—C29—C28 | -177.67 (10) |
| C10—C9—C14—C13 | 53.89 (13) | C25—C24—C29—C28 | -54.94 (13) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|----------------------------|-------------|---------------|-----------------------|-------------------------|
| C9—H9...O2 ⁱ | 1.00 | 2.57 | 3.4694 (18) | 150 |
| C24—H24...O4 ⁱⁱ | 1.00 | 2.60 | 3.5244 (18) | 154 |

Symmetry codes: (i) $x-1, y, z$; (ii) $x+1, y, z$.