

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## 4,4'-Diethyl-2,2'-[(N-cyclohexylimino)-bis(methylene)]diphenol

Worawat Wattanathana,<sup>a</sup> Chatchai Veranitisagul,<sup>b</sup> Attaphon Kaewvilai,<sup>c</sup> Apirat Laobuthee<sup>c</sup> and Nattamon Koonsaeng<sup>a\*</sup>

<sup>a</sup>Department of Chemistry, Faculty of Science, Kasetsart University, Bangkok 10900, Thailand, <sup>b</sup>Department of Materials and Metallurgical Engineering, Faculty of Engineering, Rajamangala University of Technology Thanyaburi, Pathumthani 12110, Thailand, and <sup>c</sup>Department of Materials Engineering, Faculty of Engineering and Center of Advanced Studies in Industrial Technology, Kasetsart University, Bangkok 10900, Thailand

Correspondence e-mail: fscinmk@ku.ac.th

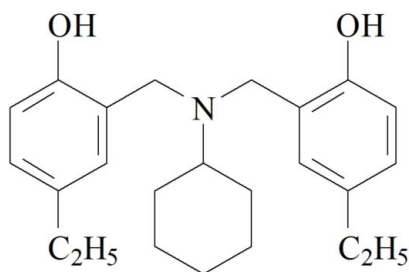
Received 7 September 2012; accepted 24 September 2012

Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.060;  $wR$  factor = 0.185; data-to-parameter ratio = 21.2.

The title compound,  $\text{C}_{24}\text{H}_{33}\text{NO}_2$ , exhibits an intramolecular hydrogen bond between a phenol  $-\text{OH}$  group and the N atom. In the crystal, molecules are connected by pairs of  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds.

## Related literature

For details of the synthesis of  $N,N$ -bis(2-hydroxybenzyl)alkylamines, see: Laobuthee *et al.* (2003). For their metal-responsive properties, see: Veranitisagul *et al.* (2011). For their use in the synthesis of macrocyclic molecules, see: Rungsimanon *et al.* (2008).



## Experimental

## Crystal data

$\text{C}_{24}\text{H}_{33}\text{NO}_2$   
 $M_r = 367.51$

Triclinic,  $P\bar{1}$   
 $a = 9.4224$  (12) Å

$b = 10.3799$  (16) Å  
 $c = 11.8143$  (18) Å  
 $\alpha = 82.140$  (4)°  
 $\beta = 73.960$  (4)°  
 $\gamma = 81.676$  (4)°  
 $V = 1093.1$  (3) Å<sup>3</sup>

$Z = 2$   
Mo  $K\alpha$  radiation  
 $\mu = 0.07$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.90 \times 0.44 \times 0.24$  mm

## Data collection

Bruker APEXII CCD  
diffractometer  
8721 measured reflections

5350 independent reflections  
3648 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.022$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.060$   
 $wR(F^2) = 0.185$   
 $S = 1.12$   
5350 reflections  
252 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.27$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.23$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1}\cdots\text{O2}$	0.94 (3)	2.52 (2)	3.156 (2)	124.8 (18)
$\text{O1}-\text{H1}\cdots\text{N}$	0.94 (3)	1.79 (3)	2.6352 (19)	147.8 (19)
$\text{O2}-\text{H2}\cdots\text{O1}^i$	0.88 (3)	1.84 (3)	2.708 (2)	168 (3)

Symmetry code: (i)  $-x + 1, -y, -z + 1$ .

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97 and PUBLICIF (Westrip, 2010).

This work was supported by the Thailand Research Fund, Office of the Higher Education Commission, and Rajamangala University of Technology Thanyaburi (grant No. MRG5480046). The authors would like to thank the Department of Chemistry, Faculty of Science, Kasetsart University, and the Center of Advanced Studies in Industrial Technology, Faculty of Engineering, Kasetsart University for partial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: GG2100).

## References

- Bruker (2007). APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.  
Laobuthee, A., Ishida, H. & Chirachanchai, S. (2003). *J. Incl. Phenom. Macro.* **47**, 179–185.  
Rungsimanon, T., Laobuthee, A., Miyata, M. & Chirachanchai, S. (2008). *J. Incl. Phenom. Macro.* **62**, 333–338.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
Veranitisagul, C., Kaewvilai, A., Sangngern, S., Wattanathana, W., Suramitr, S., Koonsaeng, N. & Laobuthee, A. (2011). *Int. J. Mol. Sci.* **12**, 4365–4377.  
Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

## supporting information

*Acta Cryst.* (2012). E68, o3050 [https://doi.org/10.1107/S1600536812040342]

**4,4'-Diethyl-2,2'-[(*N*-cyclohexylimino)bis(methylene)]diphenol**

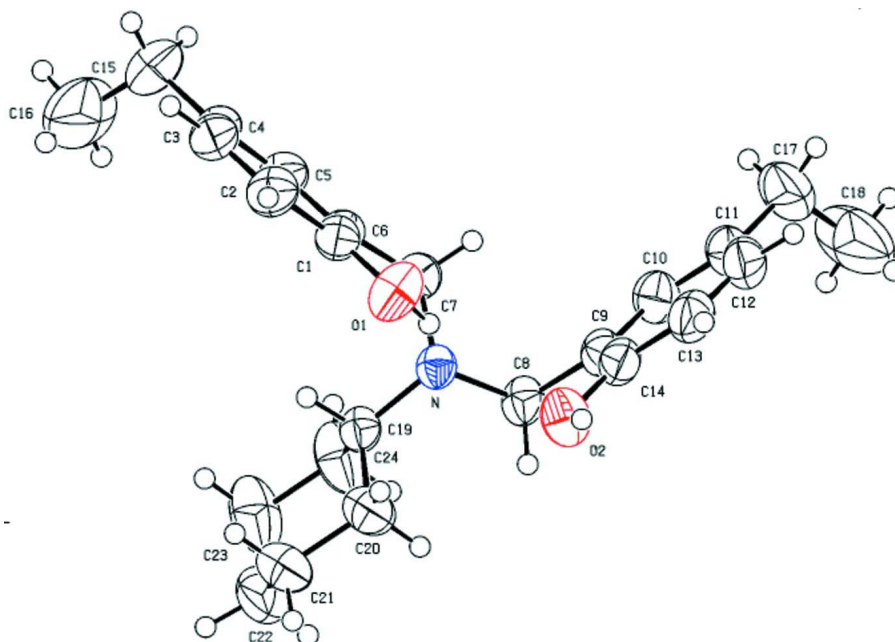
**Worawat Wattanathana, Chatchai Veranitisagul, Attaphon Kaewvilai, Apirat Laobuthee and Nattamon Koonsaeng**

**S1. Experimental**

The preparation of the title compound was reported elsewhere (Laobuthee *et al.*, 2003). Colorless blocks were recrystallized from 2-propanol solution.

**S2. Refinement**

All H atoms of the compound were placed in the calculated positions with C—H = 0.96 Å and included in the final cycles of refinement in a riding model,  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{H})$ . Except H atom of O atoms were located in different fourier map and restrained to their hosts.



**Figure 1**

Molecular structure of the title compound (arbitrary spheres for the H atoms).

**4,4'-Diethyl-2,2'-[(*N*-cyclohexylimino)bis(methylene)]diphenol***Crystal data*

$\text{C}_{24}\text{H}_{33}\text{NO}_2$

$M_r = 367.51$

Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 9.4224$  (12) Å

$b = 10.3799$  (16) Å

$c = 11.8143 (18) \text{ \AA}$   
 $\alpha = 82.140 (4)^\circ$   
 $\beta = 73.960 (4)^\circ$   
 $\gamma = 81.676 (4)^\circ$   
 $V = 1093.1 (3) \text{ \AA}^3$   
 $Z = 2$   
 $F(000) = 400.0$   
 $D_x = 1.117 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
 Cell parameters from 3314 reflections  
 $\theta = 2.6\text{--}28.2^\circ$   
 $\mu = 0.07 \text{ mm}^{-1}$   
 $T = 296 \text{ K}$   
 Block, colourless  
 $0.90 \times 0.44 \times 0.24 \text{ mm}$

*Data collection*

Bruker APEXII CCD  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 $\varphi$  and  $\omega$  scans  
 8721 measured reflections  
 5350 independent reflections

3648 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.022$   
 $\theta_{\text{max}} = 28.2^\circ$ ,  $\theta_{\text{min}} = 1.8^\circ$   
 $h = -12 \rightarrow 12$   
 $k = -13 \rightarrow 13$   
 $l = -15 \rightarrow 15$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.060$   
 $wR(F^2) = 0.185$   
 $S = 1.12$   
 5350 reflections  
 252 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H atoms treated by a mixture of independent  
 and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.085P)^2 + 0.1411P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.27 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.23 \text{ e \AA}^{-3}$

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
N	0.15859 (13)	0.23370 (12)	0.54691 (10)	0.0404 (3)
O1	0.42889 (14)	0.18278 (12)	0.57669 (13)	0.0633 (4)
O2	0.32304 (15)	-0.01044 (13)	0.44034 (11)	0.0627 (4)
C6	0.29872 (16)	0.39939 (15)	0.58608 (13)	0.0436 (3)
C8	0.09657 (17)	0.19375 (17)	0.45693 (14)	0.0487 (4)
H8A	0.0189	0.2606	0.4417	0.058*
H8B	0.0520	0.1132	0.4875	0.058*
C9	0.21273 (17)	0.17318 (17)	0.34271 (14)	0.0479 (4)
C19	0.06585 (15)	0.21269 (15)	0.67028 (13)	0.0432 (3)

H19A	0.1157	0.2486	0.7196	0.052*
C5	0.2872 (2)	0.52251 (17)	0.62205 (16)	0.0560 (4)
H5A	0.2120	0.5852	0.6064	0.067*
C1	0.41114 (16)	0.30665 (16)	0.61073 (14)	0.0468 (4)
C2	0.50701 (18)	0.3377 (2)	0.67020 (17)	0.0594 (5)
H2A	0.5813	0.2750	0.6872	0.071*
C14	0.32556 (18)	0.06974 (17)	0.33839 (15)	0.0509 (4)
C10	0.2111 (2)	0.2556 (2)	0.23952 (16)	0.0619 (5)
H10A	0.1357	0.3246	0.2419	0.074*
C13	0.4330 (2)	0.0513 (2)	0.23316 (17)	0.0639 (5)
H13A	0.5090	-0.0171	0.2303	0.077*
C4	0.3830 (2)	0.55644 (19)	0.68038 (17)	0.0634 (5)
C3	0.4923 (2)	0.4615 (2)	0.70427 (17)	0.0634 (5)
H3A	0.5574	0.4816	0.7442	0.076*
C12	0.4272 (2)	0.1345 (2)	0.13275 (18)	0.0745 (6)
H12A	0.4992	0.1200	0.0625	0.089*
C11	0.3180 (3)	0.2386 (2)	0.13297 (17)	0.0729 (6)
C20	0.0648 (3)	0.0674 (2)	0.71237 (18)	0.0735 (6)
H20A	0.1661	0.0253	0.6976	0.088*
H20B	0.0130	0.0273	0.6683	0.088*
C24	-0.0908 (2)	0.2822 (3)	0.69530 (19)	0.0811 (7)
H24A	-0.1468	0.2483	0.6505	0.097*
H24B	-0.0878	0.3750	0.6708	0.097*
C21	-0.0116 (3)	0.0471 (2)	0.8443 (2)	0.0883 (7)
H21A	-0.0157	-0.0457	0.8684	0.106*
H21B	0.0460	0.0794	0.8887	0.106*
C15	0.3654 (3)	0.6929 (2)	0.7176 (3)	0.0950 (8)
H15A	0.4601	0.7105	0.7263	0.114*
H15B	0.3409	0.7553	0.6550	0.114*
C23	-0.1674 (2)	0.2610 (3)	0.8282 (2)	0.0953 (9)
H23A	-0.1171	0.3032	0.8719	0.114*
H23B	-0.2694	0.3015	0.8428	0.114*
C17	0.3115 (4)	0.3322 (3)	0.0225 (2)	0.1067 (9)
H17A	0.2822	0.4205	0.0453	0.128*
H17B	0.4102	0.3302	-0.0312	0.128*
C22	-0.1661 (3)	0.1168 (3)	0.8725 (2)	0.0993 (9)
H22A	-0.2087	0.1074	0.9575	0.119*
H22B	-0.2272	0.0769	0.8362	0.119*
C16	0.2541 (4)	0.7143 (3)	0.8259 (3)	0.1463 (15)
H16A	0.2500	0.8027	0.8432	0.219*
H16B	0.2785	0.6548	0.8892	0.219*
H16C	0.1592	0.6998	0.8178	0.219*
C18	0.2088 (4)	0.3018 (5)	-0.0399 (3)	0.1597 (18)
H18A	0.2107	0.3637	-0.1083	0.240*
H18B	0.1100	0.3064	0.0118	0.240*
H18C	0.2379	0.2151	-0.0641	0.240*
C7	0.20041 (18)	0.36794 (15)	0.51474 (15)	0.0490 (4)
H7A	0.1110	0.4295	0.5272	0.059*

H7B	0.2520	0.3783	0.4313	0.059*
H1	0.347 (3)	0.176 (2)	0.548 (2)	0.107 (9)*
H2	0.398 (3)	-0.073 (3)	0.430 (2)	0.095 (8)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N	0.0376 (6)	0.0464 (7)	0.0397 (6)	-0.0025 (5)	-0.0143 (5)	-0.0062 (5)
O1	0.0465 (6)	0.0623 (8)	0.0889 (10)	0.0105 (5)	-0.0289 (6)	-0.0287 (7)
O2	0.0614 (8)	0.0638 (8)	0.0573 (8)	0.0116 (6)	-0.0139 (6)	-0.0103 (6)
C6	0.0435 (8)	0.0465 (8)	0.0403 (8)	-0.0091 (6)	-0.0092 (6)	-0.0023 (6)
C8	0.0416 (8)	0.0629 (10)	0.0467 (8)	-0.0035 (7)	-0.0193 (6)	-0.0096 (7)
C9	0.0465 (8)	0.0600 (10)	0.0430 (8)	-0.0093 (7)	-0.0169 (7)	-0.0105 (7)
C19	0.0360 (7)	0.0535 (9)	0.0419 (8)	-0.0032 (6)	-0.0117 (6)	-0.0092 (6)
C5	0.0613 (10)	0.0481 (9)	0.0579 (10)	-0.0094 (8)	-0.0136 (8)	-0.0038 (8)
C1	0.0371 (7)	0.0554 (9)	0.0481 (9)	-0.0062 (6)	-0.0086 (6)	-0.0098 (7)
C2	0.0418 (8)	0.0740 (12)	0.0662 (11)	-0.0100 (8)	-0.0165 (8)	-0.0117 (9)
C14	0.0499 (9)	0.0589 (10)	0.0484 (9)	-0.0097 (7)	-0.0137 (7)	-0.0156 (7)
C10	0.0668 (11)	0.0746 (12)	0.0518 (10)	-0.0144 (9)	-0.0263 (9)	-0.0028 (9)
C13	0.0523 (10)	0.0784 (13)	0.0632 (12)	-0.0129 (9)	-0.0055 (8)	-0.0291 (10)
C4	0.0730 (12)	0.0642 (11)	0.0548 (10)	-0.0314 (10)	-0.0042 (9)	-0.0128 (8)
C3	0.0548 (10)	0.0844 (14)	0.0581 (11)	-0.0291 (10)	-0.0136 (8)	-0.0114 (9)
C12	0.0691 (12)	0.1093 (18)	0.0493 (11)	-0.0365 (12)	-0.0006 (9)	-0.0245 (11)
C11	0.0801 (14)	0.1002 (16)	0.0462 (10)	-0.0369 (12)	-0.0178 (9)	-0.0033 (10)
C20	0.0908 (15)	0.0628 (12)	0.0577 (11)	-0.0216 (10)	0.0021 (10)	-0.0050 (9)
C24	0.0465 (10)	0.128 (2)	0.0611 (12)	0.0191 (11)	-0.0117 (9)	-0.0200 (12)
C21	0.1108 (19)	0.0852 (16)	0.0598 (13)	-0.0401 (14)	0.0038 (12)	0.0003 (11)
C15	0.122 (2)	0.0749 (15)	0.0982 (18)	-0.0377 (14)	-0.0226 (16)	-0.0284 (13)
C23	0.0506 (11)	0.163 (3)	0.0645 (13)	0.0154 (14)	-0.0041 (9)	-0.0361 (16)
C17	0.130 (2)	0.145 (3)	0.0532 (13)	-0.0617 (19)	-0.0274 (14)	0.0172 (14)
C22	0.0791 (16)	0.167 (3)	0.0577 (13)	-0.0605 (17)	0.0003 (11)	-0.0198 (15)
C16	0.151 (3)	0.111 (2)	0.162 (3)	-0.030 (2)	0.020 (3)	-0.077 (2)
C18	0.156 (3)	0.253 (5)	0.091 (2)	-0.105 (3)	-0.064 (2)	0.063 (3)
C7	0.0528 (9)	0.0464 (9)	0.0505 (9)	-0.0030 (7)	-0.0218 (7)	-0.0001 (7)

*Geometric parameters (Å, °)*

N—C7	1.474 (2)	C12—C11	1.380 (3)
N—C8	1.4756 (19)	C12—H12A	0.9300
N—C19	1.4833 (19)	C11—C17	1.527 (3)
O1—C1	1.373 (2)	C20—C21	1.525 (3)
O1—H1	0.94 (3)	C20—H20A	0.9700
O2—C14	1.364 (2)	C20—H20B	0.9700
O2—H2	0.88 (3)	C24—C23	1.536 (3)
C6—C5	1.381 (2)	C24—H24A	0.9700
C6—C1	1.390 (2)	C24—H24B	0.9700
C6—C7	1.506 (2)	C21—C22	1.498 (4)
C8—C9	1.503 (2)	C21—H21A	0.9700

C8—H8A	0.9700	C21—H21B	0.9700
C8—H8B	0.9700	C15—C16	1.433 (4)
C9—C10	1.393 (2)	C15—H15A	0.9700
C9—C14	1.393 (2)	C15—H15B	0.9700
C19—C24	1.515 (2)	C23—C22	1.516 (4)
C19—C20	1.522 (2)	C23—H23A	0.9700
C19—H19A	0.9800	C23—H23B	0.9700
C5—C4	1.384 (3)	C17—C18	1.458 (4)
C5—H5A	0.9300	C17—H17A	0.9700
C1—C2	1.382 (2)	C17—H17B	0.9700
C2—C3	1.375 (3)	C22—H22A	0.9700
C2—H2A	0.9300	C22—H22B	0.9700
C14—C13	1.386 (2)	C16—H16A	0.9600
C10—C11	1.393 (3)	C16—H16B	0.9600
C10—H10A	0.9300	C16—H16C	0.9600
C13—C12	1.377 (3)	C18—H18A	0.9600
C13—H13A	0.9300	C18—H18B	0.9600
C4—C3	1.379 (3)	C18—H18C	0.9600
C4—C15	1.514 (3)	C7—H7A	0.9700
C3—H3A	0.9300	C7—H7B	0.9700
C7—N—C8	111.05 (12)	C21—C20—H20B	109.5
C7—N—C19	112.92 (12)	H20A—C20—H20B	108.1
C8—N—C19	114.48 (11)	C19—C24—C23	109.84 (17)
C1—O1—H1	106.7 (16)	C19—C24—H24A	109.7
C14—O2—H2	111.5 (16)	C23—C24—H24A	109.7
C5—C6—C1	117.94 (15)	C19—C24—H24B	109.7
C5—C6—C7	121.62 (15)	C23—C24—H24B	109.7
C1—C6—C7	120.30 (14)	H24A—C24—H24B	108.2
N—C8—C9	112.25 (12)	C22—C21—C20	111.5 (2)
N—C8—H8A	109.2	C22—C21—H21A	109.3
C9—C8—H8A	109.2	C20—C21—H21A	109.3
N—C8—H8B	109.2	C22—C21—H21B	109.3
C9—C8—H8B	109.2	C20—C21—H21B	109.3
H8A—C8—H8B	107.9	H21A—C21—H21B	108.0
C10—C9—C14	118.44 (16)	C16—C15—C4	114.7 (2)
C10—C9—C8	121.57 (16)	C16—C15—H15A	108.6
C14—C9—C8	119.99 (15)	C4—C15—H15A	108.6
N—C19—C24	116.08 (14)	C16—C15—H15B	108.6
N—C19—C20	111.12 (13)	C4—C15—H15B	108.6
C24—C19—C20	110.84 (17)	H15A—C15—H15B	107.6
N—C19—H19A	106.0	C22—C23—C24	111.7 (2)
C24—C19—H19A	106.0	C22—C23—H23A	109.3
C20—C19—H19A	106.0	C24—C23—H23A	109.3
C6—C5—C4	122.68 (17)	C22—C23—H23B	109.3
C6—C5—H5A	118.7	C24—C23—H23B	109.3
C4—C5—H5A	118.7	H23A—C23—H23B	107.9
O1—C1—C2	118.84 (15)	C18—C17—C11	113.9 (2)

O1—C1—C6	120.72 (14)	C18—C17—H17A	108.8
C2—C1—C6	120.44 (16)	C11—C17—H17A	108.8
C3—C2—C1	119.93 (18)	C18—C17—H17B	108.8
C3—C2—H2A	120.0	C11—C17—H17B	108.8
C1—C2—H2A	120.0	H17A—C17—H17B	107.7
O2—C14—C13	122.99 (17)	C21—C22—C23	111.59 (18)
O2—C14—C9	117.11 (15)	C21—C22—H22A	109.3
C13—C14—C9	119.90 (17)	C23—C22—H22A	109.3
C11—C10—C9	122.5 (2)	C21—C22—H22B	109.3
C11—C10—H10A	118.8	C23—C22—H22B	109.3
C9—C10—H10A	118.8	H22A—C22—H22B	108.0
C12—C13—C14	119.99 (19)	C15—C16—H16A	109.5
C12—C13—H13A	120.0	C15—C16—H16B	109.5
C14—C13—H13A	120.0	H16A—C16—H16B	109.5
C3—C4—C5	117.73 (17)	C15—C16—H16C	109.5
C3—C4—C15	121.8 (2)	H16A—C16—H16C	109.5
C5—C4—C15	120.5 (2)	H16B—C16—H16C	109.5
C4—C3—C2	121.27 (17)	C17—C18—H18A	109.5
C4—C3—H3A	119.4	C17—C18—H18B	109.5
C2—C3—H3A	119.4	H18A—C18—H18B	109.5
C11—C12—C13	122.14 (18)	C17—C18—H18C	109.5
C11—C12—H12A	118.9	H18A—C18—H18C	109.5
C13—C12—H12A	118.9	H18B—C18—H18C	109.5
C12—C11—C10	117.07 (19)	N—C7—C6	112.32 (12)
C12—C11—C17	122.6 (2)	N—C7—H7A	109.1
C10—C11—C17	120.4 (2)	C6—C7—H7A	109.1
C19—C20—C21	110.63 (17)	N—C7—H7B	109.1
C19—C20—H20A	109.5	C6—C7—H7B	109.1
C21—C20—H20A	109.5	H7A—C7—H7B	107.9
C19—C20—H20B	109.5		
C7—N—C8—C9	70.20 (17)	C6—C5—C4—C15	179.91 (18)
C19—N—C8—C9	-160.49 (13)	C5—C4—C3—C2	0.7 (3)
N—C8—C9—C10	-112.40 (16)	C15—C4—C3—C2	179.84 (19)
N—C8—C9—C14	67.57 (19)	C1—C2—C3—C4	0.1 (3)
C7—N—C19—C24	69.65 (18)	C14—C13—C12—C11	1.0 (3)
C8—N—C19—C24	-58.74 (19)	C13—C12—C11—C10	-1.0 (3)
C7—N—C19—C20	-162.49 (14)	C13—C12—C11—C17	179.35 (19)
C8—N—C19—C20	69.13 (18)	C9—C10—C11—C12	0.6 (3)
C1—C6—C5—C4	0.4 (2)	C9—C10—C11—C17	-179.71 (18)
C7—C6—C5—C4	-175.33 (16)	N—C19—C20—C21	171.75 (17)
C5—C6—C1—O1	-179.63 (15)	C24—C19—C20—C21	-57.6 (2)
C7—C6—C1—O1	-3.9 (2)	N—C19—C24—C23	-174.90 (18)
C5—C6—C1—C2	0.5 (2)	C20—C19—C24—C23	57.1 (2)
C7—C6—C1—C2	176.22 (15)	C19—C20—C21—C22	56.1 (3)
O1—C1—C2—C3	179.41 (16)	C3—C4—C15—C16	-97.6 (3)
C6—C1—C2—C3	-0.7 (3)	C5—C4—C15—C16	81.5 (3)
C10—C9—C14—O2	-179.05 (14)	C19—C24—C23—C22	-55.7 (3)

C8—C9—C14—O2	1.0 (2)	C12—C11—C17—C18	97.7 (3)
C10—C9—C14—C13	0.3 (2)	C10—C11—C17—C18	-81.9 (4)
C8—C9—C14—C13	-179.66 (14)	C20—C21—C22—C23	-54.7 (3)
C14—C9—C10—C11	-0.3 (3)	C24—C23—C22—C21	54.8 (3)
C8—C9—C10—C11	179.68 (16)	C8—N—C7—C6	-165.28 (12)
O2—C14—C13—C12	178.65 (16)	C19—N—C7—C6	64.58 (16)
C9—C14—C13—C12	-0.7 (3)	C5—C6—C7—N	-143.79 (15)
C6—C5—C4—C3	-1.0 (3)	C1—C6—C7—N	40.6 (2)

*Hydrogen-bond geometry (Å, °)*

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
O1—H1...O2	0.94 (3)	2.52 (2)	3.156 (2)	124.8 (18)
O1—H1...N	0.94 (3)	1.79 (3)	2.6352 (19)	147.8 (19)
O2—H2...O1 <sup>i</sup>	0.88 (3)	1.84 (3)	2.708 (2)	168 (3)

Symmetry code: (i)  $-x+1, -y, -z+1$ .