

1-Methoxy-11H-benzo[b]fluoren-11-one

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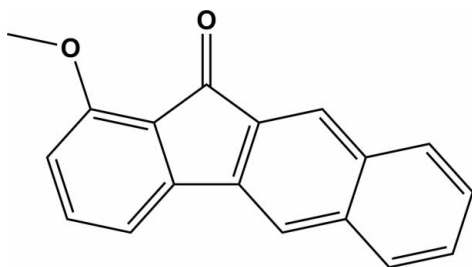
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Key indicators: single-crystal X-ray study; $T = 299$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.083; wR factor = 0.250; data-to-parameter ratio = 14.0.

In the title compound, $\text{C}_{18}\text{H}_{12}\text{O}_2$, the non-H atoms are nearly coplanar, the maximum atomic deviation being 0.113 (2) Å. π - π stacking is observed in the crystal structure, the shortest centroid-centroid distance being 3.5983 (19) Å. The molecular packing is further stabilized by weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds, forming an infinite chain along [100] and generating a $C(6)$ motif.

Related literature

For the preparation of the title compound, see: Tang *et al.* (2011). For applications of indanone derivatives, see: Borbone *et al.* (2011); Borge *et al.* (2010); Cai & Dolbier (2005); Cui *et al.* (2009); Fu & Wang (2008); Li *et al.* (2009); Rahman *et al.* (2011); Sousa *et al.* (2011); Yu *et al.* (2011). For related structures, see: Chang & Chen (2012); Chen *et al.* (2011a,b).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{12}\text{O}_2$
 $M_r = 260.28$
 Monoclinic, $P2_1/c$
 $a = 7.7202$ (3) Å
 $b = 9.2462$ (4) Å
 $c = 18.0294$ (8) Å
 $\beta = 99.935$ (2)°

$V = 1267.68$ (9) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 299$ K
 $0.50 \times 0.47 \times 0.20$ mm

Data collection

Bruker SMART CCD detector
 diffractometer
 Absorption correction: multi-scan
 (SADABS; Bruker, 2001)
 $T_{\min} = 0.957$, $T_{\max} = 0.983$

11580 measured reflections
 2559 independent reflections
 1881 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.083$
 $wR(F^2) = 0.250$
 $S = 1.11$
 2559 reflections

183 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.49$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.42$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C7}-\text{H7}\cdots\text{O1}^i$	0.93	2.57	3.299 (5)	135
$\text{C11}-\text{H11}\cdots\text{O1}^{\text{ii}}$	0.93	2.55	3.298 (4)	138

Symmetry codes: (i) $-x + 2, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x - 1, y, z$.

Data collection: *SMART* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *WinGX* (Farrugia, 2012).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5629).

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supporting information

Acta Cryst. (2013). E69, o79 [https://doi.org/10.1107/S1600536812050076]

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S1. Comment

Indanone and its derivatives are some of the most widely used organic compounds (Rahman *et al.*, 2011). They are used as pigments and dyes (Cui *et al.*, 2009; Li *et al.*, 2009), intermediates in organic synthesis (Borbone *et al.*, 2011; Borge *et al.*, 2010; Fu & Wang, 2008; Yu *et al.*, 2011) and exhibit a wide variety of biological activities (Sousa *et al.*, 2011). In addition, 1-indanones were important precursors in the regiospecific synthesis of 2-fluoro-1-naphthols (Cai *et al.*, 2005).

The molecular structure of the title compound comprises a 7-methoxy-1-indanone unit having a naphthalene ring fused on one side (Fig. 1). The 1-indanone moiety is essentially planar (r.m.s. deviation = 0.0075 Å), which is consistent with previous studies (Chang & Chen, 2012; Chen *et al.*, 2011*a,b*). π – π stacking is observed between the tetracyclic plane and its adjacent one, the closest centroid-centroid distance being 3.5983 (19) Å [symmetry code: 2 - *x*, -*y*, 2 - *z*]. The molecular packing (Fig. 2) is further stabilized by weak non-classical intermolecular C—H \cdots O hydrogen bonds (Table 1).

S2. Experimental

The title compound was synthesized according to the literature (Tang *et al.*, 2011). Yellow parallelepiped-shaped crystals suitable for the crystallographic studies reported here were isolated over a period of six weeks by slow evaporation from a chloroform solution.

S3. Refinement

The C bound H atoms were positioned geometrically (C—H = 0.93–0.96 Å) and allowed to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

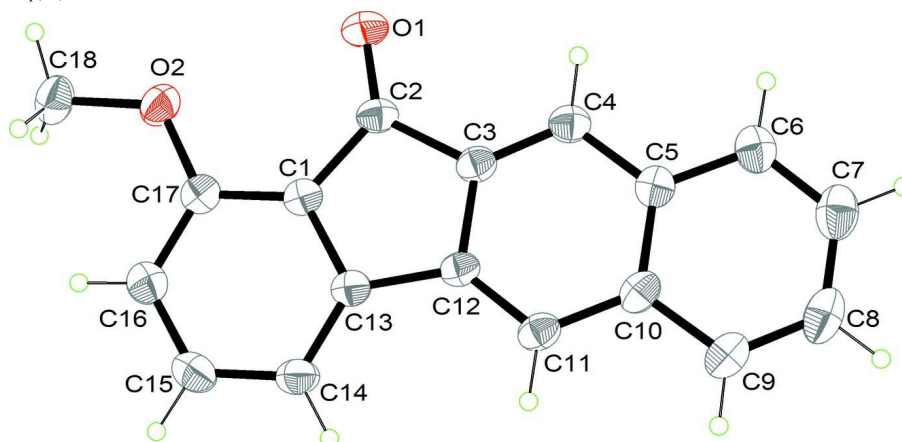


Figure 1

The molecular structure of the title compound, showing 50% probability displacement ellipsoids.

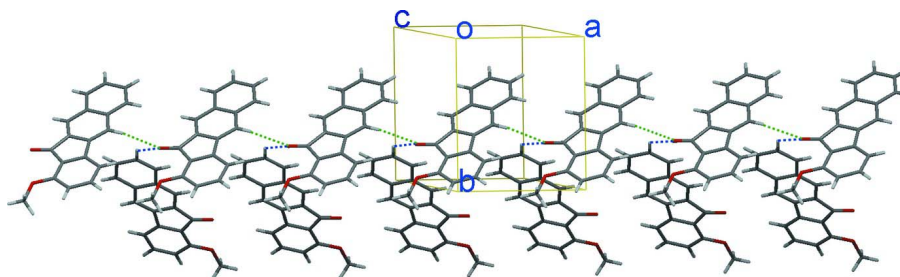


Figure 2

A section of the crystal packing of the title compound, viewed down the c axis. Blue and green dashed lines denote the intermolecular C7—H7 \cdots O1 and C11—H11 \cdots O1 hydrogen bonds, respectively.

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 $\beta = 99.935$ (2)°
 $V = 1267.68$ (9) Å³
 $Z = 4$

$F(000) = 544$
 $D_x = 1.364$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 6867 reflections
 $\theta = 2.7$ – 26.4 °
 $\mu = 0.09$ mm⁻¹
 $T = 299$ K
 Parallelepiped, yellow
 $0.50 \times 0.47 \times 0.20$ mm

Data collection

Bruker SMART CCD detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2001)
 $T_{\min} = 0.957$, $T_{\max} = 0.983$

11580 measured reflections
 2559 independent reflections
 1881 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$
 $\theta_{\max} = 26.4$ °, $\theta_{\min} = 2.3$ °
 $h = -9 \rightarrow 9$
 $k = -11 \rightarrow 10$
 $l = -22 \rightarrow 21$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.083$
 $wR(F^2) = 0.250$
 $S = 1.11$
 2559 reflections
 183 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods
 Secondary atom site location: difference Fourier
 map

Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.1046P)^2 + 1.682P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.49$ e Å⁻³
 $\Delta\rho_{\min} = -0.42$ e Å⁻³
 Extinction correction: SHELXL97 (Sheldrick,
 2008), $F_c^* = kF_c[1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.041 (7)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	1.2838 (3)	0.3051 (3)	0.92283 (15)	0.0675 (8)
O2	1.3420 (3)	0.0972 (3)	1.04942 (14)	0.0571 (7)
C1	1.0841 (4)	0.2264 (3)	1.00540 (17)	0.0428 (7)
C2	1.1419 (4)	0.3058 (4)	0.94252 (18)	0.0464 (8)
C3	0.9836 (4)	0.3911 (3)	0.90635 (17)	0.0439 (7)
C4	0.9640 (4)	0.4820 (4)	0.84656 (18)	0.0490 (8)
H4	1.0571	0.4978	0.8209	0.059*
C5	0.8006 (4)	0.5531 (3)	0.82323 (17)	0.0461 (8)
C6	0.7726 (5)	0.6517 (4)	0.7623 (2)	0.0579 (9)
H6	0.8634	0.6712	0.7360	0.069*
C7	0.6147 (6)	0.7183 (4)	0.7419 (2)	0.0683 (11)
H7	0.5990	0.7835	0.7020	0.082*
C8	0.4773 (6)	0.6901 (4)	0.7796 (2)	0.0694 (11)
H8	0.3696	0.7359	0.7646	0.083*
C9	0.4978 (5)	0.5959 (4)	0.8385 (2)	0.0578 (9)
H9	0.4038	0.5780	0.8633	0.069*
C10	0.6602 (4)	0.5249 (3)	0.86266 (19)	0.0475 (8)
C11	0.6857 (4)	0.4280 (3)	0.92502 (19)	0.0473 (8)
H11	0.5937	0.4089	0.9508	0.057*
C12	0.8438 (4)	0.3637 (3)	0.94670 (17)	0.0401 (7)
C13	0.9079 (4)	0.2608 (3)	1.00807 (17)	0.0420 (7)
C14	0.8220 (4)	0.2001 (4)	1.0611 (2)	0.0522 (8)
H14	0.7052	0.2224	1.0625	0.063*
C15	0.9155 (5)	0.1040 (4)	1.1126 (2)	0.0589 (9)
H15	0.8600	0.0626	1.1493	0.071*
C16	1.0877 (5)	0.0687 (4)	1.1106 (2)	0.0544 (9)
H16	1.1465	0.0040	1.1458	0.065*
C17	1.1749 (4)	0.1289 (4)	1.05646 (18)	0.0471 (8)
C18	1.4337 (5)	-0.0107 (4)	1.0985 (2)	0.0627 (10)
H18A	1.4545	0.0248	1.1493	0.094*
H18B	1.5440	-0.0321	1.0832	0.094*
H18C	1.3638	-0.0971	1.0958	0.094*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0431 (14)	0.087 (2)	0.0777 (18)	0.0056 (12)	0.0246 (12)	0.0121 (15)
O2	0.0438 (13)	0.0598 (16)	0.0666 (16)	0.0075 (11)	0.0062 (11)	0.0054 (12)
C1	0.0410 (16)	0.0411 (16)	0.0463 (16)	-0.0028 (13)	0.0079 (12)	-0.0059 (13)
C2	0.0366 (16)	0.0548 (19)	0.0490 (17)	-0.0035 (13)	0.0105 (13)	-0.0041 (14)
C3	0.0381 (16)	0.0436 (17)	0.0502 (17)	-0.0034 (12)	0.0086 (12)	-0.0051 (13)
C4	0.0439 (17)	0.053 (2)	0.0506 (18)	-0.0039 (14)	0.0105 (13)	-0.0026 (14)
C5	0.0506 (18)	0.0394 (17)	0.0468 (17)	-0.0019 (13)	0.0040 (13)	-0.0069 (13)
C6	0.069 (2)	0.051 (2)	0.0510 (19)	0.0002 (17)	0.0046 (16)	-0.0012 (15)
C7	0.085 (3)	0.054 (2)	0.060 (2)	0.012 (2)	-0.003 (2)	-0.0058 (18)
C8	0.070 (3)	0.057 (2)	0.073 (2)	0.0191 (19)	-0.009 (2)	-0.0058 (19)
C9	0.0480 (19)	0.056 (2)	0.067 (2)	0.0071 (16)	0.0021 (16)	-0.0088 (17)
C10	0.0422 (17)	0.0401 (17)	0.0579 (18)	0.0022 (13)	0.0023 (13)	-0.0100 (14)
C11	0.0393 (16)	0.0428 (18)	0.0606 (19)	-0.0022 (13)	0.0105 (13)	-0.0071 (14)
C12	0.0355 (15)	0.0377 (16)	0.0478 (16)	-0.0042 (12)	0.0095 (12)	-0.0070 (12)
C13	0.0426 (16)	0.0360 (16)	0.0486 (16)	-0.0038 (12)	0.0113 (12)	-0.0073 (12)
C14	0.0449 (18)	0.055 (2)	0.0601 (19)	-0.0025 (14)	0.0189 (14)	-0.0008 (16)
C15	0.065 (2)	0.058 (2)	0.058 (2)	-0.0087 (17)	0.0212 (17)	0.0047 (17)
C16	0.058 (2)	0.052 (2)	0.0530 (19)	-0.0010 (16)	0.0095 (15)	0.0011 (15)
C17	0.0447 (17)	0.0449 (18)	0.0511 (17)	-0.0018 (13)	0.0064 (13)	-0.0043 (14)
C18	0.053 (2)	0.056 (2)	0.074 (2)	0.0063 (16)	-0.0035 (17)	0.0071 (18)

Geometric parameters (Å, °)

O1—C2	1.208 (4)	C8—H8	0.9300
O2—C17	1.350 (4)	C9—C10	1.416 (5)
O2—C18	1.436 (4)	C9—H9	0.9300
C1—C17	1.389 (5)	C10—C11	1.425 (5)
C1—C13	1.406 (4)	C11—C12	1.354 (4)
C1—C2	1.483 (4)	C11—H11	0.9300
C2—C3	1.505 (4)	C12—C13	1.478 (4)
C3—C4	1.355 (5)	C13—C14	1.374 (4)
C3—C12	1.424 (4)	C14—C15	1.394 (5)
C4—C5	1.421 (5)	C14—H14	0.9300
C4—H4	0.9300	C15—C16	1.375 (5)
C5—C6	1.415 (5)	C15—H15	0.9300
C5—C10	1.419 (5)	C16—C17	1.394 (5)
C6—C7	1.359 (5)	C16—H16	0.9300
C6—H6	0.9300	C18—H18A	0.9600
C7—C8	1.379 (6)	C18—H18B	0.9600
C7—H7	0.9300	C18—H18C	0.9600
C8—C9	1.363 (6)		
C17—O2—C18	118.0 (3)	C9—C10—C11	121.9 (3)
C17—C1—C13	120.3 (3)	C5—C10—C11	119.8 (3)
C17—C1—C2	130.1 (3)	C12—C11—C10	119.9 (3)

C13—C1—C2	109.5 (3)	C12—C11—H11	120.0
O1—C2—C1	129.0 (3)	C10—C11—H11	120.0
O1—C2—C3	125.8 (3)	C11—C12—C3	120.0 (3)
C1—C2—C3	105.2 (3)	C11—C12—C13	131.9 (3)
C4—C3—C12	121.8 (3)	C3—C12—C13	108.2 (3)
C4—C3—C2	129.8 (3)	C14—C13—C1	121.2 (3)
C12—C3—C2	108.4 (3)	C14—C13—C12	130.1 (3)
C3—C4—C5	119.6 (3)	C1—C13—C12	108.7 (3)
C3—C4—H4	120.2	C13—C14—C15	117.9 (3)
C5—C4—H4	120.2	C13—C14—H14	121.1
C6—C5—C10	118.7 (3)	C15—C14—H14	121.1
C6—C5—C4	122.4 (3)	C16—C15—C14	121.8 (3)
C10—C5—C4	118.9 (3)	C16—C15—H15	119.1
C7—C6—C5	120.8 (4)	C14—C15—H15	119.1
C7—C6—H6	119.6	C15—C16—C17	120.6 (3)
C5—C6—H6	119.6	C15—C16—H16	119.7
C6—C7—C8	120.8 (4)	C17—C16—H16	119.7
C6—C7—H7	119.6	O2—C17—C1	117.4 (3)
C8—C7—H7	119.6	O2—C17—C16	124.3 (3)
C9—C8—C7	120.6 (4)	C1—C17—C16	118.3 (3)
C9—C8—H8	119.7	O2—C18—H18A	109.5
C7—C8—H8	119.7	O2—C18—H18B	109.5
C8—C9—C10	120.9 (4)	H18A—C18—H18B	109.5
C8—C9—H9	119.5	O2—C18—H18C	109.5
C10—C9—H9	119.5	H18A—C18—H18C	109.5
C9—C10—C5	118.2 (3)	H18B—C18—H18C	109.5
C17—C1—C2—O1	-1.7 (6)	C10—C11—C12—C13	179.8 (3)
C13—C1—C2—O1	179.4 (3)	C4—C3—C12—C11	0.8 (5)
C17—C1—C2—C3	178.3 (3)	C2—C3—C12—C11	-179.9 (3)
C13—C1—C2—C3	-0.6 (3)	C4—C3—C12—C13	-179.9 (3)
O1—C2—C3—C4	0.0 (6)	C2—C3—C12—C13	-0.6 (3)
C1—C2—C3—C4	180.0 (3)	C17—C1—C13—C14	0.4 (5)
O1—C2—C3—C12	-179.3 (3)	C2—C1—C13—C14	179.4 (3)
C1—C2—C3—C12	0.7 (3)	C17—C1—C13—C12	-178.7 (3)
C12—C3—C4—C5	0.3 (5)	C2—C1—C13—C12	0.2 (3)
C2—C3—C4—C5	-178.9 (3)	C11—C12—C13—C14	0.4 (6)
C3—C4—C5—C6	178.6 (3)	C3—C12—C13—C14	-178.8 (3)
C3—C4—C5—C10	-1.1 (5)	C11—C12—C13—C1	179.4 (3)
C10—C5—C6—C7	0.0 (5)	C3—C12—C13—C1	0.2 (3)
C4—C5—C6—C7	-179.6 (3)	C1—C13—C14—C15	0.5 (5)
C5—C6—C7—C8	-0.7 (6)	C12—C13—C14—C15	179.4 (3)
C6—C7—C8—C9	0.6 (6)	C13—C14—C15—C16	-0.7 (5)
C7—C8—C9—C10	0.1 (6)	C14—C15—C16—C17	0.1 (6)
C8—C9—C10—C5	-0.7 (5)	C18—O2—C17—C1	-175.9 (3)
C8—C9—C10—C11	178.8 (3)	C18—O2—C17—C16	3.1 (5)
C6—C5—C10—C9	0.7 (4)	C13—C1—C17—O2	178.0 (3)
C4—C5—C10—C9	-179.7 (3)	C2—C1—C17—O2	-0.7 (5)

C6—C5—C10—C11	-178.9 (3)	C13—C1—C17—C16	-1.0 (5)
C4—C5—C10—C11	0.7 (4)	C2—C1—C17—C16	-179.7 (3)
C9—C10—C11—C12	-179.2 (3)	C15—C16—C17—O2	-178.2 (3)
C5—C10—C11—C12	0.3 (5)	C15—C16—C17—C1	0.8 (5)
C10—C11—C12—C3	-1.1 (5)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C7—H7 \cdots O1 ⁱ	0.93	2.57	3.299 (5)	135
C11—H11 \cdots O1 ⁱⁱ	0.93	2.55	3.298 (4)	138

Symmetry codes: (i) $-x+2, y+1/2, -z+3/2$; (ii) $x-1, y, z$.