

## 5-Bromo-2-chloropyrimidin-4-amine

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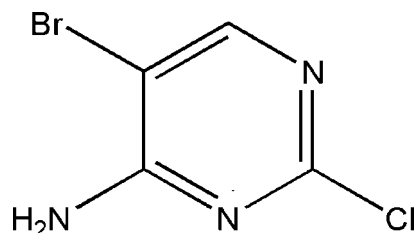
Received 1 March 2013; accepted 16 March 2013

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.024;  $wR$  factor = 0.058; data-to-parameter ratio = 14.4.

In the title compound,  $\text{C}_4\text{H}_3\text{BrClN}_3$ , the pyrimidine ring is essentially planar (r.m.s. deviation from the plane = 0.087 Å). In the crystal, pairs of  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds connect the molecules into inversion dimers; these are connected by further  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds into a two-dimensional framework parallel to the  $bc$  plane.

### Related literature

For background to pyrimidine derivatives, see: Yu *et al.* (2007). For related structures, see: van Albada *et al.* (2012); Yang *et al.* (2012).



### Experimental

#### Crystal data

$\text{C}_4\text{H}_3\text{BrClN}_3$   
 $M_r = 208.45$   
Monoclinic,  $P2_1/c$   
 $a = 6.0297$  (1) Å

$b = 8.1542$  (2) Å  
 $c = 13.4163$  (3) Å  
 $\beta = 90.491$  (2)°  
 $V = 659.62$  (2) Å<sup>3</sup>

$Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 6.54$  mm<sup>-1</sup>

$T = 293$  K  
 $0.3 \times 0.2 \times 0.1$  mm

#### Data collection

Oxford Diffraction Xcalibur  
Sapphire3 diffractometer  
Absorption correction: multi-scan  
(*CrysAlis PRO*; Oxford  
Diffraction, 2010)  
 $T_{\min} = 0.306$ ,  $T_{\max} = 1.000$

43395 measured reflections  
1297 independent reflections  
1164 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.046$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.024$   
 $wR(F^2) = 0.058$   
 $S = 1.10$   
1297 reflections  
90 parameters

H atoms treated by a mixture of  
independent and constrained  
refinement  
 $\Delta\rho_{\max} = 0.33$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.28$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$                                     | $D-H$    | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---|----------|-------------|-------------|---------------|
| $\text{N7}-\text{H71}\cdots\text{N1}^{\text{i}}$  | 0.78 (3) | 2.38 (3)    | 3.087 (3)   | 153 (3)       |
| $\text{N7}-\text{H72}\cdots\text{N3}^{\text{ii}}$ | 0.91 (4) | 2.19 (4)    | 3.088 (3)   | 171 (3)       |

Symmetry codes: (i)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (ii)  $-x, -y + 1, -z + 1$ .

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis RED* (Oxford Diffraction, 2010); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: GK2560).

### References

- Albada, G. van, Ghazzali, M., Al-Farhan, K. & Reedijk, J. (2012). *Acta Cryst.* **E68**, o302.  
Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.  
Oxford Diffraction (2010). *CrysAlis PRO* and *CrysAlis RED*. Oxford Diffraction Ltd, Yarnton, England.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.  
Yang, Q., Xu, N., Zhu, K., Lv, X. & Han, P. (2012). *Acta Cryst.* **E68**, o111.  
Yu, Z. H., Niu, C. W., Ban, S. R., Wen, X. & Xi, Z. (2007). *Chin. Sci. Bull.* **52**, 1929–1941.

## supporting information

*Acta Cryst.* (2013). E69, o583 [doi:10.1107/S1600536813007228]

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### S1. Comment

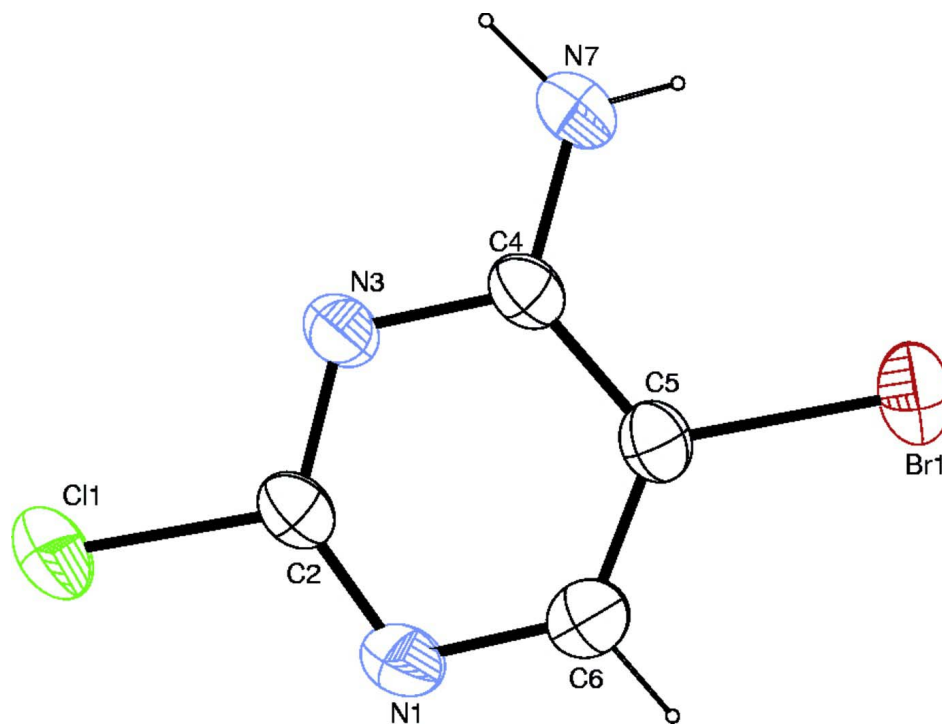
Some derivatives of pyrimidine are important chemical materials (Yu *et al.*, 2007). Here in this article, the preparation and crystal structure of the title compound is presented. Bond lengths and angles in the title compound (Fig. 1) are comparable with the similar crystal structures (van Albada *et al.*, 2012; Yang *et al.*, 2012). The pyrimidine ring is essentially planar (r.m.s. deviation from the plane 0.087 Å). The atoms Br, Cl and N7 are coplanar with the pyrimidine ring. In the crystal, molecules are linked into dimers by N7—H72···N3 hydrogen bonds and these dimers are further connected by N7—H71···N1 hydrogen bonds, forming two dimensional supramolecular network in the *bc* plane (Fig. 2, (Table 2).

### S2. Experimental

To a solution of stannous chloride dihydrate (2.8 ml, 0.012 mole) in hydrochloric acid (30 ml) cooled to 273K, 5-bromo-2-chloro-4-nitropyrimidine (2 g, 0.0083 mole) was added in portions while the suspension was vigorously stirred for 6 hrs. The mixture was then poured onto crushed ice, made alkaline with solid sodium hydroxide, and extracted three times with ethyl acetate (100 ml). The combined organic phase was dried over anhydrous sodium sulfate and the filtrate was evaporated to dryness. The compound was purified by successive recrystallization from acetonitrile (yield 90%, m. p. 460–461 K).

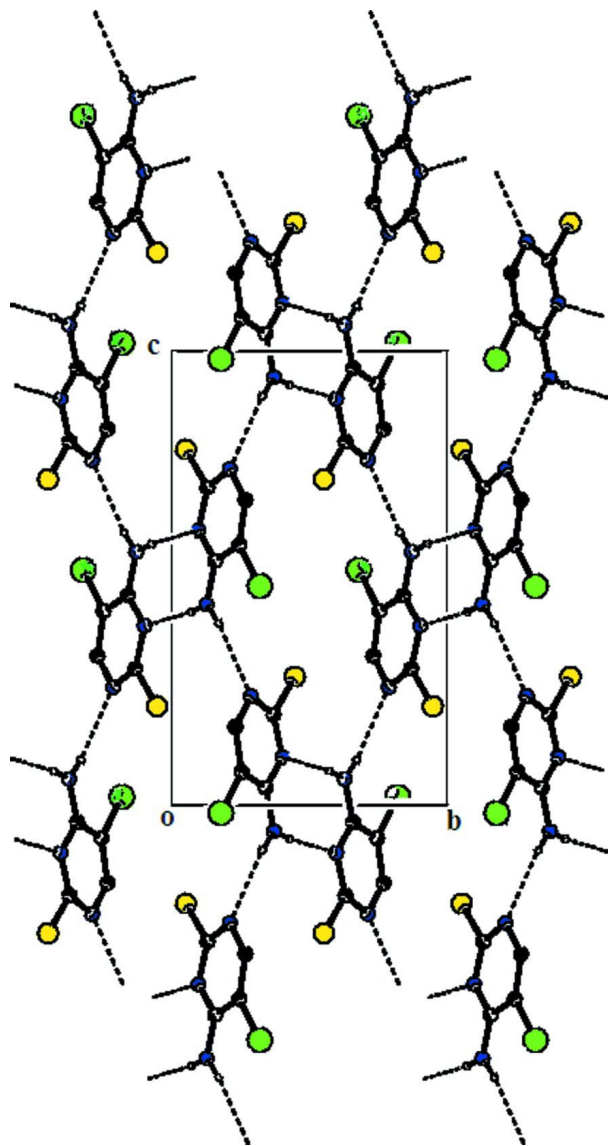
### S3. Refinement

The N-bound H atoms were located in a difference Fourier map and freely refined. All other H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C—H distances of 0.93 Å and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .



**Figure 1**

*ORTEP* view of the molecule with the atom-labeling scheme. The displacement ellipsoids are drawn at the 40% probability level. H atoms are shown as small spheres of arbitrary radii.

**Figure 2**

A molecular packing view of the title compound down the  $a$  axis, showing intermolecular interactions. The dotted lines show intermolecular N—H...N hydrogen bonds.

### 5-Bromo-2-chloropyrimidin-4-amine

#### Crystal data

$C_4H_3BrClN_3$

$M_r = 208.45$

Monoclinic,  $P2_1/c$

Hall symbol:  $-P 2_1/c$

$a = 6.0297$  (1) Å

$b = 8.1542$  (2) Å

$c = 13.4163$  (3) Å

$\beta = 90.491$  (2)°

$V = 659.62$  (2) Å<sup>3</sup>

$Z = 4$

$F(000) = 400$

$D_x = 2.099$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 20319 reflections

$\theta = 3.7$ – $29.0$ °

$\mu = 6.54$  mm<sup>-1</sup>

$T = 293$  K

Block, white

$0.3 \times 0.2 \times 0.1$  mm

*Data collection*

Oxford Diffraction Xcalibur Sapphire3  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
Detector resolution: 16.1049 pixels mm<sup>-1</sup>  
 $\omega$  scans  
Absorption correction: multi-scan  
(*CrysAlis PRO*; Oxford Diffraction, 2010)  
 $T_{\min} = 0.306$ ,  $T_{\max} = 1.000$

43395 measured reflections  
1297 independent reflections  
1164 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.046$   
 $\theta_{\max} = 26.0^\circ$ ,  $\theta_{\min} = 3.9^\circ$   
 $h = -7 \rightarrow 7$   
 $k = -10 \rightarrow 10$   
 $l = -16 \rightarrow 16$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.024$   
 $wR(F^2) = 0.058$   
 $S = 1.10$   
1297 reflections  
90 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites  
H atoms treated by a mixture of independent  
and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0275P)^2 + 0.4995P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.33 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.28 \text{ e } \text{\AA}^{-3}$

*Special details*

**Experimental.** *CrysAlis PRO*, Oxford Diffraction Ltd., Version 1.171.34.40 (release 27-08-2010 *CrysAlis171.NET*) (compiled Aug 27 2010, 11:50:40) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

|     | <i>x</i>      | <i>y</i>     | <i>z</i>     | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|---------------|--------------|--------------|----------------------------------|
| Br1 | 0.69350 (4)   | 0.17972 (4)  | 0.48221 (2)  | 0.04520 (12)                     |
| Cl1 | -0.00058 (12) | 0.44749 (10) | 0.78250 (5)  | 0.04662 (19)                     |
| N1  | 0.3588 (4)    | 0.2880 (3)   | 0.74005 (15) | 0.0397 (5)                       |
| C2  | 0.1934 (4)    | 0.3694 (3)   | 0.69923 (17) | 0.0302 (5)                       |
| N3  | 0.1489 (3)    | 0.4025 (3)   | 0.60525 (14) | 0.0310 (4)                       |
| C4  | 0.2921 (4)    | 0.3417 (3)   | 0.53797 (17) | 0.0293 (5)                       |
| C5  | 0.4802 (4)    | 0.2544 (3)   | 0.57261 (17) | 0.0308 (5)                       |
| C6  | 0.5050 (4)    | 0.2317 (4)   | 0.67197 (19) | 0.0400 (6)                       |
| H6  | 0.6288        | 0.1744       | 0.6947       | 0.048*                           |
| N7  | 0.2498 (5)    | 0.3721 (3)   | 0.44228 (16) | 0.0413 (6)                       |
| H71 | 0.317 (5)     | 0.329 (3)    | 0.401 (2)    | 0.037 (8)*                       |
| H72 | 0.125 (6)     | 0.428 (4)    | 0.425 (3)    | 0.063 (10)*                      |

Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$     | $U^{22}$     | $U^{33}$     | $U^{12}$     | $U^{13}$     | $U^{23}$      |
|-----|--------------|--------------|--------------|--------------|--------------|---------------|
| Br1 | 0.03877 (18) | 0.05169 (19) | 0.04536 (18) | 0.00220 (12) | 0.01297 (12) | -0.00698 (12) |
| Cl1 | 0.0474 (4)   | 0.0627 (5)   | 0.0300 (3)   | 0.0046 (3)   | 0.0130 (3)   | -0.0023 (3)   |
| N1  | 0.0393 (12)  | 0.0544 (14)  | 0.0254 (10)  | 0.0025 (10)  | 0.0014 (9)   | 0.0059 (9)    |
| C2  | 0.0323 (13)  | 0.0332 (12)  | 0.0251 (11)  | -0.0060 (10) | 0.0038 (9)   | -0.0011 (9)   |
| N3  | 0.0335 (11)  | 0.0357 (11)  | 0.0238 (9)   | -0.0003 (9)  | 0.0024 (8)   | -0.0001 (8)   |
| C4  | 0.0300 (12)  | 0.0331 (13)  | 0.0247 (11)  | -0.0069 (10) | 0.0022 (9)   | -0.0016 (9)   |
| C5  | 0.0303 (13)  | 0.0312 (12)  | 0.0309 (12)  | -0.0037 (10) | 0.0048 (10)  | -0.0016 (10)  |
| C6  | 0.0347 (14)  | 0.0483 (16)  | 0.0371 (13)  | 0.0052 (12)  | -0.0010 (11) | 0.0049 (12)   |
| N7  | 0.0448 (14)  | 0.0569 (15)  | 0.0221 (11)  | 0.0080 (12)  | 0.0027 (10)  | -0.0018 (10)  |

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

|              |              |              |              |
|--------------|--------------|--------------|--------------|
| Br1—C5       | 1.877 (2)    | C4—N7        | 1.330 (3)    |
| Cl1—C2       | 1.744 (2)    | C4—C5        | 1.415 (3)    |
| N1—C2        | 1.314 (3)    | C5—C6        | 1.353 (3)    |
| N1—C6        | 1.355 (3)    | C6—H6        | 0.9300       |
| C2—N3        | 1.315 (3)    | N7—H71       | 0.78 (3)     |
| N3—C4        | 1.349 (3)    | N7—H72       | 0.91 (4)     |
| C2—N1—C6     | 112.7 (2)    | C6—C5—Br1    | 121.5 (2)    |
| N1—C2—N3     | 130.6 (2)    | C4—C5—Br1    | 120.20 (17)  |
| N1—C2—Cl1    | 115.35 (18)  | C5—C6—N1     | 123.4 (2)    |
| N3—C2—Cl1    | 114.02 (18)  | C5—C6—H6     | 118.3        |
| C2—N3—C4     | 116.1 (2)    | N1—C6—H6     | 118.3        |
| N7—C4—N3     | 117.3 (2)    | C4—N7—H71    | 121 (2)      |
| N7—C4—C5     | 123.9 (2)    | C4—N7—H72    | 119 (2)      |
| N3—C4—C5     | 118.8 (2)    | H71—N7—H72   | 119 (3)      |
| C6—C5—C4     | 118.3 (2)    |              |              |
| C6—N1—C2—N3  | 0.5 (4)      | N3—C4—C5—C6  | 1.9 (4)      |
| C6—N1—C2—Cl1 | -179.43 (19) | N7—C4—C5—Br1 | 2.1 (3)      |
| N1—C2—N3—C4  | 1.4 (4)      | N3—C4—C5—Br1 | -175.96 (17) |
| Cl1—C2—N3—C4 | -178.71 (17) | C4—C5—C6—N1  | 0.1 (4)      |
| C2—N3—C4—N7  | 179.4 (2)    | Br1—C5—C6—N1 | 177.9 (2)    |
| C2—N3—C4—C5  | -2.5 (3)     | C2—N1—C6—C5  | -1.2 (4)     |
| N7—C4—C5—C6  | 179.9 (3)    |              |              |

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

| $D—H\cdots A$                    | $D—H$    | $H\cdots A$ | $D\cdots A$ | $D—H\cdots A$ |
|----------------------------------|----------|-------------|-------------|---------------|
| N7—H71 $\cdots$ N1 <sup>i</sup>  | 0.78 (3) | 2.38 (3)    | 3.087 (3)   | 153 (3)       |
| N7—H72 $\cdots$ N3 <sup>ii</sup> | 0.91 (4) | 2.19 (4)    | 3.088 (3)   | 171 (3)       |

Symmetry codes: (i)  $x, -y+1/2, z-1/2$ ; (ii)  $-x, -y+1, -z+1$ .