

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

3-Amino-1-(4-fluorophenyl)-7-methoxy-1*H*-benzo[*f*]chromene-2-carbonitrile

Abd El-Galil E. Amr,^{a,b,†} Ahmed M. El-Agrody,^{c,d}
 Mohamed A. Al-Omar,^{e,a} Seik Weng Ng^{f,g} and
 Edward R. T. Tiekink^{f,*}

^aDrug Exploration & Development Chair (DEDC), College of Pharmacy, King Saud University, Riyadh 11451, Saudi Arabia, ^bApplied Organic Chemistry Department, National Research Center, Dokki 12622, Cairo, Egypt, ^cChemistry Department, Faculty of Science, King Khalid University, Abha 61413, PO Box 9004, Saudi Arabia, ^dChemistry Department, Faculty of Science, Al-Azhar University, Nasr City, Cairo, 11884, Egypt, ^ePharmaceutical Chemistry Department, College of Pharmacy, King Saud University, Riyadh 11451, Saudi Arabia, ^fDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and ^gChemistry Department, Faculty of Science, King Abdulaziz University, PO Box 80203 Jeddah, Saudi Arabia
 Correspondence e-mail: edward.tiekink@gmail.com

Received 25 February 2013; accepted 25 February 2013

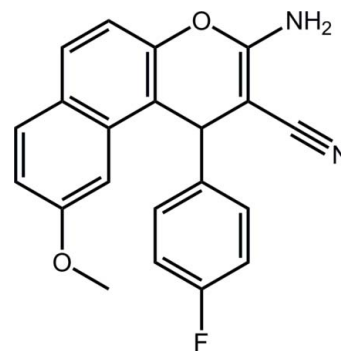
Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å;
 R factor = 0.052; wR factor = 0.145; data-to-parameter ratio = 15.9.

In the title compound, $\text{C}_{21}\text{H}_{15}\text{FN}_2\text{O}_2$, the furan ring has a flattened half-chair conformation [the methine C atom lies 0.136 (2) Å above the C_5 plane which has an r.m.s. deviation of 0.0229 Å]. Overall, the 1*H*-benzo[*f*]chromene fused-ring system approximates a plane (r.m.s. deviation of the 14 non-H atoms = 0.049 Å). The fluorobenzene ring is almost perpendicular to this plane [dihedral angle = 89.58 (8)°]. Zigzag supramolecular tapes along the b axis are the most notable feature of the crystal packing. This arises through an alternating sequence of 12-membered $\{\cdots\text{HNC}_3\text{N}\}_2$ and eight-membered $\{\cdots\text{HNCO}\}_2$ synthons. These are connected into a three-dimensional architecture by π - π [intercentroid distance for centrosymmetrically related fluorobenzene rings = 3.5181 (10) Å] and C—H $\cdots\pi$ interactions.

Related literature

For a related structure and background to 4*H*-chromene derivatives, see: El-Agrody *et al.* (2013). For related structures, see: Wang *et al.* (2008); Shekhar *et al.* (2012);

† Additional correspondence author, e-mail: aamr1963@yahoo.com.



Experimental

Crystal data

$\text{C}_{21}\text{H}_{15}\text{FN}_2\text{O}_2$
 $M_r = 346.35$
 Triclinic, $P\bar{1}$
 $a = 8.7798$ (9) Å
 $b = 9.6329$ (6) Å
 $c = 10.9130$ (11) Å
 $\alpha = 77.074$ (7)°
 $\beta = 68.414$ (10)°
 $\gamma = 87.083$ (7)°
 $V = 835.99$ (13) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.10$ mm⁻¹
 $T = 295$ K
 $0.30 \times 0.30 \times 0.10$ mm

Data collection

Agilent SuperNova Dual
 diffractometer with an Atlas
 detector
 Absorption correction: multi-scan
 (*CrysAlis PRO*; Agilent, 2011)
 $T_{\min} = 0.821$, $T_{\max} = 1.000$
 7587 measured reflections
 3868 independent reflections
 2569 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.029$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.145$
 $S = 1.02$
 3868 reflections
 244 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement
 $\Delta\rho_{\max} = 0.19$ e Å⁻³
 $\Delta\rho_{\min} = -0.16$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the $\text{C15}-\text{CC20}$ ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{N2}^{\text{i}}$	0.89 (2)	2.34 (3)	3.189 (2)	160 (2)
$\text{N1}-\text{H2}\cdots\text{O1}^{\text{ii}}$	0.87 (2)	2.36 (3)	3.219 (2)	169 (2)
$\text{C19}-\text{H19}\cdots\text{Cg1}^{\text{iii}}$	0.93	2.90	3.831 (2)	174

Symmetry codes: (i) $-x + 2, -y + 2, -z + 1$; (ii) $-x + 2, -y + 1, -z + 1$; (iii) $-x + 1, -y + 1, -z + 2$.

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

The authors extend their appreciation to the Deanship of Scientific Research at King Saud University for funding this work through the research group project No. RGP-VPP-099. We also thank the Ministry of Higher Education (Malaysia)

for funding structural studies through the High-Impact Research scheme (UM.C/HIR-MOHE/SC/12).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB7049).

References

- Agilent (2011). *CrysAlis PRO*. Agilent Technologies, Yarnton, England.
- Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
- El-Agrody, A. M., Amr, A.-G. E., Al-Omar, M. A., Ng, S. W. & Tiekink, E. R. T. (2013). *Acta Cryst.* **E69**, o476–o477.
- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
- Shekhar, A. C., Kumar, A. R., Sathaiah, G., Raju, K., Rao, P. S., Sridhar, M., Narsaiah, B., Srinivas, P. V. S. S. & Sridhar, B. (2012). *Helv. Chim. Acta*, **95**, 502–508.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Wang, X.-S., Yang, G.-S. & Zhao, G. (2008). *Tetrahedron Asymmetry*, **19**, 709–714.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

supporting information

Acta Cryst. (2013). E69, o478–o479 [doi:10.1107/S1600536813005473]

3-Amino-1-(4-fluorophenyl)-7-methoxy-1*H*-benzo[*f*]chromene-2-carbonitrile

Abd El-Galil E. Amr, Ahmed M. El-Agrody, Mohamed A. Al-Omar, Seik Weng Ng and Edward R. T. Tiekink

S1. Comment

As part of our ongoing studies of 4*H*-Chromene derivatives (El-Agrody *et al.*, 2013), the crystal and molecular structure of the title compound, (I), is described herein.

In (I), Fig. 1, the furan ring has a flattened half-chair conformation with the methine-C11 atom lying only 0.136 (2) Å above the plane of the remaining atoms (r.m.s. deviation = 0.0229 Å). In fact, the 14 non-hydrogen atoms of the 1*H*-benzo[*f*]chromene fused-ring system are co-planar with a r.m.s. deviation of the fitted atoms = 0.049 Å. The maximum deviations are 0.068 (2) Å for the aforementioned methine-C11 atom and -0.107 (2) Å for the adjacent C12 atom. The fluorobenzene ring is perpendicular to the 1*H*-benzo[*f*]chromene residue, forming a dihedral angle of 89.58 (8)°. The methoxy group is co-planar with the ring to which it is attached as manifested in the C14—O2—C7—C6 torsion angle of 177.17 (19)°. The structure described here resembles very closely those of the 4-bromo (Wang *et al.*, 2008) and 2-CF₃ (Shekhar *et al.*, 2012) derivatives of the parent compound, as well as that of the 8-methoxy analogue (El-Agrody *et al.*, 2013).

In the crystal packing, zigzag tapes are formed along the *b* axis via an alternating sequence of 12-membered {⋯HNC₃N}₂, arising from amine-*N*—*H*⋯*N*(cyano) hydrogen bonds, and eight-membered {⋯HNCO}₂, arising from amine-*N*—*H*⋯*O*(furan) hydrogen bonds, synthons, Fig. 2 and Table 1. These are connected into a layer in the *ab* plane by π — π interactions occurring between centrosymmetrically related fluorobenzene rings [inter-centroid distance = 3.5181 (10) Å for symmetry operation: 1 - *x*, 1 - *y*, 1 - *z*]. Layers are connected along the *c* axis by C—H⋯ π interactions where both the donor atom and acceptor- π system are derived from fluorobenzene rings, Fig. 3 and Table 1, highlighting the key role this residue plays in consolidating the crystal structure of (I).

S2. Experimental

A solution of 7-methoxy-2-naphthol (0.01 mol) in EtOH (30 ml) was treated with α -cyano-*p*-fluorocinnamionitrile (0.01 mol) and piperidine (0.5 ml). The reaction mixture was heated until complete precipitation occurred (reaction time: 60 min). The solid product which formed was collected by filtration and recrystallized from ethanol to give the title compound, (I), as light-brown prisms; *M*.pt: 533–534 K.

S3. Refinement

The C-bound H atoms were geometrically placed (C—H = 0.93–0.98 Å) and refined as riding with $U_{iso}(\text{H}) = 1.2U_{eq}(\text{C})$. The N-bound-H atoms were refined freely.

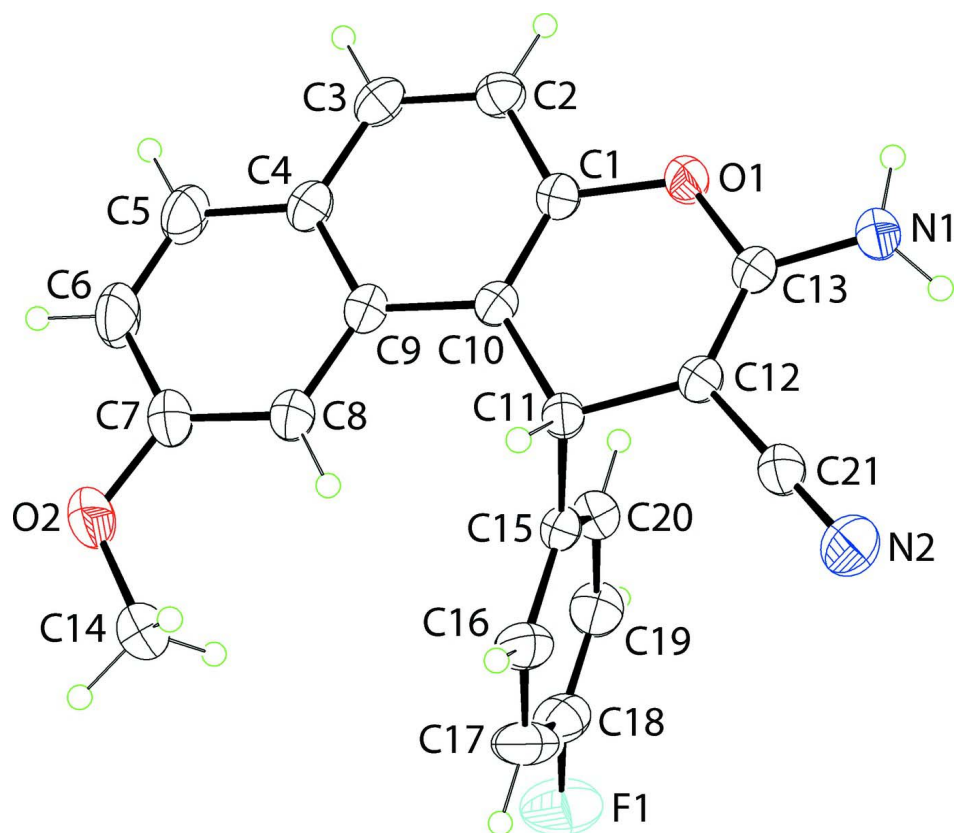


Figure 1

The molecular structure of (I) showing displacement ellipsoids at the 35% probability level.

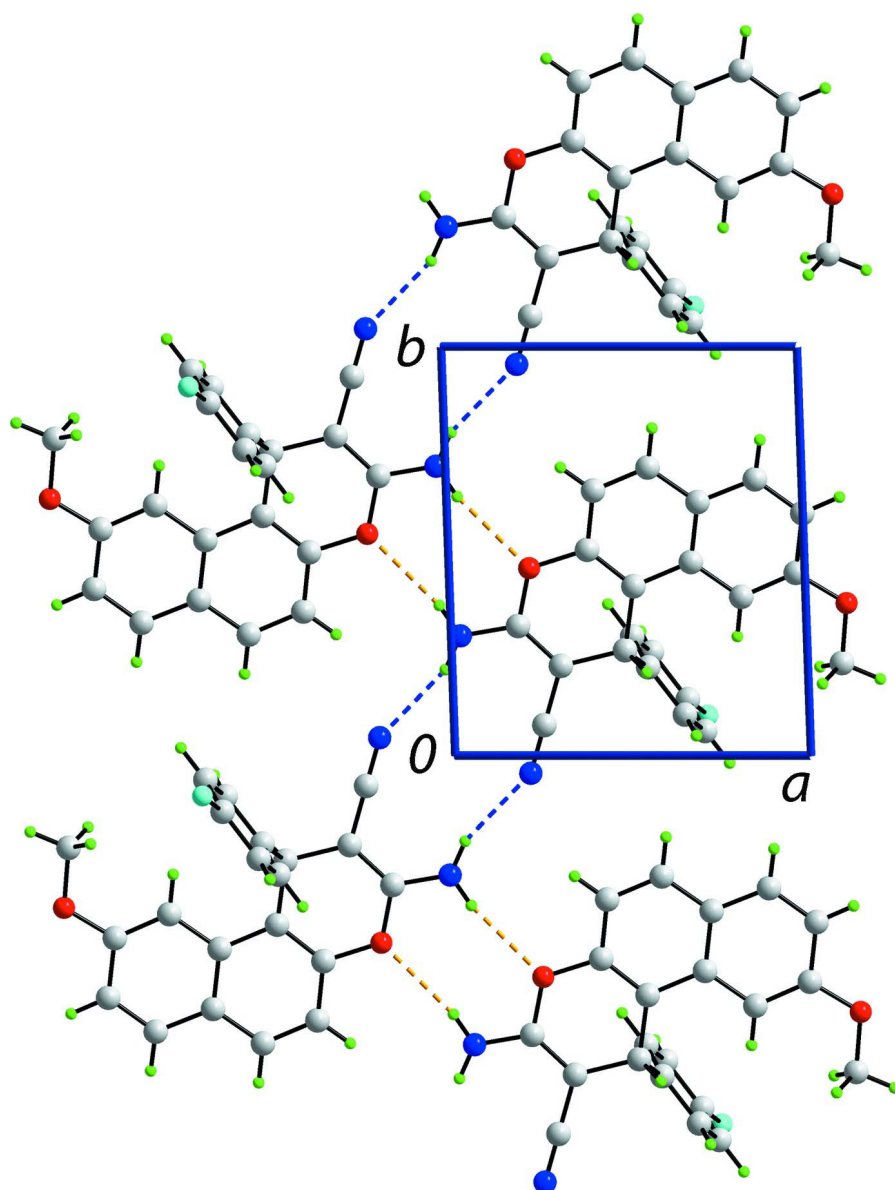


Figure 2

A view of the supramolecular zigzag tape along the b axis in (I). The $\text{N—H}\cdots\text{N}$ and $\text{N—H}\cdots\text{O}$ hydrogen bonds are shown as blue and orange dashed lines, respectively.

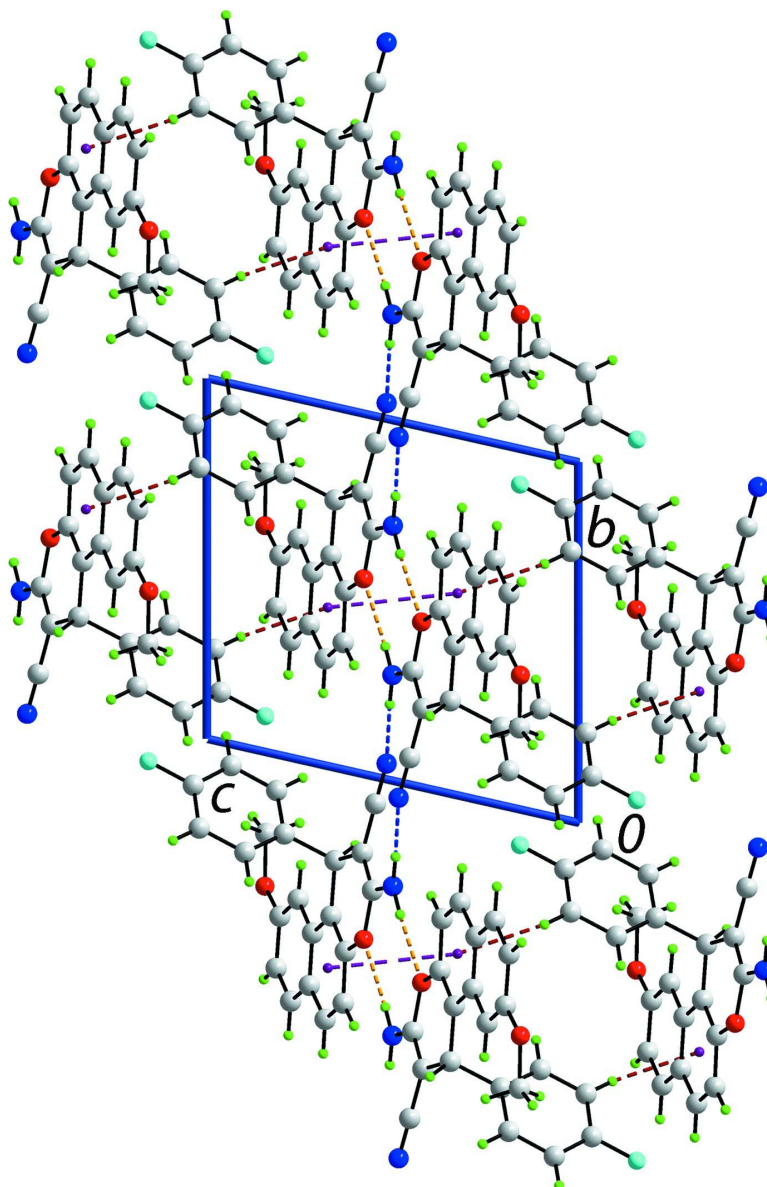


Figure 3

A view in projection down the a axis of the crystal packing in (I). The N—H \cdots N, N—H \cdots O, C—H \cdots π and π — π interactions are shown as blue, orange, brown and purple dashed lines, respectively.

3-Amino-1-(4-fluorophenyl)-7-methoxy-1*H*-benzo[*f*]chromene-2-carbonitrile

Crystal data

$C_{21}H_{15}FN_2O_2$

$M_r = 346.35$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 8.7798$ (9) Å

$b = 9.6329$ (6) Å

$c = 10.9130$ (11) Å

$\alpha = 77.074$ (7)°

$\beta = 68.414$ (10)°

$\gamma = 87.083$ (7)°

$V = 835.99$ (13) Å³

$Z = 2$

$F(000) = 360$

$D_x = 1.376$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1958 reflections

$\theta = 3.1$ – 27.5 °

$\mu = 0.10$ mm⁻¹

$T = 295$ K
Prism, light-brown

$0.30 \times 0.30 \times 0.10$ mm

Data collection

Agilent SuperNova Dual
diffractometer with an Atlas detector
Radiation source: SuperNova (Mo) X-ray
Source
Mirror monochromator
Detector resolution: 10.4041 pixels mm^{-1}
 ω scan
Absorption correction: multi-scan
(*CrysAlis PRO*; Agilent, 2011)

$T_{\min} = 0.821$, $T_{\max} = 1.000$
7587 measured reflections
3868 independent reflections
2569 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.029$
 $\theta_{\max} = 27.6^\circ$, $\theta_{\min} = 3.1^\circ$
 $h = -8 \rightarrow 11$
 $k = -12 \rightarrow 12$
 $l = -13 \rightarrow 14$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.145$
 $S = 1.02$
3868 reflections
244 parameters
0 restraints
Primary atom site location: structure-invariant
direct methods
Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites
H atoms treated by a mixture of independent
and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0577P)^2 + 0.1106P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.16 \text{ e } \text{\AA}^{-3}$
Extinction correction: *SHELXL97* (Sheldrick,
2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
Extinction coefficient: 0.011 (3)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.2857 (2)	0.90095 (14)	1.15726 (13)	0.0943 (5)
O1	0.77412 (14)	0.54184 (11)	0.57793 (13)	0.0452 (3)
O2	-0.11449 (16)	0.62729 (15)	0.84385 (16)	0.0692 (5)
N1	0.97391 (19)	0.70744 (19)	0.49939 (18)	0.0537 (5)
H1	1.020 (3)	0.790 (2)	0.492 (2)	0.078 (7)*
H2	1.030 (3)	0.633 (3)	0.480 (2)	0.080 (7)*
N2	0.7893 (2)	1.04350 (16)	0.51621 (19)	0.0677 (6)
C1	0.6096 (2)	0.49698 (16)	0.62727 (16)	0.0366 (4)
C2	0.5869 (2)	0.35217 (16)	0.63298 (17)	0.0414 (4)
H2A	0.6766	0.2958	0.6040	0.050*
C3	0.4316 (2)	0.29574 (17)	0.68165 (17)	0.0446 (4)
H3	0.4156	0.1996	0.6869	0.054*

C4	0.2945 (2)	0.38016 (17)	0.72431 (17)	0.0414 (4)
C5	0.1313 (3)	0.3245 (2)	0.7744 (2)	0.0564 (5)
H5	0.1133	0.2285	0.7807	0.068*
C6	0.0011 (3)	0.4075 (2)	0.8134 (2)	0.0633 (6)
H6	-0.1047	0.3681	0.8469	0.076*
C7	0.0254 (2)	0.5534 (2)	0.8034 (2)	0.0513 (5)
C8	0.1808 (2)	0.61196 (18)	0.75438 (18)	0.0441 (4)
H8	0.1957	0.7089	0.7465	0.053*
C9	0.3190 (2)	0.52689 (16)	0.71562 (16)	0.0372 (4)
C10	0.4828 (2)	0.58616 (15)	0.66581 (15)	0.0343 (4)
C11	0.51257 (19)	0.74259 (15)	0.65506 (16)	0.0351 (4)
H11	0.4541	0.7977	0.6002	0.042*
C12	0.6944 (2)	0.78082 (15)	0.58154 (16)	0.0373 (4)
C13	0.8109 (2)	0.68330 (16)	0.55414 (17)	0.0392 (4)
C14	-0.0958 (3)	0.7740 (2)	0.8413 (3)	0.0734 (7)
H14A	-0.2018	0.8130	0.8766	0.110*
H14B	-0.0391	0.8247	0.7500	0.110*
H14C	-0.0338	0.7832	0.8957	0.110*
C15	0.44924 (19)	0.78420 (15)	0.79176 (16)	0.0370 (4)
C16	0.3506 (2)	0.89994 (17)	0.8121 (2)	0.0509 (5)
H16	0.3206	0.9516	0.7421	0.061*
C17	0.2955 (3)	0.9403 (2)	0.9351 (2)	0.0634 (6)
H17	0.2295	1.0183	0.9483	0.076*
C18	0.3407 (3)	0.8624 (2)	1.0362 (2)	0.0575 (6)
C19	0.4377 (3)	0.7473 (2)	1.02070 (19)	0.0548 (5)
H19	0.4670	0.6963	1.0913	0.066*
C20	0.4910 (2)	0.70854 (18)	0.89766 (17)	0.0454 (4)
H20	0.5564	0.6299	0.8858	0.054*
C21	0.7466 (2)	0.92564 (17)	0.54616 (18)	0.0441 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.1272 (14)	0.0935 (10)	0.0578 (8)	0.0132 (9)	-0.0173 (8)	-0.0393 (7)
O1	0.0329 (7)	0.0349 (6)	0.0654 (8)	0.0018 (5)	-0.0135 (6)	-0.0140 (5)
O2	0.0339 (7)	0.0678 (9)	0.1011 (12)	-0.0024 (6)	-0.0146 (8)	-0.0258 (8)
N1	0.0325 (9)	0.0453 (9)	0.0791 (12)	-0.0011 (7)	-0.0103 (8)	-0.0220 (8)
N2	0.0572 (11)	0.0378 (9)	0.0939 (14)	-0.0085 (7)	-0.0128 (10)	-0.0095 (8)
C1	0.0352 (9)	0.0358 (8)	0.0379 (9)	-0.0014 (6)	-0.0123 (7)	-0.0075 (6)
C2	0.0444 (10)	0.0335 (8)	0.0458 (10)	0.0038 (7)	-0.0160 (8)	-0.0093 (7)
C3	0.0556 (12)	0.0310 (8)	0.0483 (10)	-0.0068 (7)	-0.0209 (9)	-0.0059 (7)
C4	0.0446 (10)	0.0375 (9)	0.0418 (9)	-0.0071 (7)	-0.0171 (8)	-0.0043 (7)
C5	0.0511 (12)	0.0434 (10)	0.0715 (13)	-0.0151 (8)	-0.0205 (10)	-0.0060 (9)
C6	0.0401 (11)	0.0596 (12)	0.0834 (15)	-0.0168 (9)	-0.0165 (11)	-0.0085 (10)
C7	0.0355 (10)	0.0569 (11)	0.0599 (12)	-0.0041 (8)	-0.0149 (9)	-0.0125 (9)
C8	0.0373 (10)	0.0430 (9)	0.0520 (11)	-0.0034 (7)	-0.0156 (8)	-0.0107 (7)
C9	0.0376 (9)	0.0387 (9)	0.0361 (9)	-0.0049 (7)	-0.0151 (7)	-0.0060 (6)
C10	0.0355 (9)	0.0332 (8)	0.0346 (8)	-0.0020 (6)	-0.0130 (7)	-0.0074 (6)

C11	0.0303 (8)	0.0328 (8)	0.0418 (9)	0.0004 (6)	-0.0130 (7)	-0.0076 (6)
C12	0.0353 (9)	0.0321 (8)	0.0420 (9)	-0.0023 (6)	-0.0114 (7)	-0.0076 (7)
C13	0.0361 (9)	0.0343 (8)	0.0459 (10)	-0.0027 (7)	-0.0122 (8)	-0.0104 (7)
C14	0.0436 (12)	0.0753 (15)	0.1080 (19)	0.0078 (10)	-0.0225 (12)	-0.0435 (13)
C15	0.0321 (8)	0.0325 (8)	0.0435 (9)	-0.0049 (6)	-0.0092 (7)	-0.0095 (7)
C16	0.0594 (12)	0.0367 (9)	0.0555 (11)	0.0048 (8)	-0.0190 (9)	-0.0122 (8)
C17	0.0748 (16)	0.0451 (11)	0.0654 (14)	0.0113 (9)	-0.0140 (11)	-0.0251 (9)
C18	0.0646 (14)	0.0588 (12)	0.0443 (11)	-0.0064 (10)	-0.0069 (10)	-0.0220 (9)
C19	0.0575 (13)	0.0618 (12)	0.0440 (11)	-0.0024 (9)	-0.0169 (9)	-0.0110 (9)
C20	0.0409 (10)	0.0473 (10)	0.0469 (10)	0.0013 (7)	-0.0142 (8)	-0.0115 (8)
C21	0.0345 (9)	0.0397 (9)	0.0531 (10)	-0.0008 (7)	-0.0099 (8)	-0.0106 (7)

Geometric parameters (Å, °)

F1—C18	1.359 (2)	C8—C9	1.414 (2)
O1—C13	1.3613 (18)	C8—H8	0.9300
O1—C1	1.3966 (19)	C9—C10	1.434 (2)
O2—C7	1.365 (2)	C10—C11	1.514 (2)
O2—C14	1.424 (2)	C11—C12	1.521 (2)
N1—C13	1.343 (2)	C11—C15	1.526 (2)
N1—H1	0.89 (2)	C11—H11	0.9800
N1—H2	0.87 (2)	C12—C13	1.349 (2)
N2—C21	1.151 (2)	C12—C21	1.414 (2)
C1—C10	1.369 (2)	C14—H14A	0.9600
C1—C2	1.403 (2)	C14—H14B	0.9600
C2—C3	1.360 (2)	C14—H14C	0.9600
C2—H2A	0.9300	C15—C20	1.382 (2)
C3—C4	1.409 (3)	C15—C16	1.383 (2)
C3—H3	0.9300	C16—C17	1.387 (3)
C4—C5	1.418 (2)	C16—H16	0.9300
C4—C9	1.417 (2)	C17—C18	1.363 (3)
C5—C6	1.352 (3)	C17—H17	0.9300
C5—H5	0.9300	C18—C19	1.366 (3)
C6—C7	1.405 (3)	C19—C20	1.380 (2)
C6—H6	0.9300	C19—H19	0.9300
C7—C8	1.369 (2)	C20—H20	0.9300
C13—O1—C1	118.63 (12)	C10—C11—C15	113.09 (12)
C7—O2—C14	117.06 (15)	C12—C11—C15	110.77 (13)
C13—N1—H1	122.0 (15)	C10—C11—H11	107.8
C13—N1—H2	114.6 (16)	C12—C11—H11	107.8
H1—N1—H2	123 (2)	C15—C11—H11	107.8
C10—C1—O1	123.09 (14)	C13—C12—C21	117.59 (15)
C10—C1—C2	123.33 (16)	C13—C12—C11	123.60 (13)
O1—C1—C2	113.58 (14)	C21—C12—C11	118.71 (14)
C3—C2—C1	118.86 (16)	N1—C13—C12	127.20 (15)
C3—C2—H2A	120.6	N1—C13—O1	110.29 (14)
C1—C2—H2A	120.6	C12—C13—O1	122.50 (15)

C2—C3—C4	121.27 (15)	O2—C14—H14A	109.5
C2—C3—H3	119.4	O2—C14—H14B	109.5
C4—C3—H3	119.4	H14A—C14—H14B	109.5
C3—C4—C5	122.50 (16)	O2—C14—H14C	109.5
C3—C4—C9	119.31 (16)	H14A—C14—H14C	109.5
C5—C4—C9	118.18 (17)	H14B—C14—H14C	109.5
C6—C5—C4	121.72 (17)	C20—C15—C16	118.34 (16)
C6—C5—H5	119.1	C20—C15—C11	120.85 (15)
C4—C5—H5	119.1	C16—C15—C11	120.80 (16)
C5—C6—C7	120.09 (18)	C15—C16—C17	121.12 (19)
C5—C6—H6	120.0	C15—C16—H16	119.4
C7—C6—H6	120.0	C17—C16—H16	119.4
O2—C7—C8	124.64 (17)	C18—C17—C16	118.26 (18)
O2—C7—C6	115.14 (17)	C18—C17—H17	120.9
C8—C7—C6	120.21 (18)	C16—C17—H17	120.9
C7—C8—C9	120.80 (16)	F1—C18—C19	119.1 (2)
C7—C8—H8	119.6	F1—C18—C17	118.35 (19)
C9—C8—H8	119.6	C19—C18—C17	122.58 (18)
C8—C9—C4	118.98 (15)	C18—C19—C20	118.36 (19)
C8—C9—C10	121.54 (15)	C18—C19—H19	120.8
C4—C9—C10	119.47 (15)	C20—C19—H19	120.8
C1—C10—C9	117.74 (14)	C15—C20—C19	121.33 (17)
C1—C10—C11	121.65 (14)	C15—C20—H20	119.3
C9—C10—C11	120.61 (14)	C19—C20—H20	119.3
C10—C11—C12	109.49 (13)	N2—C21—C12	179.4 (2)
C13—O1—C1—C10	-6.8 (2)	C4—C9—C10—C11	179.82 (14)
C13—O1—C1—C2	172.96 (14)	C1—C10—C11—C12	7.1 (2)
C10—C1—C2—C3	-1.4 (3)	C9—C10—C11—C12	-172.08 (14)
O1—C1—C2—C3	178.88 (15)	C1—C10—C11—C15	-116.92 (17)
C1—C2—C3—C4	0.8 (3)	C9—C10—C11—C15	63.9 (2)
C2—C3—C4—C5	179.49 (17)	C10—C11—C12—C13	-11.2 (2)
C2—C3—C4—C9	0.3 (3)	C15—C11—C12—C13	114.20 (18)
C3—C4—C5—C6	-179.2 (2)	C10—C11—C12—C21	172.48 (15)
C9—C4—C5—C6	0.0 (3)	C15—C11—C12—C21	-62.1 (2)
C4—C5—C6—C7	0.7 (4)	C21—C12—C13—N1	1.9 (3)
C14—O2—C7—C8	-3.6 (3)	C11—C12—C13—N1	-174.45 (18)
C14—O2—C7—C6	177.17 (19)	C21—C12—C13—O1	-176.76 (15)
C5—C6—C7—O2	179.2 (2)	C11—C12—C13—O1	6.9 (3)
C5—C6—C7—C8	-0.1 (3)	C1—O1—C13—N1	-176.14 (15)
O2—C7—C8—C9	179.56 (18)	C1—O1—C13—C12	2.7 (2)
C6—C7—C8—C9	-1.2 (3)	C10—C11—C15—C20	51.2 (2)
C7—C8—C9—C4	1.9 (3)	C12—C11—C15—C20	-72.18 (18)
C7—C8—C9—C10	-179.13 (17)	C10—C11—C15—C16	-129.82 (17)
C3—C4—C9—C8	177.90 (16)	C12—C11—C15—C16	106.85 (18)
C5—C4—C9—C8	-1.3 (3)	C20—C15—C16—C17	0.5 (3)
C3—C4—C9—C10	-1.1 (2)	C11—C15—C16—C17	-178.57 (17)
C5—C4—C9—C10	179.75 (16)	C15—C16—C17—C18	-0.2 (3)

O1—C1—C10—C9	-179.64 (14)	C16—C17—C18—F1	-179.39 (18)
C2—C1—C10—C9	0.6 (3)	C16—C17—C18—C19	0.1 (3)
O1—C1—C10—C11	1.1 (3)	F1—C18—C19—C20	179.25 (17)
C2—C1—C10—C11	-178.59 (15)	C17—C18—C19—C20	-0.2 (3)
C8—C9—C10—C1	-178.36 (16)	C16—C15—C20—C19	-0.6 (3)
C4—C9—C10—C1	0.6 (2)	C11—C15—C20—C19	178.43 (16)
C8—C9—C10—C11	0.9 (3)	C18—C19—C20—C15	0.5 (3)

Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C15–CC20 ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...N2 ⁱ	0.89 (2)	2.34 (3)	3.189 (2)	160 (2)
N1—H2...O1 ⁱⁱ	0.87 (2)	2.36 (3)	3.219 (2)	169 (2)
C19—H19...Cg1 ⁱⁱⁱ	0.93	2.90	3.831 (2)	174

Symmetry codes: (i) $-x+2, -y+2, -z+1$; (ii) $-x+2, -y+1, -z+1$; (iii) $-x+1, -y+1, -z+2$.