

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-(2-Fluorophenyl)-3-methylsulfanyl-5-phenyl-1-benzofuran

 Hong Dae Choi,^a Pil Ja Seo^a and Uk Lee^{b*}

^aDepartment of Chemistry, Donggeui University, San 24 Kaya-dong, Busanjin-gu, Busan 614-714, Republic of Korea, and ^bDepartment of Chemistry, Pukyong National University, 599-1 Daeyeon 3-dong, Nam-gu, Busan 608-737, Republic of Korea

Correspondence e-mail: uklee@pknu.ac.kr

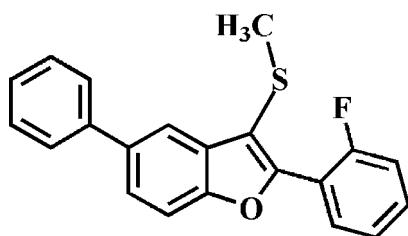
Received 1 April 2013; accepted 9 April 2013

 Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.037; wR factor = 0.101; data-to-parameter ratio = 18.2.

In the title compound, $\text{C}_{21}\text{H}_{15}\text{FOS}$, the dihedral angles between the mean plane [r.m.s. deviation = 0.041 (1) Å] of the benzofuran fragment and the pendant 2-fluorophenyl and phenyl rings are 46.09 (3) and 24.34 (5)°, respectively. In the crystal, molecules are linked by weak $\text{C}-\text{H}\cdots\pi$ interactions, forming a three-dimensional network.

Related literature

For background information and the crystal structures of related compounds, see: Choi *et al.* (2006); Seo *et al.* (2011).



Experimental

Crystal data

$\text{C}_{21}\text{H}_{15}\text{FOS}$
 $M_r = 334.39$
 Monoclinic, $P2_1/n$
 $a = 11.1257$ (2) Å

$b = 7.4232$ (1) Å
 $c = 19.4212$ (3) Å
 $\beta = 97.319$ (1)°
 $V = 1590.90$ (4) Å³

$Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.22$ mm⁻¹

$T = 173$ K
 $0.27 \times 0.19 \times 0.14$ mm

Data collection

Bruker APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2009)
 $T_{\min} = 0.689$, $T_{\max} = 0.746$

28795 measured reflections
 3967 independent reflections
 3190 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.101$
 $S = 1.05$
 3967 reflections

218 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.26$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 and Cg2 are the C9–C14 phenyl and 2-fluorophenyl rings, respectively.

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------------|-------|-------------|-------------|---------------|
| C10–H10 \cdots Cg1 ⁱ | 0.95 | 2.82 | 3.682 (2) | 131 |
| C14–H14 \cdots Cg2 ⁱⁱ | 0.95 | 2.71 | 3.528 (2) | 145 |

 Symmetry codes: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + 1, -y + 1, -z + 1$.

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012) and DIAMOND (Brandenburg, 1998); software used to prepare material for publication: SHELXL97.

This work was supported by the Blue-Bio Industry Regional Innovation Center (RIC08-06-07) at Donggeui University as an RIC program under the Ministry of Knowledge Economy and Busan city.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DS2230).

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supporting information

Acta Cryst. (2013). E69, o721 [https://doi.org/10.1107/S1600536813009768]

2-(2-Fluorophenyl)-3-methylsulfanyl-5-phenyl-1-benzofuran**Hong Dae Choi, Pil Ja Seo and Uk Lee****S1. Comment**

As a part of our continuing study of 3-methylsulfanyl-5-phenyl-1-benzofuran derivatives containing phenyl (Choi *et al.*, 2006) and 3-fluorophenyl (Seo *et al.*, 2011) substituents in 2-position, we report herein the crystal structure of the title compound.

In the title molecule (Fig. 1), the benzofuran unit is essentially planar, with a mean deviation of 0.041 (1) Å from the least-squares plane defined by the nine constituent atoms. The dihedral angles between the mean plane of the benzofuran fragment and the pendant 2-fluorophenyl and phenyl rings are 46.09 (3) and 24.34 (5), respectively. In the crystal structure (Fig. 2), molecules are connected by weak intermolecular C–H \cdots π interactions (Table 1, Cg1 and Cg2 are the centroids of the C9–C14 phenyl ring and the C15–C20 2-fluorophenyl ring, respectively).

S2. Experimental

Zinc chloride (218 mg, 1.6 mmol) was added to a stirred solution of 4-phenylphenol (272 mg, 1.6 mmol) and 2-chloro-2-methylsulfanyl-2'-fluoroacetophenone (350 mg, 1.6 mmol) in dichloromethene (20 mL) at room temperature, and stirring was continued at the same temperature for 40 min. The reaction was quenched by the addition of water and the organic layer was separated, dried over magnesium sulfate, filtered and concentrated at reduced pressure. The residue was purified by column chromatography (hexane-benzene, 5:2 v/v) to afford the title compound as a colorless solid [yield 51%, m.p. 368-369 K; R_f = 0.63 (hexane-benzene, 5:2 v/v)]. Single crystals suitable for X-ray diffraction were prepared by slow evaporation of a solution of the title compound in acetone at room temperature.

S3. Refinement

All H atoms were positioned geometrically and refined using a riding model, with C–H = 0.95 Å for aryl and 0.98 Å for methyl H atoms. $U_{iso}(H) = 1.2U_{eq}(C)$ for aryl and $1.5U_{eq}(C)$ for methyl H atoms. The positions of methyl hydrogens were optimized rotationally.

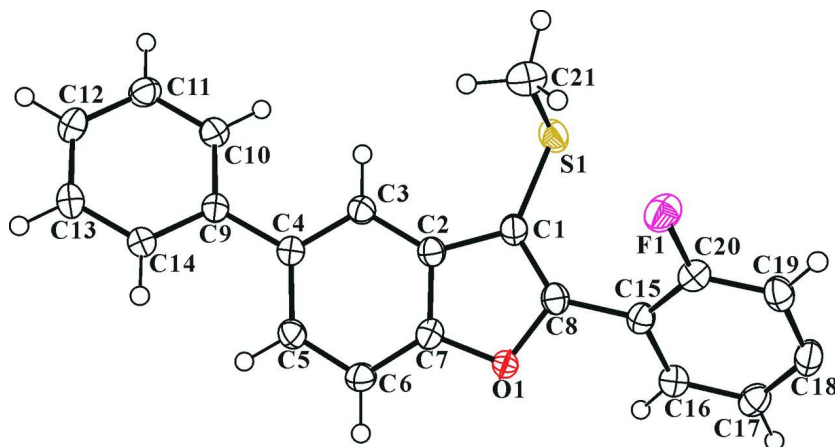


Figure 1

The molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are presented as small spheres of arbitrary radius.

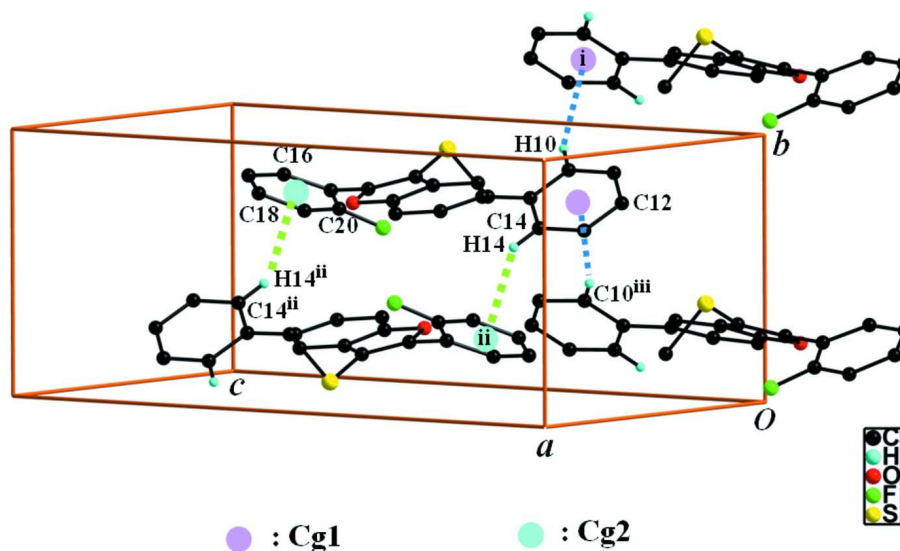


Figure 2

A view of the C–H... π interactions (dotted lines) in the crystal structure of the title compound. H atoms non-participating in hydrogen-bonding were omitted for clarity. [Symmetry codes: (i) $-x + 1/2, y + 1/2, -z + 1/2$; (ii) $-x + 1, -y + 1, -z + 1$; (iii) $-x + 1/2, y - 1/2, -z + 1/2$.]

2-(2-Fluorophenyl)-3-methylsulfanyl-5-phenyl-1-benzofuran

Crystal data

$C_{21}H_{15}FOS$
 $M_r = 334.39$
 Monoclinic, $P2_1/n$

Hall symbol: $-P 2_1n$
 $a = 11.1257 (2) \text{ \AA}$
 $b = 7.4232 (1) \text{ \AA}$

$c = 19.4212(3) \text{ \AA}$
 $\beta = 97.319(1)^\circ$
 $V = 1590.90(4) \text{ \AA}^3$
 $Z = 4$
 $F(000) = 696$
 $D_x = 1.396 \text{ Mg m}^{-3}$
 Melting point = 368–369 K

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
 Cell parameters from 7139 reflections
 $\theta = 2.2\text{--}27.4^\circ$
 $\mu = 0.22 \text{ mm}^{-1}$
 $T = 173 \text{ K}$
 Block, colourless
 $0.27 \times 0.19 \times 0.14 \text{ mm}$

Data collection

Bruker APEXII CCD
 diffractometer
 Radiation source: rotating anode
 Graphite multilayer monochromator
 Detector resolution: 10.0 pixels mm^{-1}
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2009)
 $T_{\min} = 0.689$, $T_{\max} = 0.746$

28795 measured reflections
 3967 independent reflections
 3190 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$
 $\theta_{\max} = 28.4^\circ$, $\theta_{\min} = 2.0^\circ$
 $h = -14 \rightarrow 14$
 $k = -9 \rightarrow 9$
 $l = -25 \rightarrow 25$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.101$
 $S = 1.05$
 3967 reflections
 218 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: difference Fourier map
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0483P)^2 + 0.4925P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.26 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.26 \text{ e \AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|--------------|--------------|---------------|----------------------------------|
| S1 | 0.16737 (3) | 0.90172 (5) | 0.526490 (19) | 0.03354 (12) |
| F1 | 0.19589 (8) | 0.59691 (12) | 0.63666 (4) | 0.0364 (2) |
| O1 | 0.50485 (9) | 0.72718 (13) | 0.56484 (5) | 0.0267 (2) |
| C1 | 0.31237 (13) | 0.81153 (18) | 0.52659 (7) | 0.0251 (3) |
| C2 | 0.37725 (12) | 0.79253 (18) | 0.46729 (7) | 0.0233 (3) |
| C3 | 0.34570 (12) | 0.80710 (18) | 0.39580 (7) | 0.0242 (3) |
| H3 | 0.2662 | 0.8429 | 0.3772 | 0.029* |
| C4 | 0.43212 (12) | 0.76849 (17) | 0.35187 (7) | 0.0221 (3) |
| C5 | 0.55052 (12) | 0.72200 (19) | 0.38149 (7) | 0.0258 (3) |

| | | | | |
|------|--------------|--------------|-------------|------------|
| H5 | 0.6099 | 0.6998 | 0.3515 | 0.031* |
| C6 | 0.58376 (13) | 0.7074 (2) | 0.45239 (7) | 0.0282 (3) |
| H6 | 0.6638 | 0.6757 | 0.4716 | 0.034* |
| C7 | 0.49417 (13) | 0.74160 (18) | 0.49375 (7) | 0.0244 (3) |
| C8 | 0.39147 (13) | 0.76675 (18) | 0.58301 (7) | 0.0244 (3) |
| C9 | 0.39863 (12) | 0.76923 (17) | 0.27537 (7) | 0.0218 (3) |
| C10 | 0.29923 (12) | 0.86819 (19) | 0.24410 (7) | 0.0266 (3) |
| H10 | 0.2541 | 0.9401 | 0.2721 | 0.032* |
| C11 | 0.26579 (13) | 0.8628 (2) | 0.17297 (7) | 0.0298 (3) |
| H11 | 0.1976 | 0.9300 | 0.1527 | 0.036* |
| C12 | 0.33101 (14) | 0.7602 (2) | 0.13126 (7) | 0.0327 (3) |
| H12 | 0.3076 | 0.7561 | 0.0825 | 0.039* |
| C13 | 0.43046 (14) | 0.6637 (2) | 0.16105 (7) | 0.0325 (3) |
| H13 | 0.4764 | 0.5946 | 0.1326 | 0.039* |
| C14 | 0.46343 (13) | 0.66746 (19) | 0.23219 (7) | 0.0265 (3) |
| H14 | 0.5315 | 0.5994 | 0.2520 | 0.032* |
| C15 | 0.38210 (13) | 0.75223 (17) | 0.65736 (7) | 0.0241 (3) |
| C16 | 0.47502 (13) | 0.81460 (18) | 0.70674 (7) | 0.0263 (3) |
| H16 | 0.5452 | 0.8668 | 0.6918 | 0.032* |
| C17 | 0.46672 (14) | 0.80178 (19) | 0.77690 (7) | 0.0299 (3) |
| H17 | 0.5302 | 0.8471 | 0.8097 | 0.036* |
| C18 | 0.36603 (14) | 0.7229 (2) | 0.79949 (7) | 0.0303 (3) |
| H18 | 0.3602 | 0.7153 | 0.8478 | 0.036* |
| C19 | 0.27373 (13) | 0.65489 (19) | 0.75195 (7) | 0.0284 (3) |
| H19 | 0.2050 | 0.5985 | 0.7671 | 0.034* |
| C20 | 0.28393 (13) | 0.67099 (18) | 0.68235 (7) | 0.0262 (3) |
| C21 | 0.07375 (16) | 0.7190 (3) | 0.48970 (9) | 0.0480 (4) |
| H21A | 0.1000 | 0.6826 | 0.4455 | 0.072* |
| H21B | -0.0110 | 0.7585 | 0.4818 | 0.072* |
| H21C | 0.0812 | 0.6167 | 0.5218 | 0.072* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|------------|------------|------------|--------------|--------------|--------------|
| S1 | 0.0334 (2) | 0.0392 (2) | 0.0296 (2) | 0.01220 (16) | 0.01019 (15) | 0.00356 (15) |
| F1 | 0.0344 (5) | 0.0430 (5) | 0.0304 (5) | -0.0135 (4) | -0.0006 (4) | 0.0033 (4) |
| O1 | 0.0262 (5) | 0.0344 (5) | 0.0195 (5) | -0.0012 (4) | 0.0030 (4) | 0.0012 (4) |
| C1 | 0.0285 (7) | 0.0251 (7) | 0.0223 (6) | 0.0019 (5) | 0.0058 (5) | 0.0005 (5) |
| C2 | 0.0254 (7) | 0.0215 (6) | 0.0234 (6) | 0.0003 (5) | 0.0041 (5) | -0.0007 (5) |
| C3 | 0.0247 (7) | 0.0249 (6) | 0.0231 (6) | 0.0027 (5) | 0.0036 (5) | 0.0005 (5) |
| C4 | 0.0249 (7) | 0.0194 (6) | 0.0222 (6) | -0.0018 (5) | 0.0038 (5) | -0.0001 (5) |
| C5 | 0.0234 (7) | 0.0299 (7) | 0.0250 (7) | -0.0019 (5) | 0.0059 (5) | -0.0004 (5) |
| C6 | 0.0222 (7) | 0.0361 (8) | 0.0259 (7) | -0.0013 (6) | 0.0017 (5) | 0.0016 (6) |
| C7 | 0.0271 (7) | 0.0249 (7) | 0.0210 (6) | -0.0033 (5) | 0.0026 (5) | -0.0003 (5) |
| C8 | 0.0272 (7) | 0.0227 (6) | 0.0241 (7) | -0.0016 (5) | 0.0060 (5) | -0.0004 (5) |
| C9 | 0.0232 (6) | 0.0210 (6) | 0.0216 (6) | -0.0036 (5) | 0.0040 (5) | -0.0008 (5) |
| C10 | 0.0264 (7) | 0.0268 (7) | 0.0267 (7) | 0.0008 (5) | 0.0040 (5) | -0.0009 (5) |
| C11 | 0.0290 (7) | 0.0319 (7) | 0.0274 (7) | 0.0024 (6) | -0.0008 (6) | 0.0020 (6) |

| | | | | | | |
|-----|------------|-------------|-------------|-------------|-------------|-------------|
| C12 | 0.0381 (8) | 0.0391 (8) | 0.0203 (7) | 0.0000 (7) | 0.0017 (6) | 0.0001 (6) |
| C13 | 0.0350 (8) | 0.0389 (8) | 0.0248 (7) | 0.0032 (6) | 0.0083 (6) | -0.0048 (6) |
| C14 | 0.0255 (7) | 0.0301 (7) | 0.0242 (7) | 0.0021 (6) | 0.0043 (5) | -0.0014 (5) |
| C15 | 0.0294 (7) | 0.0217 (6) | 0.0214 (6) | 0.0016 (5) | 0.0037 (5) | 0.0004 (5) |
| C16 | 0.0289 (7) | 0.0245 (7) | 0.0258 (7) | -0.0012 (5) | 0.0042 (5) | -0.0001 (5) |
| C17 | 0.0349 (8) | 0.0290 (7) | 0.0247 (7) | 0.0006 (6) | -0.0008 (6) | -0.0027 (5) |
| C18 | 0.0409 (8) | 0.0300 (7) | 0.0208 (7) | 0.0061 (6) | 0.0072 (6) | 0.0014 (5) |
| C19 | 0.0321 (8) | 0.0268 (7) | 0.0279 (7) | 0.0030 (6) | 0.0105 (6) | 0.0038 (5) |
| C20 | 0.0284 (7) | 0.0237 (6) | 0.0258 (7) | -0.0002 (5) | 0.0011 (5) | 0.0000 (5) |
| C21 | 0.0322 (9) | 0.0693 (13) | 0.0410 (10) | 0.0020 (8) | -0.0013 (7) | -0.0052 (9) |

Geometric parameters (Å, °)

| | | | |
|-----------|-------------|-------------|-------------|
| S1—C1 | 1.7464 (14) | C10—H10 | 0.9500 |
| S1—C21 | 1.8013 (19) | C11—C12 | 1.383 (2) |
| F1—C20 | 1.3521 (16) | C11—H11 | 0.9500 |
| O1—C7 | 1.3748 (16) | C12—C13 | 1.382 (2) |
| O1—C8 | 1.3840 (16) | C12—H12 | 0.9500 |
| C1—C8 | 1.3565 (19) | C13—C14 | 1.3839 (19) |
| C1—C2 | 1.4420 (18) | C13—H13 | 0.9500 |
| C2—C7 | 1.3887 (19) | C14—H14 | 0.9500 |
| C2—C3 | 1.3922 (18) | C15—C20 | 1.3882 (19) |
| C3—C4 | 1.3941 (18) | C15—C16 | 1.3964 (19) |
| C3—H3 | 0.9500 | C16—C17 | 1.381 (2) |
| C4—C5 | 1.4114 (19) | C16—H16 | 0.9500 |
| C4—C9 | 1.4851 (18) | C17—C18 | 1.384 (2) |
| C5—C6 | 1.3837 (19) | C17—H17 | 0.9500 |
| C5—H5 | 0.9500 | C18—C19 | 1.386 (2) |
| C6—C7 | 1.381 (2) | C18—H18 | 0.9500 |
| C6—H6 | 0.9500 | C19—C20 | 1.3760 (19) |
| C8—C15 | 1.4649 (18) | C19—H19 | 0.9500 |
| C9—C14 | 1.3956 (18) | C21—H21A | 0.9800 |
| C9—C10 | 1.4002 (19) | C21—H21B | 0.9800 |
| C10—C11 | 1.3847 (19) | C21—H21C | 0.9800 |
| C1—S1—C21 | 101.49 (8) | C10—C11—H11 | 119.8 |
| C7—O1—C8 | 105.94 (10) | C13—C12—C11 | 119.49 (13) |
| C8—C1—C2 | 106.37 (12) | C13—C12—H12 | 120.3 |
| C8—C1—S1 | 126.80 (11) | C11—C12—H12 | 120.3 |
| C2—C1—S1 | 126.37 (10) | C12—C13—C14 | 120.26 (13) |
| C7—C2—C3 | 119.70 (13) | C12—C13—H13 | 119.9 |
| C7—C2—C1 | 105.77 (12) | C14—C13—H13 | 119.9 |
| C3—C2—C1 | 134.47 (13) | C13—C14—C9 | 121.28 (13) |
| C2—C3—C4 | 119.23 (13) | C13—C14—H14 | 119.4 |
| C2—C3—H3 | 120.4 | C9—C14—H14 | 119.4 |
| C4—C3—H3 | 120.4 | C20—C15—C16 | 116.78 (12) |
| C3—C4—C5 | 118.78 (12) | C20—C15—C8 | 122.31 (12) |
| C3—C4—C9 | 120.45 (12) | C16—C15—C8 | 120.87 (13) |

| | | | |
|--------------|--------------|-----------------|--------------|
| C5—C4—C9 | 120.72 (12) | C17—C16—C15 | 121.21 (13) |
| C6—C5—C4 | 122.81 (13) | C17—C16—H16 | 119.4 |
| C6—C5—H5 | 118.6 | C15—C16—H16 | 119.4 |
| C4—C5—H5 | 118.6 | C16—C17—C18 | 120.04 (14) |
| C7—C6—C5 | 116.31 (13) | C16—C17—H17 | 120.0 |
| C7—C6—H6 | 121.8 | C18—C17—H17 | 120.0 |
| C5—C6—H6 | 121.8 | C17—C18—C19 | 120.26 (13) |
| O1—C7—C6 | 126.40 (13) | C17—C18—H18 | 119.9 |
| O1—C7—C2 | 110.48 (12) | C19—C18—H18 | 119.9 |
| C6—C7—C2 | 123.11 (13) | C20—C19—C18 | 118.43 (13) |
| C1—C8—O1 | 111.39 (11) | C20—C19—H19 | 120.8 |
| C1—C8—C15 | 133.94 (13) | C18—C19—H19 | 120.8 |
| O1—C8—C15 | 114.67 (12) | F1—C20—C19 | 117.78 (12) |
| C14—C9—C10 | 117.60 (12) | F1—C20—C15 | 118.92 (12) |
| C14—C9—C4 | 121.08 (12) | C19—C20—C15 | 123.24 (13) |
| C10—C9—C4 | 121.30 (12) | S1—C21—H21A | 109.5 |
| C11—C10—C9 | 120.98 (13) | S1—C21—H21B | 109.5 |
| C11—C10—H10 | 119.5 | H21A—C21—H21B | 109.5 |
| C9—C10—H10 | 119.5 | S1—C21—H21C | 109.5 |
| C12—C11—C10 | 120.38 (13) | H21A—C21—H21C | 109.5 |
| C12—C11—H11 | 119.8 | H21B—C21—H21C | 109.5 |
| | | | |
| C21—S1—C1—C8 | -112.43 (14) | C3—C4—C9—C14 | -154.56 (13) |
| C21—S1—C1—C2 | 76.40 (14) | C5—C4—C9—C14 | 22.97 (19) |
| C8—C1—C2—C7 | -1.73 (15) | C3—C4—C9—C10 | 23.80 (19) |
| S1—C1—C2—C7 | 170.91 (11) | C5—C4—C9—C10 | -158.67 (13) |
| C8—C1—C2—C3 | 175.29 (15) | C14—C9—C10—C11 | 0.9 (2) |
| S1—C1—C2—C3 | -12.1 (2) | C4—C9—C10—C11 | -177.53 (13) |
| C7—C2—C3—C4 | 0.60 (19) | C9—C10—C11—C12 | -0.6 (2) |
| C1—C2—C3—C4 | -176.10 (14) | C10—C11—C12—C13 | -0.4 (2) |
| C2—C3—C4—C5 | -2.27 (19) | C11—C12—C13—C14 | 1.0 (2) |
| C2—C3—C4—C9 | 175.31 (12) | C12—C13—C14—C9 | -0.7 (2) |
| C3—C4—C5—C6 | 2.1 (2) | C10—C9—C14—C13 | -0.3 (2) |
| C9—C4—C5—C6 | -175.46 (13) | C4—C9—C14—C13 | 178.15 (13) |
| C4—C5—C6—C7 | -0.2 (2) | C1—C8—C15—C20 | 43.6 (2) |
| C8—O1—C7—C6 | -177.46 (13) | O1—C8—C15—C20 | -136.50 (13) |
| C8—O1—C7—C2 | 1.10 (14) | C1—C8—C15—C16 | -138.77 (17) |
| C5—C6—C7—O1 | 176.77 (13) | O1—C8—C15—C16 | 41.10 (17) |
| C5—C6—C7—C2 | -1.6 (2) | C20—C15—C16—C17 | -2.4 (2) |
| C3—C2—C7—O1 | -177.18 (11) | C8—C15—C16—C17 | 179.85 (13) |
| C1—C2—C7—O1 | 0.37 (15) | C15—C16—C17—C18 | 1.3 (2) |
| C3—C2—C7—C6 | 1.4 (2) | C16—C17—C18—C19 | 0.6 (2) |
| C1—C2—C7—C6 | 178.99 (13) | C17—C18—C19—C20 | -1.2 (2) |
| C2—C1—C8—O1 | 2.52 (15) | C18—C19—C20—F1 | 177.08 (12) |
| S1—C1—C8—O1 | -170.08 (10) | C18—C19—C20—C15 | -0.1 (2) |
| C2—C1—C8—C15 | -177.61 (14) | C16—C15—C20—F1 | -175.28 (12) |
| S1—C1—C8—C15 | 9.8 (2) | C8—C15—C20—F1 | 2.4 (2) |
| C7—O1—C8—C1 | -2.29 (15) | C16—C15—C20—C19 | 1.9 (2) |

| | | | |
|--------------|-------------|----------------|-------------|
| C7—O1—C8—C15 | 177.81 (11) | C8—C15—C20—C19 | 179.55 (13) |
|--------------|-------------|----------------|-------------|

Hydrogen-bond geometry (Å, °)

Cg1 and Cg2 are the C9–C14 phenyl and 2-fluorophenyl rings, respectively.

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|------------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| C10—H10 \cdots Cg1 ⁱ | 0.95 | 2.82 | 3.682 (2) | 131 |
| C14—H14 \cdots Cg2 ⁱⁱ | 0.95 | 2.71 | 3.528 (2) | 145 |

Symmetry codes: (i) $-x+1/2, y+1/2, -z+1/2$; (ii) $-x+1, -y+1, -z+1$.