

5-[4'-(5-Benzyl-2H-tetrazol-2-yl)-methyl]biphenyl-2-yl]-1H-tetrazole monohydrate

Gangadhar Y. Meti,^a S. Jeyaseelan,^b R. R. Kamble,^{a*} Atakuri Dorababu^a and H. C. Devarajgowda^{c*}

^aDepartment of Studies in Chemistry, Karnataka University, Dharwad 580 003, Karnataka, India, ^bDepartment of Physics, St. Philomena's College, Mysore 570 006, Karnataka, India, and ^cDepartment of Physics, Yuvaraja's College (Constituent College), University of Mysore, Mysore 570 005, Karnataka, India
Correspondence e-mail: kamchem9@gmail.com, devarajgowda@yahoo.com

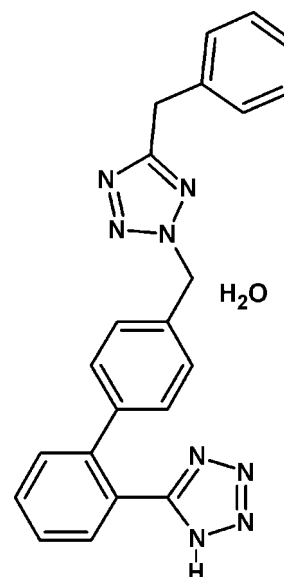
Received 23 March 2013; accepted 11 April 2013

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.048; wR factor = 0.107; data-to-parameter ratio = 13.8.

In the title compound, $\text{C}_{22}\text{H}_{18}\text{N}_8\cdot\text{H}_2\text{O}$, the dihedral angle between the tetrazole rings is $69.58(1)^\circ$ while the terminal phenyl ring makes dihedral angles of $26.98(8)$ and $39.75(8)^\circ$ with the other benzene rings. The rings of the biphenyl unit subtend a dihedral angle of $55.23(8)^\circ$. In the crystal, the solvent water molecule is linked to the main molecule *via* an $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond. In addition, $\text{C}-\text{H}\cdots\text{N}$ and $\text{O}-\text{H}\cdots\text{N}$ hydrogen bonds link the components into chains along [010]. The crystal structure also features $\text{C}-\text{H}\cdots\pi$ and $\pi-\pi$ interactions, with centroid-centroid distances of $3.6556(9)$ and $3.826(1)$ Å.

Related literature

For general background to biphenyl derivatives, see: Li *et al.* (2011); Tomori *et al.* (2000). For the synthesis and biological activity of tetrazole derivatives, see: Kamble *et al.* (2011); Rao & Babu (2011). For biological properties of tetrazole-derivatized biphenyl moieties, see: Zhang *et al.* (2008); Wang *et al.* (2010); Reddy *et al.* (2007). For related structures, see: Zhang *et al.* (2004). For the extinction correction, see: Larson (1970).



Experimental

Crystal data

$\text{C}_{22}\text{H}_{18}\text{N}_8\cdot\text{H}_2\text{O}$
 $M_r = 412.45$
Monoclinic, $P2_1/c$
 $a = 14.7659(4)$ Å
 $b = 7.6507(3)$ Å
 $c = 18.2922(5)$ Å
 $\beta = 97.153(2)^\circ$

$V = 2050.38(11)$ Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 293$ K
 $0.24 \times 0.20 \times 0.12$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: ψ scan (SADABS; Sheldrick, 2007)
 $T_{\min} = 0.770$, $T_{\max} = 1.000$

18220 measured reflections
3923 independent reflections
3469 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.107$
 $S = 1.00$
3881 reflections

281 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.49$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.41$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

C_{g1} , C_{g3} , C_{g4} and C_{g5} are the centroids of the C_6/N_2-N_5 tetrazole ring, the C_8-C13 benzene ring, the $C15-C20$ benzene ring and the $C21/C22/C28-C31$ benzene ring, respectively.

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------|-------|-------------|-------------|---------------|
| $C7-H71\cdots N25^i$ | 0.99 | 2.59 | 3.514 (2) | 156 |
| $N24-H241\cdots O1^{ii}$ | 0.92 | 1.79 | 2.703 (2) | 173 |
| $O1-H11\cdots N27$ | 0.83 | 2.35 | 2.950 (2) | 130 |
| $C9-H91\cdots C_{g4}^{iii}$ | 0.96 | 2.85 | 3.418 (1) | 119 |
| $C12-H121\cdots C_{g1}^i$ | 0.94 | 2.82 | 3.602 (1) | 141 |
| $C14-H142\cdots C_{g3}^{iv}$ | 0.96 | 2.72 | 3.676 (1) | 171 |
| $C29-H291\cdots C_{g5}^v$ | 0.96 | 2.80 | 3.682 (2) | 153 |
| $C31-H311\cdots C_{g3}^{vi}$ | 0.95 | 2.96 | 3.582 (1) | 125 |

Symmetry codes: (i) $x, y+1, z$; (ii) $x, y-1, z$; (iii) $x, -y+\frac{1}{2}, z+\frac{1}{2}$; (iv) $-x+2, -y+1, -z+1$; (v) $-x+1, y-\frac{1}{2}, -z+\frac{1}{2}$; (vi) $x, -y+\frac{1}{2}, z-\frac{1}{2}$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve

structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *ORTEP-3* (Farrugia, 2012); software used to prepare material for publication: *CAMERON* (Watkin *et al.*, 1996).

The authors thank the University Sophisticated Instrumental Centre, Karnatak University, Dharwad, for the data collection and Professor T. N. Guru Row, Solid State and Structural Chemistry Unit, Indian Institute of Science, Bangalore, for his constant support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5301).

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). *J. Appl. Cryst.* **36**, 1487.
- Bruker (2001). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
- Kamble, R. R., Biradar, D. B., Meti, G. Y., Taj, T., Gireesh, T., Khazi, I. M., Vaidynathan, S. T., Mohandoss, R., Sridhar, B. & Parthasarathi, V. (2011). *J. Chem. Sci.* **123**, 393–401.
- Larson, A. C. (1970). *Crystallographic Computing*, edited by F. R. Ahmed, S. R. Hall & C. P. Huber, pp. 291–294. Copenhagen: Munksgaard.
- Li, W., Xu, Z., Sun, P., Jiang, X. & Fang, M. (2011). *Org. Lett.* **13**, 1286–1289.
- Rao, S. N. & Babu, K. S. (2011). *Org. Commun.* **4**, 105–111.
- Reddy, K. S., Srinivasan, N., Reddy, C. R., Kolla, N., Anjaneyulu, Y., Venkatraman, S., Bhattacharya, A. & Mathad, V. T. (2007). *Org. Process Res. Dev.* **11**, 81–85.
- Sheldrick, G. M. (2007). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Tomori, H., Fox, J. M. & Buchwald, S. L. (2000). *J. Org. Chem.* **65**, 5334–5341.
- Wang, P., Zheng, G., Wang, Y., Wang, X., Li, Y. & Xiang, W. (2010). *Tetrahedron*, **66**, 5402–5406.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). *CAMERON*. Chemical Crystallography Laboratory, Oxford, England.
- Zhang, H., Yang, B., Zheng, Y., Yang, G., Ye, L., Ma, Y., Chen, X., Cheng, G. & Liu, S. (2004). *J. Phys. Chem.* **108**, 9571–9573.
- Zhang, C. X., Zheng, G. J., Bi, F. Q. & Li, Y. L. (2008). *Chin. Chem. Lett.* **19**, 759–761.

supporting information

Acta Cryst. (2013). E69, o743–o744 [https://doi.org/10.1107/S1600536813009963]

5-{4'-[(5-Benzyl-2*H*-tetrazol-2-yl)methyl]biphenyl-2-yl}-1*H*-tetrazole monohydrate

Gangadhar Y. Meti, S. Jeyaseelan, R. R. Kamble, Atakuri Dorababu and H. C. Devarajgowda

S1. Comment

Biphenyl and tetrazole and derivatives are present in many of the bioactive heterocyclic compounds which are of wide interest because of their diverse pharmaceutical and clinical applications (Li *et al.*, 2011; Tomori *et al.*, 2000). As the tetrazole moiety functions as a carboxylic acid biostere that imparts the greater metabolic stability and increased absorption relative to the carboxylic acid. Tetrazole linked biphenyl moiety is the building block of all the Antihypertensive saratans (Rao & Babu, 2011; Reddy *et al.*, 2007; Wang *et al.*, 2010; Zhang *et al.*, 2008).

The asymmetric unit of 5-{4'-[(5-Benzyl-2*H*-tetrazol-2-yl) methyl] biphenyl -2-yl}-1*H*-tetrazole is shown in Fig. 1. the dihedral angle between the tetrazole (C6/N2/N3–N5, C23/C24/N25–N27) rings is 69.58 (1)° while the terminal phenyl ring (C8–C13) makes dihedral angles of 26.98 (8)° and 39.75 (8)° with the other benzene rings (C15–C20 & C21/C22/C28/C29–C31) and also dihedral angle for biphenyl (C15–C20 & C21\C22\C28\...C31 is 55.23 (8)°. In the crystal, the solvent water molecule is linked to the main molecule via N27—H12...O1 and O1—H11...N27 hydrogen bonds. In addition, there are intermolecular C7—H71...N25 hydrogen. The structure contains π - π , with a centroid–centroid ($C_g(1)$, C6/N2/N3–N5) and ($C_g(2)$, C23/24/N25–N27) distance of 3.6556 (9) Å and 3.826 (1) Å° respectively and also C—H... π interactions (Table 1). In the structure, all bond lengths and angles are within normal ranges (Kamble *et al.*, 2011; Zhang *et al.*, 2004). The crystal packing shows stack the molecules along the *b* axis (Fig. 2).

S2. Experimental

Prepared using 4'-[(5-benzyl-2*H*-tetrazol-2-yl) methyl] biphenyl-2-carbonitrile according the method reported by Kamble *et al.* (2011). (mp. 393 K). Spectral data IR (KBr) cm^{-1} 3421 (N—H), 3125, 2985 (C—H of CH_2), 1616 (C=C, str). ^1H NMR (CDCl_3): 7.3–8.00 (13 H, m Ar—H) 5.7 (2*H*, s, biphenyl CH_2) 4.2 (2*H*, s, benzyl CH_2). MS (*m/z*, 70 eV): 394 (*M*⁺), 382, 318, 235, 205, 192, 178 (base peak), 165, 91 and 77.

S3. Refinement

The H atoms were all located in a difference map, but those attached to carbon atoms were repositioned geometrically, with N—H = 0.91 Å, O—H = 0.83 Å and 0.96 Å, C—H = 0.93–0.98 Å for aromatic H and C—H = 0.97–0.99 Å for methylene H and refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic and methylene H.

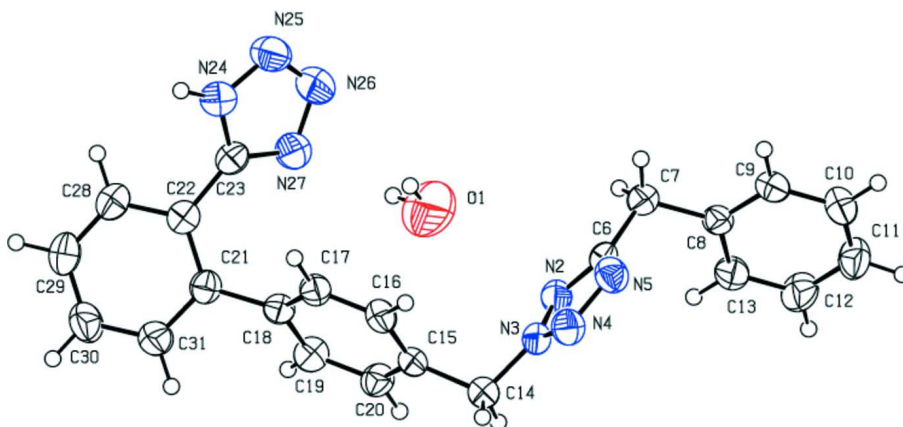


Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms are shown as spheres of arbitrary radius.

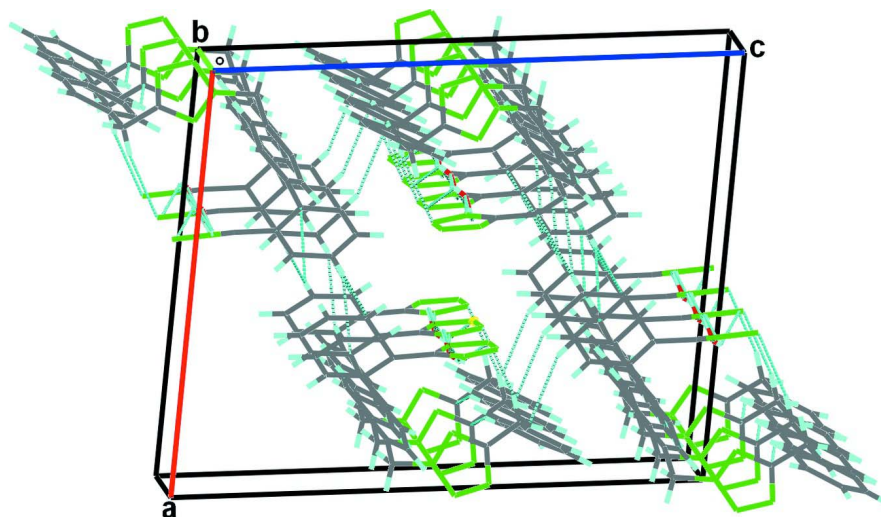


Figure 2

Crystal packing for the title compound with hydrogen bonds drawn as dashed lines.

5-{4'-[(5-Benzyl-2H-tetrazol-2-yl)methyl]biphenyl-2-yl}-1H-tetrazole monohydrate

Crystal data

$C_{22}H_{18}N_8 \cdot H_2O$

$M_r = 412.45$

Monoclinic, $P2_1/c$

Hall symbol: $-P 2_1/c$

$a = 14.7659 (4) \text{ \AA}$

$b = 7.6507 (3) \text{ \AA}$

$c = 18.2922 (5) \text{ \AA}$

$\beta = 97.153 (2)^\circ$

$V = 2050.38 (11) \text{ \AA}^3$

$Z = 4$

$F(000) = 864$

$D_x = 1.336 \text{ Mg m}^{-3}$

Melting point: 393 K

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 3923 reflections

$\theta = 1.4\text{--}25.9^\circ$

$\mu = 0.09 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Plate, colourless

$0.24 \times 0.20 \times 0.12 \text{ mm}$

Data collection

Bruker SMART CCD area-detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 $\omega/2\theta$ scans
 Absorption correction: ψ scan
 (SADABS; Sheldrick, 2007)
 $T_{\min} = 0.770$, $T_{\max} = 1.000$

18220 measured reflections
 3923 independent reflections
 3469 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$
 $\theta_{\max} = 25.9^\circ$, $\theta_{\min} = 1.4^\circ$
 $h = -18 \rightarrow 18$
 $k = -9 \rightarrow 4$
 $l = -22 \rightarrow 22$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.107$
 $S = 1.00$
 3881 reflections
 281 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods
 Hydrogen site location: difference Fourier map
 H-atom parameters constrained

Method, part 1, Chebychev polynomial,
 (Watkin, 1994, Prince, 1982) [weight] =
 $1.0/[A_0 * T_0(x) + A_1 * T_1(x) \dots + A_{n-1} * T_{n-1}(x)]$
 where A_i are the Chebychev coefficients listed
 below and $x = F / F_{\max}$ Method = Robust
 Weighting (Prince, 1982) $W = [\text{weight}] * [1 - (\Delta F / 6 * \sigma F)^2]^2$ A_i are: 2.08 2.65 0.641
 $(\Delta/\sigma)_{\max} = 0.0003916$
 $\Delta\rho_{\max} = 0.49 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.41 \text{ e } \text{\AA}^{-3}$
 Extinction correction: Larson (1970), eq. 22
 Extinction coefficient: 194 (8)

Special details

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|--------------|--------------|----------------------------------|
| O1 | 0.6409 (2) | 0.1472 (2) | 0.47303 (14) | 0.1327 |
| N2 | 0.83657 (9) | 0.38128 (17) | 0.49755 (7) | 0.0418 |
| N3 | 0.90644 (9) | 0.31880 (16) | 0.46640 (7) | 0.0393 |
| N4 | 0.98138 (9) | 0.29937 (19) | 0.51215 (7) | 0.0458 |
| N5 | 0.96150 (10) | 0.35174 (19) | 0.57699 (7) | 0.0467 |
| C6 | 0.87317 (10) | 0.40044 (19) | 0.56678 (8) | 0.0376 |
| C7 | 0.82354 (12) | 0.4669 (2) | 0.62728 (9) | 0.0450 |
| C8 | 0.84982 (10) | 0.6537 (2) | 0.64806 (8) | 0.0379 |
| C9 | 0.89345 (11) | 0.6936 (2) | 0.71693 (8) | 0.0405 |
| C10 | 0.91827 (12) | 0.8634 (2) | 0.73543 (10) | 0.0497 |
| C11 | 0.89843 (14) | 0.9950 (2) | 0.68486 (11) | 0.0559 |
| C12 | 0.85406 (15) | 0.9565 (2) | 0.61566 (11) | 0.0583 |
| C13 | 0.82998 (13) | 0.7868 (2) | 0.59720 (9) | 0.0490 |
| C14 | 0.90005 (12) | 0.2733 (2) | 0.38826 (9) | 0.0469 |
| C15 | 0.83532 (10) | 0.1230 (2) | 0.36911 (8) | 0.0376 |
| C16 | 0.85021 (11) | -0.0387 (2) | 0.40347 (8) | 0.0413 |
| C17 | 0.79254 (11) | -0.1773 (2) | 0.38405 (8) | 0.0396 |
| C18 | 0.71895 (10) | -0.1599 (2) | 0.32939 (8) | 0.0377 |

| | | | | |
|------|--------------|-------------|--------------|---------|
| C19 | 0.70374 (11) | 0.0023 (2) | 0.29551 (9) | 0.0441 |
| C20 | 0.76118 (11) | 0.1420 (2) | 0.31558 (9) | 0.0435 |
| C21 | 0.66114 (10) | -0.3118 (2) | 0.30243 (8) | 0.0396 |
| C22 | 0.61489 (10) | -0.4190 (2) | 0.34797 (8) | 0.0393 |
| C23 | 0.61203 (10) | -0.3818 (2) | 0.42664 (9) | 0.0419 |
| N24 | 0.61342 (10) | -0.5040 (2) | 0.47864 (8) | 0.0503 |
| N25 | 0.60308 (12) | -0.4293 (3) | 0.54318 (8) | 0.0616 |
| N26 | 0.59523 (12) | -0.2645 (3) | 0.53016 (9) | 0.0658 |
| N27 | 0.60071 (11) | -0.2297 (2) | 0.45804 (9) | 0.0574 |
| C28 | 0.56465 (11) | -0.5620 (2) | 0.31777 (10) | 0.0479 |
| C29 | 0.55904 (12) | -0.5987 (3) | 0.24391 (10) | 0.0559 |
| C30 | 0.60313 (13) | -0.4927 (3) | 0.19858 (10) | 0.0575 |
| C31 | 0.65331 (12) | -0.3514 (2) | 0.2277 (1) | 0.0513 |
| H72 | 0.8360 | 0.3888 | 0.6694 | 0.0557* |
| H71 | 0.7576 | 0.4619 | 0.6101 | 0.0561* |
| H91 | 0.9083 | 0.6017 | 0.7535 | 0.0474* |
| H101 | 0.9499 | 0.8872 | 0.7832 | 0.0609* |
| H111 | 0.9198 | 1.1134 | 0.6966 | 0.0685* |
| H121 | 0.8397 | 1.0480 | 0.5800 | 0.0720* |
| H131 | 0.7977 | 0.7624 | 0.5483 | 0.0586* |
| H142 | 0.9617 | 0.2406 | 0.3806 | 0.0582* |
| H141 | 0.8769 | 0.3783 | 0.3599 | 0.0583* |
| H161 | 0.9003 | -0.0516 | 0.4409 | 0.0493* |
| H171 | 0.8033 | -0.2867 | 0.4087 | 0.0477* |
| H191 | 0.6537 | 0.0202 | 0.2569 | 0.0536* |
| H201 | 0.7493 | 0.2551 | 0.2919 | 0.0532* |
| H281 | 0.5325 | -0.6371 | 0.3493 | 0.0572* |
| H291 | 0.5220 | -0.6962 | 0.2232 | 0.0682* |
| H301 | 0.6007 | -0.5195 | 0.1476 | 0.0699* |
| H311 | 0.6853 | -0.2767 | 0.1966 | 0.0623* |
| H12 | 0.5816 | 0.1027 | 0.4540 | 0.1934* |
| H241 | 0.6242 | -0.6206 | 0.4728 | 0.0639* |
| H11 | 0.6563 | 0.0508 | 0.4913 | 0.1953* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|-------------|-------------|-------------|
| O1 | 0.203 (3) | 0.0529 (11) | 0.150 (2) | 0.0060 (13) | 0.0539 (19) | 0.0055 (11) |
| N2 | 0.0411 (7) | 0.0413 (7) | 0.0432 (7) | -0.0038 (6) | 0.0065 (5) | -0.0056 (6) |
| N3 | 0.0433 (7) | 0.0357 (7) | 0.0398 (7) | -0.0059 (5) | 0.0082 (5) | -0.0056 (5) |
| N4 | 0.0450 (7) | 0.0446 (8) | 0.0475 (8) | 0.0014 (6) | 0.0041 (6) | -0.0028 (6) |
| N5 | 0.0513 (8) | 0.0472 (8) | 0.0403 (7) | 0.0009 (6) | 0.0000 (6) | -0.0014 (6) |
| C6 | 0.0474 (8) | 0.0267 (7) | 0.0391 (8) | -0.0051 (6) | 0.0066 (6) | 0.0003 (6) |
| C7 | 0.0564 (9) | 0.0375 (8) | 0.0431 (8) | -0.0063 (7) | 0.0147 (7) | -0.0016 (7) |
| C8 | 0.0441 (8) | 0.0344 (8) | 0.0368 (7) | 0.0008 (6) | 0.0118 (6) | -0.0013 (6) |
| C9 | 0.0450 (8) | 0.0420 (9) | 0.0353 (7) | 0.0063 (7) | 0.0081 (6) | 0.0020 (6) |
| C10 | 0.0564 (10) | 0.0497 (10) | 0.0419 (8) | 0.0007 (8) | 0.0013 (7) | -0.0093 (7) |
| C11 | 0.0668 (11) | 0.0369 (9) | 0.0628 (11) | -0.0026 (8) | 0.0030 (9) | -0.0073 (8) |

| | | | | | | |
|-----|-------------|-------------|-------------|-------------|-------------|-------------|
| C12 | 0.0781 (13) | 0.0379 (9) | 0.0565 (10) | -0.0010 (9) | -0.0011 (9) | 0.0108 (8) |
| C13 | 0.0627 (10) | 0.0459 (9) | 0.0369 (8) | -0.0045 (8) | 0.0007 (7) | 0.0036 (7) |
| C14 | 0.0559 (9) | 0.0470 (9) | 0.0396 (8) | -0.0124 (8) | 0.0128 (7) | -0.0063 (7) |
| C15 | 0.0427 (8) | 0.0384 (8) | 0.0335 (7) | -0.0038 (6) | 0.0116 (6) | -0.0059 (6) |
| C16 | 0.0439 (8) | 0.0441 (9) | 0.0346 (7) | -0.0004 (7) | 0.0006 (6) | -0.0040 (6) |
| C17 | 0.0463 (8) | 0.0362 (8) | 0.0361 (7) | 0.0005 (6) | 0.0039 (6) | 0.0005 (6) |
| C18 | 0.0382 (7) | 0.0384 (8) | 0.0374 (7) | -0.0011 (6) | 0.0084 (6) | -0.0013 (6) |
| C19 | 0.0398 (8) | 0.0453 (9) | 0.0458 (9) | 0.0023 (7) | -0.0002 (6) | 0.0038 (7) |
| C20 | 0.0489 (9) | 0.0352 (8) | 0.0467 (8) | 0.0012 (7) | 0.0074 (7) | 0.0033 (7) |
| C21 | 0.0350 (7) | 0.0399 (8) | 0.0433 (8) | -0.0007 (6) | 0.0027 (6) | -0.0014 (7) |
| C22 | 0.0347 (7) | 0.0385 (8) | 0.0442 (8) | 0.0028 (6) | 0.0037 (6) | 0.0013 (7) |
| C23 | 0.0368 (7) | 0.0419 (9) | 0.0473 (9) | -0.0043 (6) | 0.0067 (6) | 0.0016 (7) |
| N24 | 0.0527 (8) | 0.0508 (8) | 0.0479 (8) | -0.0012 (7) | 0.0082 (6) | 0.0028 (6) |
| N25 | 0.0634 (10) | 0.0743 (12) | 0.0490 (9) | -0.0070 (8) | 0.0143 (7) | -0.0005 (8) |
| N26 | 0.0736 (11) | 0.0714 (12) | 0.0564 (9) | -0.0080 (9) | 0.0236 (8) | -0.0117 (8) |
| N27 | 0.0702 (10) | 0.0481 (9) | 0.0574 (9) | -0.0025 (7) | 0.0211 (8) | -0.0061 (7) |
| C28 | 0.0438 (8) | 0.0432 (9) | 0.0557 (10) | -0.0061 (7) | 0.0030 (7) | 0.0012 (7) |
| C29 | 0.0502 (10) | 0.0518 (10) | 0.0629 (11) | -0.0108 (8) | -0.0038 (8) | -0.0100 (9) |
| C30 | 0.0583 (10) | 0.0660 (12) | 0.0460 (9) | -0.0093 (9) | -0.0017 (8) | -0.0125 (9) |
| C31 | 0.0519 (9) | 0.0587 (11) | 0.0434 (8) | -0.0105 (8) | 0.0067 (7) | -0.0026 (8) |

Geometric parameters (Å, °)

| | | | |
|----------|-------------|----------|-----------|
| O1—H12 | 0.963 | C15—C20 | 1.383 (2) |
| O1—H11 | 0.830 | C16—C17 | 1.379 (2) |
| N2—N3 | 1.3281 (18) | C16—H161 | 0.947 |
| N2—C6 | 1.3218 (19) | C17—C18 | 1.388 (2) |
| N3—N4 | 1.3101 (19) | C17—H171 | 0.954 |
| N3—C14 | 1.4626 (19) | C18—C19 | 1.393 (2) |
| N4—N5 | 1.3191 (19) | C18—C21 | 1.489 (2) |
| N5—C6 | 1.347 (2) | C19—C20 | 1.385 (2) |
| C6—C7 | 1.491 (2) | C19—H191 | 0.966 |
| C7—C8 | 1.517 (2) | C20—H201 | 0.974 |
| C7—H72 | 0.975 | C21—C22 | 1.405 (2) |
| C7—H71 | 0.985 | C21—C31 | 1.390 (2) |
| C8—C9 | 1.376 (2) | C22—C23 | 1.473 (2) |
| C8—C13 | 1.386 (2) | C22—C28 | 1.396 (2) |
| C9—C10 | 1.381 (2) | C23—N24 | 1.332 (2) |
| C9—H91 | 0.977 | C23—N27 | 1.317 (2) |
| C10—C11 | 1.374 (3) | N24—N25 | 1.337 (2) |
| C10—H101 | 0.955 | N24—H241 | 0.915 |
| C11—C12 | 1.382 (3) | N25—N26 | 1.286 (3) |
| C11—H111 | 0.975 | N26—N27 | 1.358 (2) |
| C12—C13 | 1.377 (3) | C28—C29 | 1.372 (3) |
| C12—H121 | 0.962 | C28—H281 | 0.978 |
| C13—H131 | 0.978 | C29—C30 | 1.380 (3) |
| C14—C15 | 1.508 (2) | C29—H291 | 0.973 |
| C14—H142 | 0.971 | C30—C31 | 1.380 (3) |

| | | | |
|---------------|-------------|--------------|-------------|
| C14—H141 | 0.994 | C30—H301 | 0.952 |
| C15—C16 | 1.392 (2) | C31—H311 | 0.970 |
| H12—O1—H11 | 91.3 | C15—C16—C17 | 120.61 (14) |
| N3—N2—C6 | 101.78 (12) | C15—C16—H161 | 118.9 |
| N2—N3—N4 | 113.97 (12) | C17—C16—H161 | 120.5 |
| N2—N3—C14 | 123.09 (13) | C16—C17—C18 | 120.98 (14) |
| N4—N3—C14 | 122.94 (13) | C16—C17—H171 | 119.5 |
| N3—N4—N5 | 105.97 (12) | C18—C17—H171 | 119.5 |
| N4—N5—C6 | 106.23 (13) | C17—C18—C19 | 118.31 (14) |
| N5—C6—N2 | 112.06 (14) | C17—C18—C21 | 121.92 (14) |
| N5—C6—C7 | 123.27 (14) | C19—C18—C21 | 119.60 (13) |
| N2—C6—C7 | 124.67 (14) | C18—C19—C20 | 120.66 (14) |
| C6—C7—C8 | 111.93 (13) | C18—C19—H191 | 121.2 |
| C6—C7—H72 | 108.4 | C20—C19—H191 | 118.2 |
| C8—C7—H72 | 111.3 | C19—C20—C15 | 120.74 (14) |
| C6—C7—H71 | 107.9 | C19—C20—H201 | 119.8 |
| C8—C7—H71 | 109.2 | C15—C20—H201 | 119.4 |
| H72—C7—H71 | 108.0 | C18—C21—C22 | 124.06 (14) |
| C7—C8—C9 | 120.94 (14) | C18—C21—C31 | 118.03 (14) |
| C7—C8—C13 | 119.87 (14) | C22—C21—C31 | 117.91 (14) |
| C9—C8—C13 | 119.19 (15) | C21—C22—C23 | 122.78 (14) |
| C8—C9—C10 | 120.84 (15) | C21—C22—C28 | 119.75 (15) |
| C8—C9—H91 | 120.5 | C23—C22—C28 | 117.37 (14) |
| C10—C9—H91 | 118.6 | C22—C23—N24 | 124.20 (15) |
| C9—C10—C11 | 119.77 (16) | C22—C23—N27 | 128.14 (15) |
| C9—C10—H101 | 119.2 | N24—C23—N27 | 107.47 (15) |
| C11—C10—H101 | 121.0 | C23—N24—N25 | 109.70 (16) |
| C10—C11—C12 | 119.85 (17) | C23—N24—H241 | 126.0 |
| C10—C11—H111 | 119.8 | N25—N24—H241 | 124.1 |
| C12—C11—H111 | 120.2 | N24—N25—N26 | 105.85 (15) |
| C11—C12—C13 | 120.23 (17) | N25—N26—N27 | 110.93 (15) |
| C11—C12—H121 | 120.3 | N26—N27—C23 | 106.06 (15) |
| C13—C12—H121 | 119.4 | C22—C28—C29 | 120.93 (16) |
| C8—C13—C12 | 120.11 (16) | C22—C28—H281 | 120.0 |
| C8—C13—H131 | 120.8 | C29—C28—H281 | 119.0 |
| C12—C13—H131 | 119.1 | C28—C29—C30 | 119.74 (16) |
| N3—C14—C15 | 111.74 (12) | C28—C29—H291 | 120.1 |
| N3—C14—H142 | 104.8 | C30—C29—H291 | 120.1 |
| C15—C14—H142 | 110.4 | C29—C30—C31 | 119.95 (17) |
| N3—C14—H141 | 107.1 | C29—C30—H301 | 120.0 |
| C15—C14—H141 | 109.2 | C31—C30—H301 | 120.0 |
| H142—C14—H141 | 113.6 | C21—C31—C30 | 121.71 (17) |
| C14—C15—C16 | 120.83 (14) | C21—C31—H311 | 117.4 |
| C14—C15—C20 | 120.47 (14) | C30—C31—H311 | 120.8 |
| C16—C15—C20 | 118.68 (14) | | |

Hydrogen-bond geometry (Å, °)

Cg1, Cg3, Cg4 and Cg5 are the centroids of the C6/N2–N5 tetrazole ring, the C8–C13 benzene ring, the C15–C20 benzene ring and the C21/C22/C28–C31 benzene ring, respectively.

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|------------------------------|-------------|---------------|-----------------------|-------------------------|
| C7—H71...N25 ⁱ | 0.99 | 2.59 | 3.514 (2) | 156 |
| N24—H241...O1 ⁱⁱ | 0.92 | 1.79 | 2.703 (2) | 173 |
| O1—H11...N27 | 0.83 | 2.35 | 2.950 (2) | 130 |
| C9—H91...Cg4 ⁱⁱⁱ | 0.96 | 2.85 | 3.418 (1) | 119 |
| C12—H121...Cg1 ⁱ | 0.94 | 2.82 | 3.602 (1) | 141 |
| C14—H142...Cg3 ^{iv} | 0.96 | 2.72 | 3.676 (1) | 171 |
| C29—H291...Cg5 ^v | 0.96 | 2.80 | 3.682 (2) | 153 |
| C31—H311...Cg3 ^{vi} | 0.95 | 2.96 | 3.582 (1) | 125 |

Symmetry codes: (i) $x, y+1, z$; (ii) $x, y-1, z$; (iii) $x, -y+1/2, z+1/2$; (iv) $-x+2, -y+1, -z+1$; (v) $-x+1, y-1/2, -z+1/2$; (vi) $x, -y+1/2, z-1/2$.