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## Structure Reports

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## 2-(4-Fluorophenyl)-5-iodo-7-methyl-3-phenylsulfinyl-1-benzofuran

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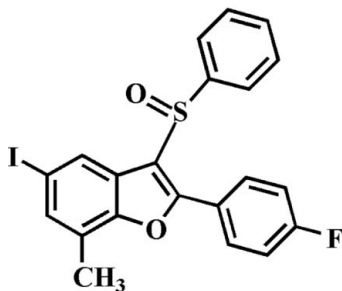
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Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.023;  $wR$  factor = 0.062; data-to-parameter ratio = 18.7.

In the title compound,  $\text{C}_{21}\text{H}_{14}\text{FIO}_2\text{S}$ , the dihedral angles between the mean plane [r.m.s. deviation = 0.007 (1) Å] of the benzofuran fragment and the pendant 4-fluorophenyl and phenyl rings are 3.66 (7) and 82.37 (6)°, respectively. In the crystal, molecules are linked by pairs of  $\text{C}-\text{H}\cdots\text{I}$  hydrogen bonds into centrosymmetric dimers, which are further packed into stacks along the  $b$  axis by  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds. In addition, the stacked molecules exhibit inversion-related  $\text{S}\cdots\text{O}$  contacts [2.9627 (14) Å] involving the sulfinyl groups.

## Related literature

For background information and the crystal structures of related compounds, see: Choi *et al.* (2009, 2012). For details of sulfinyl-sulfinyl interactions, see: Choi *et al.* (2008) and for a review of carbonyl-carbonyl interactions, see: Allen *et al.* (1998).



## Experimental

## Crystal data

 $\text{C}_{21}\text{H}_{14}\text{FIO}_2\text{S}$  $M_r = 476.28$ 

Triclinic,  $P\bar{1}$   
 $a = 8.1060$  (3) Å  
 $b = 10.7771$  (4) Å  
 $c = 11.2639$  (4) Å  
 $\alpha = 71.814$  (2)°  
 $\beta = 74.496$  (2)°  
 $\gamma = 79.038$  (2)°

$V = 894.69$  (6) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 1.93$  mm<sup>-1</sup>  
 $T = 173$  K  
 $0.36 \times 0.28 \times 0.19$  mm

## Data collection

Bruker SMART APEXII CCD diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker, 2009)  
 $T_{\min} = 0.562$ ,  $T_{\max} = 0.746$

16292 measured reflections  
 4404 independent reflections  
 4125 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.036$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.023$   
 $wR(F^2) = 0.062$   
 $S = 1.05$   
 4404 reflections

236 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.37$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.63$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}20-\text{H}20\cdots\text{O}2^{\text{i}}$	0.95	2.36	3.314 (3)	177
$\text{C}17-\text{H}17\cdots\text{I}^{\text{ii}}$	0.95	2.97	3.705 (2)	135

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $-x + 2, -y, -z + 2$ .

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012) and DIAMOND (Brandenburg, 1998); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LD2102).

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## supporting information

*Acta Cryst.* (2013). E69, o820 [doi:10.1107/S1600536813011793]

**2-(4-Fluorophenyl)-5-iodo-7-methyl-3-phenylsulfinyl-1-benzofuran**

Hong Dae Choi, Pil Ja Seo and Uk Lee

**S1. Comment**

As a part of our continuing study of 5-iodo-3-phenylsulfinyl-1-benzofuran derivatives containing phenyl (Choi *et al.*, 2009) and 4-fluorophenyl (Choi *et al.*, 2012) substituents in 2-position, we report herein the crystal structure of the title compound.

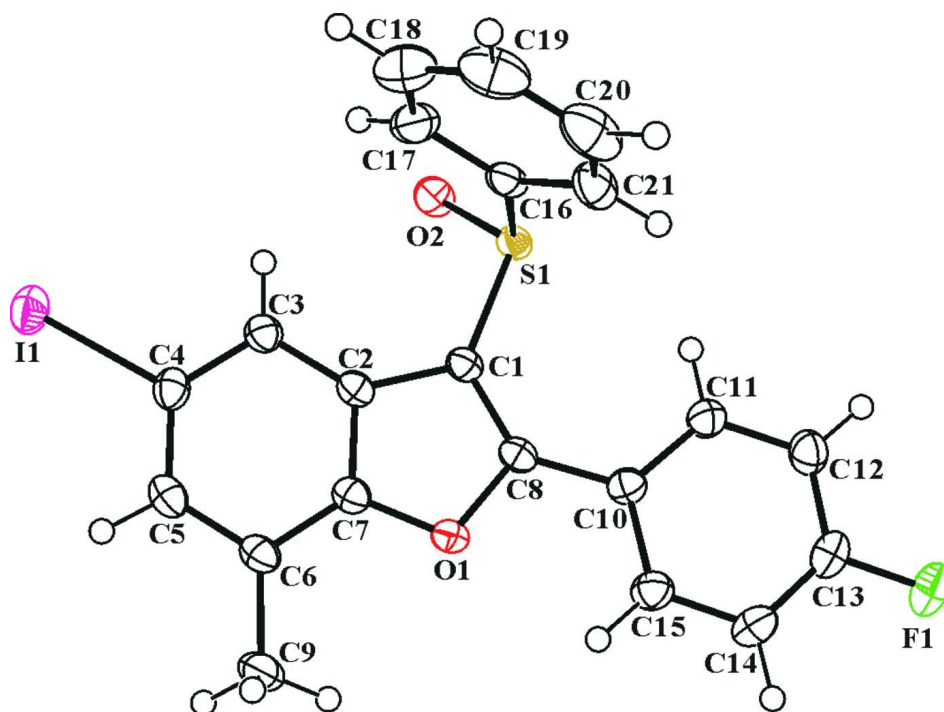
In the title molecule (Fig. 1), the benzofuran unit is essentially planar, with a mean deviation of 0.007 (1) Å from the least-squares plane defined by the nine constituent atoms. The dihedral angles between the mean plane of the benzofuran fragment and the pendant 4-fluorophenyl and phenyl rings are 3.66 (7) and 82.37 (6)°, respectively. In the crystal structure (Fig. 2), molecules are linked by pairs of C—H⋯I hydrogen bonds into centrosymmetric dimers, which are further packed into stacks along the *b* axis by C—H⋯O hydrogen bonds (Table 1). In addition, the crystal packing (Fig. 2) exhibits a sulfinyl–sulfinyl interaction (Choi *et al.*, 2008) interpreted as similar to a type-II carbonyl–carbonyl interaction (Allen *et al.*, 1998), with S1⋯O2<sup>iii</sup> and O2..S1<sup>iii</sup> distance of 2.9627 (14) Å.

**S2. Experimental**

3-Chloroperoxybenzoic acid (77%, 179 mg, 0.8 mmol) was added in small portions to a stirred solution of 2-(4-fluorophenyl)-5-iodo-7-methyl-3-phenylsulfonyl-1-benzofuran (322 mg, 0.7 mmol) in dichloromethane (30 mL) at 273 K. After being stirred at room temperature for 4h, the mixture was washed with saturated sodium bicarbonate solution and the organic layer was separated, dried over magnesium sulfate, filtered and concentrated at reduced pressure. The residue was purified by column chromatography (hexane–ethyl acetate, 2:1 v/v) to afford the title compound as a colorless solid [yield 70%, m.p. 473–474 K;  $R_f$  = 0.74 (hexane–ethyl acetate, 2:1 v/v)]. Single crystals suitable for X-ray diffraction were prepared by slow evaporation of a solution of the title compound in benzene at room temperature.

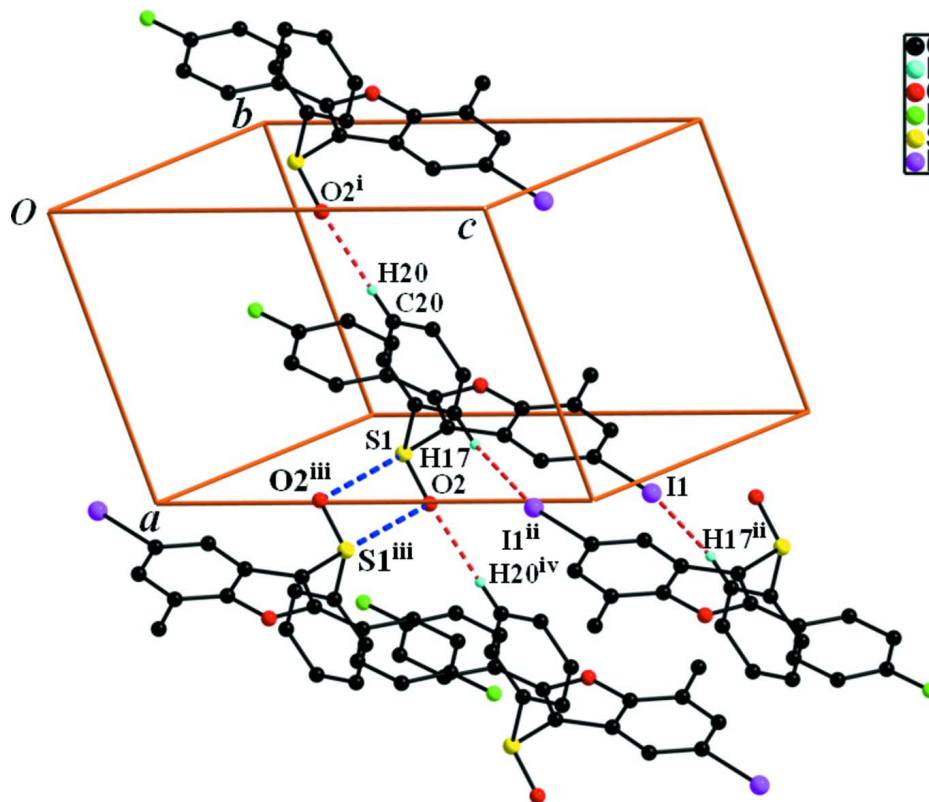
**S3. Refinement**

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.95 Å for aryl and 0.98 Å for methyl H atoms.  $U_{iso}(H) = 1.2U_{eq}(C)$  for aryl and  $1.5U_{eq}(C)$  for methyl H atoms. The positions of methyl hydrogens were optimized rotationally.



**Figure 1**

The molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are presented as small spheres of arbitrary radius.

**Figure 2**

A view of the C—H...I, C—H...O and S...O interactions (dotted lines) in the crystal structure of the title compound. H atoms non-participating in hydrogen-bonding were omitted for clarity. [Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $-x + 2, -y, -z + 2$ ; (iii)  $-x + 2, -y, -z + 1$ ; (iv)  $x + 1, y, z$ .]

### 2-(4-Fluorophenyl)-5-iodo-7-methyl-3-phenylsulfinyl-1-benzofuran

#### Crystal data

$C_{21}H_{14}FO_2S$

$M_r = 476.28$

Triclinic,  $P\bar{1}$

Hall symbol:  $-P\ 1$

$a = 8.1060$  (3) Å

$b = 10.7771$  (4) Å

$c = 11.2639$  (4) Å

$\alpha = 71.814$  (2)°

$\beta = 74.496$  (2)°

$\gamma = 79.038$  (2)°

$V = 894.69$  (6) Å<sup>3</sup>

$Z = 2$

$F(000) = 468$

$D_x = 1.768$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 6652 reflections

$\theta = 2.6$ – $28.4$ °

$\mu = 1.93$  mm<sup>-1</sup>

$T = 173$  K

Block, colourless

$0.36 \times 0.28 \times 0.19$  mm

#### Data collection

Bruker SMART APEXII CCD  
diffractometer

Radiation source: rotating anode

Graphite multilayer monochromator

Detector resolution: 10.0 pixels mm<sup>-1</sup>

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan  
(*SADABS*; Bruker, 2009)

$T_{\min} = 0.562$ ,  $T_{\max} = 0.746$

16292 measured reflections

4404 independent reflections

4125 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.036$

$\theta_{\max} = 28.3^\circ$ ,  $\theta_{\min} = 2.0^\circ$   
 $h = -10 \rightarrow 10$

$k = -14 \rightarrow 14$   
 $l = -14 \rightarrow 14$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.023$   
 $wR(F^2) = 0.062$   
 $S = 1.05$   
 4404 reflections  
 236 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: difference Fourier map  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0291P)^2 + 0.4238P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.002$   
 $\Delta\rho_{\max} = 0.37 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.63 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional R-factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
I1	1.038283 (17)	0.207761 (13)	1.022854 (12)	0.03450 (6)
S1	0.88164 (6)	0.15272 (4)	0.52014 (4)	0.02267 (9)
F1	0.52793 (19)	0.62193 (14)	0.03604 (12)	0.0432 (3)
O1	0.76006 (17)	0.52135 (12)	0.54415 (12)	0.0240 (3)
O2	1.03200 (17)	0.08292 (13)	0.57680 (14)	0.0297 (3)
C1	0.8377 (2)	0.30667 (17)	0.55547 (17)	0.0219 (3)
C2	0.8726 (2)	0.32977 (17)	0.66597 (17)	0.0226 (3)
C3	0.9380 (2)	0.25313 (18)	0.77187 (18)	0.0256 (4)
H3	0.9754	0.1619	0.7836	0.031*
C4	0.9455 (2)	0.31737 (19)	0.85932 (18)	0.0263 (4)
C5	0.8945 (3)	0.45161 (19)	0.84379 (18)	0.0274 (4)
H5	0.9038	0.4903	0.9065	0.033*
C6	0.8304 (2)	0.52953 (18)	0.73849 (18)	0.0256 (4)
C7	0.8216 (2)	0.46343 (17)	0.65298 (17)	0.0232 (3)
C8	0.7708 (2)	0.42461 (17)	0.48521 (17)	0.0224 (3)
C9	0.7725 (3)	0.67397 (19)	0.7182 (2)	0.0336 (4)
H9A	0.7783	0.7163	0.6267	0.050*
H9B	0.8479	0.7132	0.7480	0.050*
H9C	0.6536	0.6868	0.7665	0.050*
C10	0.7072 (2)	0.47139 (17)	0.36758 (17)	0.0234 (3)
C11	0.7151 (3)	0.39187 (19)	0.28875 (19)	0.0285 (4)
H11	0.7626	0.3024	0.3120	0.034*
C12	0.6542 (3)	0.4420 (2)	0.17688 (19)	0.0313 (4)

H12	0.6595	0.3880	0.1233	0.038*
C13	0.5866 (3)	0.5714 (2)	0.14591 (18)	0.0305 (4)
C14	0.5757 (3)	0.6538 (2)	0.22019 (19)	0.0315 (4)
H14	0.5279	0.7431	0.1958	0.038*
C15	0.6363 (3)	0.60277 (19)	0.33112 (18)	0.0287 (4)
H15	0.6298	0.6580	0.3839	0.034*
C16	0.6935 (2)	0.08351 (17)	0.62724 (19)	0.0253 (4)
C17	0.7012 (3)	0.0107 (2)	0.7511 (2)	0.0351 (4)
H17	0.8053	-0.0028	0.7794	0.042*
C18	0.5561 (3)	-0.0422 (2)	0.8336 (3)	0.0457 (6)
H18	0.5598	-0.0921	0.9191	0.055*
C19	0.4053 (3)	-0.0223 (2)	0.7911 (3)	0.0484 (6)
H19	0.3053	-0.0579	0.8481	0.058*
C20	0.3990 (3)	0.0487 (3)	0.6670 (3)	0.0465 (6)
H20	0.2952	0.0611	0.6385	0.056*
C21	0.5442 (3)	0.1020 (2)	0.5834 (2)	0.0343 (4)
H21	0.5411	0.1505	0.4975	0.041*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
I1	0.03624 (9)	0.04058 (9)	0.02868 (8)	-0.00585 (6)	-0.01368 (6)	-0.00611 (6)
S1	0.0228 (2)	0.01980 (19)	0.0283 (2)	0.00242 (16)	-0.00949 (17)	-0.01045 (16)
F1	0.0501 (8)	0.0477 (7)	0.0309 (6)	-0.0013 (6)	-0.0206 (6)	-0.0025 (5)
O1	0.0290 (7)	0.0186 (6)	0.0250 (6)	-0.0013 (5)	-0.0063 (5)	-0.0078 (5)
O2	0.0234 (6)	0.0282 (7)	0.0422 (8)	0.0057 (5)	-0.0148 (6)	-0.0153 (6)
C1	0.0216 (8)	0.0206 (8)	0.0248 (8)	-0.0013 (6)	-0.0050 (7)	-0.0091 (6)
C2	0.0213 (8)	0.0227 (8)	0.0259 (8)	-0.0038 (6)	-0.0048 (7)	-0.0092 (7)
C3	0.0254 (9)	0.0249 (8)	0.0288 (9)	-0.0039 (7)	-0.0085 (7)	-0.0077 (7)
C4	0.0248 (9)	0.0311 (9)	0.0241 (9)	-0.0067 (7)	-0.0074 (7)	-0.0056 (7)
C5	0.0277 (9)	0.0314 (9)	0.0269 (9)	-0.0091 (8)	-0.0030 (7)	-0.0126 (7)
C6	0.0275 (9)	0.0243 (8)	0.0264 (9)	-0.0079 (7)	-0.0016 (7)	-0.0098 (7)
C7	0.0235 (8)	0.0226 (8)	0.0237 (8)	-0.0042 (7)	-0.0031 (7)	-0.0076 (7)
C8	0.0218 (8)	0.0217 (8)	0.0246 (8)	-0.0024 (6)	-0.0025 (7)	-0.0099 (7)
C9	0.0434 (12)	0.0249 (9)	0.0352 (10)	-0.0051 (8)	-0.0064 (9)	-0.0135 (8)
C10	0.0221 (8)	0.0227 (8)	0.0237 (8)	-0.0007 (7)	-0.0031 (7)	-0.0067 (7)
C11	0.0322 (10)	0.0255 (9)	0.0281 (9)	0.0015 (7)	-0.0095 (8)	-0.0083 (7)
C12	0.0342 (10)	0.0354 (10)	0.0260 (9)	-0.0029 (8)	-0.0076 (8)	-0.0110 (8)
C13	0.0279 (9)	0.0365 (10)	0.0236 (9)	-0.0027 (8)	-0.0078 (8)	-0.0022 (8)
C14	0.0312 (10)	0.0271 (9)	0.0303 (10)	0.0030 (8)	-0.0073 (8)	-0.0031 (8)
C15	0.0312 (10)	0.0255 (9)	0.0273 (9)	0.0001 (7)	-0.0050 (8)	-0.0076 (7)
C16	0.0244 (9)	0.0195 (8)	0.0358 (10)	0.0003 (7)	-0.0100 (8)	-0.0121 (7)
C17	0.0366 (11)	0.0256 (9)	0.0420 (12)	-0.0042 (8)	-0.0141 (9)	-0.0031 (8)
C18	0.0474 (14)	0.0339 (11)	0.0486 (14)	-0.0100 (10)	-0.0034 (11)	-0.0043 (10)
C19	0.0365 (12)	0.0406 (13)	0.0661 (17)	-0.0133 (10)	0.0050 (12)	-0.0207 (12)
C20	0.0274 (11)	0.0477 (13)	0.0755 (18)	-0.0037 (9)	-0.0134 (11)	-0.0312 (13)
C21	0.0285 (10)	0.0350 (10)	0.0471 (12)	0.0010 (8)	-0.0162 (9)	-0.0186 (9)

*Geometric parameters (Å, °)*

I1—C4	2.0974 (19)	C9—H9C	0.9800
S1—O2 <sup>i</sup>	2.9627 (14)	C10—C11	1.397 (3)
S1—O2	1.4920 (14)	C10—C15	1.399 (3)
S1—C1	1.7748 (17)	C11—C12	1.389 (3)
S1—C16	1.796 (2)	C11—H11	0.9500
F1—C13	1.360 (2)	C12—C13	1.369 (3)
O1—C7	1.368 (2)	C12—H12	0.9500
O1—C8	1.379 (2)	C13—C14	1.375 (3)
C1—C8	1.369 (2)	C14—C15	1.379 (3)
C1—C2	1.449 (2)	C14—H14	0.9500
C2—C7	1.393 (2)	C15—H15	0.9500
C2—C3	1.394 (3)	C16—C17	1.382 (3)
C3—C4	1.388 (3)	C16—C21	1.383 (3)
C3—H3	0.9500	C17—C18	1.383 (3)
C4—C5	1.396 (3)	C17—H17	0.9500
C5—C6	1.387 (3)	C18—C19	1.385 (4)
C5—H5	0.9500	C18—H18	0.9500
C6—C7	1.387 (2)	C19—C20	1.378 (4)
C6—C9	1.501 (3)	C19—H19	0.9500
C8—C10	1.458 (3)	C20—C21	1.390 (3)
C9—H9A	0.9800	C20—H20	0.9500
C9—H9B	0.9800	C21—H21	0.9500
O2—S1—C1	105.04 (8)	C11—C10—C15	118.34 (17)
O2—S1—C16	106.95 (9)	C11—C10—C8	123.24 (16)
C1—S1—C16	97.02 (8)	C15—C10—C8	118.41 (16)
O2—S1—O2 <sup>i</sup>	78.36 (6)	C12—C11—C10	120.83 (18)
C1—S1—O2 <sup>i</sup>	172.01 (7)	C12—C11—H11	119.6
C16—S1—O2 <sup>i</sup>	88.76 (6)	C10—C11—H11	119.6
C7—O1—C8	107.21 (13)	C13—C12—C11	118.25 (19)
C8—C1—C2	107.21 (15)	C13—C12—H12	120.9
C8—C1—S1	127.99 (14)	C11—C12—H12	120.9
C2—C1—S1	124.78 (13)	F1—C13—C12	118.80 (19)
C7—C2—C3	119.35 (17)	F1—C13—C14	117.99 (18)
C7—C2—C1	104.84 (16)	C12—C13—C14	123.21 (19)
C3—C2—C1	135.81 (17)	C13—C14—C15	117.99 (18)
C4—C3—C2	116.34 (17)	C13—C14—H14	121.0
C4—C3—H3	121.8	C15—C14—H14	121.0
C2—C3—H3	121.8	C14—C15—C10	121.38 (18)
C3—C4—C5	123.24 (18)	C14—C15—H15	119.3
C3—C4—I1	118.53 (14)	C10—C15—H15	119.3
C5—C4—I1	118.23 (14)	C17—C16—C21	121.16 (19)
C6—C5—C4	121.18 (17)	C17—C16—S1	119.54 (15)
C6—C5—H5	119.4	C21—C16—S1	119.29 (16)
C4—C5—H5	119.4	C16—C17—C18	119.5 (2)
C5—C6—C7	114.83 (17)	C16—C17—H17	120.3

C5—C6—C9	123.14 (17)	C18—C17—H17	120.3
C7—C6—C9	122.02 (18)	C17—C18—C19	119.7 (2)
O1—C7—C6	124.22 (16)	C17—C18—H18	120.1
O1—C7—C2	110.72 (15)	C19—C18—H18	120.1
C6—C7—C2	125.06 (18)	C20—C19—C18	120.6 (2)
C1—C8—O1	110.02 (16)	C20—C19—H19	119.7
C1—C8—C10	136.31 (16)	C18—C19—H19	119.7
O1—C8—C10	113.67 (15)	C19—C20—C21	120.1 (2)
C6—C9—H9A	109.5	C19—C20—H20	120.0
C6—C9—H9B	109.5	C21—C20—H20	120.0
H9A—C9—H9B	109.5	C16—C21—C20	119.0 (2)
C6—C9—H9C	109.5	C16—C21—H21	120.5
H9A—C9—H9C	109.5	C20—C21—H21	120.5
H9B—C9—H9C	109.5		
O2—S1—C1—C8	149.12 (17)	S1—C1—C8—C10	2.6 (3)
C16—S1—C1—C8	-101.17 (18)	C7—O1—C8—C1	0.17 (19)
O2—S1—C1—C2	-28.97 (18)	C7—O1—C8—C10	179.24 (15)
C16—S1—C1—C2	80.74 (17)	C1—C8—C10—C11	-5.0 (3)
C8—C1—C2—C7	0.3 (2)	O1—C8—C10—C11	176.23 (17)
S1—C1—C2—C7	178.71 (13)	C1—C8—C10—C15	175.6 (2)
C8—C1—C2—C3	179.5 (2)	O1—C8—C10—C15	-3.1 (2)
S1—C1—C2—C3	-2.1 (3)	C15—C10—C11—C12	0.1 (3)
C7—C2—C3—C4	0.8 (3)	C8—C10—C11—C12	-179.26 (18)
C1—C2—C3—C4	-178.3 (2)	C10—C11—C12—C13	0.1 (3)
C2—C3—C4—C5	-1.2 (3)	C11—C12—C13—F1	179.46 (18)
C2—C3—C4—I1	178.91 (13)	C11—C12—C13—C14	-0.1 (3)
C3—C4—C5—C6	0.7 (3)	F1—C13—C14—C15	-179.55 (18)
I1—C4—C5—C6	-179.40 (14)	C12—C13—C14—C15	0.0 (3)
C4—C5—C6—C7	0.2 (3)	C13—C14—C15—C10	0.1 (3)
C4—C5—C6—C9	179.44 (18)	C11—C10—C15—C14	-0.2 (3)
C8—O1—C7—C6	-179.63 (17)	C8—C10—C15—C14	179.19 (18)
C8—O1—C7—C2	0.02 (19)	O2—S1—C16—C17	16.79 (18)
C5—C6—C7—O1	178.99 (16)	C1—S1—C16—C17	-91.31 (16)
C9—C6—C7—O1	-0.2 (3)	O2—S1—C16—C21	-162.03 (15)
C5—C6—C7—C2	-0.6 (3)	C1—S1—C16—C21	89.87 (16)
C9—C6—C7—C2	-179.84 (18)	C21—C16—C17—C18	-1.4 (3)
C3—C2—C7—O1	-179.54 (16)	S1—C16—C17—C18	179.85 (17)
C1—C2—C7—O1	-0.2 (2)	C16—C17—C18—C19	0.3 (3)
C3—C2—C7—C6	0.1 (3)	C17—C18—C19—C20	0.7 (4)
C1—C2—C7—C6	179.46 (17)	C18—C19—C20—C21	-0.6 (4)
C2—C1—C8—O1	-0.3 (2)	C17—C16—C21—C20	1.5 (3)
S1—C1—C8—O1	-178.64 (13)	S1—C16—C21—C20	-179.72 (16)
C2—C1—C8—C10	-179.0 (2)	C19—C20—C21—C16	-0.5 (3)

Symmetry code: (i)  $-x+2, -y, -z+1$ .



*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
C20—H20 $\cdots$ O2 <sup>ii</sup>	0.95	2.36	3.314 (3)	177
C17—H17 $\cdots$ I1 <sup>iii</sup>	0.95	2.97	3.705 (2)	135

Symmetry codes: (ii)  $x-1, y, z$ ; (iii)  $-x+2, -y, -z+2$ .