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1-(2,6-Diisopropylphenyl)-1*H*-benzimidazole

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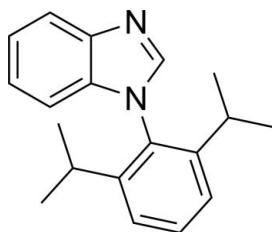
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.035; wR factor = 0.088; data-to-parameter ratio = 18.3.

In the title compound, $\text{C}_{19}\text{H}_{22}\text{N}_2$, both the benzimidazole unit and the 2,6-diisopropylphenyl group are essentially planar [maximum deviations from the least-squares planes of 0.005 (1) and 0.009 (1) Å, respectively]. The dihedral angle between the two planes is 79.6 (7)°. In the crystal, molecules are linked into chains along the a -axis direction by weak $\text{C}-\text{H}\cdots\text{N}$ interactions. The crystal structure also features $\text{C}-\text{H}\cdots\pi$ interactions, which link the chains into a three-dimensional network.

Related literature

For the properties of related compounds, see: Shi *et al.* (2013); Cross *et al.* (1995); Akpinar *et al.* (2010); Wang *et al.* (2007); Mason *et al.* (1999). For bond lengths and angles in related structures, see: Jayamoorthy *et al.* (2013); Fathima *et al.* (2013); Geiger & Nellist (2013).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{22}\text{N}_2$
 $M_r = 278.39$
Orthorhombic, $P2_12_12_1$
 $a = 6.6471$ (8) Å

$b = 14.1216$ (18) Å
 $c = 17.285$ (2) Å
 $V = 1622.5$ (4) Å³
 $Z = 4$

Mo $K\alpha$ radiation
 $\mu = 0.07$ mm⁻¹

$T = 173$ K
 $0.44 \times 0.42 \times 0.28$ mm

Data collection

Bruker APEXII CCD area-detector diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2008)
 $T_{\min} = 0.971$, $T_{\max} = 0.982$

9716 measured reflections
3551 independent reflections
3133 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.088$
 $S = 1.08$
3551 reflections

194 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.16$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.16$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 and Cg2 are the centroids of the $\text{C2}-\text{C7}$ and $\text{C8}/\text{C9}/\text{C13}-\text{C16}$ rings, respectively.

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--|--------------|--------------------|-------------|----------------------|
| $\text{C6}-\text{H6}\cdots\text{N1}^i$ | 0.95 | 2.47 | 3.4040 (18) | 168 |
| $\text{C14}-\text{H14}\cdots\text{Cg3}^{ii}$ | 0.95 | 2.68 | 3.5908 (16) | 150 |
| $\text{C18}-\text{H18B}\cdots\text{Cg2}^{iii}$ | 0.98 | 2.79 | 3.5314 (17) | 125 |

Symmetry codes: (i) $x - 1, y, z$; (ii) $x - \frac{1}{2}, -y + \frac{3}{2}, -z + 1$; (iii) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BG2510).

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supporting information

Acta Cryst. (2013). E69, o1330 [doi:10.1107/S1600536813020473]

1-(2,6-Diisopropylphenyl)-1*H*-benzimidazole

Sha Lai, Yong Chen, Yan Li, Yonggang Shen and Hongwei Wu

S1. Comment

Benzimidazole is one of the most important organic intermediates in drug design, in light of the affinity it displays towards some enzymes and protein receptors (Mason *et al.*, 1999). Our interest is focused on the design and synthesis of benzimidazole derivatives with various ancillary ligands, and their application in antioxidant activities. Herein, we report the synthesis and structure of the title compound (I). Its molecular structure is shown in Fig.1. Bond lengths and angles of the benzimidazole group are in good agreement with those observed in other benzimidazole derivatives (Jayamoorthy *et al.*, 2013; Fathima *et al.*, 2013; Geiger *et al.*, 2013). Both the benzimidazole unit and the 2, 6-diisopropylphenyl groups are essentially planar (max. deviations from the L.S. plane: 0.005 (1) and 0.009 (1) Å, for atoms C7 and C9, respectively). The dihedral angle between both planes is 100.4° (7). In the crystal structure, the molecules are linked into chains along the *a* axis by intermolecular C—H···N hydrogen bonds (Table 1). The structure is further stabilized by weak intermolecular C—H···Cg interactions linking chains into a 3D network (Table 1 and Fig 2).

S2. Experimental

N-(2-bromophenyl)-*N'*-(2, 6-diisopropylphenyl)-Methanimidamide, (1.65 g, 4.58 mmol, 1 eq.) was dissolved in 18.5 ml DMSO. CuI (174 mg, 0.92 mmol, 20 mol%) and DBU (1.37 ml, 1.39 g, 9.16 mmol, 2 eq.) were added and the reaction was stirred for 3 h at 110 °C. H₂O(160 ml) and acetoacetate (160 ml) were added and the layers were separated. The aqueous layer was extracted with acetoacetate (2 times 20 ml), the combined organic layers were dried over Na₂SO₄. The crude mixture was purified by column chromatography to afford the benzimidazole (1.22 g, 96%). Single crystals were grown in ethanol as a solvent at room temperature.

S3. Refinement

C-bound H-atoms were geometrically positioned (C—H 0.93 or 0.98 Å for aromatic or methyl C atoms respectively) and refined using a riding model, with $U_{\text{iso}} = 1.2/1.5U_{\text{eq}}$ (C), respectively.

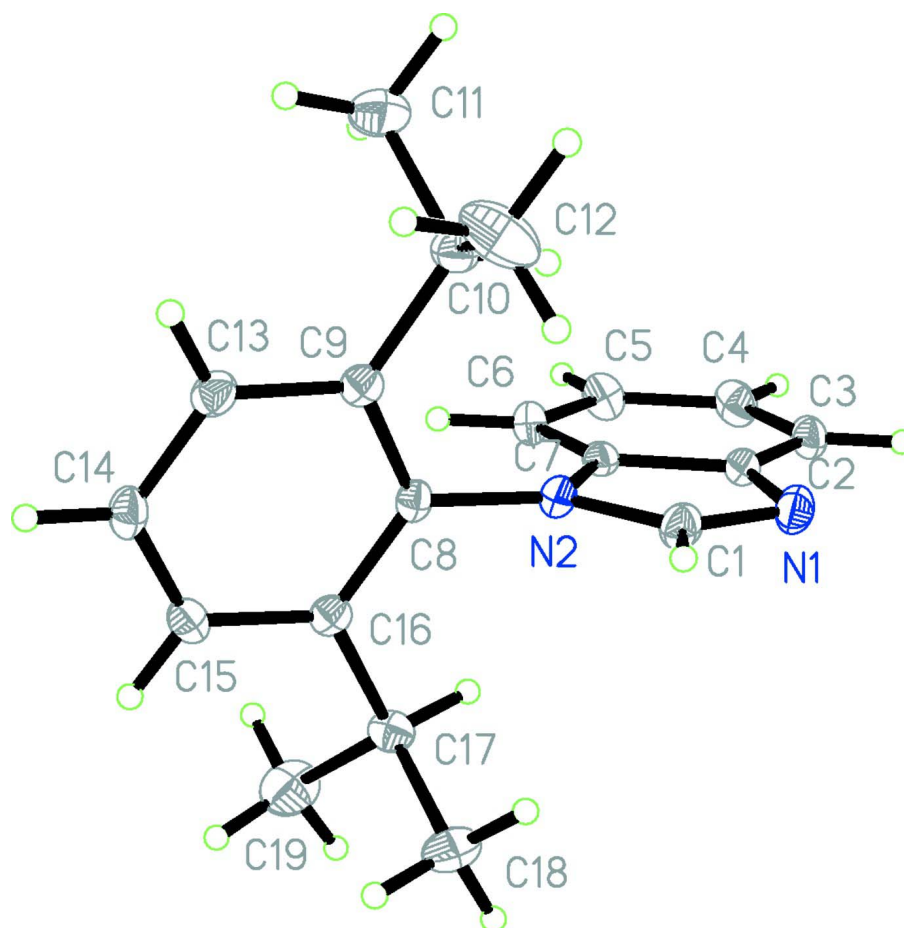
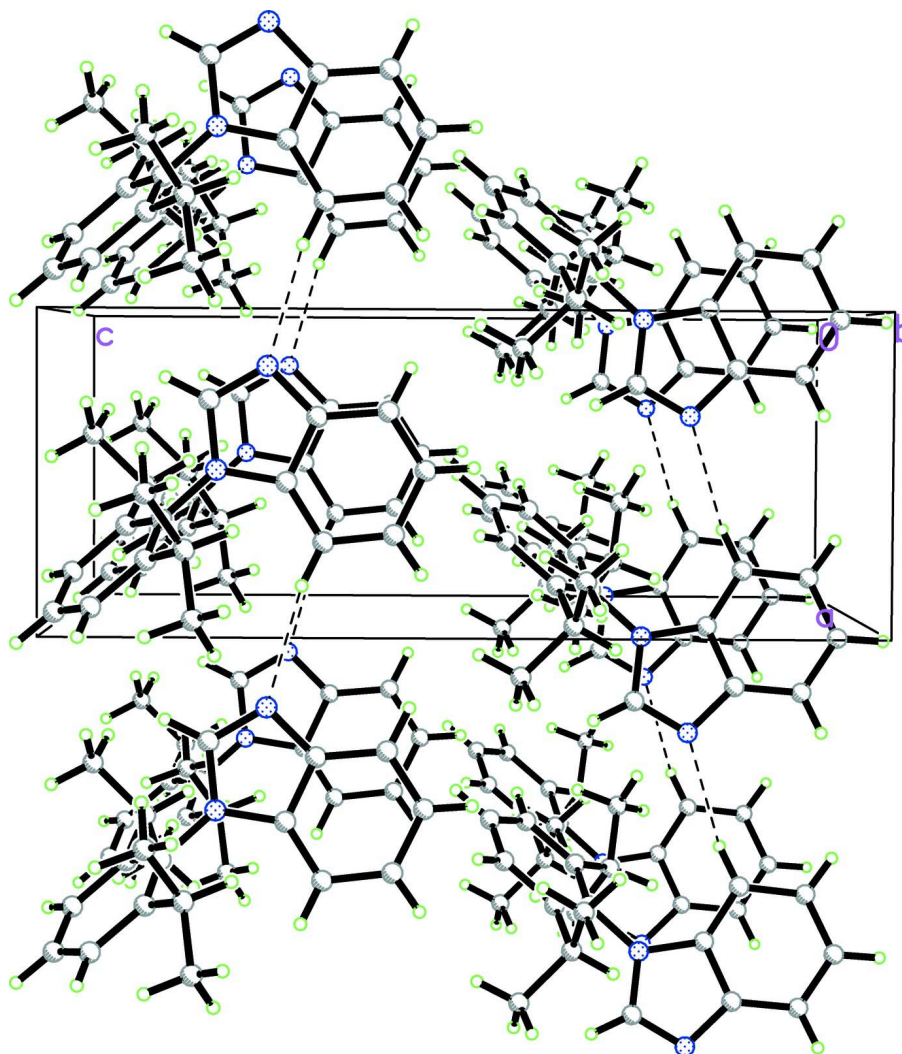


Figure 1

Molecular structure of the title compound. Displacement ellipsoids are shown at the 40% probability level. H atoms are presented as small spheres of arbitrary radius.

**Figure 2**

Packing diagram of (I). Dashed lines indicate intermolecular hydrogen bonding interactions.

1-(2,6-Diisopropylphenyl)-1*H*-benzimidazole

Crystal data

$C_{19}H_{22}N_2$

$M_r = 278.39$

Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 6.6471$ (8) Å

$b = 14.1216$ (18) Å

$c = 17.285$ (2) Å

$V = 1622.5$ (4) Å³

$Z = 4$

$F(000) = 600$

$D_x = 1.140$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 5376 reflections

$\theta = 2.4$ – 27.1°

$\mu = 0.07$ mm⁻¹

$T = 173$ K

Block, colorless

$0.44 \times 0.42 \times 0.28$ mm

Data collection

Bruker APEXII CCD area-detector
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 0 pixels mm⁻¹
phi and ω scans
Absorption correction: multi-scan
(SADABS; Bruker, 2008)
 $T_{\min} = 0.971$, $T_{\max} = 0.982$

9716 measured reflections
3551 independent reflections
3133 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$
 $\theta_{\max} = 27.1^\circ$, $\theta_{\min} = 1.9^\circ$
 $h = -8 \rightarrow 8$
 $k = -18 \rightarrow 16$
 $l = -18 \rightarrow 22$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.088$
 $S = 1.08$
3551 reflections
194 parameters
0 restraints
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0411P)^2 + 0.2264P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.16 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.16 \text{ e } \text{Å}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|--------------|--------------|----------------------------------|
| C1 | 1.2274 (2) | 0.79599 (10) | 0.29080 (8) | 0.0303 (3) |
| H1 | 1.2920 | 0.7885 | 0.3394 | 0.036* |
| N1 | 1.32623 (17) | 0.80167 (9) | 0.22581 (7) | 0.0335 (3) |
| C2 | 1.17791 (19) | 0.81252 (9) | 0.17010 (8) | 0.0262 (3) |
| N2 | 1.02227 (16) | 0.80177 (8) | 0.28307 (6) | 0.0243 (2) |
| C3 | 1.1940 (2) | 0.82181 (10) | 0.08970 (8) | 0.0327 (3) |
| H3 | 1.3217 | 0.8218 | 0.0650 | 0.039* |
| C4 | 1.0203 (2) | 0.83093 (11) | 0.04764 (8) | 0.0366 (3) |
| H4 | 1.0285 | 0.8375 | -0.0070 | 0.044* |
| C5 | 0.8308 (2) | 0.83080 (11) | 0.08327 (8) | 0.0347 (3) |
| H5 | 0.7138 | 0.8373 | 0.0522 | 0.042* |
| C6 | 0.8096 (2) | 0.82147 (10) | 0.16281 (8) | 0.0284 (3) |
| H6 | 0.6814 | 0.8211 | 0.1872 | 0.034* |
| C7 | 0.9866 (2) | 0.81270 (8) | 0.20455 (7) | 0.0234 (3) |
| C8 | 0.87636 (19) | 0.80084 (9) | 0.34461 (7) | 0.0227 (3) |
| C9 | 0.8510 (2) | 0.88320 (9) | 0.38878 (7) | 0.0254 (3) |
| C10 | 0.9634 (2) | 0.97431 (10) | 0.37135 (8) | 0.0324 (3) |
| H10 | 1.0380 | 0.9652 | 0.3217 | 0.039* |
| C11 | 0.8185 (3) | 1.05673 (13) | 0.36039 (13) | 0.0604 (6) |

| | | | | |
|------|--------------|--------------|--------------|------------|
| H11A | 0.7243 | 1.0419 | 0.3185 | 0.091* |
| H11B | 0.8946 | 1.1140 | 0.3472 | 0.091* |
| H11C | 0.7436 | 1.0674 | 0.4084 | 0.091* |
| C12 | 1.1172 (4) | 0.99708 (15) | 0.43400 (12) | 0.0645 (6) |
| H12A | 1.0480 | 1.0063 | 0.4835 | 0.097* |
| H12B | 1.1898 | 1.0550 | 0.4202 | 0.097* |
| H12C | 1.2127 | 0.9445 | 0.4386 | 0.097* |
| C13 | 0.7136 (2) | 0.87915 (10) | 0.44978 (8) | 0.0303 (3) |
| H13 | 0.6941 | 0.9334 | 0.4815 | 0.036* |
| C14 | 0.6056 (2) | 0.79775 (11) | 0.46485 (8) | 0.0315 (3) |
| H14 | 0.5136 | 0.7963 | 0.5069 | 0.038* |
| C15 | 0.6306 (2) | 0.71811 (10) | 0.41892 (8) | 0.0292 (3) |
| H15 | 0.5532 | 0.6630 | 0.4292 | 0.035* |
| C16 | 0.76747 (19) | 0.71788 (10) | 0.35795 (7) | 0.0249 (3) |
| C17 | 0.8059 (2) | 0.62843 (10) | 0.31125 (8) | 0.0303 (3) |
| H17 | 0.8567 | 0.6479 | 0.2592 | 0.036* |
| C18 | 0.9704 (3) | 0.56985 (11) | 0.35044 (10) | 0.0421 (4) |
| H18A | 1.0924 | 0.6083 | 0.3557 | 0.063* |
| H18B | 0.9999 | 0.5138 | 0.3189 | 0.063* |
| H18C | 0.9244 | 0.5499 | 0.4018 | 0.063* |
| C19 | 0.6166 (3) | 0.56884 (13) | 0.29914 (12) | 0.0534 (5) |
| H19A | 0.5719 | 0.5432 | 0.3489 | 0.080* |
| H19B | 0.6465 | 0.5166 | 0.2636 | 0.080* |
| H19C | 0.5101 | 0.6085 | 0.2771 | 0.080* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|-------------|
| C1 | 0.0239 (6) | 0.0333 (8) | 0.0338 (7) | 0.0025 (6) | -0.0026 (6) | 0.0005 (6) |
| N1 | 0.0241 (6) | 0.0388 (7) | 0.0375 (7) | 0.0032 (5) | 0.0041 (5) | 0.0009 (5) |
| C2 | 0.0238 (6) | 0.0226 (6) | 0.0324 (7) | -0.0001 (5) | 0.0043 (5) | -0.0012 (5) |
| N2 | 0.0211 (5) | 0.0284 (6) | 0.0234 (5) | 0.0027 (5) | 0.0009 (4) | -0.0005 (4) |
| C3 | 0.0331 (7) | 0.0323 (8) | 0.0325 (7) | -0.0025 (6) | 0.0124 (6) | -0.0023 (6) |
| C4 | 0.0451 (9) | 0.0407 (8) | 0.0239 (7) | -0.0053 (7) | 0.0070 (7) | -0.0005 (6) |
| C5 | 0.0328 (8) | 0.0425 (8) | 0.0286 (7) | -0.0065 (7) | -0.0041 (6) | 0.0019 (6) |
| C6 | 0.0232 (6) | 0.0348 (8) | 0.0271 (7) | -0.0021 (6) | 0.0015 (5) | -0.0005 (6) |
| C7 | 0.0265 (6) | 0.0193 (6) | 0.0244 (6) | -0.0011 (5) | 0.0041 (5) | -0.0013 (5) |
| C8 | 0.0197 (6) | 0.0294 (7) | 0.0189 (6) | 0.0034 (5) | -0.0001 (5) | 0.0015 (5) |
| C9 | 0.0260 (6) | 0.0271 (7) | 0.0232 (6) | 0.0022 (5) | -0.0026 (5) | 0.0014 (5) |
| C10 | 0.0404 (8) | 0.0262 (7) | 0.0305 (7) | -0.0017 (6) | 0.0044 (6) | -0.0009 (5) |
| C11 | 0.0707 (13) | 0.0352 (9) | 0.0754 (13) | 0.0121 (9) | 0.0173 (11) | 0.0173 (9) |
| C12 | 0.0756 (14) | 0.0544 (12) | 0.0634 (13) | -0.0329 (11) | -0.0177 (11) | 0.0042 (9) |
| C13 | 0.0354 (7) | 0.0303 (7) | 0.0254 (7) | 0.0052 (6) | 0.0029 (6) | -0.0026 (6) |
| C14 | 0.0283 (7) | 0.0431 (8) | 0.0232 (7) | 0.0030 (6) | 0.0054 (5) | 0.0025 (6) |
| C15 | 0.0271 (7) | 0.0322 (7) | 0.0283 (7) | -0.0034 (6) | -0.0003 (6) | 0.0056 (6) |
| C16 | 0.0249 (6) | 0.0274 (7) | 0.0226 (6) | 0.0028 (5) | -0.0039 (5) | 0.0014 (5) |
| C17 | 0.0374 (8) | 0.0274 (7) | 0.0262 (7) | -0.0015 (6) | -0.0019 (6) | -0.0018 (5) |
| C18 | 0.0470 (10) | 0.0309 (8) | 0.0485 (9) | 0.0066 (7) | -0.0081 (8) | -0.0062 (7) |

| | | | | | | |
|-----|-------------|-------------|-------------|-------------|--------------|-------------|
| C19 | 0.0466 (10) | 0.0429 (10) | 0.0706 (13) | -0.0067 (8) | -0.0106 (10) | -0.0192 (9) |
|-----|-------------|-------------|-------------|-------------|--------------|-------------|

Geometric parameters (Å, °)

| | | | |
|----------|-------------|---------------|-------------|
| C1—N1 | 1.3038 (18) | C11—H11A | 0.9800 |
| C1—N2 | 1.3725 (16) | C11—H11B | 0.9800 |
| C1—H1 | 0.9500 | C11—H11C | 0.9800 |
| N1—C2 | 1.3868 (18) | C12—H12A | 0.9800 |
| C2—C3 | 1.400 (2) | C12—H12B | 0.9800 |
| C2—C7 | 1.4044 (18) | C12—H12C | 0.9800 |
| N2—C7 | 1.3865 (16) | C13—C14 | 1.380 (2) |
| N2—C8 | 1.4396 (15) | C13—H13 | 0.9500 |
| C3—C4 | 1.370 (2) | C14—C15 | 1.387 (2) |
| C3—H3 | 0.9500 | C14—H14 | 0.9500 |
| C4—C5 | 1.402 (2) | C15—C16 | 1.3924 (19) |
| C4—H4 | 0.9500 | C15—H15 | 0.9500 |
| C5—C6 | 1.388 (2) | C16—C17 | 1.5207 (19) |
| C5—H5 | 0.9500 | C17—C19 | 1.528 (2) |
| C6—C7 | 1.3857 (18) | C17—C18 | 1.529 (2) |
| C6—H6 | 0.9500 | C17—H17 | 1.0000 |
| C8—C16 | 1.3963 (19) | C18—H18A | 0.9800 |
| C8—C9 | 1.4013 (18) | C18—H18B | 0.9800 |
| C9—C13 | 1.3961 (19) | C18—H18C | 0.9800 |
| C9—C10 | 1.5181 (19) | C19—H19A | 0.9800 |
| C10—C11 | 1.522 (2) | C19—H19B | 0.9800 |
| C10—C12 | 1.524 (2) | C19—H19C | 0.9800 |
| C10—H10 | 1.0000 | | |
| | | | |
| N1—C1—N2 | 114.39 (12) | C10—C11—H11C | 109.5 |
| N1—C1—H1 | 122.8 | H11A—C11—H11C | 109.5 |
| N2—C1—H1 | 122.8 | H11B—C11—H11C | 109.5 |
| C1—N1—C2 | 104.30 (11) | C10—C12—H12A | 109.5 |
| N1—C2—C3 | 130.20 (12) | C10—C12—H12B | 109.5 |
| N1—C2—C7 | 110.46 (11) | H12A—C12—H12B | 109.5 |
| C3—C2—C7 | 119.33 (12) | C10—C12—H12C | 109.5 |
| C1—N2—C7 | 105.78 (11) | H12A—C12—H12C | 109.5 |
| C1—N2—C8 | 126.64 (11) | H12B—C12—H12C | 109.5 |
| C7—N2—C8 | 127.52 (11) | C14—C13—C9 | 121.13 (13) |
| C4—C3—C2 | 118.11 (13) | C14—C13—H13 | 119.4 |
| C4—C3—H3 | 120.9 | C9—C13—H13 | 119.4 |
| C2—C3—H3 | 120.9 | C13—C14—C15 | 120.34 (12) |
| C3—C4—C5 | 121.57 (13) | C13—C14—H14 | 119.8 |
| C3—C4—H4 | 119.2 | C15—C14—H14 | 119.8 |
| C5—C4—H4 | 119.2 | C14—C15—C16 | 120.90 (13) |
| C6—C5—C4 | 121.77 (14) | C14—C15—H15 | 119.5 |
| C6—C5—H5 | 119.1 | C16—C15—H15 | 119.5 |
| C4—C5—H5 | 119.1 | C15—C16—C8 | 117.50 (12) |
| C7—C6—C5 | 115.96 (12) | C15—C16—C17 | 120.90 (12) |

| | | | |
|---------------|--------------|-----------------|--------------|
| C7—C6—H6 | 122.0 | C8—C16—C17 | 121.48 (11) |
| C5—C6—H6 | 122.0 | C16—C17—C19 | 113.06 (12) |
| C6—C7—N2 | 131.68 (12) | C16—C17—C18 | 109.53 (11) |
| C6—C7—C2 | 123.26 (11) | C19—C17—C18 | 110.57 (13) |
| N2—C7—C2 | 105.06 (11) | C16—C17—H17 | 107.8 |
| C16—C8—C9 | 122.96 (11) | C19—C17—H17 | 107.8 |
| C16—C8—N2 | 118.61 (11) | C18—C17—H17 | 107.8 |
| C9—C8—N2 | 118.43 (11) | C17—C18—H18A | 109.5 |
| C13—C9—C8 | 117.14 (12) | C17—C18—H18B | 109.5 |
| C13—C9—C10 | 120.43 (12) | H18A—C18—H18B | 109.5 |
| C8—C9—C10 | 122.42 (12) | C17—C18—H18C | 109.5 |
| C9—C10—C11 | 111.19 (13) | H18A—C18—H18C | 109.5 |
| C9—C10—C12 | 111.60 (13) | H18B—C18—H18C | 109.5 |
| C11—C10—C12 | 110.58 (15) | C17—C19—H19A | 109.5 |
| C9—C10—H10 | 107.8 | C17—C19—H19B | 109.5 |
| C11—C10—H10 | 107.8 | H19A—C19—H19B | 109.5 |
| C12—C10—H10 | 107.8 | C17—C19—H19C | 109.5 |
| C10—C11—H11A | 109.5 | H19A—C19—H19C | 109.5 |
| C10—C11—H11B | 109.5 | H19B—C19—H19C | 109.5 |
| H11A—C11—H11B | 109.5 | | |
| | | | |
| N2—C1—N1—C2 | 0.30 (16) | C7—N2—C8—C9 | -100.77 (14) |
| C1—N1—C2—C3 | -179.65 (15) | C16—C8—C9—C13 | 1.78 (19) |
| C1—N1—C2—C7 | -0.23 (15) | N2—C8—C9—C13 | -177.79 (12) |
| N1—C1—N2—C7 | -0.25 (16) | C16—C8—C9—C10 | -177.13 (12) |
| N1—C1—N2—C8 | -177.75 (12) | N2—C8—C9—C10 | 3.30 (18) |
| N1—C2—C3—C4 | 179.48 (14) | C13—C9—C10—C11 | -54.09 (19) |
| C7—C2—C3—C4 | 0.1 (2) | C8—C9—C10—C11 | 124.78 (15) |
| C2—C3—C4—C5 | -0.2 (2) | C13—C9—C10—C12 | 69.89 (19) |
| C3—C4—C5—C6 | 0.0 (2) | C8—C9—C10—C12 | -111.24 (16) |
| C4—C5—C6—C7 | 0.2 (2) | C8—C9—C13—C14 | -1.0 (2) |
| C5—C6—C7—N2 | -179.57 (13) | C10—C9—C13—C14 | 177.94 (13) |
| C5—C6—C7—C2 | -0.32 (19) | C9—C13—C14—C15 | -0.5 (2) |
| C1—N2—C7—C6 | 179.44 (14) | C13—C14—C15—C16 | 1.4 (2) |
| C8—N2—C7—C6 | -3.1 (2) | C14—C15—C16—C8 | -0.60 (19) |
| C1—N2—C7—C2 | 0.09 (14) | C14—C15—C16—C17 | 175.54 (13) |
| C8—N2—C7—C2 | 177.56 (12) | C9—C8—C16—C15 | -1.00 (18) |
| N1—C2—C7—C6 | -179.34 (13) | N2—C8—C16—C15 | 178.57 (11) |
| C3—C2—C7—C6 | 0.16 (19) | C9—C8—C16—C17 | -177.12 (12) |
| N1—C2—C7—N2 | 0.09 (14) | N2—C8—C16—C17 | 2.45 (18) |
| C3—C2—C7—N2 | 179.58 (12) | C15—C16—C17—C19 | 36.52 (19) |
| C1—N2—C8—C16 | -103.40 (15) | C8—C16—C17—C19 | -147.49 (14) |
| C7—N2—C8—C16 | 79.63 (16) | C15—C16—C17—C18 | -87.28 (16) |
| C1—N2—C8—C9 | 76.19 (17) | C8—C16—C17—C18 | 88.71 (15) |

Hydrogen-bond geometry (Å, °)

Cg1 and Cg2 are the centroids of the C2–C7 and C8/C9/C13–C16 rings, respectively.

| <i>D</i> —H··· <i>A</i> | <i>D</i> —H | H··· <i>A</i> | <i>D</i> ··· <i>A</i> | <i>D</i> —H··· <i>A</i> |
|--|-------------|---------------|-----------------------|-------------------------|
| C6—H6···N1 ⁱ | 0.95 | 2.47 | 3.4040 (18) | 168 |
| C14—H14···Cg3 ⁱⁱ | 0.95 | 2.68 | 3.5908 (16) | 150 |
| C18—H18 <i>B</i> ···Cg2 ⁱⁱⁱ | 0.98 | 2.79 | 3.5314 (17) | 125 |

Symmetry codes: (i) $x-1, y, z$; (ii) $x-1/2, -y+3/2, -z+1$; (iii) $-x+2, y-1/2, -z+1/2$.