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2-Oxo-2-(2-oxo-2*H*-chromen-3-yl)ethyl pyrrolidine-1-carbodithioate

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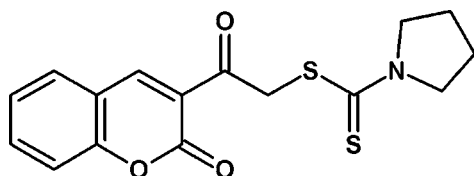
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.039; wR factor = 0.109; data-to-parameter ratio = 13.7.

There are two independent molecules in the asymmetric unit of the title compound, $\text{C}_{16}\text{H}_{15}\text{NO}_3\text{S}_2$, in which the pyrrolidine rings adopt envelope conformations, with a methylene C atom as the flap. The dihedral angles between the near-planar 2*H*-chromene ring systems [maximum deviations = 0.0167 (20) and 0.0136 (19) Å] and the pyrrolidine rings (all atoms) are 83.83 (11) and 82.43 (11)°. In the crystal, inversion dimers linked by pairs of C—H···O hydrogen bonds occur for one of the molecules. Further C—H···O links involving both molecules generate a three-dimensional network.

Related literature

For a related structure and the synthesis of the title compound, see: Mahabaleshwaraiah *et al.* (2012).



Experimental

Crystal data

$\text{C}_{16}\text{H}_{15}\text{NO}_3\text{S}_2$
 $M_r = 333.41$

Triclinic, $P\bar{1}$
 $a = 9.7158$ (2) Å

$b = 12.5040$ (2) Å
 $c = 13.5925$ (2) Å
 $\alpha = 106.415$ (1)°
 $\beta = 100.882$ (1)°
 $\gamma = 94.854$ (1)°
 $V = 1538.74$ (5) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.36$ mm⁻¹
 $T = 296$ K
 $0.24 \times 0.20 \times 0.12$ mm

Data collection

Bruker SMART CCD
diffractometer
Absorption correction: multi-scan
(*SADABS*; Bruker, 2001)
 $T_{\min} = 0.770$, $T_{\max} = 1.000$

21575 measured reflections
5420 independent reflections
4477 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.109$
 $S = 1.09$
5420 reflections

397 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.34$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.38$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C9 <i>A</i> —H9 <i>A</i> ···O5 <i>B</i> ⁱ	0.93	2.48	3.307 (2)	149
C15 <i>A</i> —H15 <i>A</i> ···O5 <i>B</i> ⁱ	0.93	2.50	3.319 (3)	147
C9 <i>B</i> —H9 <i>B</i> ···O5 <i>A</i> ⁱ	0.93	2.46	3.288 (3)	149
C15 <i>B</i> —H15 <i>B</i> ···O5 <i>A</i> ⁱ	0.93	2.50	3.319 (3)	146
C17 <i>B</i> —H17 <i>B</i> ···O3 <i>B</i> ⁱⁱ	0.97	2.47	3.432 (3)	170

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x + 1, -y + 1, -z + 2$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB7109).

References

- Bruker (2001). *SMART*, *SAINTE* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
Mahabaleshwaraiah, N. M., Kumar, K. M., Kotresh, O., Al-eryani, W. F. A. & Devarajegowda, H. C. (2012). *Acta Cryst.* **E68**, o1566.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

supporting information

Acta Cryst. (2013). E69, o1382 [doi:10.1107/S1600536813020965]

2-Oxo-2-(2-oxo-2H-chromen-3-yl)ethyl pyrrolidine-1-carbodithioate

K. Mahesh Kumar, N. M. Mahabaleshwaraiah, O. Kotresh, K. R. Roopashree and H. C. Devarajegowda

S1. Comment

As part of our ongoing structural studies of coumarin derivatives with possible biological activity (Mahabaleshwaraiah *et al.*, 2012), we now describe the structure of the title compound, (I).

There are two independent molecules in the asymmetric unit of 2-oxo-2-(2-oxo-2H-chromen-3-yl)ethyl pyrrolidine-1-carbodithioate is shown in Fig. 1. The 2H-chromene ring systems (O3a/C7a–C15a) and (O3b/C7b–C15b) are nearly planar, with a maximum deviation of 0.0167 (20) Å and 0.0136 (19) Å respectively. The pyrrolidine rings (N6a/C19a–C22a) and (N6b/C19b–C22b) are not coplanar with the 2H-chromene ring systems (O3a/C7a–C15a) and (O3b/C7b–C15b); the dihedral angles between two planes being 83.83 (11)° and 82.43 (11)° in the two molecules. In the crystal, inversion related C17B—H17B···O3B hydrogen bonds generate $R_2^2(12)$ loops.

S2. Experimental

This compound was prepared according to the reported method (Mahabaleshwaraiah *et al.*, 2012). Colourless needles were grown from a mixed solution of EtOH/CHCl₃ (v/v = 2/1) by slow evaporation at room temperature. Yield = 88%, m.p. 441 K.

S3. Refinement

All H atoms were positioned geometrically, with C—H = 0.93 Å for aromatic H, C—H = 0.97 Å for methylene H and C—H = 0.96 Å for methyl H and refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for all other H.

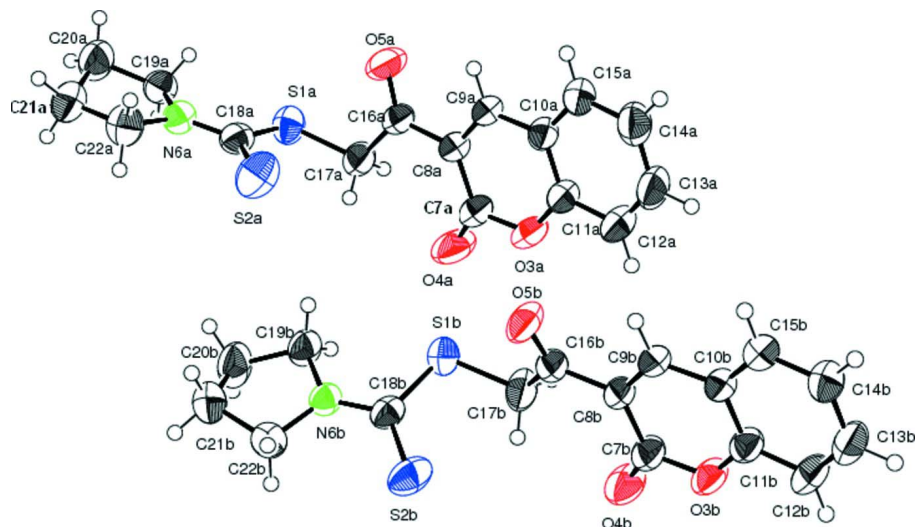


Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

2-Oxo-2-(2-oxo-2H-chromen-3-yl)ethyl pyrrolidine-1-carbodithioate

Crystal data

$C_{16}H_{15}NO_3S_2$

$M_r = 333.41$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 9.7158(2)\ \text{\AA}$

$b = 12.5040(2)\ \text{\AA}$

$c = 13.5925(2)\ \text{\AA}$

$\alpha = 106.415(1)^\circ$

$\beta = 100.882(1)^\circ$

$\gamma = 94.854(1)^\circ$

$V = 1538.74(5)\ \text{\AA}^3$

$Z = 4$

$F(000) = 696$

$D_x = 1.439\ \text{Mg m}^{-3}$

Melting point: 441 K

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 5420 reflections

$\theta = 1.7\text{--}25.0^\circ$

$\mu = 0.36\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Plate, colourless

$0.24 \times 0.20 \times 0.12\ \text{mm}$

Data collection

Bruker SMART CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω and ϕ scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2001)

$T_{\min} = 0.770$, $T_{\max} = 1.000$

21575 measured reflections

5420 independent reflections

4477 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$

$\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 1.7^\circ$

$h = -11 \rightarrow 11$

$k = -14 \rightarrow 14$

$l = -16 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.039$

$wR(F^2) = 0.109$

$S = 1.09$

5420 reflections

397 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0528P)^2 + 0.3725P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$

$$\Delta\rho_{\max} = 0.34 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.38 \text{ e } \text{\AA}^{-3}$$

Special details

Experimental. IR (KBr): 636 cm^{-1} (C—S), 1254 cm^{-1} (C=S), 1070 cm^{-1} (C—O), 854 cm^{-1} (C—N), 1125 cm^{-1} (C—O—C), 1694 cm^{-1} (C=O), 1731 cm^{-1} (Coumarin C=O). GCMS: m/e: 335. ¹H NMR (400 MHz, CDCl₃, δ , p.p.m): 2.11(m, 2H, Pyrrolidine-CH₂), 2.16(m, 2H, Pyrrolidine-CH₂), 3.68(t, 2H, Pyrrolidine-CH₂), 3.91(t, 2H, Pyrrolidine-CH₂), 4.59(s, 2H, Methylene-CH₂), 7.24(t, 1H, Ar—H), 7.41(d, 1H, Ar—H), 7.49(t, 1H, Ar—H), 7.65(d, 1H, Ar—H), 8.16(s, 1H, Ar—H). Elemental analysis for C₁₆H₁₅NO₃S₂: C, 57.58; H, 4.46; N, 4.13.

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1B	0.46924 (6)	0.77672 (4)	0.81900 (5)	0.05885 (16)
S2B	0.72066 (6)	0.92077 (6)	0.98610 (5)	0.07162 (19)
O3B	0.74479 (16)	0.48705 (12)	1.07902 (11)	0.0598 (4)
O4B	0.6049 (2)	0.61413 (15)	1.07471 (13)	0.0813 (5)
O5B	0.68855 (18)	0.62686 (13)	0.78589 (12)	0.0704 (4)
N6B	0.55263 (15)	0.98720 (12)	0.84624 (12)	0.0455 (4)
C7B	0.6776 (2)	0.55533 (17)	1.02806 (16)	0.0566 (5)
C8B	0.7036 (2)	0.54840 (15)	0.92381 (15)	0.0470 (4)
C9B	0.7859 (2)	0.47448 (15)	0.88210 (15)	0.0474 (4)
H9B	0.8008	0.4702	0.8156	0.057*
C10B	0.85098 (19)	0.40263 (15)	0.93565 (15)	0.0461 (4)
C11B	0.8291 (2)	0.41211 (16)	1.03604 (15)	0.0502 (5)
C12B	0.8900 (2)	0.34839 (19)	1.09553 (18)	0.0627 (6)
H12B	0.8739	0.3559	1.1623	0.075*
C13B	0.9745 (2)	0.27392 (19)	1.05399 (19)	0.0652 (6)
H13B	1.0165	0.2304	1.0931	0.078*
C14B	0.9988 (2)	0.26217 (17)	0.95419 (18)	0.0612 (5)
H14B	1.0565	0.2110	0.9272	0.073*
C15B	0.9378 (2)	0.32573 (16)	0.89536 (16)	0.0549 (5)
H15B	0.9543	0.3176	0.8286	0.066*
C16B	0.6446 (2)	0.62295 (15)	0.86268 (16)	0.0513 (5)
C17B	0.5292 (3)	0.68791 (18)	0.89563 (19)	0.0650 (6)
H17A	0.5630	0.7342	0.9683	0.078*
H17B	0.4489	0.6347	0.8928	0.078*
C18B	0.58542 (19)	0.90472 (16)	0.88551 (15)	0.0464 (4)
C19B	0.4308 (2)	0.97956 (16)	0.76104 (17)	0.0568 (5)
H19A	0.3446	0.9454	0.7730	0.068*
H19B	0.4458	0.9359	0.6936	0.068*

C20B	0.4238 (3)	1.10145 (19)	0.7655 (2)	0.0776 (7)
H20A	0.3921	1.1074	0.6954	0.093*
H20B	0.3588	1.1328	0.8082	0.093*
C21B	0.5699 (3)	1.16160 (18)	0.8132 (2)	0.0697 (6)
H21A	0.5687	1.2402	0.8496	0.084*
H21B	0.6245	1.1574	0.7597	0.084*
C22B	0.6319 (2)	1.10183 (16)	0.88939 (17)	0.0573 (5)
H22A	0.7323	1.1007	0.8926	0.069*
H22B	0.6185	1.1381	0.9594	0.069*
S1A	-0.04479 (6)	0.82131 (5)	0.38879 (5)	0.06212 (17)
S2A	0.26976 (7)	0.90519 (6)	0.46861 (6)	0.0836 (2)
O3A	0.24069 (18)	0.53930 (13)	0.65250 (11)	0.0711 (4)
O4A	0.1038 (2)	0.66877 (16)	0.64994 (13)	0.0907 (6)
O5A	0.0825 (2)	0.61136 (15)	0.32787 (13)	0.0935 (6)
N6A	0.09101 (16)	0.98182 (13)	0.34318 (13)	0.0503 (4)
C7A	0.1645 (3)	0.60259 (19)	0.59960 (17)	0.0620 (5)
C8A	0.1661 (2)	0.57919 (16)	0.48808 (15)	0.0512 (5)
C9A	0.2381 (2)	0.49904 (15)	0.44312 (15)	0.0518 (5)
H9A	0.2358	0.4836	0.3717	0.062*
C10A	0.3180 (2)	0.43666 (15)	0.50018 (16)	0.0507 (5)
C11A	0.3162 (2)	0.45940 (17)	0.60620 (16)	0.0585 (5)
C12A	0.3893 (3)	0.4023 (2)	0.6678 (2)	0.0774 (7)
H12A	0.3854	0.4167	0.7381	0.093*
C13A	0.4674 (3)	0.3242 (2)	0.6226 (2)	0.0787 (7)
H13A	0.5183	0.2861	0.6634	0.094*
C14A	0.4726 (3)	0.30032 (18)	0.5179 (2)	0.0698 (6)
H14A	0.5268	0.2470	0.4891	0.084*
C15A	0.3977 (2)	0.35547 (16)	0.45664 (18)	0.0598 (5)
H15A	0.4001	0.3387	0.3859	0.072*
C16A	0.0888 (2)	0.64165 (17)	0.42166 (16)	0.0582 (5)
C17A	0.0179 (3)	0.73924 (18)	0.47142 (18)	0.0632 (6)
H17C	0.0847	0.7885	0.5333	0.076*
H17D	-0.0616	0.7101	0.4948	0.076*
C18A	0.1125 (2)	0.91016 (16)	0.39811 (16)	0.0524 (5)
C19A	-0.0455 (2)	0.99070 (17)	0.28054 (16)	0.0546 (5)
H19C	-0.1066	1.0253	0.3253	0.065*
H19D	-0.0935	0.9172	0.2351	0.065*
C20A	-0.0040 (3)	1.0652 (2)	0.2166 (2)	0.0765 (7)
H20C	0.0106	1.0197	0.1497	0.092*
H20D	-0.0773	1.1107	0.2035	0.092*
C21A	0.1297 (2)	1.1377 (2)	0.2816 (2)	0.0690 (6)
H21C	0.1099	1.2057	0.3291	0.083*
H21D	0.1877	1.1587	0.2373	0.083*
C22A	0.2031 (2)	1.06729 (19)	0.3421 (2)	0.0672 (6)
H22C	0.2740	1.0322	0.3078	0.081*
H22D	0.2489	1.1131	0.4132	0.081*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1B	0.0681 (3)	0.0498 (3)	0.0632 (4)	0.0148 (2)	0.0129 (3)	0.0236 (2)
S2B	0.0666 (4)	0.0952 (4)	0.0564 (4)	0.0269 (3)	-0.0002 (3)	0.0330 (3)
O3B	0.0772 (9)	0.0722 (9)	0.0451 (8)	0.0197 (7)	0.0246 (7)	0.0319 (7)
O4B	0.1223 (14)	0.0904 (11)	0.0576 (10)	0.0506 (11)	0.0490 (10)	0.0349 (9)
O5B	0.1029 (12)	0.0805 (10)	0.0563 (9)	0.0452 (9)	0.0402 (9)	0.0419 (8)
N6B	0.0439 (8)	0.0489 (8)	0.0451 (9)	0.0114 (6)	0.0063 (7)	0.0177 (7)
C7B	0.0768 (14)	0.0566 (11)	0.0469 (12)	0.0160 (10)	0.0236 (10)	0.0238 (10)
C8B	0.0619 (11)	0.0454 (10)	0.0391 (10)	0.0085 (8)	0.0164 (9)	0.0181 (8)
C9B	0.0622 (11)	0.0472 (10)	0.0358 (10)	0.0058 (8)	0.0132 (8)	0.0168 (8)
C10B	0.0550 (10)	0.0449 (9)	0.0404 (10)	0.0045 (8)	0.0087 (8)	0.0180 (8)
C11B	0.0556 (11)	0.0541 (11)	0.0460 (11)	0.0041 (9)	0.0133 (9)	0.0233 (9)
C12B	0.0713 (13)	0.0773 (14)	0.0518 (13)	0.0088 (11)	0.0137 (10)	0.0398 (11)
C13B	0.0685 (13)	0.0698 (14)	0.0658 (15)	0.0113 (11)	0.0043 (11)	0.0407 (12)
C14B	0.0667 (13)	0.0572 (12)	0.0635 (14)	0.0165 (10)	0.0085 (11)	0.0260 (11)
C15B	0.0667 (12)	0.0539 (11)	0.0458 (12)	0.0113 (9)	0.0098 (9)	0.0191 (9)
C16B	0.0695 (12)	0.0454 (10)	0.0445 (11)	0.0122 (9)	0.0195 (10)	0.0168 (9)
C17B	0.0857 (15)	0.0598 (12)	0.0717 (15)	0.0276 (11)	0.0369 (12)	0.0374 (11)
C18B	0.0490 (10)	0.0569 (11)	0.0417 (10)	0.0206 (8)	0.0177 (8)	0.0195 (9)
C19B	0.0535 (11)	0.0582 (11)	0.0579 (13)	0.0092 (9)	-0.0022 (9)	0.0259 (10)
C20B	0.0760 (15)	0.0692 (14)	0.0923 (19)	0.0148 (12)	-0.0030 (13)	0.0449 (14)
C21B	0.0829 (16)	0.0533 (12)	0.0761 (16)	0.0101 (11)	0.0146 (13)	0.0267 (11)
C22B	0.0589 (12)	0.0525 (11)	0.0567 (13)	0.0043 (9)	0.0088 (10)	0.0146 (10)
S1A	0.0642 (3)	0.0604 (3)	0.0718 (4)	0.0186 (2)	0.0199 (3)	0.0301 (3)
S2A	0.0641 (4)	0.1074 (5)	0.0888 (5)	0.0289 (3)	0.0023 (3)	0.0495 (4)
O3A	0.1042 (12)	0.0781 (10)	0.0399 (8)	0.0224 (9)	0.0178 (8)	0.0279 (8)
O4A	0.1331 (16)	0.1087 (14)	0.0493 (10)	0.0566 (12)	0.0395 (10)	0.0302 (10)
O5A	0.1646 (18)	0.0908 (12)	0.0433 (10)	0.0714 (12)	0.0306 (10)	0.0290 (9)
N6A	0.0475 (8)	0.0547 (9)	0.0514 (10)	0.0140 (7)	0.0102 (7)	0.0191 (8)
C7A	0.0842 (15)	0.0650 (13)	0.0432 (12)	0.0146 (11)	0.0165 (11)	0.0240 (10)
C8A	0.0678 (12)	0.0491 (10)	0.0382 (11)	0.0061 (9)	0.0128 (9)	0.0161 (9)
C9A	0.0715 (13)	0.0500 (10)	0.0363 (10)	0.0061 (9)	0.0115 (9)	0.0183 (9)
C10A	0.0598 (11)	0.0456 (10)	0.0464 (11)	-0.0001 (8)	0.0059 (9)	0.0195 (9)
C11A	0.0740 (13)	0.0546 (11)	0.0458 (12)	0.0018 (10)	0.0046 (10)	0.0216 (10)
C12A	0.1071 (19)	0.0735 (15)	0.0520 (14)	0.0074 (14)	-0.0004 (13)	0.0332 (12)
C13A	0.0972 (18)	0.0650 (14)	0.0736 (18)	0.0093 (13)	-0.0082 (14)	0.0393 (13)
C14A	0.0748 (14)	0.0555 (12)	0.0796 (18)	0.0122 (10)	0.0044 (12)	0.0292 (12)
C15A	0.0723 (13)	0.0510 (11)	0.0575 (13)	0.0070 (10)	0.0085 (11)	0.0233 (10)
C16A	0.0842 (15)	0.0558 (11)	0.0418 (12)	0.0196 (10)	0.0203 (10)	0.0192 (9)
C17A	0.0853 (15)	0.0601 (12)	0.0555 (13)	0.0197 (11)	0.0271 (11)	0.0256 (11)
C18A	0.0589 (11)	0.0550 (11)	0.0463 (11)	0.0226 (9)	0.0147 (9)	0.0142 (9)
C19A	0.0530 (11)	0.0556 (11)	0.0536 (12)	0.0135 (9)	0.0046 (9)	0.0175 (10)
C20A	0.0835 (16)	0.0745 (15)	0.0727 (17)	0.0067 (12)	0.0010 (13)	0.0364 (13)
C21A	0.0718 (14)	0.0744 (14)	0.0741 (16)	0.0134 (11)	0.0256 (12)	0.0365 (13)
C22A	0.0547 (12)	0.0714 (14)	0.0800 (17)	0.0060 (10)	0.0163 (11)	0.0301 (12)

Geometric parameters (Å, °)

S1B—C18B	1.773 (2)	S1A—C18A	1.774 (2)
S1B—C17B	1.787 (2)	S1A—C17A	1.785 (2)
S2B—C18B	1.6608 (19)	S2A—C18A	1.659 (2)
O3B—C11B	1.375 (2)	O3A—C11A	1.370 (3)
O3B—C7B	1.376 (2)	O3A—C7A	1.382 (2)
O4B—C7B	1.198 (2)	O4A—C7A	1.190 (3)
O5B—C16B	1.211 (2)	O5A—C16A	1.211 (2)
N6B—C18B	1.324 (2)	N6A—C18A	1.325 (2)
N6B—C19B	1.467 (2)	N6A—C22A	1.465 (3)
N6B—C22B	1.469 (2)	N6A—C19A	1.466 (2)
C7B—C8B	1.466 (3)	C7A—C8A	1.464 (3)
C8B—C9B	1.348 (3)	C8A—C9A	1.342 (3)
C8B—C16B	1.495 (3)	C8A—C16A	1.494 (3)
C9B—C10B	1.424 (2)	C9A—C10A	1.425 (3)
C9B—H9B	0.9300	C9A—H9A	0.9300
C10B—C11B	1.394 (3)	C10A—C11A	1.392 (3)
C10B—C15B	1.396 (3)	C10A—C15A	1.395 (3)
C11B—C12B	1.379 (3)	C11A—C12A	1.383 (3)
C12B—C13B	1.367 (3)	C12A—C13A	1.368 (4)
C12B—H12B	0.9300	C12A—H12A	0.9300
C13B—C14B	1.390 (3)	C13A—C14A	1.381 (4)
C13B—H13B	0.9300	C13A—H13A	0.9300
C14B—C15B	1.372 (3)	C14A—C15A	1.371 (3)
C14B—H14B	0.9300	C14A—H14A	0.9300
C15B—H15B	0.9300	C15A—H15A	0.9300
C16B—C17B	1.502 (3)	C16A—C17A	1.507 (3)
C17B—H17A	0.9700	C17A—H17C	0.9700
C17B—H17B	0.9700	C17A—H17D	0.9700
C19B—C20B	1.516 (3)	C19A—C20A	1.520 (3)
C19B—H19A	0.9700	C19A—H19C	0.9700
C19B—H19B	0.9700	C19A—H19D	0.9700
C20B—C21B	1.483 (3)	C20A—C21A	1.485 (3)
C20B—H20A	0.9700	C20A—H20C	0.9700
C20B—H20B	0.9700	C20A—H20D	0.9700
C21B—C22B	1.507 (3)	C21A—C22A	1.501 (3)
C21B—H21A	0.9700	C21A—H21C	0.9700
C21B—H21B	0.9700	C21A—H21D	0.9700
C22B—H22A	0.9700	C22A—H22C	0.9700
C22B—H22B	0.9700	C22A—H22D	0.9700
C18B—S1B—C17B	102.01 (11)	C18A—S1A—C17A	101.77 (10)
C11B—O3B—C7B	123.64 (15)	C11A—O3A—C7A	123.44 (16)
C18B—N6B—C19B	125.64 (16)	C18A—N6A—C22A	123.09 (17)
C18B—N6B—C22B	122.64 (16)	C18A—N6A—C19A	125.71 (17)
C19B—N6B—C22B	111.60 (14)	C22A—N6A—C19A	111.15 (16)
O4B—C7B—O3B	115.83 (18)	O4A—C7A—O3A	116.01 (19)

O4B—C7B—C8B	127.55 (19)	O4A—C7A—C8A	127.7 (2)
O3B—C7B—C8B	116.62 (17)	O3A—C7A—C8A	116.33 (19)
C9B—C8B—C7B	119.33 (17)	C9A—C8A—C7A	119.63 (18)
C9B—C8B—C16B	118.46 (16)	C9A—C8A—C16A	118.44 (17)
C7B—C8B—C16B	122.20 (17)	C7A—C8A—C16A	121.92 (18)
C8B—C9B—C10B	122.74 (17)	C8A—C9A—C10A	122.76 (18)
C8B—C9B—H9B	118.6	C8A—C9A—H9A	118.6
C10B—C9B—H9B	118.6	C10A—C9A—H9A	118.6
C11B—C10B—C15B	118.09 (17)	C11A—C10A—C15A	118.51 (19)
C11B—C10B—C9B	117.53 (17)	C11A—C10A—C9A	117.30 (19)
C15B—C10B—C9B	124.36 (17)	C15A—C10A—C9A	124.19 (19)
O3B—C11B—C12B	117.76 (18)	O3A—C11A—C12A	118.0 (2)
O3B—C11B—C10B	120.10 (16)	O3A—C11A—C10A	120.50 (18)
C12B—C11B—C10B	122.14 (19)	C12A—C11A—C10A	121.5 (2)
C13B—C12B—C11B	118.4 (2)	C13A—C12A—C11A	118.2 (2)
C13B—C12B—H12B	120.8	C13A—C12A—H12A	120.9
C11B—C12B—H12B	120.8	C11A—C12A—H12A	120.9
C12B—C13B—C14B	121.07 (19)	C12A—C13A—C14A	121.7 (2)
C12B—C13B—H13B	119.5	C12A—C13A—H13A	119.1
C14B—C13B—H13B	119.5	C14A—C13A—H13A	119.1
C15B—C14B—C13B	120.3 (2)	C15A—C14A—C13A	119.8 (2)
C15B—C14B—H14B	119.9	C15A—C14A—H14A	120.1
C13B—C14B—H14B	119.9	C13A—C14A—H14A	120.1
C14B—C15B—C10B	120.04 (19)	C14A—C15A—C10A	120.2 (2)
C14B—C15B—H15B	120.0	C14A—C15A—H15A	119.9
C10B—C15B—H15B	120.0	C10A—C15A—H15A	119.9
O5B—C16B—C8B	118.99 (17)	O5A—C16A—C8A	118.90 (18)
O5B—C16B—C17B	121.65 (18)	O5A—C16A—C17A	121.16 (19)
C8B—C16B—C17B	119.33 (17)	C8A—C16A—C17A	119.93 (17)
C16B—C17B—S1B	115.66 (15)	C16A—C17A—S1A	115.49 (15)
C16B—C17B—H17A	108.4	C16A—C17A—H17C	108.4
S1B—C17B—H17A	108.4	S1A—C17A—H17C	108.4
C16B—C17B—H17B	108.4	C16A—C17A—H17D	108.4
S1B—C17B—H17B	108.4	S1A—C17A—H17D	108.4
H17A—C17B—H17B	107.4	H17C—C17A—H17D	107.5
N6B—C18B—S2B	123.21 (15)	N6A—C18A—S2A	123.39 (16)
N6B—C18B—S1B	112.52 (14)	N6A—C18A—S1A	112.74 (14)
S2B—C18B—S1B	124.27 (11)	S2A—C18A—S1A	123.87 (12)
N6B—C19B—C20B	103.42 (16)	N6A—C19A—C20A	103.19 (17)
N6B—C19B—H19A	111.1	N6A—C19A—H19C	111.1
C20B—C19B—H19A	111.1	C20A—C19A—H19C	111.1
N6B—C19B—H19B	111.1	N6A—C19A—H19D	111.1
C20B—C19B—H19B	111.1	C20A—C19A—H19D	111.1
H19A—C19B—H19B	109.0	H19C—C19A—H19D	109.1
C21B—C20B—C19B	105.75 (17)	C21A—C20A—C19A	105.50 (19)
C21B—C20B—H20A	110.6	C21A—C20A—H20C	110.6
C19B—C20B—H20A	110.6	C19A—C20A—H20C	110.6
C21B—C20B—H20B	110.6	C21A—C20A—H20D	110.6

C19B—C20B—H20B	110.6	C19A—C20A—H20D	110.6
H20A—C20B—H20B	108.7	H20C—C20A—H20D	108.8
C20B—C21B—C22B	105.07 (18)	C20A—C21A—C22A	105.23 (18)
C20B—C21B—H21A	110.7	C20A—C21A—H21C	110.7
C22B—C21B—H21A	110.7	C22A—C21A—H21C	110.7
C20B—C21B—H21B	110.7	C20A—C21A—H21D	110.7
C22B—C21B—H21B	110.7	C22A—C21A—H21D	110.7
H21A—C21B—H21B	108.8	H21C—C21A—H21D	108.8
N6B—C22B—C21B	104.04 (16)	N6A—C22A—C21A	105.22 (17)
N6B—C22B—H22A	110.9	N6A—C22A—H22C	110.7
C21B—C22B—H22A	110.9	C21A—C22A—H22C	110.7
N6B—C22B—H22B	110.9	N6A—C22A—H22D	110.7
C21B—C22B—H22B	110.9	C21A—C22A—H22D	110.7
H22A—C22B—H22B	109.0	H22C—C22A—H22D	108.8
C11B—O3B—C7B—O4B	178.2 (2)	C11A—O3A—C7A—O4A	179.9 (2)
C11B—O3B—C7B—C8B	-2.1 (3)	C11A—O3A—C7A—C8A	0.8 (3)
O4B—C7B—C8B—C9B	-178.1 (2)	O4A—C7A—C8A—C9A	-178.3 (2)
O3B—C7B—C8B—C9B	2.3 (3)	O3A—C7A—C8A—C9A	0.6 (3)
O4B—C7B—C8B—C16B	3.5 (4)	O4A—C7A—C8A—C16A	1.3 (4)
O3B—C7B—C8B—C16B	-176.11 (17)	O3A—C7A—C8A—C16A	-179.78 (18)
C7B—C8B—C9B—C10B	-0.7 (3)	C7A—C8A—C9A—C10A	-1.9 (3)
C16B—C8B—C9B—C10B	177.76 (17)	C16A—C8A—C9A—C10A	178.45 (18)
C8B—C9B—C10B—C11B	-1.1 (3)	C8A—C9A—C10A—C11A	1.8 (3)
C8B—C9B—C10B—C15B	-179.45 (19)	C8A—C9A—C10A—C15A	-177.80 (19)
C7B—O3B—C11B—C12B	-179.84 (18)	C7A—O3A—C11A—C12A	179.4 (2)
C7B—O3B—C11B—C10B	0.3 (3)	C7A—O3A—C11A—C10A	-0.9 (3)
C15B—C10B—C11B—O3B	179.76 (17)	C15A—C10A—C11A—O3A	179.27 (18)
C9B—C10B—C11B—O3B	1.3 (3)	C9A—C10A—C11A—O3A	-0.4 (3)
C15B—C10B—C11B—C12B	0.0 (3)	C15A—C10A—C11A—C12A	-1.0 (3)
C9B—C10B—C11B—C12B	-178.46 (18)	C9A—C10A—C11A—C12A	179.3 (2)
O3B—C11B—C12B—C13B	-179.70 (19)	O3A—C11A—C12A—C13A	-178.6 (2)
C10B—C11B—C12B—C13B	0.1 (3)	C10A—C11A—C12A—C13A	1.7 (4)
C11B—C12B—C13B—C14B	-0.1 (3)	C11A—C12A—C13A—C14A	-1.1 (4)
C12B—C13B—C14B—C15B	0.1 (3)	C12A—C13A—C14A—C15A	-0.2 (4)
C13B—C14B—C15B—C10B	0.0 (3)	C13A—C14A—C15A—C10A	0.9 (3)
C11B—C10B—C15B—C14B	0.0 (3)	C11A—C10A—C15A—C14A	-0.3 (3)
C9B—C10B—C15B—C14B	178.29 (18)	C9A—C10A—C15A—C14A	179.32 (19)
C9B—C8B—C16B—O5B	-10.7 (3)	C9A—C8A—C16A—O5A	6.7 (3)
C7B—C8B—C16B—O5B	167.7 (2)	C7A—C8A—C16A—O5A	-172.9 (2)
C9B—C8B—C16B—C17B	167.16 (19)	C9A—C8A—C16A—C17A	-174.44 (19)
C7B—C8B—C16B—C17B	-14.4 (3)	C7A—C8A—C16A—C17A	6.0 (3)
O5B—C16B—C17B—S1B	-3.5 (3)	O5A—C16A—C17A—S1A	-11.3 (3)
C8B—C16B—C17B—S1B	178.68 (15)	C8A—C16A—C17A—S1A	169.88 (16)
C18B—S1B—C17B—C16B	-89.01 (19)	C18A—S1A—C17A—C16A	-80.90 (19)
C19B—N6B—C18B—S2B	-176.91 (15)	C22A—N6A—C18A—S2A	-2.4 (3)
C22B—N6B—C18B—S2B	-1.1 (3)	C19A—N6A—C18A—S2A	-179.34 (15)
C19B—N6B—C18B—S1B	3.0 (2)	C22A—N6A—C18A—S1A	177.18 (16)

C22B—N6B—C18B—S1B	178.87 (14)	C19A—N6A—C18A—S1A	0.2 (2)
C17B—S1B—C18B—N6B	-175.02 (14)	C17A—S1A—C18A—N6A	-178.87 (15)
C17B—S1B—C18B—S2B	4.90 (15)	C17A—S1A—C18A—S2A	0.69 (16)
C18B—N6B—C19B—C20B	167.00 (19)	C18A—N6A—C19A—C20A	-167.79 (19)
C22B—N6B—C19B—C20B	-9.2 (2)	C22A—N6A—C19A—C20A	15.0 (2)
N6B—C19B—C20B—C21B	25.7 (3)	N6A—C19A—C20A—C21A	-28.4 (2)
C19B—C20B—C21B—C22B	-32.7 (3)	C19A—C20A—C21A—C22A	31.4 (3)
C18B—N6B—C22B—C21B	173.23 (18)	C18A—N6A—C22A—C21A	-173.35 (19)
C19B—N6B—C22B—C21B	-10.4 (2)	C19A—N6A—C22A—C21A	4.0 (2)
C20B—C21B—C22B—N6B	26.3 (2)	C20A—C21A—C22A—N6A	-21.9 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C9A—H9A \cdots O5B ⁱ	0.93	2.48	3.307 (2)	149
C15A—H15A \cdots O5B ⁱ	0.93	2.50	3.319 (3)	147
C9B—H9B \cdots O5A ⁱ	0.93	2.46	3.288 (3)	149
C15B—H15B \cdots O5A ⁱ	0.93	2.50	3.319 (3)	146
C17B—H17B \cdots O3B ⁱⁱ	0.97	2.47	3.432 (3)	170

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+1, -y+1, -z+2$.