

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

1-(4-Methoxyphenyl)pyrrolidine-2,5-dione

Muhammad Sirajuddin* and Saqib Ali

 Department of Chemistry, Quaid-i-Azam University, Islamabad, Pakistan
 Correspondence e-mail: m.siraj09@yahoo.com

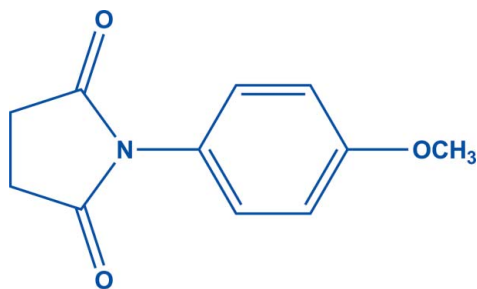
Received 5 August 2013; accepted 9 August 2013

 Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å;
 R factor = 0.034; wR factor = 0.094; data-to-parameter ratio = 14.0.

In the title compound, $\text{C}_{11}\text{H}_{11}\text{NO}_3$, the dihydrofuran-2,5-dione ring has a shallow envelope conformation, with one of the methylene C atoms displaced by 0.216 (1) Å from the other atoms. These near-planar atoms subtend a dihedral angle of 55.88 (8)° with the benzene ring. In the crystal, $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into [010] chains.

Related literature

For related structures, see: Sirajuddin *et al.* (2012); Tahir *et al.* (2012).



Experimental

Crystal data

 $\text{C}_{11}\text{H}_{11}\text{NO}_3$
 $M_r = 205.21$
 Monoclinic, $P2_1/n$
 $a = 9.3684$ (7) Å

 $b = 6.6146$ (4) Å
 $c = 16.0720$ (11) Å
 $\beta = 99.939$ (4)°
 $V = 981.01$ (12) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.10$ mm⁻¹
 $T = 296$ K
 $0.32 \times 0.25 \times 0.22$ mm

Data collection

 Bruker Kappa APEXII CCD
 diffractometer
 Absorption correction: multi-scan
 (SADABS; Bruker, 2005)
 $T_{\min} = 0.968$, $T_{\max} = 0.978$

 7585 measured reflections
 1927 independent reflections
 1626 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.018$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.094$
 $S = 1.04$
 1927 reflections

 138 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.14$ e Å⁻³
 $\Delta\rho_{\min} = -0.14$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C2}-\text{H2}\cdots\text{O3}^i$	0.93	2.50	3.1666 (17)	129
$\text{C5}-\text{H5}\cdots\text{O2}^{ii}$	0.93	2.47	3.3245 (17)	152

 Symmetry codes: (i) $x, y - 1, z$; (ii) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL2012 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012) and PLATON (Spek, 2009); software used to prepare material for publication: WinGX (Farrugia, 2012) and PLATON.

The authors acknowledge Quaid-i-Azam University, Islamabad, Pakistan, for providing research facilities, and the University Research Fund (URF) for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB7117).

References

- Bruker (2005). APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Sirajuddin, M., Ali, S. & Tahir, M. N. (2012). *Acta Cryst.* **E68**, o2282.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Tahir, M. N., Sirajuddin, M., Ali, S. & Munawar, K. S. (2012). *Acta Cryst.* **E68**, o2589.

supporting information

Acta Cryst. (2013). E69, o1439 [doi:10.1107/S1600536813022460]

1-(4-Methoxyphenyl)pyrrolidine-2,5-dione

Muhammad Sirajuddin and Saqib Ali

S1. Experimental

Equimolar quantities of 4-methoxyaniline and dihydrofuran-2,5-dione were stirred and refluxed in acetic acid for 4 h. The solution was kept at room temperature for 24 h which afforded colourless prisms of the title compound.

S2. Refinement

The H atoms were positioned geometrically (C—H = 0.93–0.96 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$, where $x = 1.5$ for methyl and $x = 1.2$ for other H atoms.

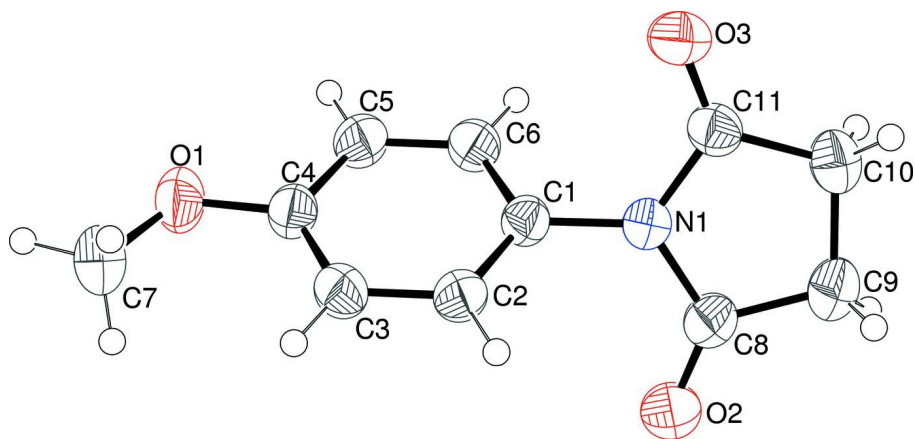
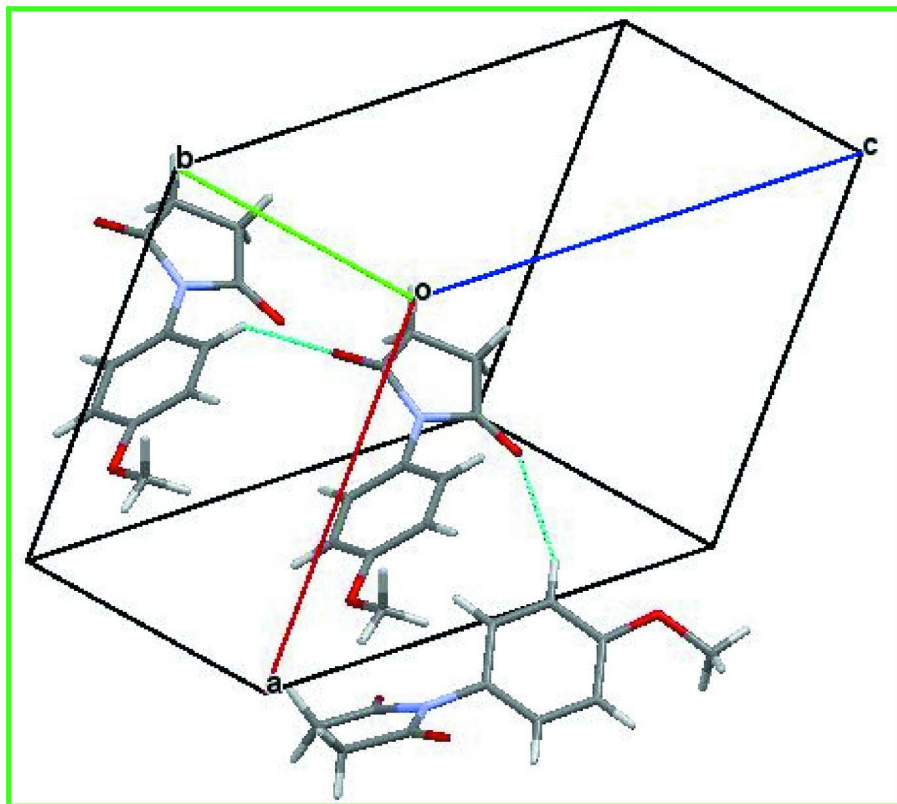


Figure 1

View of the title compound with displacement ellipsoids drawn at the 50% probability level.

**Figure 2**

The partial packing, which shows that molecules form dimers due to C—H···O bondings.

1-(4-Methoxyphenyl)pyrrolidine-2,5-dione

Crystal data

$C_{11}H_{11}NO_3$

$M_r = 205.21$

Monoclinic, $P2_1/n$

$a = 9.3684 (7) \text{ \AA}$

$b = 6.6146 (4) \text{ \AA}$

$c = 16.0720 (11) \text{ \AA}$

$\beta = 99.939 (4)^\circ$

$V = 981.01 (12) \text{ \AA}^3$

$Z = 4$

$F(000) = 432$

$D_x = 1.389 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

$\mu = 0.10 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Prism, colourless

$0.32 \times 0.25 \times 0.22 \text{ mm}$

Data collection

Bruker Kappa APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube

Detector resolution: 8 pixels mm^{-1}

ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2005)

$T_{\min} = 0.968$, $T_{\max} = 0.978$

7585 measured reflections

1927 independent reflections

1626 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.018$

$\theta_{\max} = 26.0^\circ$, $\theta_{\min} = 2.4^\circ$

$h = -11 \rightarrow 11$

$k = -8 \rightarrow 7$

$l = -19 \rightarrow 19$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.034$

$wR(F^2) = 0.094$

$S = 1.04$

1927 reflections

138 parameters

0 restraints

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0409P)^2 + 0.2498P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.14 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$

Extinction correction: *SHELXL2012* (Sheldrick,
2012), $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.047 (4)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.73798 (12)	-0.12759 (16)	0.03467 (7)	0.0594 (3)
O2	0.52321 (14)	-0.03671 (17)	0.38782 (7)	0.0685 (4)
O3	0.38050 (12)	0.48906 (15)	0.21179 (6)	0.0588 (3)
N1	0.48031 (11)	0.21443 (15)	0.28804 (6)	0.0397 (3)
C1	0.54572 (13)	0.12100 (19)	0.22333 (7)	0.0378 (3)
C2	0.50543 (14)	-0.0707 (2)	0.19526 (8)	0.0418 (3)
H2	0.4367	-0.1412	0.2192	0.050*
C3	0.56658 (14)	-0.1592 (2)	0.13165 (8)	0.0441 (3)
H3	0.5385	-0.2882	0.1125	0.053*
C4	0.66963 (14)	-0.0546 (2)	0.09689 (8)	0.0427 (3)
C5	0.71239 (14)	0.1368 (2)	0.12663 (9)	0.0461 (3)
H5	0.7830	0.2061	0.1039	0.055*
C6	0.65080 (14)	0.2246 (2)	0.18962 (8)	0.0427 (3)
H6	0.6797	0.3529	0.2094	0.051*
C7	0.6833 (2)	-0.3084 (2)	-0.00666 (10)	0.0639 (4)
H7A	0.6941	-0.4177	0.0333	0.096*
H7B	0.7362	-0.3386	-0.0512	0.096*
H7C	0.5825	-0.2911	-0.0299	0.096*
C8	0.47536 (15)	0.1275 (2)	0.36638 (8)	0.0471 (3)
C9	0.39873 (18)	0.2721 (2)	0.41557 (9)	0.0556 (4)
H9A	0.4574	0.3009	0.4701	0.067*
H9B	0.3067	0.2164	0.4244	0.067*
C10	0.37563 (16)	0.4609 (2)	0.36233 (9)	0.0494 (4)
H10A	0.2764	0.5078	0.3572	0.059*
H10B	0.4402	0.5679	0.3869	0.059*
C11	0.40880 (14)	0.39934 (19)	0.27787 (8)	0.0418 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0720 (7)	0.0563 (6)	0.0574 (6)	-0.0022 (5)	0.0327 (5)	-0.0108 (5)
O2	0.0974 (9)	0.0604 (7)	0.0501 (6)	0.0251 (6)	0.0192 (6)	0.0146 (5)
O3	0.0802 (7)	0.0454 (6)	0.0510 (6)	0.0083 (5)	0.0116 (5)	0.0074 (5)
N1	0.0446 (6)	0.0388 (6)	0.0366 (6)	0.0017 (4)	0.0099 (4)	-0.0006 (4)
C1	0.0399 (6)	0.0397 (6)	0.0337 (6)	0.0018 (5)	0.0065 (5)	-0.0006 (5)
C2	0.0427 (7)	0.0406 (7)	0.0440 (7)	-0.0038 (5)	0.0125 (5)	0.0014 (5)
C3	0.0488 (7)	0.0380 (7)	0.0458 (7)	-0.0034 (6)	0.0090 (6)	-0.0048 (6)
C4	0.0447 (7)	0.0458 (7)	0.0388 (7)	0.0036 (6)	0.0110 (5)	-0.0010 (6)
C5	0.0450 (7)	0.0468 (8)	0.0494 (8)	-0.0058 (6)	0.0163 (6)	0.0017 (6)
C6	0.0453 (7)	0.0392 (7)	0.0436 (7)	-0.0054 (5)	0.0079 (5)	-0.0033 (5)
C7	0.0837 (11)	0.0598 (10)	0.0514 (9)	0.0052 (8)	0.0207 (8)	-0.0129 (7)
C8	0.0540 (8)	0.0498 (8)	0.0377 (7)	0.0036 (6)	0.0088 (6)	0.0021 (6)
C9	0.0657 (9)	0.0614 (9)	0.0430 (8)	0.0048 (7)	0.0183 (7)	-0.0035 (7)
C10	0.0496 (8)	0.0492 (8)	0.0502 (8)	0.0003 (6)	0.0110 (6)	-0.0116 (6)
C11	0.0440 (7)	0.0363 (6)	0.0447 (7)	-0.0034 (5)	0.0066 (5)	-0.0023 (6)

Geometric parameters (\AA , $^\circ$)

O1—C4	1.3654 (16)	C5—C6	1.3773 (18)
O1—C7	1.4203 (18)	C5—H5	0.9300
O2—C8	1.2028 (17)	C6—H6	0.9300
O3—C11	1.2055 (16)	C7—H7A	0.9600
N1—C11	1.3905 (16)	C7—H7B	0.9600
N1—C8	1.3920 (17)	C7—H7C	0.9600
N1—C1	1.4350 (15)	C8—C9	1.5002 (19)
C1—C2	1.3766 (18)	C9—C10	1.508 (2)
C1—C6	1.3844 (18)	C9—H9A	0.9700
C2—C3	1.3850 (18)	C9—H9B	0.9700
C2—H2	0.9300	C10—C11	1.5003 (18)
C3—C4	1.3819 (18)	C10—H10A	0.9700
C3—H3	0.9300	C10—H10B	0.9700
C4—C5	1.3873 (19)		
C4—O1—C7	117.61 (11)	O1—C7—H7B	109.5
C11—N1—C8	112.27 (11)	H7A—C7—H7B	109.5
C11—N1—C1	123.43 (10)	O1—C7—H7C	109.5
C8—N1—C1	124.25 (11)	H7A—C7—H7C	109.5
C2—C1—C6	120.06 (11)	H7B—C7—H7C	109.5
C2—C1—N1	120.46 (11)	O2—C8—N1	124.27 (13)
C6—C1—N1	119.47 (11)	O2—C8—C9	127.83 (13)
C1—C2—C3	120.42 (12)	N1—C8—C9	107.89 (12)
C1—C2—H2	119.8	C8—C9—C10	105.29 (11)
C3—C2—H2	119.8	C8—C9—H9A	110.7
C4—C3—C2	119.58 (12)	C10—C9—H9A	110.7
C4—C3—H3	120.2	C8—C9—H9B	110.7

C2—C3—H3	120.2	C10—C9—H9B	110.7
O1—C4—C3	124.58 (12)	H9A—C9—H9B	108.8
O1—C4—C5	115.53 (12)	C11—C10—C9	104.87 (11)
C3—C4—C5	119.87 (12)	C11—C10—H10A	110.8
C6—C5—C4	120.36 (12)	C9—C10—H10A	110.8
C6—C5—H5	119.8	C11—C10—H10B	110.8
C4—C5—H5	119.8	C9—C10—H10B	110.8
C5—C6—C1	119.69 (12)	H10A—C10—H10B	108.8
C5—C6—H6	120.2	O3—C11—N1	124.39 (12)
C1—C6—H6	120.2	O3—C11—C10	127.96 (13)
O1—C7—H7A	109.5	N1—C11—C10	107.65 (11)
C11—N1—C1—C2	-123.12 (13)	N1—C1—C6—C5	-179.05 (11)
C8—N1—C1—C2	53.95 (17)	C11—N1—C8—O2	176.06 (14)
C11—N1—C1—C6	57.37 (16)	C1—N1—C8—O2	-1.3 (2)
C8—N1—C1—C6	-125.55 (14)	C11—N1—C8—C9	-2.76 (16)
C6—C1—C2—C3	-1.76 (19)	C1—N1—C8—C9	179.88 (12)
N1—C1—C2—C3	178.74 (11)	O2—C8—C9—C10	174.72 (15)
C1—C2—C3—C4	0.6 (2)	N1—C8—C9—C10	-6.52 (16)
C7—O1—C4—C3	10.8 (2)	C8—C9—C10—C11	12.49 (15)
C7—O1—C4—C5	-170.72 (13)	C8—N1—C11—O3	-168.93 (13)
C2—C3—C4—O1	179.31 (12)	C1—N1—C11—O3	8.5 (2)
C2—C3—C4—C5	0.8 (2)	C8—N1—C11—C10	11.02 (15)
O1—C4—C5—C6	-179.76 (12)	C1—N1—C11—C10	-171.59 (11)
C3—C4—C5—C6	-1.2 (2)	C9—C10—C11—O3	165.50 (14)
C4—C5—C6—C1	0.0 (2)	C9—C10—C11—N1	-14.44 (14)
C2—C1—C6—C5	1.44 (19)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C2—H2...O3 ⁱ	0.93	2.50	3.1666 (17)	129
C5—H5...O2 ⁱⁱ	0.93	2.47	3.3245 (17)	152

Symmetry codes: (i) *x*, *y*-1, *z*; (ii) -*x*+3/2, *y*+1/2, -*z*+1/2.