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(E)-1-[2-(2-Nitrostyryl)-1-phenylsulfonyl]-1H-indol-3-yl]propan-1-oneJ. Kanchanadevi,^a G. Anbalagan,^b V. Saravanan,^c
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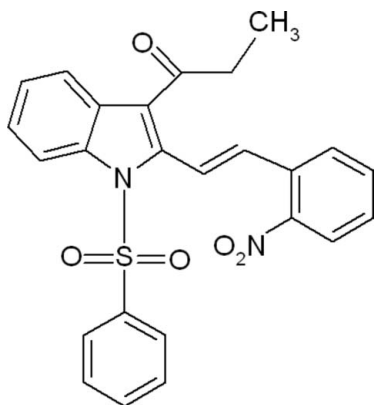
Received 3 January 2014; accepted 8 January 2014

Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.051; wR factor = 0.151; data-to-parameter ratio = 17.9.

In the title compound, $\text{C}_{25}\text{H}_{20}\text{N}_2\text{O}_5\text{S}$, the phenyl ring makes dihedral angles of 89.88 (8) and 13.98 (8)°, respectively, with the indole ring system and the nitrobenzene ring. The dihedral angle between the indole ring system and the nitrobenzene ring is 88.48 (11)°. The molecular structure is stabilized by a weak intramolecular C—H...O interaction. In the crystal, π – π interactions, with centroid–centroid distances of 3.6741 (18) and 3.8873 (17) Å, link the molecules into layers parallel to the *ab* plane.

Related literature

For biological activity of indole derivatives, see: Andreani *et al.* (2001); Quetin-Leclercq (1994); Mukhopadhyay *et al.* (1981); Singh *et al.* (2000). For related structures, see: Umadevi *et al.* (2013); Kanchanadevi *et al.* (2014).



Experimental

Crystal data

$\text{C}_{25}\text{H}_{20}\text{N}_2\text{O}_5\text{S}$
 $M_r = 460.49$
Monoclinic, $C2/c$
 $a = 21.6100$ (18) Å
 $b = 8.3072$ (7) Å
 $c = 25.898$ (3) Å
 $\beta = 112.379$ (2)°
 $V = 4299.1$ (7) Å³
 $Z = 8$
Mo $K\alpha$ radiation
 $\mu = 0.19$ mm⁻¹
 $T = 295$ K
0.30 × 0.25 × 0.25 mm

Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.945$, $T_{\max} = 0.955$
40633 measured reflections
5343 independent reflections
3700 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.044$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.151$
 $S = 1.04$
5343 reflections
299 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.66$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.67$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
C2—H2...O1	0.93	2.32	2.912 (3)	121

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: SHELXL97.

Supporting information for this paper is available from the IUCr electronic archives (Reference: IS5331).

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supporting information

Acta Cryst. (2014). E70, o148 [doi:10.1107/S1600536814000506]

(*E*)-1-[2-(2-Nitrostyryl)-1-phenylsulfonyl-1*H*-indol-3-yl]propan-1-one

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S1. Comment

Indole derivatives exhibits antibacterial, antifungal (Singh *et al.*, 2000) and antitumour activities (Andreani *et al.*, 2001). Some of the indole alkaloids extracted from plants possess interesting cytotoxic and antiparasitic properties (Quetin-Leclercq, 1994; Mukhopadhyay *et al.*, 1981).

The geometric parameters of the title molecule (Fig. 1) agree well with reported similar structure (Umadevi *et al.*, 2013). The phenyl ring makes a dihedral angles of 89.88 (8) and 13.98 (8)° with the indole ring system and the nitro benzene ring, respectively. The sum of bond angles around the atom N1 [356.03°] indicates sp^2 hybridized. The molecular structure is stabilized by a weak intramolecular C—H \cdots O interaction.

S2. Experimental

A solution of 1-[2-(bromomethyl)-1-(phenylsulfonyl)-1*H*-indol-3-yl] propan-1-one (5 g, 12.31 mmol) and triphenylphosphine (3.5 g, 13.54 mmol) in dry THF (100 ml) was refluxed for 6 h. After consumption of the starting material, the solvent was removed under vacuum and the solid was washed with diethyl ether to give the phosphonium salt. Then, the mixture of phosphonium salt (8 g, 11.97 mmol), 2-nitrobenzaldehydes (1.99 g, 13.17 mmol) and K_2CO_3 (3.30 g, 23.95 mmol) in DCM (70 ml) was stirred at room temperature for 24 h. After completion of the reaction (monitored by TLC), it was diluted using DCM (30 ml), washed with water (2×100 ml) and dried (Na_2SO_4). Removal of solvent followed by trituration of the crude product with MeOH (20 ml) afforded (*E*)-1-[2-(2-nitrostyryl)-1-(phenylsulfonyl)-1*H*-indol-3-yl]propan-1-one as yellow solid (4.30 g, 76%) with melting point 168–170 °C.

S3. Refinement

H atoms were positioned geometrically and refined using riding model with C—H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic CH, C—H = 0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for CH_2 , and C—H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$ for CH_3 .

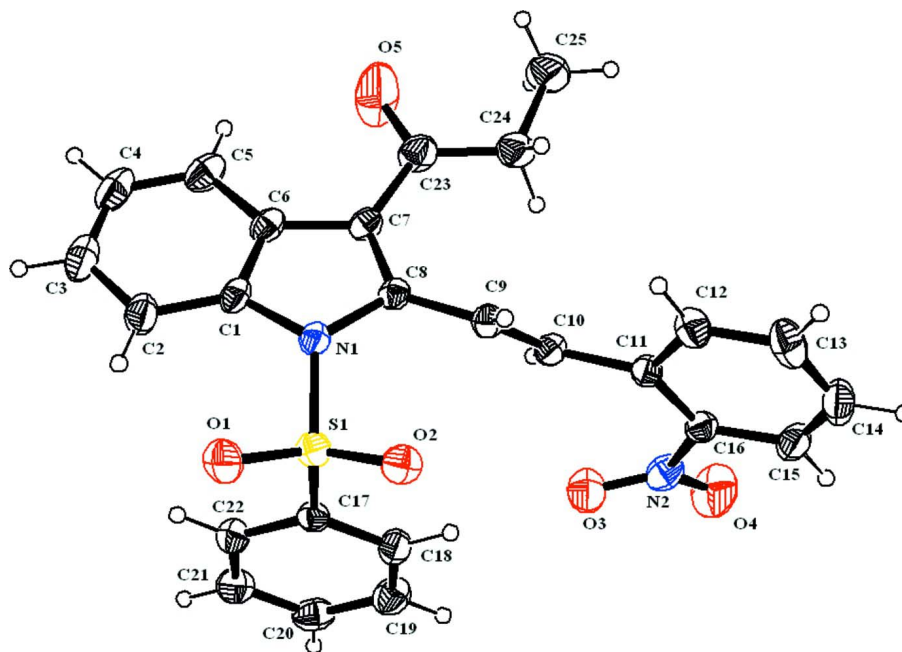


Figure 1

The molecular structure of the title compound, with atom labels and 30% probability displacement ellipsoids for non-H atoms.

(E)-1-[2-(2-Nitrostyryl)-1-phenylsulfonyl-1H-indol-3-yl]propan-1-one

Crystal data

$C_{25}H_{20}N_2O_5S$

$M_r = 460.49$

Monoclinic, $C2/c$

Hall symbol: $-C\ 2yc$

$a = 21.6100$ (18) Å

$b = 8.3072$ (7) Å

$c = 25.898$ (3) Å

$\beta = 112.379$ (2)°

$V = 4299.1$ (7) Å³

$Z = 8$

$F(000) = 1920$

$D_x = 1.423$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 5350 reflections

$\theta = 2.0$ – 28.3 °

$\mu = 0.19$ mm⁻¹

$T = 295$ K

Block, yellow

$0.30 \times 0.25 \times 0.25$ mm

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 0 pixels mm⁻¹

ω and φ scans

Absorption correction: multi-scan

(*SADABS*; Sheldrick, 1996)

$T_{\min} = 0.945$, $T_{\max} = 0.955$

40633 measured reflections

5343 independent reflections

3700 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.044$

$\theta_{\max} = 28.3$ °, $\theta_{\min} = 2.0$ °

$h = -28 \rightarrow 28$

$k = -11 \rightarrow 11$

$l = -32 \rightarrow 34$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.051$ $wR(F^2) = 0.151$ $S = 1.04$

5343 reflections

299 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0587P)^2 + 6.0066P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.66 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.67 \text{ e } \text{\AA}^{-3}$ *Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.15190 (11)	-0.0496 (3)	0.50378 (9)	0.0411 (5)
C2	0.08840 (13)	-0.1098 (3)	0.47316 (10)	0.0544 (7)
H2	0.0544	-0.1071	0.4869	0.065*
C3	0.07820 (16)	-0.1738 (3)	0.42139 (11)	0.0649 (8)
H3	0.0364	-0.2157	0.3997	0.078*
C4	0.12856 (17)	-0.1769 (3)	0.40107 (11)	0.0647 (8)
H4	0.1199	-0.2207	0.3659	0.078*
C5	0.19136 (15)	-0.1172 (3)	0.43131 (10)	0.0561 (7)
H5	0.2249	-0.1208	0.4171	0.067*
C6	0.20352 (12)	-0.0506 (3)	0.48417 (9)	0.0433 (5)
C7	0.26149 (11)	0.0260 (3)	0.52583 (9)	0.0400 (5)
C8	0.24458 (10)	0.0690 (3)	0.56984 (8)	0.0356 (5)
C9	0.28201 (11)	0.1582 (3)	0.62103 (9)	0.0384 (5)
H9	0.2654	0.2575	0.6262	0.046*
C10	0.33832 (11)	0.1058 (3)	0.66058 (8)	0.0384 (5)
H10	0.3533	0.0032	0.6568	0.046*
C11	0.37820 (10)	0.2010 (3)	0.71009 (9)	0.0374 (5)
C12	0.37603 (13)	0.3685 (3)	0.70753 (11)	0.0503 (6)
H12	0.3482	0.4178	0.6746	0.060*
C13	0.41344 (15)	0.4635 (4)	0.75177 (14)	0.0651 (8)
H13	0.4098	0.5749	0.7487	0.078*
C14	0.45614 (14)	0.3952 (4)	0.80052 (12)	0.0661 (8)
H14	0.4814	0.4598	0.8304	0.079*
C15	0.46118 (13)	0.2320 (4)	0.80475 (10)	0.0566 (7)
H15	0.4906	0.1847	0.8374	0.068*
C16	0.42243 (11)	0.1363 (3)	0.76039 (9)	0.0415 (5)
C17	0.16595 (11)	-0.1823 (3)	0.63322 (9)	0.0372 (5)
C18	0.22218 (12)	-0.1958 (3)	0.68196 (10)	0.0479 (6)
H18	0.2470	-0.1051	0.6987	0.057*
C19	0.24071 (14)	-0.3460 (3)	0.70527 (11)	0.0578 (7)
H19	0.2782	-0.3571	0.7381	0.069*
C20	0.20406 (15)	-0.4794 (3)	0.68021 (11)	0.0549 (7)
H20	0.2169	-0.5802	0.6964	0.066*
C21	0.14871 (15)	-0.4656 (3)	0.63161 (11)	0.0548 (7)

H21	0.1243	-0.5567	0.6149	0.066*
C22	0.12933 (12)	-0.3159 (3)	0.60758 (10)	0.0473 (6)
H22	0.0920	-0.3055	0.5745	0.057*
C23	0.32263 (14)	0.0616 (4)	0.51526 (11)	0.0565 (7)
C25	0.43883 (15)	0.1576 (5)	0.53443 (13)	0.0788 (10)
H25A	0.4556	0.0538	0.5299	0.118*
H25B	0.4744	0.2201	0.5605	0.118*
H25C	0.4216	0.2120	0.4991	0.118*
N1	0.17663 (9)	0.0287 (2)	0.55666 (7)	0.0387 (4)
N2	0.42953 (12)	-0.0372 (3)	0.76957 (9)	0.0551 (6)
O1	0.07086 (8)	0.0081 (2)	0.57260 (8)	0.0582 (5)
O2	0.16992 (9)	0.1277 (2)	0.64547 (7)	0.0495 (4)
O3	0.38239 (11)	-0.1241 (2)	0.74363 (8)	0.0685 (6)
O4	0.48249 (12)	-0.0875 (3)	0.80380 (10)	0.0950 (8)
O5	0.32135 (15)	0.0336 (6)	0.46959 (12)	0.1606 (19)
C24	0.38311 (13)	0.1372 (4)	0.55629 (11)	0.0596 (7)
H24A	0.3994	0.0716	0.5898	0.072*
H24B	0.3714	0.2420	0.5664	0.072*
S1	0.14097 (3)	0.00862 (7)	0.60389 (2)	0.04073 (16)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0433 (12)	0.0396 (12)	0.0310 (10)	0.0040 (9)	0.0034 (9)	0.0030 (9)
C2	0.0478 (14)	0.0557 (15)	0.0461 (13)	-0.0036 (12)	0.0025 (11)	0.0019 (12)
C3	0.0667 (18)	0.0546 (16)	0.0486 (15)	-0.0027 (14)	-0.0058 (14)	-0.0016 (13)
C4	0.089 (2)	0.0498 (15)	0.0366 (13)	0.0053 (15)	0.0028 (14)	-0.0063 (11)
C5	0.0748 (18)	0.0494 (14)	0.0405 (13)	0.0120 (13)	0.0180 (13)	-0.0017 (11)
C6	0.0512 (13)	0.0389 (12)	0.0339 (11)	0.0084 (10)	0.0095 (10)	0.0037 (9)
C7	0.0407 (12)	0.0423 (12)	0.0352 (11)	0.0064 (9)	0.0123 (9)	0.0031 (9)
C8	0.0356 (11)	0.0343 (10)	0.0332 (10)	0.0040 (8)	0.0090 (8)	0.0050 (8)
C9	0.0389 (11)	0.0376 (11)	0.0385 (11)	0.0001 (9)	0.0145 (9)	-0.0019 (9)
C10	0.0428 (12)	0.0363 (11)	0.0347 (10)	0.0010 (9)	0.0131 (9)	0.0001 (9)
C11	0.0344 (10)	0.0423 (12)	0.0367 (11)	-0.0020 (9)	0.0147 (9)	-0.0035 (9)
C12	0.0466 (13)	0.0431 (13)	0.0569 (15)	-0.0023 (11)	0.0149 (11)	-0.0004 (11)
C13	0.0616 (18)	0.0473 (15)	0.086 (2)	-0.0141 (13)	0.0275 (16)	-0.0191 (14)
C14	0.0574 (17)	0.078 (2)	0.0618 (17)	-0.0217 (15)	0.0214 (14)	-0.0309 (16)
C15	0.0452 (14)	0.085 (2)	0.0363 (12)	-0.0074 (13)	0.0113 (11)	-0.0076 (13)
C16	0.0363 (11)	0.0537 (14)	0.0349 (11)	-0.0011 (10)	0.0141 (9)	-0.0005 (10)
C17	0.0365 (11)	0.0407 (11)	0.0379 (11)	-0.0014 (9)	0.0182 (9)	-0.0015 (9)
C18	0.0489 (13)	0.0472 (13)	0.0415 (12)	-0.0043 (11)	0.0105 (11)	0.0003 (10)
C19	0.0615 (16)	0.0576 (16)	0.0476 (14)	0.0064 (13)	0.0131 (12)	0.0084 (12)
C20	0.0778 (19)	0.0422 (13)	0.0542 (15)	0.0059 (13)	0.0358 (14)	0.0061 (11)
C21	0.0716 (18)	0.0438 (14)	0.0562 (15)	-0.0095 (12)	0.0324 (14)	-0.0099 (11)
C22	0.0475 (13)	0.0507 (14)	0.0432 (13)	-0.0071 (11)	0.0168 (11)	-0.0063 (11)
C23	0.0539 (15)	0.0727 (18)	0.0490 (14)	0.0045 (13)	0.0263 (12)	-0.0045 (13)
C25	0.0535 (17)	0.117 (3)	0.071 (2)	-0.0062 (18)	0.0306 (15)	0.0107 (19)
N1	0.0353 (9)	0.0444 (10)	0.0335 (9)	-0.0002 (8)	0.0096 (7)	0.0003 (8)

N2	0.0550 (13)	0.0605 (14)	0.0427 (11)	0.0093 (11)	0.0106 (10)	0.0101 (10)
O1	0.0334 (9)	0.0724 (12)	0.0652 (11)	0.0094 (8)	0.0146 (8)	0.0086 (10)
O2	0.0594 (10)	0.0413 (9)	0.0533 (10)	0.0017 (8)	0.0276 (8)	-0.0066 (7)
O3	0.0716 (13)	0.0500 (11)	0.0676 (12)	-0.0019 (10)	0.0083 (10)	0.0098 (10)
O4	0.0792 (15)	0.0911 (17)	0.0785 (15)	0.0235 (13)	-0.0106 (12)	0.0267 (13)
O5	0.099 (2)	0.318 (5)	0.0942 (19)	-0.081 (3)	0.0701 (17)	-0.103 (3)
C24	0.0500 (15)	0.0769 (19)	0.0548 (15)	-0.0035 (13)	0.0232 (12)	0.0066 (14)
S1	0.0360 (3)	0.0430 (3)	0.0434 (3)	0.0043 (2)	0.0154 (2)	0.0012 (2)

Geometric parameters (Å, °)

C1—C6	1.390 (3)	C15—C16	1.386 (3)
C1—C2	1.390 (3)	C15—H15	0.9300
C1—N1	1.424 (3)	C16—N2	1.460 (3)
C2—C3	1.380 (4)	C17—C22	1.378 (3)
C2—H2	0.9300	C17—C18	1.384 (3)
C3—C4	1.378 (5)	C17—S1	1.753 (2)
C3—H3	0.9300	C18—C19	1.378 (4)
C4—C5	1.376 (4)	C18—H18	0.9300
C4—H4	0.9300	C19—C20	1.373 (4)
C5—C6	1.405 (3)	C19—H19	0.9300
C5—H5	0.9300	C20—C21	1.372 (4)
C6—C7	1.452 (3)	C20—H20	0.9300
C7—C8	1.370 (3)	C21—C22	1.382 (4)
C7—C23	1.478 (4)	C21—H21	0.9300
C8—N1	1.415 (3)	C22—H22	0.9300
C8—C9	1.464 (3)	C23—O5	1.195 (3)
C9—C10	1.331 (3)	C23—C24	1.475 (4)
C9—H9	0.9300	C25—C24	1.523 (4)
C10—C11	1.474 (3)	C25—H25A	0.9600
C10—H10	0.9300	C25—H25B	0.9600
C11—C12	1.393 (3)	C25—H25C	0.9600
C11—C16	1.397 (3)	N1—S1	1.6838 (19)
C12—C13	1.372 (4)	N2—O3	1.220 (3)
C12—H12	0.9300	N2—O4	1.224 (3)
C13—C14	1.371 (4)	O1—S1	1.4204 (17)
C13—H13	0.9300	O2—S1	1.4198 (17)
C14—C15	1.361 (4)	C24—H24A	0.9700
C14—H14	0.9300	C24—H24B	0.9700
C6—C1—C2	123.0 (2)	C11—C16—N2	121.6 (2)
C6—C1—N1	107.12 (19)	C22—C17—C18	121.2 (2)
C2—C1—N1	129.8 (2)	C22—C17—S1	119.64 (17)
C3—C2—C1	116.8 (3)	C18—C17—S1	119.14 (17)
C3—C2—H2	121.6	C19—C18—C17	118.7 (2)
C1—C2—H2	121.6	C19—C18—H18	120.6
C4—C3—C2	121.4 (3)	C17—C18—H18	120.6
C4—C3—H3	119.3	C20—C19—C18	120.3 (2)

C2—C3—H3	119.3	C20—C19—H19	119.8
C5—C4—C3	121.8 (3)	C18—C19—H19	119.8
C5—C4—H4	119.1	C21—C20—C19	120.7 (2)
C3—C4—H4	119.1	C21—C20—H20	119.6
C4—C5—C6	118.3 (3)	C19—C20—H20	119.6
C4—C5—H5	120.8	C20—C21—C22	119.8 (2)
C6—C5—H5	120.8	C20—C21—H21	120.1
C1—C6—C5	118.6 (2)	C22—C21—H21	120.1
C1—C6—C7	108.1 (2)	C17—C22—C21	119.2 (2)
C5—C6—C7	133.2 (2)	C17—C22—H22	120.4
C8—C7—C6	107.7 (2)	C21—C22—H22	120.4
C8—C7—C23	130.0 (2)	O5—C23—C24	118.4 (3)
C6—C7—C23	122.0 (2)	O5—C23—C7	117.3 (3)
C7—C8—N1	108.57 (18)	C24—C23—C7	124.2 (2)
C7—C8—C9	130.9 (2)	C24—C25—H25A	109.5
N1—C8—C9	120.19 (19)	C24—C25—H25B	109.5
C10—C9—C8	124.0 (2)	H25A—C25—H25B	109.5
C10—C9—H9	118.0	C24—C25—H25C	109.5
C8—C9—H9	118.0	H25A—C25—H25C	109.5
C9—C10—C11	123.7 (2)	H25B—C25—H25C	109.5
C9—C10—H10	118.2	C8—N1—C1	108.36 (18)
C11—C10—H10	118.2	C8—N1—S1	124.61 (14)
C12—C11—C16	115.3 (2)	C1—N1—S1	123.06 (16)
C12—C11—C10	119.8 (2)	O3—N2—O4	123.5 (3)
C16—C11—C10	124.8 (2)	O3—N2—C16	118.7 (2)
C13—C12—C11	122.4 (3)	O4—N2—C16	117.8 (2)
C13—C12—H12	118.8	C23—C24—C25	112.9 (2)
C11—C12—H12	118.8	C23—C24—H24A	109.0
C14—C13—C12	120.4 (3)	C25—C24—H24A	109.0
C14—C13—H13	119.8	C23—C24—H24B	109.0
C12—C13—H13	119.8	C25—C24—H24B	109.0
C15—C14—C13	119.4 (3)	H24A—C24—H24B	107.8
C15—C14—H14	120.3	O2—S1—O1	120.06 (11)
C13—C14—H14	120.3	O2—S1—N1	106.89 (10)
C14—C15—C16	120.0 (3)	O1—S1—N1	105.65 (10)
C14—C15—H15	120.0	O2—S1—C17	109.13 (10)
C16—C15—H15	120.0	O1—S1—C17	109.28 (11)
C15—C16—C11	122.4 (2)	N1—S1—C17	104.71 (10)
C15—C16—N2	116.0 (2)		
C6—C1—C2—C3	0.6 (4)	C17—C18—C19—C20	0.4 (4)
N1—C1—C2—C3	177.1 (2)	C18—C19—C20—C21	0.2 (4)
C1—C2—C3—C4	-0.3 (4)	C19—C20—C21—C22	-0.2 (4)
C2—C3—C4—C5	0.2 (4)	C18—C17—C22—C21	1.0 (4)
C3—C4—C5—C6	-0.4 (4)	S1—C17—C22—C21	-178.78 (19)
C2—C1—C6—C5	-0.8 (4)	C20—C21—C22—C17	-0.4 (4)
N1—C1—C6—C5	-178.0 (2)	C8—C7—C23—O5	-167.9 (4)
C2—C1—C6—C7	178.1 (2)	C6—C7—C23—O5	4.3 (5)

N1—C1—C6—C7	0.9 (2)	C8—C7—C23—C24	8.5 (4)
C4—C5—C6—C1	0.7 (4)	C6—C7—C23—C24	-179.2 (3)
C4—C5—C6—C7	-177.9 (2)	C7—C8—N1—C1	3.1 (2)
C1—C6—C7—C8	1.0 (3)	C9—C8—N1—C1	177.08 (18)
C5—C6—C7—C8	179.7 (2)	C7—C8—N1—S1	161.12 (16)
C1—C6—C7—C23	-172.8 (2)	C9—C8—N1—S1	-24.9 (3)
C5—C6—C7—C23	5.9 (4)	C6—C1—N1—C8	-2.4 (2)
C6—C7—C8—N1	-2.5 (2)	C2—C1—N1—C8	-179.3 (2)
C23—C7—C8—N1	170.6 (2)	C6—C1—N1—S1	-160.86 (16)
C6—C7—C8—C9	-175.6 (2)	C2—C1—N1—S1	22.2 (3)
C23—C7—C8—C9	-2.5 (4)	C15—C16—N2—O3	152.5 (2)
C7—C8—C9—C10	-64.8 (3)	C11—C16—N2—O3	-26.4 (4)
N1—C8—C9—C10	122.7 (2)	C15—C16—N2—O4	-26.3 (4)
C8—C9—C10—C11	175.6 (2)	C11—C16—N2—O4	154.8 (2)
C9—C10—C11—C12	-27.0 (3)	O5—C23—C24—C25	-4.3 (5)
C9—C10—C11—C16	156.7 (2)	C7—C23—C24—C25	179.3 (3)
C16—C11—C12—C13	-1.6 (4)	C8—N1—S1—O2	36.9 (2)
C10—C11—C12—C13	-178.2 (2)	C1—N1—S1—O2	-168.17 (17)
C11—C12—C13—C14	1.4 (4)	C8—N1—S1—O1	165.85 (18)
C12—C13—C14—C15	0.0 (5)	C1—N1—S1—O1	-39.2 (2)
C13—C14—C15—C16	-1.2 (4)	C8—N1—S1—C17	-78.81 (19)
C14—C15—C16—C11	1.1 (4)	C1—N1—S1—C17	76.13 (19)
C14—C15—C16—N2	-177.9 (2)	C22—C17—S1—O2	159.83 (18)
C12—C11—C16—C15	0.3 (3)	C18—C17—S1—O2	-20.0 (2)
C10—C11—C16—C15	176.8 (2)	C22—C17—S1—O1	26.7 (2)
C12—C11—C16—N2	179.2 (2)	C18—C17—S1—O1	-153.07 (19)
C10—C11—C16—N2	-4.4 (3)	C22—C17—S1—N1	-86.0 (2)
C22—C17—C18—C19	-1.0 (4)	C18—C17—S1—N1	94.2 (2)
S1—C17—C18—C19	178.8 (2)		

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C2—H2 \cdots O1	0.93	2.32	2.912 (3)	121