

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

An amide cyclophane

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Received 19 May 2014; accepted 4 July 2014

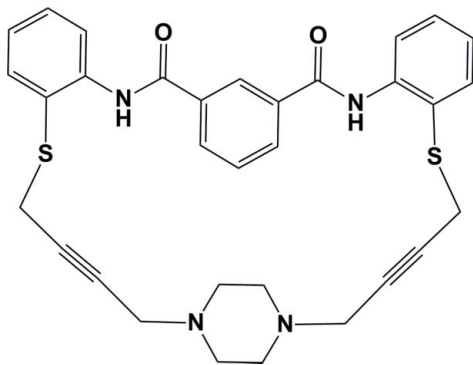
Edited by H. Stoeckli-Evans, University of Neuchâtel, Switzerland

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.042; wR factor = 0.117; data-to-parameter ratio = 19.3.

The title compound, 8,18-dithia-2,6-diaza-13(1,4)-piperidina-1(1,2),4(1,3),7(1,2)-tribenzenaoctadecaphane-10,15-diyne-3,6-dione, $\text{C}_{32}\text{H}_{30}\text{N}_4\text{O}_2\text{S}_2$, is composed of a relatively planar bis(2-mercaptophenyl)isophthalamide unit linked to a bridging 1,4-di(but-2-yn-1-yl)piperazine unit, forming a macrocycle. The isophthalamide ring is inclined to the outer mercaptophenyl rings by 8.18 (11) and 5.59 (10)°, while these two rings are inclined to one another by 9.10 (12)°. The piperazine ring adopts a chair conformation. There are two intramolecular $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds generating $S(5)$ ring motifs. In the crystal, molecules are linked *via* $\text{C}-\text{H}\cdots\text{S}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds, forming slabs lying parallel to (001). An O atom in the isophthalamide group is disordered over two positions with an occupancy ratio of 0.41 (6):0.59 (6).

Related literature

For the biological activity of piperazine derivatives, see: Fun *et al.* (2011); Kavitha *et al.* (2013). For standard bond lengths, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{32}\text{H}_{30}\text{N}_4\text{O}_2\text{S}_2$
 $M_r = 566.72$
 Monoclinic, $C2/c$
 $a = 23.0760$ (5) Å
 $b = 9.9380$ (3) Å
 $c = 27.0341$ (7) Å
 $\beta = 108.428$ (2)°

$V = 5881.8$ (3) Å³
 $Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.22$ mm⁻¹
 $T = 293$ K
 $0.30 \times 0.25 \times 0.20$ mm

Data collection

Bruker SMART APEXII area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2008)
 $T_{\min} = 0.380$, $T_{\max} = 0.745$

27488 measured reflections
 7294 independent reflections
 4040 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.036$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.117$
 $S = 1.01$
 7294 reflections
 377 parameters
 2 restraints

H atoms treated by a mixture of independent and constrained refinement

$\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N}2-\text{H}2A\cdots\text{S}1$	0.83 (2)	2.46 (2)	2.9621 (18)	120 (2)
$\text{N}3-\text{H}3A\cdots\text{S}2$	0.83 (2)	2.49 (2)	2.9905 (19)	120 (2)
$\text{C}21-\text{H}21B\cdots\text{O}2A^i$	0.97	2.57	3.43 (3)	148
$\text{C}21-\text{H}21B\cdots\text{O}2B^i$	0.97	2.34	3.230 (11)	152
$\text{C}24-\text{H}24A\cdots\text{O}1^{\text{ii}}$	0.97	2.54	3.505 (3)	171
$\text{C}26-\text{H}26B\cdots\text{S}2^{\text{ii}}$	0.97	2.85	3.632 (2)	139
$\text{C}32-\text{H}32B\cdots\text{O}1^{\text{iii}}$	0.97	2.49	3.146 (3)	125

Symmetry codes: (i) $x - \frac{1}{2}, y - \frac{1}{2}, z$; (ii) $x, y - 1, z$; (iii) $-x + 1, y, -z + \frac{3}{2}$.

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2009).

The authors thank the TBI X-ray facility, CAS in Crystallography and Biophysics, University of Madras, India, for the data collection.

Supporting information for this paper is available from the IUCr electronic archives (Reference: SU2737).

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supporting information

Acta Cryst. (2014). E70, o865 [doi:10.1107/S1600536814015621]

An amide cyclophane

Vijayan Viswanathan, Ayyavu Thirunarayanan, Perumal Rajakumar and Devadasan Velmurugan

S1. Comment

Piperazine derivatives are popular in organic synthesis and for their applications in biology and medicine. They are known to exhibit antibacterial and antimicrobial activities (Fun *et al.*, 2011). They are also amongst the most important building blocks found in biologically active compounds in a number of different therapeutic areas and a review about the current pharmacological and toxicological information for piperazine derivatives is available (Kavitha *et al.*, 2013).

The molecular structure of the title molecule is shown in Fig. 1. It is composed of a relatively planar bis(2-mercaptophenyl)isophthalamide moiety linked to a bridging 1,4-di(but-2-yn-1-yl)piperazine unit forming a macrocycle. The isophthalamide ring (C8-C13) is inclined to the outer mercaptophenyl rings (C1-C6 and C15-C20) by 8.18 (11) and 5.59 (10) °, respectively. The two mercaptophenyl rings (C1-C6 and C15-C20) are inclined to one another by 9.10 (12) °. The S2-C1 and S1-C20 bond lengths are 1.773 (2) Å and 1.777 (2) Å, respectively, similar to the standard value of 1.769 (9) Å (Allen *et al.*, 1987). There are two intramolecular N—H···S hydrogen bonds generating S(5) ring motifs (Fig. 1 and Table 1).

The but-2-yne group is linear and connects the piperazine group to the bis(2-mercaptophenyl)isophthalamide moiety. The piperazine ring (N1/N4/C25-C28) adopts a chair conformation.

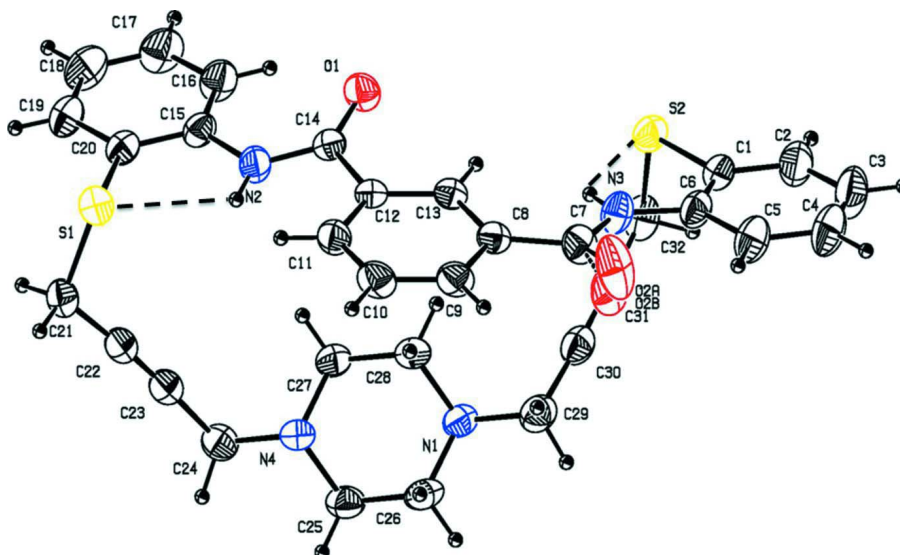
In the crystal, molecules are linked via C—H···S and C—H···O hydrogen bonds forming slabs lying parallel to (001); see Table 1 and Fig. 2.

S2. Experimental

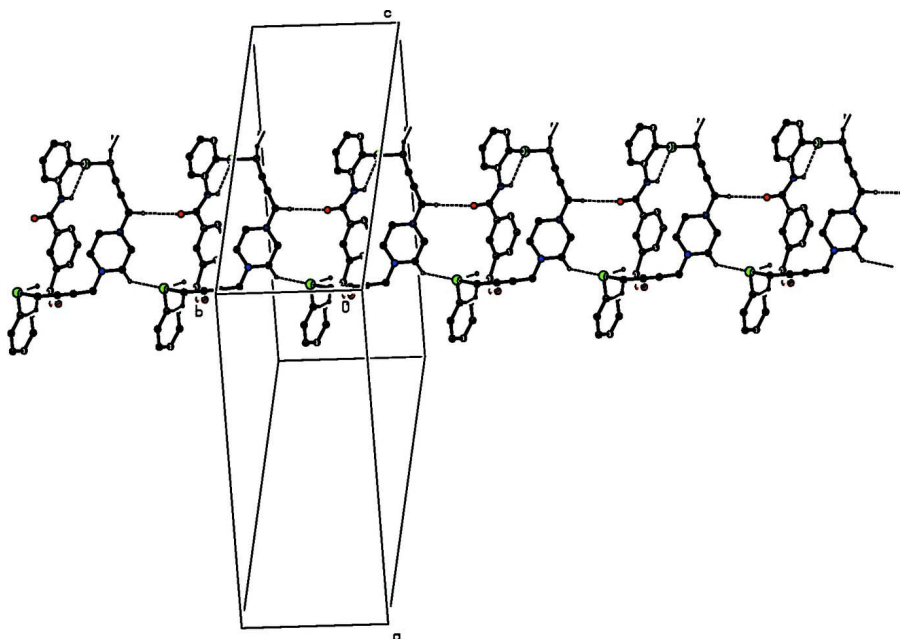
A mixture of precyclophane diyne (0.2 g, 3.98 mmol), piperazine (0.04 g, 3.98 mmol), formaldehyde (0.02 g, 7.96 mmol) from 37-41% formalin solution and CuCl (0.04 g, 3.98 mmol) in dioxane (30 mL) was refluxed at 363 K for 2 h under a nitrogen atmosphere by the Multi Components Reaction (MCR) technique. After the reaction was complete, the solvent was removed under reduced pressure and the residue was extracted with CHCl₃ (3 × 100 mL), washed with water (2 × 100 mL), brine (150 mL) and dried over anhydrous Na₂SO₄. The solvent was removed and the crude product was purified by column chromatography on silica gel using CHCl₃/MeOH (24:1) as eluent. After purification the piperazinophane was recrystallised in MeOH by slow evaporation yielding block-like colourless crystals.

S3. Refinement

N-bound H atoms were refined with distance restraints: N-H = 0.86 (2) Å with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$. The C-bound H atoms were positioned geometrically (C-H = 0.93–0.97 Å) and allowed to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C-methyl})$ and $= 1.2U_{\text{eq}}(\text{C})$ for other H atoms. In the isophthalamide group atom O2 is disordered over two positions with an occupancy ratio of 0.41 (6):0.59 (6).

**Figure 1**

The molecular structure of the title molecule, with atom labelling. Displacement ellipsoids are drawn at the 50% probability level. The intramolecular N-H...S hydrogen bonds are shown as dashed lines (see Table 1 for details).

**Figure 2**

A partial view of the crystal packing of the title compound. Hydrogen bonds are shown as dashed lines (see Table 1 for details; H-atoms not involved in hydrogen bonding have been omitted for clarity).

8,18-Dithia-2,6-diaza-13(1,4)-piperidina-1(1,2),4(1,3),7(1,2)-tribenzenaoctadecaphane-10,15-diyne-3,6-dione

Crystal data

$C_{32}H_{30}N_4O_2S_2$

$M_r = 566.72$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 23.0760 (5) \text{ \AA}$

$b = 9.9380 (3) \text{ \AA}$

$c = 27.0341$ (7) Å
 $\beta = 108.428$ (2)°
 $V = 5881.8$ (3) Å³
 $Z = 8$
 $F(000) = 2384$
 $D_x = 1.280$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 7294 reflections
 $\theta = 1.6$ – 28.3 °
 $\mu = 0.22$ mm⁻¹
 $T = 293$ K
 Block, colourless
 $0.30 \times 0.25 \times 0.20$ mm

Data collection

Bruker SMART APEXII area-detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 ω and ϕ scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2008)
 $T_{\min} = 0.380$, $T_{\max} = 0.745$

27488 measured reflections
 7294 independent reflections
 4040 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.036$
 $\theta_{\max} = 28.3$ °, $\theta_{\min} = 1.6$ °
 $h = -30 \rightarrow 30$
 $k = -13 \rightarrow 13$
 $l = -34 \rightarrow 35$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.117$
 $S = 1.01$
 7294 reflections
 377 parameters
 2 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0438P)^2 + 2.0132P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.20$ e Å⁻³
 $\Delta\rho_{\min} = -0.23$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.41059 (2)	0.21672 (6)	1.01916 (2)	0.06329 (16)	
S2	0.58670 (2)	0.64936 (6)	0.76454 (2)	0.06810 (17)	
O1	0.47117 (6)	0.54558 (14)	0.89655 (5)	0.0671 (4)	
O2A	0.7513 (7)	0.440 (3)	0.9253 (10)	0.108 (5)	0.41 (6)
O2B	0.7544 (4)	0.482 (3)	0.9314 (5)	0.115 (4)	0.59 (6)
N1	0.57053 (7)	0.08419 (16)	0.81521 (6)	0.0584 (4)	
N2	0.45152 (7)	0.37111 (18)	0.94254 (7)	0.0634 (5)	
H2A	0.4674 (9)	0.3112 (19)	0.9642 (8)	0.076*	
N3	0.67548 (7)	0.5408 (2)	0.86248 (7)	0.0650 (5)	

H3A	0.6377 (7)	0.544 (2)	0.8508 (8)	0.078*
N4	0.50216 (7)	−0.03836 (15)	0.87441 (6)	0.0541 (4)
C1	0.66745 (9)	0.6485 (2)	0.78117 (9)	0.0625 (5)
C2	0.69417 (11)	0.7074 (3)	0.74746 (11)	0.0866 (7)
H2	0.6695	0.7458	0.7166	0.104*
C3	0.75672 (13)	0.7100 (3)	0.75881 (13)	0.1039 (9)
H3	0.7744	0.7507	0.7361	0.125*
C4	0.79235 (11)	0.6519 (3)	0.80401 (13)	0.0995 (9)
H4	0.8346	0.6519	0.8115	0.119*
C5	0.76738 (10)	0.5931 (3)	0.83875 (10)	0.0825 (7)
H5	0.7925	0.5540	0.8693	0.099*
C6	0.70420 (8)	0.5926 (2)	0.82788 (9)	0.0618 (5)
C7	0.69931 (9)	0.4820 (2)	0.90941 (9)	0.0660 (6)
C8	0.65534 (8)	0.43753 (18)	0.93660 (8)	0.0536 (5)
C9	0.67931 (9)	0.3618 (2)	0.98150 (9)	0.0657 (6)
H9	0.7211	0.3448	0.9937	0.079*
C10	0.64258 (10)	0.3116 (2)	1.00816 (9)	0.0709 (6)
H10	0.6594	0.2596	1.0379	0.085*
C11	0.58065 (9)	0.3380 (2)	0.99107 (8)	0.0622 (5)
H11	0.5558	0.3034	1.0092	0.075*
C12	0.55554 (8)	0.41599 (17)	0.94693 (7)	0.0484 (4)
C13	0.59316 (8)	0.46595 (18)	0.92026 (7)	0.0497 (4)
H13	0.5766	0.5194	0.8909	0.060*
C14	0.48890 (8)	0.45151 (19)	0.92630 (7)	0.0517 (4)
C15	0.38767 (8)	0.3776 (2)	0.93183 (8)	0.0578 (5)
C16	0.34978 (10)	0.4520 (2)	0.89092 (9)	0.0770 (7)
H16	0.3663	0.5021	0.8696	0.092*
C17	0.28782 (11)	0.4512 (3)	0.88210 (11)	0.0917 (8)
H17	0.2625	0.5006	0.8545	0.110*
C18	0.26290 (11)	0.3792 (3)	0.91326 (12)	0.0942 (8)
H18	0.2209	0.3793	0.9068	0.113*
C19	0.30033 (10)	0.3061 (2)	0.95441 (10)	0.0763 (6)
H19	0.2834	0.2577	0.9759	0.092*
C20	0.36274 (8)	0.3043 (2)	0.96403 (8)	0.0575 (5)
C21	0.39422 (9)	0.0419 (2)	1.00030 (8)	0.0623 (5)
H21A	0.4140	−0.0144	1.0302	0.075*
H21B	0.3505	0.0277	0.9915	0.075*
C22	0.41344 (9)	−0.0032 (2)	0.95642 (9)	0.0573 (5)
C23	0.42885 (9)	−0.0462 (2)	0.92174 (9)	0.0590 (5)
C24	0.44899 (10)	−0.1063 (2)	0.88040 (8)	0.0651 (5)
H24A	0.4587	−0.2003	0.8884	0.078*
H24B	0.4158	−0.1020	0.8477	0.078*
C25	0.53008 (10)	−0.1175 (2)	0.84244 (9)	0.0685 (6)
H25A	0.5012	−0.1287	0.8077	0.082*
H25B	0.5408	−0.2061	0.8577	0.082*
C26	0.58637 (10)	−0.0470 (2)	0.83922 (9)	0.0695 (6)
H26A	0.6153	−0.0366	0.8739	0.083*
H26B	0.6056	−0.1007	0.8188	0.083*

C27	0.48647 (9)	0.0944 (2)	0.85116 (9)	0.0638 (5)
H27A	0.4686	0.1480	0.8726	0.077*
H27B	0.4564	0.0854	0.8169	0.077*
C28	0.54223 (9)	0.1640 (2)	0.84643 (9)	0.0648 (6)
H28A	0.5309	0.2514	0.8303	0.078*
H28B	0.5713	0.1779	0.8809	0.078*
C29	0.62297 (9)	0.1516 (2)	0.80817 (9)	0.0725 (6)
H29A	0.6431	0.0916	0.7905	0.087*
H29B	0.6518	0.1733	0.8420	0.087*
C30	0.60525 (10)	0.2749 (3)	0.77769 (9)	0.0668 (6)
C31	0.58869 (9)	0.3744 (3)	0.75342 (9)	0.0651 (6)
C32	0.56795 (10)	0.4989 (2)	0.72467 (8)	0.0748 (6)
H32A	0.5240	0.4946	0.7086	0.090*
H32B	0.5861	0.5057	0.6969	0.090*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0621 (3)	0.0687 (3)	0.0562 (3)	−0.0058 (3)	0.0146 (2)	0.0010 (3)
S2	0.0538 (3)	0.0752 (4)	0.0786 (4)	0.0148 (3)	0.0256 (3)	0.0111 (3)
O1	0.0593 (8)	0.0693 (9)	0.0674 (9)	0.0087 (7)	0.0123 (7)	0.0201 (8)
O2A	0.035 (5)	0.131 (11)	0.160 (10)	0.023 (6)	0.032 (6)	0.061 (7)
O2B	0.050 (3)	0.167 (9)	0.104 (5)	−0.038 (5)	−0.010 (3)	0.051 (5)
N1	0.0518 (9)	0.0574 (10)	0.0665 (11)	0.0045 (8)	0.0195 (8)	−0.0074 (8)
N2	0.0491 (9)	0.0689 (12)	0.0744 (12)	0.0115 (8)	0.0229 (9)	0.0232 (9)
N3	0.0378 (8)	0.0918 (13)	0.0638 (11)	−0.0020 (9)	0.0140 (8)	0.0025 (10)
N4	0.0568 (9)	0.0477 (9)	0.0555 (10)	0.0057 (7)	0.0146 (8)	−0.0030 (8)
C1	0.0536 (11)	0.0650 (13)	0.0762 (15)	0.0019 (10)	0.0308 (11)	−0.0003 (11)
C2	0.0780 (16)	0.1015 (19)	0.0924 (18)	0.0038 (14)	0.0444 (14)	0.0149 (15)
C3	0.0777 (18)	0.134 (3)	0.118 (2)	−0.0075 (17)	0.0565 (18)	0.016 (2)
C4	0.0535 (14)	0.137 (3)	0.119 (2)	−0.0141 (15)	0.0430 (16)	−0.006 (2)
C5	0.0482 (12)	0.114 (2)	0.0867 (17)	−0.0087 (13)	0.0239 (12)	−0.0074 (15)
C6	0.0466 (11)	0.0719 (13)	0.0715 (14)	−0.0064 (10)	0.0250 (10)	−0.0109 (11)
C7	0.0454 (11)	0.0712 (14)	0.0750 (15)	−0.0064 (11)	0.0098 (11)	0.0034 (12)
C8	0.0461 (10)	0.0476 (10)	0.0620 (12)	−0.0014 (8)	0.0101 (9)	−0.0022 (9)
C9	0.0483 (11)	0.0615 (13)	0.0798 (15)	0.0079 (10)	0.0094 (10)	0.0108 (11)
C10	0.0644 (13)	0.0672 (14)	0.0718 (15)	0.0120 (11)	0.0082 (11)	0.0211 (11)
C11	0.0606 (12)	0.0617 (12)	0.0640 (13)	0.0072 (10)	0.0194 (10)	0.0130 (10)
C12	0.0501 (10)	0.0419 (10)	0.0515 (11)	0.0050 (8)	0.0138 (9)	−0.0006 (8)
C13	0.0491 (10)	0.0452 (10)	0.0500 (11)	−0.0003 (8)	0.0087 (8)	−0.0014 (8)
C14	0.0541 (11)	0.0515 (11)	0.0490 (11)	0.0065 (9)	0.0156 (9)	−0.0003 (9)
C15	0.0459 (10)	0.0674 (13)	0.0605 (12)	0.0093 (9)	0.0176 (9)	0.0024 (10)
C16	0.0611 (13)	0.0964 (17)	0.0723 (15)	0.0196 (12)	0.0192 (11)	0.0208 (13)
C17	0.0590 (14)	0.116 (2)	0.0890 (18)	0.0270 (14)	0.0080 (13)	0.0195 (17)
C18	0.0475 (13)	0.118 (2)	0.112 (2)	0.0165 (14)	0.0184 (14)	0.0008 (18)
C19	0.0549 (13)	0.0916 (17)	0.0885 (17)	0.0025 (12)	0.0313 (12)	−0.0010 (14)
C20	0.0491 (10)	0.0642 (12)	0.0596 (12)	0.0033 (9)	0.0176 (9)	−0.0051 (10)
C21	0.0569 (12)	0.0657 (13)	0.0643 (13)	−0.0067 (10)	0.0192 (10)	0.0061 (11)

C22	0.0478 (11)	0.0565 (12)	0.0624 (13)	-0.0020 (9)	0.0101 (10)	0.0099 (10)
C23	0.0534 (11)	0.0564 (12)	0.0627 (14)	-0.0017 (9)	0.0117 (10)	0.0056 (11)
C24	0.0688 (13)	0.0605 (13)	0.0627 (13)	-0.0048 (11)	0.0159 (11)	-0.0036 (11)
C25	0.0812 (15)	0.0525 (12)	0.0730 (15)	0.0038 (11)	0.0259 (12)	-0.0107 (11)
C26	0.0676 (13)	0.0595 (13)	0.0838 (16)	0.0143 (11)	0.0274 (12)	-0.0113 (12)
C27	0.0598 (12)	0.0596 (12)	0.0746 (14)	0.0140 (10)	0.0249 (11)	0.0068 (11)
C28	0.0679 (13)	0.0508 (11)	0.0794 (15)	0.0084 (10)	0.0287 (11)	-0.0017 (11)
C29	0.0562 (12)	0.0752 (15)	0.0877 (17)	0.0028 (11)	0.0252 (12)	-0.0067 (13)
C30	0.0542 (12)	0.0773 (16)	0.0728 (15)	-0.0072 (12)	0.0257 (11)	-0.0155 (13)
C31	0.0503 (12)	0.0834 (17)	0.0638 (14)	-0.0069 (12)	0.0213 (10)	-0.0138 (13)
C32	0.0618 (13)	0.1047 (18)	0.0559 (13)	-0.0025 (12)	0.0157 (11)	0.0029 (13)

Geometric parameters (Å, °)

S1—C20	1.777 (2)	C12—C13	1.384 (3)
S1—C21	1.817 (2)	C12—C14	1.503 (2)
S2—C1	1.773 (2)	C13—H13	0.9300
S2—C32	1.814 (2)	C15—C16	1.387 (3)
O1—C14	1.217 (2)	C15—C20	1.391 (3)
O2A—C7	1.215 (11)	C16—C17	1.373 (3)
O2B—C7	1.222 (9)	C16—H16	0.9300
N1—C29	1.448 (3)	C17—C18	1.363 (4)
N1—C26	1.451 (3)	C17—H17	0.9300
N1—C28	1.454 (2)	C18—C19	1.380 (3)
N2—C14	1.347 (2)	C18—H18	0.9300
N2—C15	1.411 (2)	C19—C20	1.380 (3)
N2—H2A	0.833 (15)	C19—H19	0.9300
N3—C7	1.346 (3)	C21—C22	1.462 (3)
N3—C6	1.404 (3)	C21—H21A	0.9700
N3—H3A	0.828 (15)	C21—H21B	0.9700
N4—C24	1.454 (3)	C22—C23	1.183 (3)
N4—C27	1.457 (2)	C23—C24	1.466 (3)
N4—C25	1.461 (2)	C24—H24A	0.9700
C1—C2	1.382 (3)	C24—H24B	0.9700
C1—C6	1.395 (3)	C25—C26	1.503 (3)
C2—C3	1.378 (3)	C25—H25A	0.9700
C2—H2	0.9300	C25—H25B	0.9700
C3—C4	1.367 (4)	C26—H26A	0.9700
C3—H3	0.9300	C26—H26B	0.9700
C4—C5	1.378 (4)	C27—C28	1.502 (3)
C4—H4	0.9300	C27—H27A	0.9700
C5—C6	1.393 (3)	C27—H27B	0.9700
C5—H5	0.9300	C28—H28A	0.9700
C7—C8	1.496 (3)	C28—H28B	0.9700
C8—C9	1.386 (3)	C29—C30	1.461 (3)
C8—C13	1.390 (2)	C29—H29A	0.9700
C9—C10	1.369 (3)	C29—H29B	0.9700
C9—H9	0.9300	C30—C31	1.181 (3)

C10—C11	1.381 (3)	C31—C32	1.459 (3)
C10—H10	0.9300	C32—H32A	0.9700
C11—C12	1.387 (3)	C32—H32B	0.9700
C11—H11	0.9300		
C20—S1—C21	102.37 (10)	C18—C17—C16	121.0 (2)
C1—S2—C32	100.43 (10)	C18—C17—H17	119.5
C29—N1—C26	111.83 (16)	C16—C17—H17	119.5
C29—N1—C28	111.80 (16)	C17—C18—C19	119.7 (2)
C26—N1—C28	109.05 (16)	C17—C18—H18	120.1
C14—N2—C15	129.86 (17)	C19—C18—H18	120.1
C14—N2—H2A	117.9 (15)	C18—C19—C20	120.6 (2)
C15—N2—H2A	112.1 (15)	C18—C19—H19	119.7
C7—N3—C6	130.45 (17)	C20—C19—H19	119.7
C7—N3—H3A	116.3 (16)	C19—C20—C15	119.24 (19)
C6—N3—H3A	113.1 (16)	C19—C20—S1	119.93 (17)
C24—N4—C27	111.41 (15)	C15—C20—S1	120.74 (14)
C24—N4—C25	111.20 (15)	C22—C21—S1	115.54 (14)
C27—N4—C25	109.16 (16)	C22—C21—H21A	108.4
C2—C1—C6	119.7 (2)	S1—C21—H21A	108.4
C2—C1—S2	118.89 (18)	C22—C21—H21B	108.4
C6—C1—S2	121.39 (15)	S1—C21—H21B	108.4
C3—C2—C1	121.1 (3)	H21A—C21—H21B	107.5
C3—C2—H2	119.5	C23—C22—C21	176.6 (2)
C1—C2—H2	119.5	C22—C23—C24	176.9 (2)
C4—C3—C2	118.9 (2)	N4—C24—C23	112.22 (16)
C4—C3—H3	120.6	N4—C24—H24A	109.2
C2—C3—H3	120.6	C23—C24—H24A	109.2
C3—C4—C5	121.7 (2)	N4—C24—H24B	109.2
C3—C4—H4	119.1	C23—C24—H24B	109.2
C5—C4—H4	119.1	H24A—C24—H24B	107.9
C4—C5—C6	119.5 (2)	N4—C25—C26	109.52 (16)
C4—C5—H5	120.2	N4—C25—H25A	109.8
C6—C5—H5	120.2	C26—C25—H25A	109.8
C5—C6—C1	119.1 (2)	N4—C25—H25B	109.8
C5—C6—N3	122.9 (2)	C26—C25—H25B	109.8
C1—C6—N3	117.99 (17)	H25A—C25—H25B	108.2
O2A—C7—N3	123.2 (9)	N1—C26—C25	110.17 (17)
O2B—C7—N3	120.8 (7)	N1—C26—H26A	109.6
O2A—C7—C8	118.2 (9)	C25—C26—H26A	109.6
O2B—C7—C8	121.5 (6)	N1—C26—H26B	109.6
N3—C7—C8	116.93 (17)	C25—C26—H26B	109.6
C9—C8—C13	118.33 (18)	H26A—C26—H26B	108.1
C9—C8—C7	116.48 (17)	N4—C27—C28	110.57 (16)
C13—C8—C7	125.19 (18)	N4—C27—H27A	109.5
C10—C9—C8	121.13 (19)	C28—C27—H27A	109.5
C10—C9—H9	119.4	N4—C27—H27B	109.5
C8—C9—H9	119.4	C28—C27—H27B	109.5

C9—C10—C11	120.15 (19)	H27A—C27—H27B	108.1
C9—C10—H10	119.9	N1—C28—C27	110.61 (17)
C11—C10—H10	119.9	N1—C28—H28A	109.5
C10—C11—C12	120.04 (19)	C27—C28—H28A	109.5
C10—C11—H11	120.0	N1—C28—H28B	109.5
C12—C11—H11	120.0	C27—C28—H28B	109.5
C13—C12—C11	119.24 (17)	H28A—C28—H28B	108.1
C13—C12—C14	117.33 (16)	N1—C29—C30	111.46 (17)
C11—C12—C14	123.43 (17)	N1—C29—H29A	109.3
C12—C13—C8	121.06 (17)	C30—C29—H29A	109.3
C12—C13—H13	119.5	N1—C29—H29B	109.3
C8—C13—H13	119.5	C30—C29—H29B	109.3
O1—C14—N2	123.71 (18)	H29A—C29—H29B	108.0
O1—C14—C12	121.12 (18)	C31—C30—C29	177.3 (2)
N2—C14—C12	115.16 (16)	C30—C31—C32	178.5 (2)
C16—C15—C20	119.81 (18)	C31—C32—S2	113.96 (15)
C16—C15—N2	122.85 (19)	C31—C32—H32A	108.8
C20—C15—N2	117.34 (17)	S2—C32—H32A	108.8
C17—C16—C15	119.7 (2)	C31—C32—H32B	108.8
C17—C16—H16	120.2	S2—C32—H32B	108.8
C15—C16—H16	120.2	H32A—C32—H32B	107.7
C32—S2—C1—C2	86.2 (2)	C13—C12—C14—O1	-18.3 (3)
C32—S2—C1—C6	-95.30 (19)	C11—C12—C14—O1	160.92 (19)
C6—C1—C2—C3	1.0 (4)	C13—C12—C14—N2	160.74 (17)
S2—C1—C2—C3	179.6 (2)	C11—C12—C14—N2	-20.0 (3)
C1—C2—C3—C4	0.7 (4)	C14—N2—C15—C16	17.0 (3)
C2—C3—C4—C5	-1.2 (5)	C14—N2—C15—C20	-163.6 (2)
C3—C4—C5—C6	-0.1 (4)	C20—C15—C16—C17	-0.8 (3)
C4—C5—C6—C1	1.8 (4)	N2—C15—C16—C17	178.5 (2)
C4—C5—C6—N3	-176.7 (2)	C15—C16—C17—C18	0.5 (4)
C2—C1—C6—C5	-2.2 (3)	C16—C17—C18—C19	0.2 (4)
S2—C1—C6—C5	179.22 (17)	C17—C18—C19—C20	-0.6 (4)
C2—C1—C6—N3	176.3 (2)	C18—C19—C20—C15	0.3 (3)
S2—C1—C6—N3	-2.2 (3)	C18—C19—C20—S1	176.77 (19)
C7—N3—C6—C5	-1.6 (4)	C16—C15—C20—C19	0.4 (3)
C7—N3—C6—C1	179.9 (2)	N2—C15—C20—C19	-178.96 (19)
C6—N3—C7—O2A	-15 (2)	C16—C15—C20—S1	-176.02 (17)
C6—N3—C7—O2B	9.3 (15)	N2—C15—C20—S1	4.6 (3)
C6—N3—C7—C8	180.0 (2)	C21—S1—C20—C19	72.16 (19)
O2A—C7—C8—C9	6 (2)	C21—S1—C20—C15	-111.40 (17)
O2B—C7—C8—C9	-17.5 (15)	C20—S1—C21—C22	66.21 (17)
N3—C7—C8—C9	171.90 (19)	C27—N4—C24—C23	-71.5 (2)
O2A—C7—C8—C13	-173 (2)	C25—N4—C24—C23	166.44 (17)
O2B—C7—C8—C13	162.6 (15)	C24—N4—C25—C26	-177.46 (17)
N3—C7—C8—C13	-7.9 (3)	C27—N4—C25—C26	59.2 (2)
C13—C8—C9—C10	2.5 (3)	C29—N1—C26—C25	-176.08 (18)
C7—C8—C9—C10	-177.4 (2)	C28—N1—C26—C25	59.8 (2)

C8—C9—C10—C11	-1.2 (3)	N4—C25—C26—N1	-60.9 (2)
C9—C10—C11—C12	-0.3 (3)	C24—N4—C27—C28	178.72 (17)
C10—C11—C12—C13	0.4 (3)	C25—N4—C27—C28	-58.1 (2)
C10—C11—C12—C14	-178.84 (18)	C29—N1—C28—C27	177.58 (17)
C11—C12—C13—C8	1.0 (3)	C26—N1—C28—C27	-58.2 (2)
C14—C12—C13—C8	-179.76 (16)	N4—C27—C28—N1	58.1 (2)
C9—C8—C13—C12	-2.4 (3)	C26—N1—C29—C30	172.24 (18)
C7—C8—C13—C12	177.49 (18)	C28—N1—C29—C30	-65.2 (2)
C15—N2—C14—O1	-2.6 (3)	C1—S2—C32—C31	66.79 (18)
C15—N2—C14—C12	178.37 (19)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2 <i>A</i> ...S1	0.83 (2)	2.46 (2)	2.9621 (18)	120 (2)
N3—H3 <i>A</i> ...S2	0.83 (2)	2.49 (2)	2.9905 (19)	120 (2)
C21—H21 <i>B</i> ...O2 <i>A</i> ⁱ	0.97	2.57	3.43 (3)	148
C21—H21 <i>B</i> ...O2 <i>B</i> ⁱ	0.97	2.34	3.230 (11)	152
C24—H24 <i>A</i> ...O1 ⁱⁱ	0.97	2.54	3.505 (3)	171
C26—H26 <i>B</i> ...S2 ⁱⁱ	0.97	2.85	3.632 (2)	139
C32—H32 <i>B</i> ...O1 ⁱⁱⁱ	0.97	2.49	3.146 (3)	125

Symmetry codes: (i) $x-1/2, y-1/2, z$; (ii) $x, y-1, z$; (iii) $-x+1, y, -z+3/2$.