

Crystal structure of *N*-[4-amino-5-cyano-6-(methylsulfanyl)pyridin-2-yl]-2-(cyclohexylsulfanyl)acetamide

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In the title molecule, $C_{15}H_{20}N_4OS_2$, the acetamido fragment is nearly coplanar with the pyridyl ring [C–N–C–C torsion angle = $-4.1(2)^\circ$], while the cyclohexylsulfanyl portion protrudes from this plane [N–C–C–S torsion angle = $-40.8(6)^\circ$]. In the crystal, alternating pairwise N–H \cdots O and N–H \cdots N hydrogen bonds across inversion centres form chains along [101], which are associated into stepped layers *via* offset π – π stacking between pyridyl rings [centroid–centroid distance = $3.566(1) \text{ \AA}$]. The cyclohexyl group and the two atoms of the S–C bond attached to it are disordered over two sets of sites with site-occupancy factors of 0.8845 (18) and 0.1155 (18).

Keywords: crystal structure; acetamido; cyclohexylsulfanyl; hydrogen bonds; π – π stacking.

CCDC reference: 1019463

1. Related literature

For the diverse biological properties of pyridine-containing compounds see: Patrick & Kinsman (1996); Hishmat *et al.* (1990); Paronikyan *et al.* (2002); Bernardino *et al.* (2007); Doshi *et al.* (1999); Jemmezi *et al.* (2014); Mamolo *et al.* (2004); Bhatt *et al.* (2001). For the structure of a related compound, see: Akkurt *et al.* (2014).

2. Experimental

2.1. Crystal data

$C_{15}H_{20}N_4OS_2$
 $M_r = 336.47$
 Monoclinic, $P2_1/n$
 $a = 7.2269(8) \text{ \AA}$
 $b = 24.655(3) \text{ \AA}$
 $c = 9.6933(11) \text{ \AA}$
 $\beta = 92.5330(17)^\circ$
 $V = 1725.5(3) \text{ \AA}^3$
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.32 \text{ mm}^{-1}$
 $T = 150 \text{ K}$
 $0.23 \times 0.18 \times 0.08 \text{ mm}$

2.2. Data collection

Bruker SMART APEX CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2013)
 $T_{\min} = 0.93$, $T_{\max} = 0.97$
 31462 measured reflections
 4538 independent reflections
 3713 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.047$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.106$
 $S = 1.02$
 4538 reflections
 225 parameters
 68 restraints
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.36 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.26 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N3-H3B\cdots O1^i$	0.91	1.97	2.8792 (17)	179
$N3-H3A\cdots N1^{ii}$	0.91	2.22	3.0640 (19)	155

Symmetry codes: (i) $-x + 2, -y + 1, -z + 1$; (ii) $-x + 1, -y + 1, -z$.

Data collection: *APEX2* (Bruker, 2013); cell refinement: *SAINT* (Bruker, 2013); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS2014* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2014* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supporting information for this paper is available from the IUCr electronic archives (Reference: BH2502).

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supporting information

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Crystal structure of *N*-[4-amino-5-cyano-6-(methylsulfanyl)pyridin-2-yl]-2-(cyclohexylsulfanyl)acetamide

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S1. Comment

A large number of heterocyclic compounds containing pyridine rings are associated with diverse pharmacological properties such as antifungal (Patrick & Kinsman, 1996; Hishmat *et al.*, 1990), anticancer, anticonvulsant (Paronikyan *et al.*, 2002), antiviral (Bernardino *et al.*, 2007), antibacterial, antimicrobial (Doshi *et al.*, 1999; Jemmezi *et al.*, 2014), antimycobacterial (Mamolo *et al.*, 2004) and insecticidal activities (Bhatt *et al.*, 2001). In this connection, and as part of our on-going study on synthesis of bio-active heterocyclic molecules, we herein report the synthesis and crystal determination of the title compound.

In the title molecule (Fig. 1), the acetamido fragment is nearly coplanar with the pyridyl ring [C8—N4—C5—C4 torsion angle = $-4.1(2)^\circ$], possibly aided by a weak C4—H4 \cdots O1 interaction (Table 1), while the cyclohexylsulfanyl portion protrudes from this plane [N4—C8—C9—S2 torsion angle = $-40.8(6)^\circ$]. The main disordered part of the cyclohexyl group adopts the chair conformation with puckering parameters $Q = 0.578(3) \text{ \AA}$, $\theta = 177.6(3)^\circ$, and $\varphi = 311(3)^\circ$. The minor disordered part of the cyclohexyl group exhibits a distorted chair conformation with puckering parameters $Q = 0.60(2) \text{ \AA}$, $\theta = 11.3(19)^\circ$, and $\varphi = 173(10)^\circ$. All the bond lengths and bond angles are normal and comparable with those reported for a related compound (Akkurt *et al.*, 2014).

Alternating, pairwise N3—H3B \cdots O1 and N3—H3A \cdots N1 hydrogen bonds across centres of symmetry form chains (Fig. 2 and Table 1) which are associated into stepped layers *via* offset π -stacking between pyridyl rings [Fig. 3; interplanar distance = $3.384(1) \text{ \AA}$; $Cg^i \cdots Cg^j = 3.566(1) \text{ \AA}$, where Cg is the centroid of the C1 \cdots C5/N2 ring; $i: 1 - x, 1 - y, 1 - z$]. Adjacent stacks are inclined to one another by approximately 62° .

S2. Experimental

A mixture of 1 mmol (257 mg) of *N*-[4-amino-5-cyano-6-(methylthio)pyridin-2-yl]-2-chloroacetamide and 1 mmol (116 mg) of cyclohexanethiol in 30 ml ethanol along with few drops of triethylamine (TEA) as a catalyst was refluxed for 3 h at 350 K. The reaction mixture was allowed to cool down at room temperature and the excess solvent was evaporated under reduced pressure. The resulting solid was filtered off, dried and recrystallized from benzene, to afford colourless crystals (92% yield) suitable for X-ray diffraction. *M.p.* 463 – 465 K.

IR (ν_{\max} , cm^{-1}): 3470, 3325, 3215, (NH₂+NH), 2928 (CH_{aliph.}), 2210 (C \equiv N), 1689 (C=O_{amidic}); ¹H-NMR (DMSO-*d*₆), δ , p.p.m.: 10.27 (s, 1H, NH; exchanged by D₂O), 7.30 (s, 1H, CH_{pyridyl}), 6.99 (s, 2H, NH₂; exchanged by D₂O), 3.41 (s, 2H, COCH₂), 2.84–2.82 (m, 1H, CH_{cyclohexyl}), 2.52 (s, 3H, SCH₃), 1.95–1.93 (m, 2H, CH_{2 cyclohexyl}), 1.69 (m, 2H, CH_{2 cyclohexyl}), 1.56–1.54 (m, 1H, CH_{cyclohexyl}), 1.28–1.24 (m, 5H, 2CH₂+CH_{cyclohexyl}).

S3. Refinement

H atoms attached to carbon were placed in calculated positions ($C-H = 0.95 - 0.99 \text{ \AA}$) while those attached to nitrogen were placed in locations derived from a difference map and their parameters adjusted to give $N-H = 0.91 \text{ \AA}$. All H atoms were included as riding contributions with isotropic displacement parameters 1.2 - 1.5 times those of the attached atoms. The cyclohexyl group, S2 and C9 are disordered over two sites [site-occupancy factors are 0.8845 (18) and 0.1155 (18)]. The components of the disorder were refined with restraints that their geometries be the same (Sheldrick, 2008)

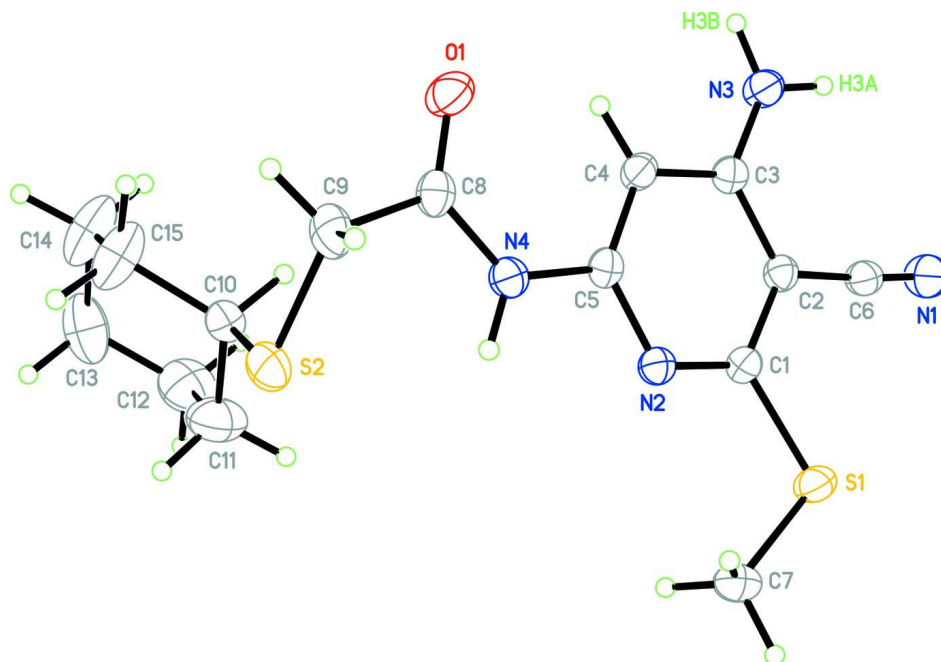
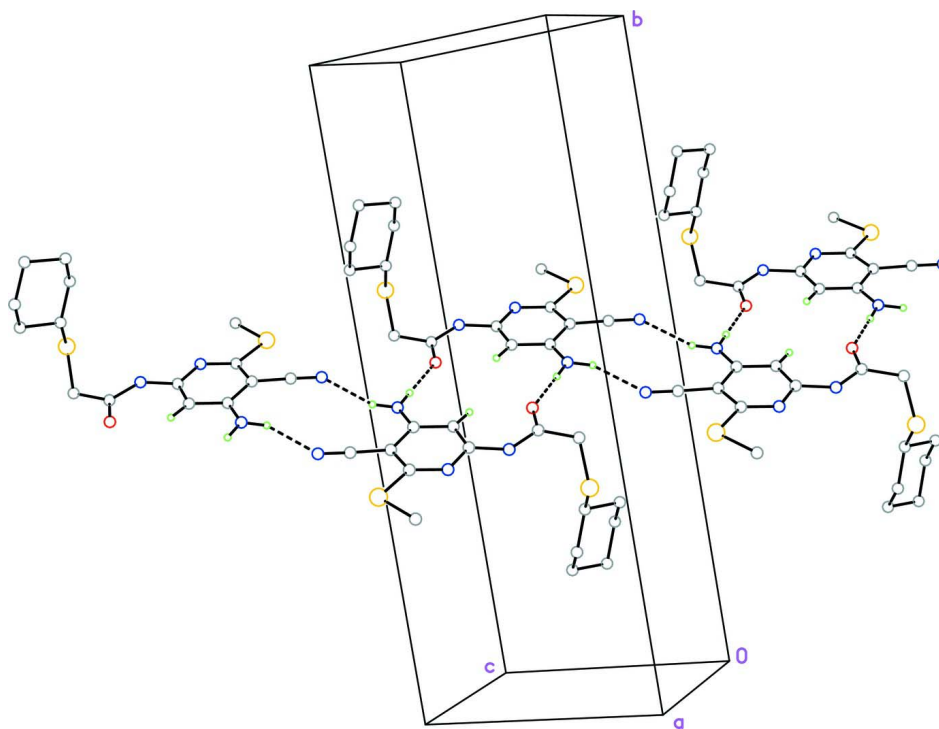
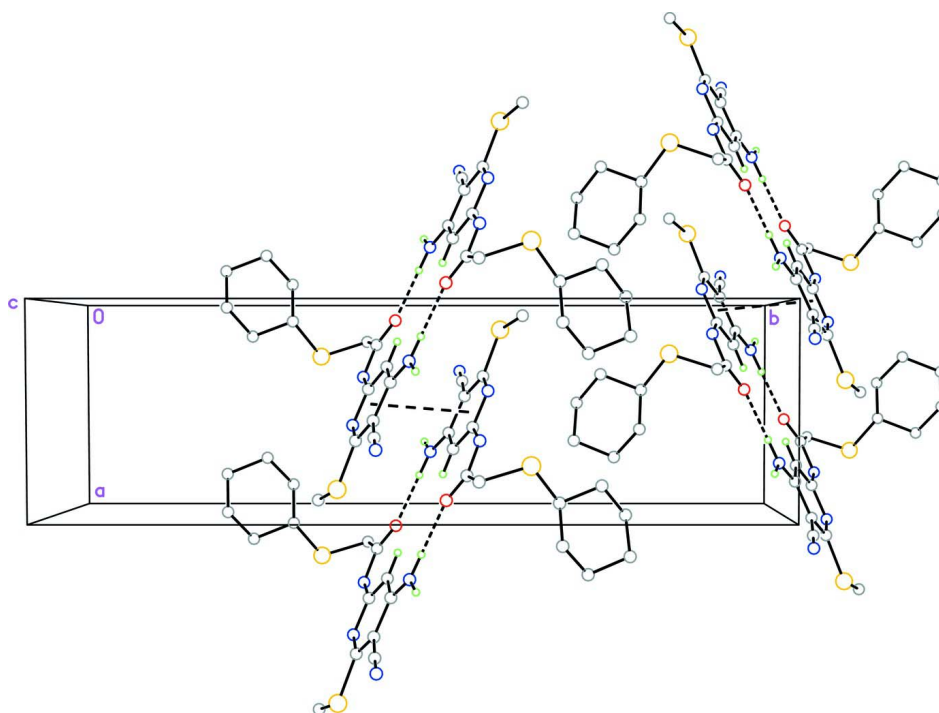


Figure 1

The title molecule showing displacement ellipsoids at the 50% probability level. Only the major component of the disorder is shown.

**Figure 2**

Perspective view of a portion of the hydrogen-bonded chain.

**Figure 3**

Packing viewed down the *c* axis, showing the pairwise N—H...O hydrogen bonding and the offset π -stacking interactions as dashed lines.

N-[4-amino-5-cyano-6-(methylsulfanyl)pyridin-2-yl]-2-(cyclohexylsulfanyl)acetamide*Crystal data*C₁₅H₂₀N₄OS₂ $M_r = 336.47$ Monoclinic, $P2_1/n$ $a = 7.2269$ (8) Å $b = 24.655$ (3) Å $c = 9.6933$ (11) Å $\beta = 92.5330$ (17)° $V = 1725.5$ (3) Å³ $Z = 4$ $F(000) = 712$ $D_x = 1.295$ Mg m⁻³

Melting point: 463 K

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 9975 reflections

 $\theta = 2.3$ – 29.0° $\mu = 0.32$ mm⁻¹ $T = 150$ K

Plate, colourless

 $0.23 \times 0.18 \times 0.08$ mm*Data collection*

Bruker SMART APEX CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 8.3660 pixels mm⁻¹ φ and ω scans

Absorption correction: multi-scan

(SADABS; Bruker, 2013)

 $T_{\min} = 0.93$, $T_{\max} = 0.97$

31462 measured reflections

4538 independent reflections

3713 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.047$ $\theta_{\text{max}} = 29.0^\circ$, $\theta_{\text{min}} = 2.3^\circ$ $h = -9 \rightarrow 9$ $k = -33 \rightarrow 33$ $l = -13 \rightarrow 13$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.040$ $wR(F^2) = 0.106$ $S = 1.02$

4538 reflections

225 parameters

68 restraints

0 constraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: mixed

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0492P)^2 + 0.7171P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.006$ $\Delta\rho_{\text{max}} = 0.36$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³*Special details*

Experimental. The diffraction data were obtained from 3 sets of 400 frames, each of width 0.5° in ω , collected at $\varphi = 0.00$, 90.00 and 180.00° and 2 sets of 800 frames, each of width 0.45° in φ , collected at $\omega = -30.00$ and 210.00° . The scan time was 40 sec/frame.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.12721 (5)	0.61034 (2)	0.34743 (4)	0.03110 (11)	
O1	0.91811 (16)	0.53918 (5)	0.71493 (12)	0.0356 (3)	
N1	0.3292 (2)	0.54925 (6)	0.04030 (14)	0.0378 (3)	
N2	0.41788 (16)	0.59429 (5)	0.52327 (12)	0.0235 (2)	
N3	0.72278 (17)	0.50958 (5)	0.23714 (13)	0.0287 (3)	
H3A	0.6773	0.4994	0.1520	0.034*	
H3B	0.8363	0.4940	0.2514	0.034*	
N4	0.64424 (17)	0.58430 (5)	0.69315 (12)	0.0264 (3)	

H4A	0.5680	0.6065	0.7397	0.032*	
C1	0.35074 (19)	0.58641 (5)	0.39522 (15)	0.0227 (3)	
C2	0.44875 (19)	0.55937 (5)	0.29318 (14)	0.0218 (3)	
C3	0.62544 (18)	0.53707 (5)	0.32894 (14)	0.0222 (3)	
C4	0.69559 (19)	0.54512 (6)	0.46531 (14)	0.0237 (3)	
H4	0.8137	0.5314	0.4948	0.028*	
C5	0.58845 (19)	0.57337 (6)	0.55479 (14)	0.0226 (3)	
C6	0.37713 (19)	0.55438 (6)	0.15424 (15)	0.0258 (3)	
C7	0.0634 (2)	0.64195 (7)	0.50548 (17)	0.0332 (3)	
H7A	0.0792	0.6160	0.5817	0.050*	
H7B	-0.0664	0.6535	0.4968	0.050*	
H7C	0.1425	0.6736	0.5242	0.050*	
C8	0.8003 (2)	0.56755 (6)	0.76461 (15)	0.0258 (3)	
S2	0.74307 (10)	0.65352 (3)	0.94799 (6)	0.03506 (15)	0.8845 (18)
C9	0.8162 (10)	0.5850 (2)	0.9147 (11)	0.0327 (9)	0.8845 (18)
H9A	0.9470	0.5811	0.9481	0.039*	0.8845 (18)
H9B	0.7413	0.5600	0.9693	0.039*	0.8845 (18)
C10	0.9173 (2)	0.69109 (7)	0.85620 (18)	0.0303 (4)	0.8845 (18)
H10	0.9378	0.6717	0.7675	0.036*	0.8845 (18)
C11	0.8422 (3)	0.74748 (9)	0.8213 (3)	0.0504 (6)	0.8845 (18)
H11A	0.8143	0.7667	0.9075	0.061*	0.8845 (18)
H11B	0.7255	0.7441	0.7645	0.061*	0.8845 (18)
C12	0.9827 (4)	0.78041 (10)	0.7421 (3)	0.0535 (6)	0.8845 (18)
H12A	1.0040	0.7625	0.6527	0.064*	0.8845 (18)
H12B	0.9323	0.8171	0.7225	0.064*	0.8845 (18)
C13	1.1633 (4)	0.78515 (11)	0.8244 (3)	0.0600 (7)	0.8845 (18)
H13A	1.1437	0.8053	0.9110	0.072*	0.8845 (18)
H13B	1.2532	0.8057	0.7707	0.072*	0.8845 (18)
C14	1.2412 (3)	0.72892 (12)	0.8587 (3)	0.0629 (7)	0.8845 (18)
H14A	1.3574	0.7327	0.9159	0.075*	0.8845 (18)
H14B	1.2708	0.7102	0.7721	0.075*	0.8845 (18)
C15	1.1027 (3)	0.69463 (11)	0.9368 (3)	0.0555 (6)	0.8845 (18)
H15A	1.1534	0.6577	0.9517	0.067*	0.8845 (18)
H15B	1.0842	0.7111	1.0284	0.067*	0.8845 (18)
S2A	0.6907 (8)	0.6447 (2)	0.9744 (5)	0.03506 (15)	0.1155 (18)
C9A	0.833 (9)	0.589 (2)	0.918 (9)	0.0327 (9)	0.1155 (18)
H9A1	0.9638	0.6007	0.9302	0.039*	0.1155 (18)
H9A2	0.8151	0.5584	0.9817	0.039*	0.1155 (18)
C10A	0.8290 (17)	0.7034 (5)	0.9292 (12)	0.0303 (4)	0.1155 (18)
H10A	0.7526	0.7352	0.9560	0.036*	0.1155 (18)
C11A	0.854 (3)	0.7113 (6)	0.7781 (15)	0.0504 (6)	0.1155 (18)
H11C	0.7317	0.7120	0.7283	0.061*	0.1155 (18)
H11D	0.9259	0.6806	0.7420	0.061*	0.1155 (18)
C12A	0.955 (3)	0.7639 (8)	0.753 (3)	0.0535 (6)	0.1155 (18)
H12C	0.8818	0.7945	0.7879	0.064*	0.1155 (18)
H12D	0.9664	0.7690	0.6524	0.064*	0.1155 (18)
C13A	1.148 (3)	0.7645 (10)	0.8242 (19)	0.0600 (7)	0.1155 (18)
H13C	1.2163	0.7980	0.8013	0.072*	0.1155 (18)

H13D	1.2214	0.7326	0.7971	0.072*	0.1155 (18)
C14A	1.109 (2)	0.7627 (8)	0.9781 (18)	0.0629 (7)	0.1155 (18)
H14C	1.2265	0.7651	1.0337	0.075*	0.1155 (18)
H14D	1.0308	0.7941	1.0018	0.075*	0.1155 (18)
C15A	1.009 (2)	0.7104 (8)	1.013 (2)	0.0555 (6)	0.1155 (18)
H15C	1.0912	0.6792	0.9947	0.067*	0.1155 (18)
H15D	0.9836	0.7102	1.1120	0.067*	0.1155 (18)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.02353 (18)	0.0410 (2)	0.0283 (2)	0.00880 (15)	-0.00329 (14)	-0.00609 (15)
O1	0.0297 (6)	0.0485 (7)	0.0281 (6)	0.0102 (5)	-0.0041 (4)	-0.0026 (5)
N1	0.0409 (8)	0.0456 (8)	0.0261 (7)	0.0097 (6)	-0.0061 (6)	-0.0053 (6)
N2	0.0222 (6)	0.0265 (6)	0.0217 (6)	0.0017 (4)	-0.0002 (4)	-0.0023 (5)
N3	0.0253 (6)	0.0402 (7)	0.0206 (6)	0.0068 (5)	-0.0003 (5)	-0.0054 (5)
N4	0.0270 (6)	0.0320 (6)	0.0201 (6)	0.0037 (5)	-0.0010 (5)	-0.0048 (5)
C1	0.0201 (6)	0.0235 (6)	0.0245 (7)	-0.0005 (5)	0.0002 (5)	-0.0004 (5)
C2	0.0222 (6)	0.0233 (6)	0.0199 (6)	0.0000 (5)	-0.0004 (5)	-0.0003 (5)
C3	0.0216 (6)	0.0239 (6)	0.0212 (6)	-0.0013 (5)	0.0023 (5)	0.0002 (5)
C4	0.0195 (6)	0.0288 (7)	0.0226 (7)	0.0012 (5)	-0.0007 (5)	-0.0013 (5)
C5	0.0225 (6)	0.0248 (6)	0.0204 (6)	-0.0018 (5)	-0.0012 (5)	-0.0007 (5)
C6	0.0242 (7)	0.0277 (7)	0.0253 (7)	0.0043 (5)	-0.0006 (5)	-0.0019 (5)
C7	0.0285 (7)	0.0363 (8)	0.0353 (8)	0.0062 (6)	0.0058 (6)	-0.0048 (7)
C8	0.0280 (7)	0.0270 (7)	0.0221 (7)	-0.0040 (5)	-0.0021 (5)	0.0018 (5)
S2	0.0358 (3)	0.0437 (3)	0.0258 (3)	0.0014 (2)	0.0031 (2)	-0.0104 (2)
C9	0.0430 (19)	0.0336 (14)	0.0210 (10)	-0.0013 (13)	-0.0052 (14)	0.0001 (11)
C10	0.0315 (9)	0.0308 (8)	0.0283 (9)	0.0031 (7)	-0.0019 (7)	-0.0059 (7)
C11	0.0486 (12)	0.0362 (10)	0.0669 (15)	0.0101 (9)	0.0078 (11)	-0.0019 (10)
C12	0.0639 (16)	0.0350 (14)	0.0615 (15)	0.0021 (12)	0.0024 (12)	0.0057 (13)
C13	0.0766 (17)	0.0525 (16)	0.0504 (13)	-0.0250 (15)	-0.0012 (12)	-0.0093 (13)
C14	0.0373 (12)	0.0873 (19)	0.0622 (16)	-0.0158 (12)	-0.0173 (11)	0.0164 (14)
C15	0.0405 (12)	0.0721 (16)	0.0521 (14)	-0.0090 (11)	-0.0170 (10)	0.0152 (12)
S2A	0.0358 (3)	0.0437 (3)	0.0258 (3)	0.0014 (2)	0.0031 (2)	-0.0104 (2)
C9A	0.0430 (19)	0.0336 (14)	0.0210 (10)	-0.0013 (13)	-0.0052 (14)	0.0001 (11)
C10A	0.0315 (9)	0.0308 (8)	0.0283 (9)	0.0031 (7)	-0.0019 (7)	-0.0059 (7)
C11A	0.0486 (12)	0.0362 (10)	0.0669 (15)	0.0101 (9)	0.0078 (11)	-0.0019 (10)
C12A	0.0639 (16)	0.0350 (14)	0.0615 (15)	0.0021 (12)	0.0024 (12)	0.0057 (13)
C13A	0.0766 (17)	0.0525 (16)	0.0504 (13)	-0.0250 (15)	-0.0012 (12)	-0.0093 (13)
C14A	0.0373 (12)	0.0873 (19)	0.0622 (16)	-0.0158 (12)	-0.0173 (11)	0.0164 (14)
C15A	0.0405 (12)	0.0721 (16)	0.0521 (14)	-0.0090 (11)	-0.0170 (10)	0.0152 (12)

Geometric parameters (Å, °)

S1—C1	1.7624 (14)	C12—C13	1.505 (4)
S1—C7	1.7972 (16)	C12—H12A	0.9900
O1—C8	1.2173 (18)	C12—H12B	0.9900
N1—C6	1.150 (2)	C13—C14	1.527 (4)

N2—C1	1.3269 (18)	C13—H13A	0.9900
N2—C5	1.3586 (18)	C13—H13B	0.9900
N3—C3	1.3422 (18)	C14—C15	1.535 (3)
N3—H3A	0.9101	C14—H14A	0.9900
N3—H3B	0.9104	C14—H14B	0.9900
N4—C8	1.3616 (19)	C15—H15A	0.9900
N4—C5	1.4093 (18)	C15—H15B	0.9900
N4—H4A	0.9098	S2A—C9A	1.807 (18)
C1—C2	1.4094 (19)	S2A—C10A	1.823 (12)
C2—C3	1.4190 (19)	C9A—H9A1	0.9900
C2—C6	1.427 (2)	C9A—H9A2	0.9900
C3—C4	1.4087 (19)	C10A—C11A	1.496 (13)
C4—C5	1.3770 (19)	C10A—C15A	1.511 (12)
C4—H4	0.9500	C10A—H10A	1.0000
C7—H7A	0.9800	C11A—C12A	1.512 (13)
C7—H7B	0.9800	C11A—H11C	0.9900
C7—H7C	0.9800	C11A—H11D	0.9900
C8—C9	1.516 (10)	C12A—C13A	1.529 (14)
C8—C9A	1.59 (8)	C12A—H12C	0.9900
S2—C9	1.803 (4)	C12A—H12D	0.9900
S2—C10	1.826 (2)	C13A—C14A	1.533 (14)
C9—H9A	0.9900	C13A—H13C	0.9900
C9—H9B	0.9900	C13A—H13D	0.9900
C10—C15	1.523 (3)	C14A—C15A	1.520 (13)
C10—C11	1.525 (3)	C14A—H14C	0.9900
C10—H10	1.0000	C14A—H14D	0.9900
C11—C12	1.532 (4)	C15A—H15C	0.9900
C11—H11A	0.9900	C15A—H15D	0.9900
C11—H11B	0.9900		
C1—S1—C7	100.83 (7)	C12—C13—H13A	109.6
C1—N2—C5	116.45 (12)	C14—C13—H13A	109.6
C3—N3—H3A	124.2	C12—C13—H13B	109.6
C3—N3—H3B	127.4	C14—C13—H13B	109.6
H3A—N3—H3B	107.9	H13A—C13—H13B	108.1
C8—N4—C5	128.45 (12)	C13—C14—C15	111.4 (2)
C8—N4—H4A	115.9	C13—C14—H14A	109.3
C5—N4—H4A	115.6	C15—C14—H14A	109.3
N2—C1—C2	123.46 (13)	C13—C14—H14B	109.3
N2—C1—S1	119.33 (10)	C15—C14—H14B	109.3
C2—C1—S1	117.21 (11)	H14A—C14—H14B	108.0
C1—C2—C3	119.11 (12)	C10—C15—C14	110.86 (18)
C1—C2—C6	122.05 (12)	C10—C15—H15A	109.5
C3—C2—C6	118.83 (12)	C14—C15—H15A	109.5
N3—C3—C4	121.05 (13)	C10—C15—H15B	109.5
N3—C3—C2	121.74 (13)	C14—C15—H15B	109.5
C4—C3—C2	117.21 (12)	H15A—C15—H15B	108.1
C5—C4—C3	118.24 (13)	C9A—S2A—C10A	102 (3)

C5—C4—H4	120.9	C8—C9A—S2A	118 (4)
C3—C4—H4	120.9	C8—C9A—H9A1	107.7
N2—C5—C4	125.47 (13)	S2A—C9A—H9A1	107.7
N2—C5—N4	111.10 (12)	C8—C9A—H9A2	107.7
C4—C5—N4	123.44 (13)	S2A—C9A—H9A2	107.7
N1—C6—C2	176.06 (16)	H9A1—C9A—H9A2	107.1
S1—C7—H7A	109.5	C11A—C10A—C15A	111.7 (13)
S1—C7—H7B	109.5	C11A—C10A—S2A	115.5 (9)
H7A—C7—H7B	109.5	C15A—C10A—S2A	115.5 (9)
S1—C7—H7C	109.5	C11A—C10A—H10A	104.2
H7A—C7—H7C	109.5	C15A—C10A—H10A	104.2
H7B—C7—H7C	109.5	S2A—C10A—H10A	104.2
O1—C8—N4	123.44 (14)	C10A—C11A—C12A	110.5 (17)
O1—C8—C9	121.2 (3)	C10A—C11A—H11C	109.6
N4—C8—C9	115.3 (3)	C12A—C11A—H11C	109.6
O1—C8—C9A	119.2 (18)	C10A—C11A—H11D	109.6
N4—C8—C9A	117.3 (17)	C12A—C11A—H11D	109.6
C9—S2—C10	100.0 (3)	H11C—C11A—H11D	108.1
C8—C9—S2	115.3 (6)	C11A—C12A—C13A	111.9 (17)
C8—C9—H9A	108.4	C11A—C12A—H12C	109.2
S2—C9—H9A	108.4	C13A—C12A—H12C	109.2
C8—C9—H9B	108.4	C11A—C12A—H12D	109.2
S2—C9—H9B	108.4	C13A—C12A—H12D	109.2
H9A—C9—H9B	107.5	H12C—C12A—H12D	107.9
C15—C10—C11	110.98 (18)	C12A—C13A—C14A	103 (2)
C15—C10—S2	112.88 (14)	C12A—C13A—H13C	111.1
C11—C10—S2	108.92 (14)	C14A—C13A—H13C	111.1
C15—C10—H10	108.0	C12A—C13A—H13D	111.1
C11—C10—H10	108.0	C14A—C13A—H13D	111.1
S2—C10—H10	108.0	H13C—C13A—H13D	109.0
C10—C11—C12	110.92 (19)	C15A—C14A—C13A	110.5 (17)
C10—C11—H11A	109.5	C15A—C14A—H14C	109.6
C12—C11—H11A	109.5	C13A—C14A—H14C	109.6
C10—C11—H11B	109.5	C15A—C14A—H14D	109.6
C12—C11—H11B	109.5	C13A—C14A—H14D	109.6
H11A—C11—H11B	108.0	H14C—C14A—H14D	108.1
C13—C12—C11	110.7 (2)	C10A—C15A—C14A	112.5 (13)
C13—C12—H12A	109.5	C10A—C15A—H15C	109.1
C11—C12—H12A	109.5	C14A—C15A—H15C	109.1
C13—C12—H12B	109.5	C10A—C15A—H15D	109.1
C11—C12—H12B	109.5	C14A—C15A—H15D	109.1
H12A—C12—H12B	108.1	H15C—C15A—H15D	107.8
C12—C13—C14	110.3 (2)		
C5—N2—C1—C2	-1.8 (2)	C10—S2—C9—C8	-65.2 (5)
C5—N2—C1—S1	177.97 (10)	C9—S2—C10—C15	-78.6 (4)
C7—S1—C1—N2	1.40 (13)	C9—S2—C10—C11	157.7 (4)
C7—S1—C1—C2	-178.84 (11)	C15—C10—C11—C12	56.0 (3)

N2—C1—C2—C3	2.8 (2)	S2—C10—C11—C12	-179.18 (18)
S1—C1—C2—C3	-176.92 (10)	C10—C11—C12—C13	-57.8 (3)
N2—C1—C2—C6	-176.20 (13)	C11—C12—C13—C14	57.8 (3)
S1—C1—C2—C6	4.05 (18)	C12—C13—C14—C15	-56.8 (3)
C1—C2—C3—N3	178.10 (13)	C11—C10—C15—C14	-54.6 (3)
C6—C2—C3—N3	-2.8 (2)	S2—C10—C15—C14	-177.19 (19)
C1—C2—C3—C4	-2.14 (19)	C13—C14—C15—C10	55.2 (3)
C6—C2—C3—C4	176.93 (13)	O1—C8—C9A—S2A	164 (3)
N3—C3—C4—C5	-179.59 (13)	N4—C8—C9A—S2A	-13 (6)
C2—C3—C4—C5	0.6 (2)	C10A—S2A—C9A—C8	-90 (5)
C1—N2—C5—C4	0.1 (2)	C9A—S2A—C10A—C11A	65 (3)
C1—N2—C5—N4	-179.93 (12)	C9A—S2A—C10A—C15A	-68 (3)
C3—C4—C5—N2	0.4 (2)	C15A—C10A—C11A—C12A	-51 (2)
C3—C4—C5—N4	-179.53 (13)	S2A—C10A—C11A—C12A	174.6 (14)
C8—N4—C5—N2	176.01 (14)	C10A—C11A—C12A—C13A	61 (3)
C8—N4—C5—C4	-4.1 (2)	C11A—C12A—C13A—C14A	-65 (3)
C5—N4—C8—O1	-0.2 (2)	C12A—C13A—C14A—C15A	62 (2)
C5—N4—C8—C9	-178.1 (4)	C11A—C10A—C15A—C14A	51 (2)
C5—N4—C8—C9A	176 (3)	S2A—C10A—C15A—C14A	-174.9 (14)
O1—C8—C9—S2	141.3 (3)	C13A—C14A—C15A—C10A	-58 (2)
N4—C8—C9—S2	-40.8 (6)		

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N3—H3B \cdots O1 ⁱ	0.91	1.97	2.8792 (17)	179
N3—H3A \cdots N1 ⁱⁱ	0.91	2.22	3.0640 (19)	155
C4—H4 \cdots O1	0.95	2.24	2.8493 (18)	121
C7—H7A \cdots O1 ⁱⁱⁱ	0.98	2.60	3.440 (2)	144

Symmetry codes: (i) $-x+2, -y+1, -z+1$; (ii) $-x+1, -y+1, -z$; (iii) $x-1, y, z$.