

Crystal structure of tolylfluanid

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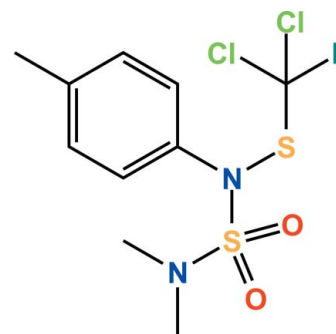
The title compound, $C_{10}H_{13}Cl_2FN_2O_2S_2$ {systematic name: *N*-[(dichlorofluoromethyl)sulfanyl]-*N,N'*-dimethyl-*N-p*-tolylsulfamide}, is a well known fungicide. The dihedral angle between the mean plane of the dimethylamino group and that of the benzene ring is $32.3(3)^\circ$. One Cl atom and one F atom of the dichlorofluoromethylthio group are disordered over two sets of sites with an occupancy ratio of 0.605(9):0.395(9). In the crystal structure, two C—H \cdots Cl hydrogen bonds link adjacent molecules, forming dimers with $R_2^2(14)$ loops. C—H \cdots O hydrogen bonds link pairs of dimers into chains along the *b*-axis direction. These chains are joined by an additional C—H \cdots O contact, generating a sheet in the *ab* plane.

Keywords: crystal structure; tolylfluanid; fungicide; hydrogen bonds.

CCDC reference: 1024472

1. Related literature

For information on the toxicity and fungicidal properties of the title compound, see: Sargis *et al.* (2012); Stajnbaher & Zupancic-Kralj (2008). For a related crystal structure, see: Ogata *et al.* (1986).



2. Experimental

2.1. Crystal data

$C_{10}H_{13}Cl_2FN_2O_2S_2$

$M_r = 347.24$

Monoclinic, $C2/c$

$a = 23.7638(19) \text{ \AA}$

$b = 8.7046(7) \text{ \AA}$

$c = 14.6559(11) \text{ \AA}$

$\beta = 102.133(3)^\circ$

$V = 2963.9(4) \text{ \AA}^3$

$Z = 8$

Mo $K\alpha$ radiation

$\mu = 0.73 \text{ mm}^{-1}$

$T = 173 \text{ K}$

$0.19 \times 0.10 \times 0.08 \text{ mm}$

2.2. Data collection

Bruker APEXII CCD

diffractometer

Absorption correction: multi-scan

(*SADABS*; Bruker, 2009)

$T_{\min} = 0.874$, $T_{\max} = 0.944$

18246 measured reflections

2913 independent reflections

2236 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.059$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.058$

$wR(F^2) = 0.135$

$S = 1.14$

2913 reflections

194 parameters

H-atom parameters constrained

$\Delta\rho_{\text{max}} = 0.48 \text{ e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.50 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C1-H1B\cdots O1^i$	0.98	2.59	3.469(7)	150
$C6-H6\cdots Cl1^{ii}$	0.95	2.77	3.457(6)	130
$C4-H4\cdots O2^{iii}$	0.95	2.62	3.563(5)	171

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+1, y, -z+\frac{1}{2}$; (iii) $-x+\frac{1}{2}, y-\frac{1}{2}, -z+\frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *DIAMOND* (Brandenburg, 2010); software used to prepare material for publication: *SHELXTL*.

Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: SJ5426).

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supporting information

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Crystal structure of tolylfluamid

Seonghwa Cho, Jineun Kim, Gihaeng Kang and Tae Ho Kim

S1. Comment

Tolylfluamid, C₁₀H₁₃Cl₂FN₂O₂S₂, is a member of the phenylsulfamide group of fungicides and has been applied in agriculture for control of fungal diseases during thinning, pruning, and harvesting of fruits and vegetables (Sargis *et al.*, 2012; Stajnbaher & Zupancic-Kralj, 2008). Its crystal structure is reported herein. In this compound (Scheme 1, Fig. 1), the dihedral angle between the dimethylamino group plane and that of the phenyl ring is 32.3 (3)°. Disorder was modeled for one Cl atom (Cl1) and one F atom (F1) of the dichlorofluoromethylthio group over two sets of sites with an occupancy ratio of 0.605 (9):0.395 (9). All bond lengths and bond angles are normal and comparable to those observed in the crystal structure of a similar compound (Ogata *et al.*, 1986).

In the crystal structure, two C6–H6..Cl1 hydrogen bonds link adjacent molecules, forming dimers with $R_2^2(14)$ loops. C1–H1B...O1 hydrogen bonds link pairs of dimers into chains along the *b* axis direction. These chains are joined by an additional slightly weaker C4–H4...O2 contact generating a two-dimensional sheet in the *ab* plane.

S2. Experimental

The title compound was purchased from the Dr. Ehrenstorfer GmbH Company. Slow evaporation from a solution in CHCl₃ gave single crystals suitable for X-ray analysis.

S3. Refinement

All H-atoms were positioned geometrically and refined using a riding model with $d(\text{C—H}) = 0.98 \text{ \AA}$, $U_{\text{iso}} = 1.5U_{\text{eq}}(\text{C})$ for methyl group and $d(\text{C—H}) = 0.95 \text{ \AA}$, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for $\text{Csp}^2\text{—H}$.

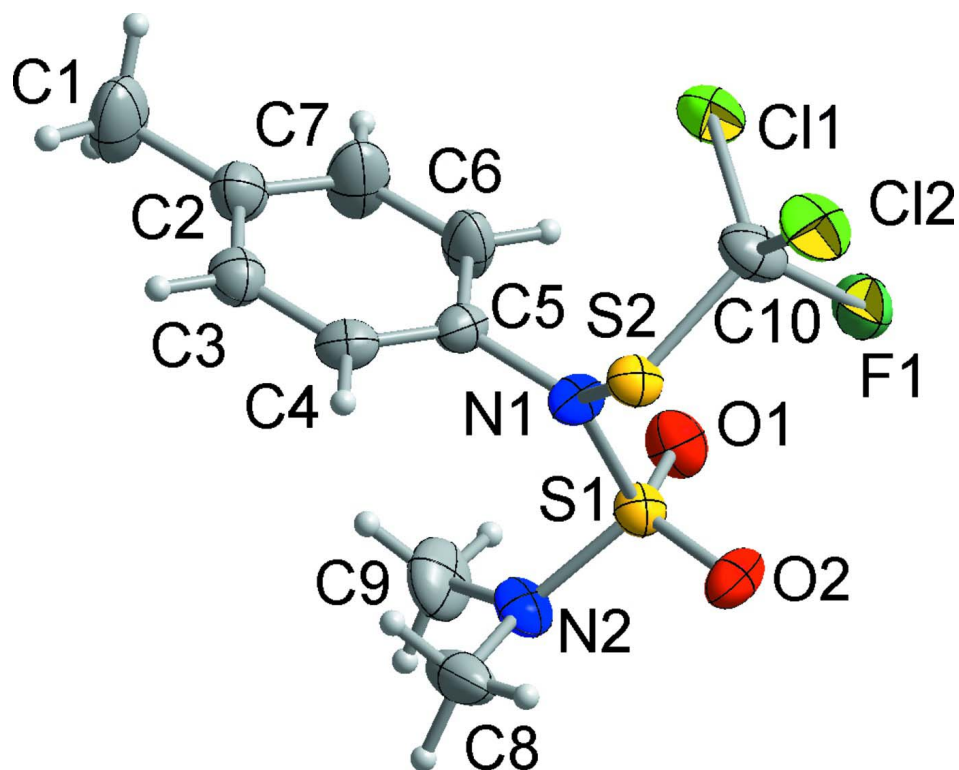
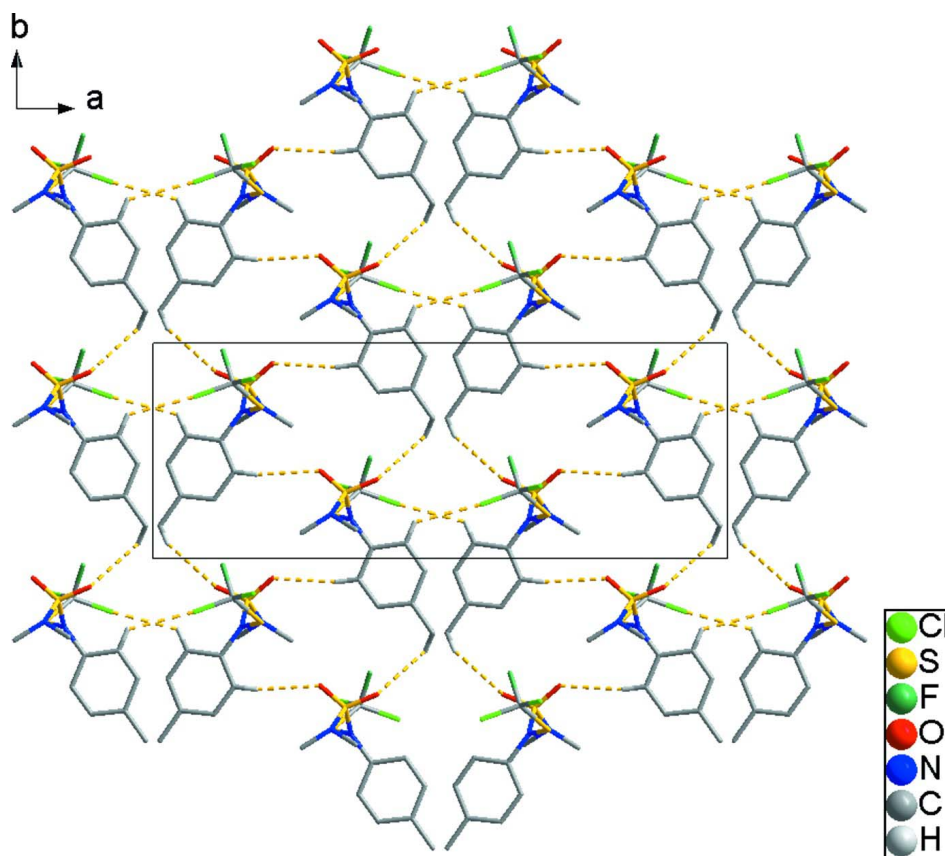


Figure 1

The asymmetric unit of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radius. Only atoms of the major disorder components are shown.

**Figure 2**

Crystal packing of the title compound with C–H···Cl and C–H···O hydrogen bonds are shown as dashed lines. H atoms bonded to C atoms have been omitted for clarity, except H atoms of hydrogen bonds. Only atoms of the major disorder components are shown.

N-[(Dichlorofluoromethyl)sulfanyl]-*N*',*N*'-dimethyl- *N*-*p*-tolylsulfamide

Crystal data

$C_{10}H_{13}Cl_2FN_2O_2S_2$

$M_r = 347.24$

Monoclinic, $C2/c$

Hall symbol: $-C\ 2yc$

$a = 23.7638$ (19) Å

$b = 8.7046$ (7) Å

$c = 14.6559$ (11) Å

$\beta = 102.133$ (3)°

$V = 2963.9$ (4) Å³

$Z = 8$

$F(000) = 1424$

$D_x = 1.556$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 6794 reflections

$\theta = 2.8$ – 27.1 °

$\mu = 0.73$ mm⁻¹

$T = 173$ K

Block, colourless

$0.19 \times 0.10 \times 0.08$ mm

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2009)

$T_{\min} = 0.874$, $T_{\max} = 0.944$

18246 measured reflections

2913 independent reflections

2236 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.059$
 $\theta_{\text{max}} = 26.0^\circ$, $\theta_{\text{min}} = 2.5^\circ$
 $h = -29 \rightarrow 26$

$k = -10 \rightarrow 10$
 $l = -18 \rightarrow 18$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.058$
 $wR(F^2) = 0.135$
 $S = 1.14$
 2913 reflections
 194 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0259P)^2 + 15.3569P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.48 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.50 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
Cl1	0.42965 (16)	0.2602 (5)	0.1267 (3)	0.0592 (11)	0.605 (9)
Cl1'	0.3886 (3)	0.4591 (10)	0.1620 (5)	0.0549 (18)	0.395 (9)
Cl2	0.32595 (6)	0.33538 (16)	-0.00564 (8)	0.0545 (4)	
S1	0.33566 (4)	0.30575 (13)	0.35656 (7)	0.0302 (3)	
S2	0.31078 (4)	0.20191 (13)	0.16220 (7)	0.0307 (3)	
F1	0.3625 (4)	0.4708 (12)	0.1530 (7)	0.052 (2)	0.605 (9)
F1'	0.4106 (7)	0.1951 (19)	0.1104 (9)	0.066 (4)	0.395 (9)
O1	0.38994 (13)	0.3659 (4)	0.4004 (2)	0.0438 (8)	
O2	0.29116 (13)	0.4060 (4)	0.31161 (19)	0.0402 (8)	
N1	0.34588 (14)	0.1819 (4)	0.2728 (2)	0.0304 (8)	
N2	0.31214 (16)	0.2043 (4)	0.4323 (2)	0.0376 (9)	
C1	0.4891 (2)	-0.3490 (7)	0.3831 (4)	0.0685 (17)	
H1A	0.5214	-0.3487	0.3510	0.103*	
H1B	0.4661	-0.4421	0.3663	0.103*	
H1C	0.5040	-0.3467	0.4507	0.103*	
C2	0.4519 (2)	-0.2090 (6)	0.3541 (3)	0.0428 (11)	
C3	0.39365 (19)	-0.2240 (6)	0.3152 (3)	0.0385 (11)	
H3	0.3771	-0.3236	0.3064	0.046*	
C4	0.35902 (18)	-0.0958 (5)	0.2887 (3)	0.0334 (10)	
H4	0.3192	-0.1084	0.2623	0.040*	
C5	0.38234 (17)	0.0485 (5)	0.3007 (3)	0.0308 (9)	
C6	0.4402 (2)	0.0684 (6)	0.3386 (4)	0.0474 (12)	

H6	0.4565	0.1684	0.3458	0.057*
C7	0.4738 (2)	-0.0593 (6)	0.3658 (4)	0.0585 (15)
H7	0.5133	-0.0452	0.3937	0.070*
C8	0.2585 (2)	0.1181 (6)	0.3994 (4)	0.0546 (14)
H8A	0.2674	0.0187	0.3743	0.082*
H8B	0.2331	0.1768	0.3503	0.082*
H8C	0.2393	0.1014	0.4515	0.082*
C9	0.3532 (3)	0.1351 (7)	0.5110 (3)	0.0612 (16)
H9A	0.3347	0.1233	0.5644	0.092*
H9B	0.3870	0.2018	0.5285	0.092*
H9C	0.3652	0.0341	0.4925	0.092*
C10	0.3611 (2)	0.3152 (6)	0.1127 (3)	0.0458 (12)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0428 (17)	0.057 (2)	0.084 (2)	-0.0029 (13)	0.0277 (15)	0.0112 (17)
C11'	0.070 (4)	0.052 (3)	0.048 (2)	-0.015 (3)	0.025 (3)	-0.0026 (19)
C12	0.0701 (9)	0.0687 (9)	0.0275 (6)	0.0036 (7)	0.0163 (6)	0.0102 (6)
S1	0.0345 (6)	0.0318 (6)	0.0237 (5)	-0.0008 (5)	0.0050 (4)	-0.0004 (4)
S2	0.0316 (6)	0.0378 (6)	0.0221 (5)	0.0014 (5)	0.0044 (4)	0.0015 (4)
F1	0.065 (5)	0.038 (4)	0.056 (4)	-0.010 (5)	0.025 (5)	-0.009 (3)
F1'	0.066 (8)	0.079 (10)	0.066 (7)	0.057 (7)	0.047 (7)	0.028 (7)
O1	0.0412 (19)	0.044 (2)	0.0430 (18)	-0.0107 (15)	0.0016 (14)	-0.0031 (15)
O2	0.0482 (19)	0.0385 (19)	0.0328 (16)	0.0149 (15)	0.0062 (14)	-0.0027 (14)
N1	0.0344 (19)	0.036 (2)	0.0203 (16)	0.0061 (16)	0.0038 (14)	-0.0011 (15)
N2	0.050 (2)	0.040 (2)	0.0251 (17)	-0.0075 (18)	0.0119 (16)	-0.0017 (16)
C1	0.058 (4)	0.050 (4)	0.091 (4)	0.011 (3)	-0.001 (3)	0.024 (3)
C2	0.041 (3)	0.043 (3)	0.045 (3)	0.003 (2)	0.010 (2)	0.009 (2)
C3	0.044 (3)	0.038 (3)	0.035 (2)	-0.001 (2)	0.010 (2)	0.000 (2)
C4	0.029 (2)	0.043 (3)	0.027 (2)	0.0004 (19)	0.0039 (17)	-0.0059 (19)
C5	0.032 (2)	0.036 (3)	0.025 (2)	0.0080 (19)	0.0069 (17)	0.0035 (18)
C6	0.034 (3)	0.035 (3)	0.069 (3)	-0.004 (2)	0.003 (2)	0.010 (2)
C7	0.034 (3)	0.050 (3)	0.083 (4)	0.000 (2)	-0.007 (3)	0.014 (3)
C8	0.061 (3)	0.052 (3)	0.062 (3)	-0.019 (3)	0.038 (3)	-0.013 (3)
C9	0.089 (4)	0.060 (4)	0.033 (3)	0.007 (3)	0.010 (3)	0.019 (3)
C10	0.044 (3)	0.062 (4)	0.033 (2)	-0.002 (2)	0.011 (2)	0.010 (2)

Geometric parameters (Å, °)

C11—C10	1.670 (6)	C1—H1C	0.9800
C11'—C10	1.523 (10)	C2—C3	1.389 (6)
C12—C10	1.768 (5)	C2—C7	1.400 (7)
S1—O1	1.415 (3)	C3—C4	1.392 (6)
S1—O2	1.422 (3)	C3—H3	0.9500
S1—N2	1.608 (3)	C4—C5	1.369 (6)
S1—N1	1.690 (3)	C4—H4	0.9500
S2—N1	1.669 (3)	C5—C6	1.381 (6)

S2—C10	1.814 (5)	C6—C7	1.378 (7)
F1—C10	1.474 (12)	C6—H6	0.9500
F1'—C10	1.581 (11)	C7—H7	0.9500
N1—C5	1.455 (5)	C8—H8A	0.9800
N2—C8	1.470 (6)	C8—H8B	0.9800
N2—C9	1.473 (6)	C8—H8C	0.9800
C1—C2	1.513 (7)	C9—H9A	0.9800
C1—H1A	0.9800	C9—H9B	0.9800
C1—H1B	0.9800	C9—H9C	0.9800
O1—S1—O2	120.1 (2)	C7—C6—H6	120.6
O1—S1—N2	107.78 (19)	C5—C6—H6	120.6
O2—S1—N2	108.94 (19)	C6—C7—C2	122.5 (5)
O1—S1—N1	108.03 (18)	C6—C7—H7	118.8
O2—S1—N1	105.17 (17)	C2—C7—H7	118.8
N2—S1—N1	105.96 (18)	N2—C8—H8A	109.5
N1—S2—C10	102.03 (19)	N2—C8—H8B	109.5
C5—N1—S2	120.2 (3)	H8A—C8—H8B	109.5
C5—N1—S1	118.3 (2)	N2—C8—H8C	109.5
S2—N1—S1	121.3 (2)	H8A—C8—H8C	109.5
C8—N2—C9	115.7 (4)	H8B—C8—H8C	109.5
C8—N2—S1	117.2 (3)	N2—C9—H9A	109.5
C9—N2—S1	119.7 (3)	N2—C9—H9B	109.5
C2—C1—H1A	109.5	H9A—C9—H9B	109.5
C2—C1—H1B	109.5	N2—C9—H9C	109.5
H1A—C1—H1B	109.5	H9A—C9—H9C	109.5
C2—C1—H1C	109.5	H9B—C9—H9C	109.5
H1A—C1—H1C	109.5	F1—C10—C11'	23.6 (3)
H1B—C1—H1C	109.5	F1—C10—F1'	131.4 (7)
C3—C2—C7	116.8 (4)	C11'—C10—F1'	107.8 (7)
C3—C2—C1	120.9 (5)	F1—C10—C11	106.0 (5)
C7—C2—C1	122.4 (4)	C11'—C10—C11	82.3 (4)
C2—C3—C4	121.3 (4)	F1'—C10—C11	25.9 (7)
C2—C3—H3	119.4	F1—C10—C12	105.5 (5)
C4—C3—H3	119.4	C11'—C10—C12	116.8 (4)
C5—C4—C3	120.0 (4)	F1'—C10—C12	104.1 (6)
C5—C4—H4	120.0	C11—C10—C12	113.3 (3)
C3—C4—H4	120.0	F1—C10—S2	107.5 (5)
C4—C5—C6	120.6 (4)	C11'—C10—S2	120.7 (4)
C4—C5—N1	119.7 (4)	F1'—C10—S2	101.9 (7)
C6—C5—N1	119.8 (4)	C11—C10—S2	120.2 (3)
C7—C6—C5	118.9 (4)	C12—C10—S2	103.4 (2)
C10—S2—N1—C5	-92.6 (3)	C3—C4—C5—C6	0.2 (6)
C10—S2—N1—S1	92.9 (3)	C3—C4—C5—N1	180.0 (4)
O1—S1—N1—C5	60.6 (3)	S2—N1—C5—C4	-59.3 (5)
O2—S1—N1—C5	-170.0 (3)	S1—N1—C5—C4	115.4 (4)
N2—S1—N1—C5	-54.7 (3)	S2—N1—C5—C6	120.5 (4)

O1—S1—N1—S2	-124.8 (2)	S1—N1—C5—C6	-64.9 (5)
O2—S1—N1—S2	4.6 (3)	C4—C5—C6—C7	-1.2 (7)
N2—S1—N1—S2	119.9 (2)	N1—C5—C6—C7	179.0 (4)
O1—S1—N2—C8	-174.3 (3)	C5—C6—C7—C2	1.9 (8)
O2—S1—N2—C8	53.9 (4)	C3—C2—C7—C6	-1.5 (8)
N1—S1—N2—C8	-58.8 (4)	C1—C2—C7—C6	179.3 (5)
O1—S1—N2—C9	-25.5 (4)	N1—S2—C10—F1	-69.9 (5)
O2—S1—N2—C9	-157.4 (4)	N1—S2—C10—C11'	-48.4 (5)
N1—S1—N2—C9	89.9 (4)	N1—S2—C10—F1'	70.9 (7)
C7—C2—C3—C4	0.4 (7)	N1—S2—C10—C11	51.2 (4)
C1—C2—C3—C4	179.7 (5)	N1—S2—C10—C12	178.7 (2)
C2—C3—C4—C5	0.2 (6)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C1—H1B \cdots O1 ⁱ	0.98	2.59	3.469 (7)	150
C6—H6 \cdots C11 ⁱⁱ	0.95	2.77	3.457 (6)	130
C4—H4 \cdots O2 ⁱⁱⁱ	0.95	2.62	3.563 (5)	171

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+1, y, -z+1/2$; (iii) $-x+1/2, y-1/2, -z+1/2$.