Crystal structure of diethyl (E)-2-[(benzofuran-2-yl)methylidene]succinate

Marie-Luis Schirmer, Anke Spannenberg and Thomas Werner*

Leibniz-Institut für Katalyse e. V. an der Universität Rostock, Albert-Einstein-Strasse 29a, 18059 Rostock, Germany. *Correspondence e-mail: Thomas.Werner@catalysis.de
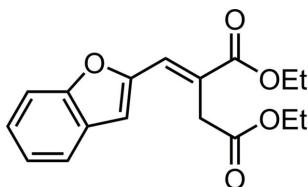
The title compound, $C_{17}H_{18}O_5$, was synthesized by a base-free catalytic Wittig reaction. The molecule consists of a diethyl itaconate unit, which is connected *via* the C=C double bond to a benzofuran moiety. The benzofuran ring system (r.m.s. deviation = 0.007 Å) forms dihedral angles of 79.58 (4) and 12.12 (10)° with the mean planes through the *cis* and *trans* ethoxycarbonyl groups, respectively. An intramolecular C—H···O hydrogen bond involving the O atom of the benzofuran moiety is observed. In the crystal, molecules are linked into ribbons running parallel to the *b* axis by C—H···O hydrogen bonds.

1. Related literature

For the synthesis of the title compound and related structures, see: Schirmer *et al.* (2015). For related crystal structures of similar compounds corresponding to (benzofuran)-CH=CR^1R^2, which only differ in R^1 and R^2 with at least one electron-withdrawing group, see: Penthala *et al.* (2012); Wei *et al.* (2011).

2. Experimental

2.1. Crystal data

$C_{17}H_{18}O_5$
$M_r = 302.31$
Orthorhombic, $P2_12_12_1$
$a = 7.0974$ (4) Å
$b = 8.0898$ (4) Å
$c = 26.9429$ (15) Å

$V = 1546.97$ (14) Å^3
$Z = 4$
Mo $K\alpha$ radiation
$\mu = 0.10$ mm^{-1}
$T = 150$ K
$0.46 \times 0.35 \times 0.20$ mm

2.2. Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2014)
$T_{min} = 0.84$, $T_{max} = 0.98$

11102 measured reflections
4020 independent reflections
3645 reflections with $I > 2\sigma(I)$
$R_{int} = 0.021$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
$wR(F^2) = 0.087$
$S = 1.03$
4020 reflections

201 parameters
H-atom parameters constrained
$\Delta\rho_{max} = 0.25$ e Å^{-3}
$\Delta\rho_{min} = -0.16$ e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C2—H2···O3^i	0.95	2.47	3.412 (2)	170
C11—H11A···O5^{ii}	0.99	2.52	3.458 (2)	158
C11—H11B···O1	0.99	2.30	3.044 (2)	131

Symmetry codes: (i) $x + 1, y, z$; (ii) $x + \frac{1}{2}, -y + \frac{3}{2}, -z$.

Data collection: *APEX2* (Bruker, 2014); cell refinement: *SAINT* (Bruker, 2013); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2014* (Sheldrick, 2015); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

References

Bruker (2013). *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Bruker (2014). *APEX2* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
Penthala, N. R., Parkin, S. & Crooks, P. A. (2012). *Acta Cryst.* E**68**, o731.
Schirmer, M.-L., Adomeit, S. & Werner, T. (2015). *Org. Lett.* **17**, 3078–3081.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Sheldrick, G. M. (2015). *Acta Cryst.* C**71**, 3–8.
Wei, S., Wei, X.-G., Su, X., You, J. & Ren, Y. (2011). *Chem. Eur. J.* **17**, 5965–5971.

supporting information

Crystal structure of diethyl (*E*)-2-[(benzofuran-2-yl)methylidene]succinate

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S1. Synthesis and crystallization

Diethyl (*E*)-2-(benzofuran-2-ylmethylene) succinate was prepared as previously described by Schirmer *et al.* (2015). All reagents were purchased from commercial sources and used as received without further purification. The reaction was performed in 5 mL Wheaton screw-top V-Vials® with solid-top cap. The vial was dried in an oven at 120°C before use. Toluene was freshly distilled from sodium/benzophenone. Thin layer chromatography was performed on Merck TLC-plates with fluorescence indication (silica type 60, F254), spots were visualized using UV-light. Flash chromatography was performed using silica with a grain size of 40–63 μm from Macherey-Nagel. Benzofuran-2-carbaldehyde (151 mg, 1.03 mmol) was dissolved in toluene (2 mL). Diethyl maleate (197 mg, 1.14 mmol), tri-*n*-butylphosphine (11 mg, 0.054 mmol) and phenylsilane (113 mg, 1.04 mmol) were added successively to the solution. After stirring for 24 h at 125°C, the reaction mixture was cooled to ambient temperature. The crude product was purified by column chromatography over silica (SiO_2) with cyclohexane/ethyl acetate (20:1 *v/v*) as eluents. All residues and volatiles were removed in vacuum (≤1 mbar, 60°C) to afford the olefination product (219 mg, 0.724 mmol, yield 70%, *E/Z* = 95:5) as colourless solid. Single crystals could be obtained by slow evaporation of a petroleum/dichloromethane (4:1 *v/v*) solution.

S2. Refinement

H atoms were placed in idealized positions with d(C—H) = 0.95 Å (CH), 0.99 Å (CH_2), 0.98 Å (CH_3) and refined using a riding model with U_{iso}(H) fixed at 1.2 U_{eq}(C) for CH and CH_2 and 1.5 U_{eq}(C) for CH_3. A rotating model was used for the methyl groups.
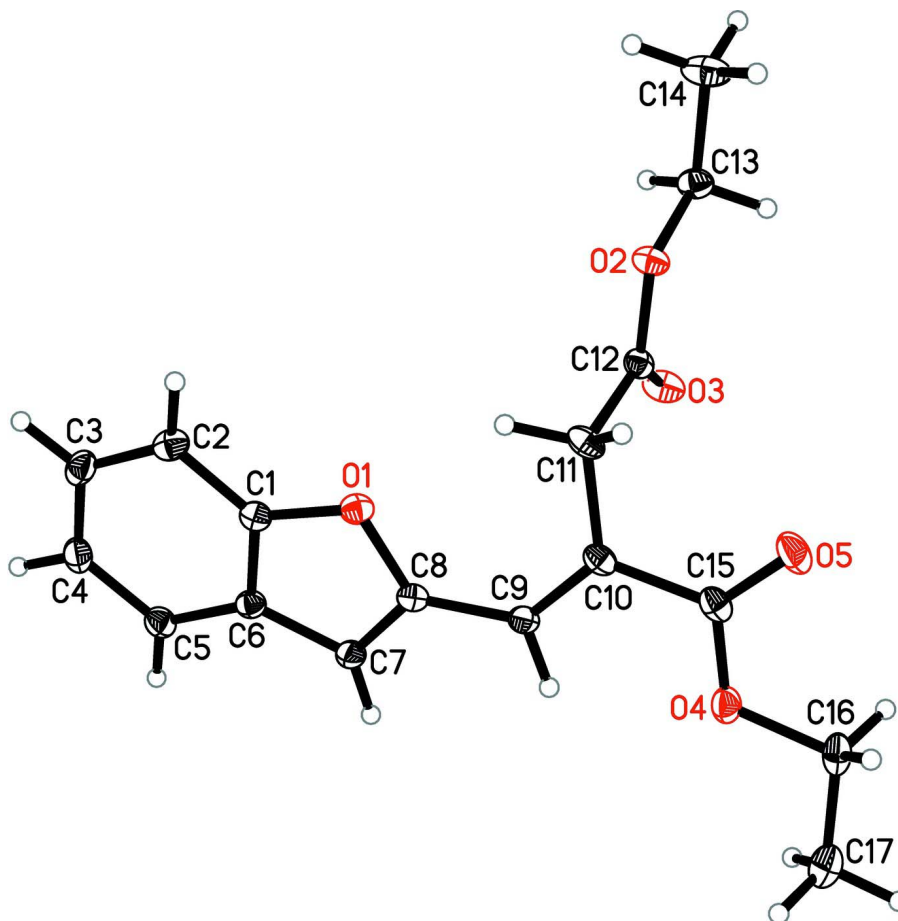
Figure 1

The molecular structure of the title compound, showing the atom labelling and displacement ellipsoids drawn at 30% probability level.

Diethyl (*E*)-2-[(1-benzofuran-2-yl)methylidene]succinate

Crystal data

$C_{17}H_{18}O_5$	$D_x = 1.298$ Mg m^{-3}
$M_r = 302.31$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Orthorhombic, $P2_12_12_1$	Cell parameters from 4958 reflections
$a = 7.0974$ (4) Å	$\theta = 2.6–28.6°$
$b = 8.0898$ (4) Å	$\mu = 0.10$ mm^{-1}
$c = 26.9429$ (15) Å	$T = 150$ K
$V = 1546.97$ (14) Å^3	Prism, colourless
$Z = 4$	$0.46 \times 0.35 \times 0.20$ mm
$F(000) = 640$	

Data collection

Bruker APEXII CCD diffractometer	Absorption correction: multi-scan (*SADABS*; Bruker, 2014)
Radiation source: fine-focus sealed tube	$T_{min} = 0.84$, $T_{max} = 0.98$
Detector resolution: 8.3333 pixels mm^{-1}	11102 measured reflections
φ and ω scans	4020 independent reflections
	3645 reflections with $I > 2\sigma(I)$

$R_{int} = 0.021$

$\theta_{max} = 28.8°$, $\theta_{min} = 1.5°$

$h = -9 \rightarrow 8$

$k = -9 \rightarrow 10$

$l = -35 \rightarrow 36$

Refinement

Refinement on F^2	H-atom parameters constrained
Least-squares matrix: full	$w = 1/[\sigma^2(F_o^2) + (0.0447P)^2 + 0.2481P]$
$R[F^2 > 2\sigma(F^2)] = 0.035$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.087$	$(\Delta/\sigma)_{max} < 0.001$
$S = 1.03$	$\Delta\rho_{max} = 0.25$ e Å^{-3}
4020 reflections	$\Delta\rho_{min} = -0.16$ e Å^{-3}
201 parameters	Absolute structure: Flack x determined using
0 restraints	1441 quotients $[(I^+)-(I^-)]/[(I^+)+(I^-)]$ (Parsons *et*
Hydrogen site location: inferred from	*al.*, 2013)
neighbouring sites	Absolute structure parameter: −0.4 (3)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
C1	1.1845 (2)	0.8258 (2)	0.20175 (6)	0.0226 (3)
C2	1.3594 (3)	0.9003 (2)	0.20151 (7)	0.0283 (4)
H2	1.4074	0.9551	0.1730	0.034*
C3	1.4612 (3)	0.8902 (2)	0.24552 (8)	0.0317 (4)
H3	1.5820	0.9403	0.2473	0.038*
C4	1.3901 (3)	0.8079 (2)	0.28718 (7)	0.0308 (4)
H4	1.4637	0.8033	0.3166	0.037*
C5	1.2152 (3)	0.7334 (2)	0.28647 (6)	0.0282 (4)
H5	1.1677	0.6777	0.3149	0.034*
C6	1.1097 (2)	0.7422 (2)	0.24256 (6)	0.0235 (3)
C7	0.9292 (3)	0.6833 (2)	0.22697 (6)	0.0251 (3)
H7	0.8431	0.6206	0.2463	0.030*
C8	0.9040 (3)	0.7338 (2)	0.17946 (6)	0.0237 (3)
C9	0.7443 (3)	0.7013 (2)	0.14801 (6)	0.0231 (3)
H9	0.6562	0.6249	0.1613	0.028*
C10	0.6987 (3)	0.7607 (2)	0.10299 (6)	0.0238 (3)
C11	0.8049 (3)	0.8887 (2)	0.07369 (6)	0.0266 (4)
H11A	0.8143	0.8530	0.0386	0.032*
H11B	0.9342	0.8994	0.0871	0.032*
C12	0.7066 (3)	1.0538 (2)	0.07631 (6)	0.0221 (3)
C13	0.6907 (3)	1.3237 (2)	0.04296 (7)	0.0275 (4)
H13A	0.7166	1.3773	0.0753	0.033*
H13B	0.5525	1.3171	0.0384	0.033*
C14	0.7776 (4)	1.4218 (3)	0.00172 (8)	0.0411 (5)
H14A	0.9137	1.4307	0.0072	0.062*

H14B	0.7219	1.5326	0.0010	0.062*
H14C	0.7541	1.3662	−0.0300	0.062*
C15	0.5193 (3)	0.7059 (2)	0.07940 (6)	0.0264 (4)
C16	0.2480 (3)	0.5350 (3)	0.08479 (7)	0.0333 (4)
H16A	0.2709	0.4904	0.0511	0.040*
H16B	0.1552	0.6259	0.0823	0.040*
C17	0.1757 (3)	0.4019 (3)	0.11828 (9)	0.0423 (5)
H17A	0.2665	0.3106	0.1193	0.063*
H17B	0.0547	0.3612	0.1057	0.063*
H17C	0.1586	0.4465	0.1518	0.063*
O1	1.05847 (18)	0.82331 (16)	0.16268 (4)	0.0251 (3)
O2	0.77290 (18)	1.15850 (15)	0.04184 (4)	0.0255 (3)
O3	0.5870 (2)	1.09132 (17)	0.10586 (5)	0.0325 (3)
O4	0.42296 (19)	0.59499 (17)	0.10627 (5)	0.0294 (3)
O5	0.4678 (2)	0.75572 (19)	0.03924 (5)	0.0382 (4)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0260 (8)	0.0202 (8)	0.0217 (7)	0.0027 (7)	0.0028 (6)	−0.0030 (6)
C2	0.0274 (9)	0.0263 (9)	0.0314 (9)	−0.0004 (8)	0.0098 (7)	−0.0024 (7)
C3	0.0213 (8)	0.0279 (9)	0.0460 (10)	0.0003 (8)	0.0010 (8)	−0.0071 (8)
C4	0.0287 (9)	0.0290 (9)	0.0349 (9)	0.0047 (8)	−0.0086 (7)	−0.0045 (8)
C5	0.0316 (9)	0.0282 (9)	0.0247 (8)	0.0011 (8)	−0.0019 (7)	0.0020 (7)
C6	0.0244 (8)	0.0210 (8)	0.0250 (8)	0.0004 (7)	0.0018 (6)	−0.0007 (7)
C7	0.0263 (8)	0.0256 (8)	0.0234 (7)	−0.0035 (8)	−0.0011 (6)	0.0041 (7)
C8	0.0273 (9)	0.0205 (8)	0.0232 (7)	−0.0006 (7)	0.0013 (6)	0.0011 (6)
C9	0.0272 (8)	0.0196 (8)	0.0227 (7)	−0.0003 (7)	−0.0006 (6)	0.0014 (6)
C10	0.0307 (9)	0.0203 (8)	0.0205 (7)	0.0061 (7)	0.0005 (6)	−0.0011 (6)
C11	0.0345 (9)	0.0251 (9)	0.0202 (7)	0.0088 (8)	0.0052 (7)	0.0034 (7)
C12	0.0271 (9)	0.0218 (8)	0.0176 (7)	0.0003 (7)	−0.0019 (6)	−0.0007 (6)
C13	0.0286 (9)	0.0196 (8)	0.0343 (9)	0.0029 (7)	0.0057 (7)	0.0000 (7)
C14	0.0496 (13)	0.0254 (10)	0.0483 (12)	0.0056 (10)	0.0184 (10)	0.0096 (9)
C15	0.0338 (9)	0.0224 (9)	0.0230 (8)	0.0098 (7)	−0.0032 (7)	−0.0045 (6)
C16	0.0259 (9)	0.0382 (11)	0.0356 (10)	0.0064 (9)	−0.0093 (8)	−0.0103 (8)
C17	0.0294 (10)	0.0500 (14)	0.0475 (12)	−0.0045 (10)	−0.0053 (9)	−0.0038 (10)
O1	0.0295 (6)	0.0262 (6)	0.0197 (5)	−0.0021 (6)	0.0032 (5)	0.0010 (5)
O2	0.0296 (7)	0.0208 (6)	0.0261 (6)	0.0035 (5)	0.0068 (5)	0.0031 (5)
O3	0.0433 (8)	0.0269 (7)	0.0272 (6)	0.0057 (6)	0.0126 (6)	0.0008 (5)
O4	0.0296 (7)	0.0320 (7)	0.0266 (6)	0.0006 (6)	−0.0073 (5)	−0.0028 (5)
O5	0.0503 (9)	0.0375 (8)	0.0267 (6)	0.0075 (7)	−0.0135 (6)	0.0018 (6)

Geometric parameters (Å, º)

C1—C2	1.380 (3)	C11—H11A	0.9900
C1—O1	1.381 (2)	C11—H11B	0.9900
C1—C6	1.396 (2)	C12—O3	1.203 (2)
C2—C3	1.391 (3)	C12—O2	1.342 (2)

C2—H2	0.9500	C13—O2	1.459 (2)
C3—C4	1.399 (3)	C13—C14	1.498 (3)
C3—H3	0.9500	C13—H13A	0.9900
C4—C5	1.380 (3)	C13—H13B	0.9900
C4—H4	0.9500	C14—H14A	0.9800
C5—C6	1.402 (2)	C14—H14B	0.9800
C5—H5	0.9500	C14—H14C	0.9800
C6—C7	1.430 (2)	C15—O5	1.211 (2)
C7—C8	1.355 (2)	C15—O4	1.340 (2)
C7—H7	0.9500	C16—O4	1.453 (2)
C8—O1	1.389 (2)	C16—C17	1.496 (3)
C8—C9	1.439 (2)	C16—H16A	0.9900
C9—C10	1.344 (2)	C16—H16B	0.9900
C9—H9	0.9500	C17—H17A	0.9800
C10—C15	1.491 (3)	C17—H17B	0.9800
C10—C11	1.504 (3)	C17—H17C	0.9800
C11—C12	1.509 (2)		
C2—C1—O1	125.84 (16)	H11A—C11—H11B	108.1
C2—C1—C6	123.88 (16)	O3—C12—O2	123.10 (16)
O1—C1—C6	110.28 (15)	O3—C12—C11	125.51 (16)
C1—C2—C3	115.95 (17)	O2—C12—C11	111.37 (14)
C1—C2—H2	122.0	O2—C13—C14	107.76 (15)
C3—C2—H2	122.0	O2—C13—H13A	110.2
C2—C3—C4	121.64 (17)	C14—C13—H13A	110.2
C2—C3—H3	119.2	O2—C13—H13B	110.2
C4—C3—H3	119.2	C14—C13—H13B	110.2
C5—C4—C3	121.42 (17)	H13A—C13—H13B	108.5
C5—C4—H4	119.3	C13—C14—H14A	109.5
C3—C4—H4	119.3	C13—C14—H14B	109.5
C4—C5—C6	118.00 (17)	H14A—C14—H14B	109.5
C4—C5—H5	121.0	C13—C14—H14C	109.5
C6—C5—H5	121.0	H14A—C14—H14C	109.5
C1—C6—C5	119.10 (16)	H14B—C14—H14C	109.5
C1—C6—C7	105.72 (15)	O5—C15—O4	123.46 (18)
C5—C6—C7	135.18 (16)	O5—C15—C10	122.67 (18)
C8—C7—C6	107.16 (16)	O4—C15—C10	113.86 (15)
C8—C7—H7	126.4	O4—C16—C17	107.07 (16)
C6—C7—H7	126.4	O4—C16—H16A	110.3
C7—C8—O1	111.12 (16)	C17—C16—H16A	110.3
C7—C8—C9	127.21 (17)	O4—C16—H16B	110.3
O1—C8—C9	121.66 (14)	C17—C16—H16B	110.3
C10—C9—C8	130.98 (17)	H16A—C16—H16B	108.6
C10—C9—H9	114.5	C16—C17—H17A	109.5
C8—C9—H9	114.5	C16—C17—H17B	109.5
C9—C10—C15	118.98 (17)	H17A—C17—H17B	109.5
C9—C10—C11	126.78 (17)	C16—C17—H17C	109.5
C15—C10—C11	114.12 (15)	H17A—C17—H17C	109.5

C10—C11—C12	110.69 (15)	H17B—C17—H17C	109.5
C10—C11—H11A	109.5	C1—O1—C8	105.71 (13)
C12—C11—H11A	109.5	C12—O2—C13	115.07 (13)
C10—C11—H11B	109.5	C15—O4—C16	116.38 (14)
C12—C11—H11B	109.5		
O1—C1—C2—C3	178.71 (17)	C9—C10—C11—C12	−103.4 (2)
C6—C1—C2—C3	−0.8 (3)	C15—C10—C11—C12	72.54 (18)
C1—C2—C3—C4	0.5 (3)	C10—C11—C12—O3	15.9 (3)
C2—C3—C4—C5	−0.1 (3)	C10—C11—C12—O2	−165.74 (14)
C3—C4—C5—C6	0.0 (3)	C9—C10—C15—O5	178.99 (17)
C2—C1—C6—C5	0.7 (3)	C11—C10—C15—O5	2.7 (2)
O1—C1—C6—C5	−178.89 (15)	C9—C10—C15—O4	−2.1 (2)
C2—C1—C6—C7	−179.37 (17)	C11—C10—C15—O4	−178.38 (15)
O1—C1—C6—C7	1.0 (2)	C2—C1—O1—C8	179.37 (17)
C4—C5—C6—C1	−0.3 (3)	C6—C1—O1—C8	−1.04 (19)
C4—C5—C6—C7	179.9 (2)	C7—C8—O1—C1	0.65 (19)
C1—C6—C7—C8	−0.6 (2)	C9—C8—O1—C1	−178.34 (16)
C5—C6—C7—C8	179.3 (2)	O3—C12—O2—C13	0.7 (2)
C6—C7—C8—O1	0.0 (2)	C11—C12—O2—C13	−177.72 (14)
C6—C7—C8—C9	178.90 (17)	C14—C13—O2—C12	−178.92 (16)
C7—C8—C9—C10	171.75 (19)	O5—C15—O4—C16	−0.2 (3)
O1—C8—C9—C10	−9.4 (3)	C10—C15—O4—C16	−179.08 (14)
C8—C9—C10—C15	−178.91 (17)	C17—C16—O4—C15	174.43 (16)
C8—C9—C10—C11	−3.1 (3)		

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C2—H2···O3[i]	0.95	2.47	3.412 (2)	170
C11—H11*A*···O5[ii]	0.99	2.52	3.458 (2)	158
C11—H11*B*···O1	0.99	2.30	3.044 (2)	131

Symmetry codes: (i) *x*+1, *y*, *z*; (ii) *x*+1/2, −*y*+3/2, −*z*.