Crystal structure of 3-benzyl-1-[(1,2,3,4-tetrahydronaphthalen-1-ylidene)amino]-thiourea

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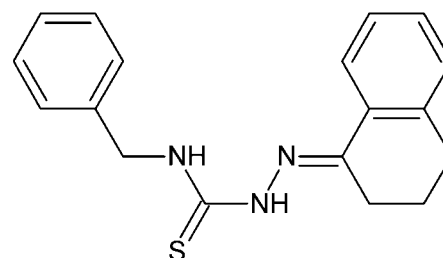
In the title compound, $C_{18}H_{19}N_3S$, the dihedral angle between the planes of the benzene rings is 58.63 (8)°. The six-membered ring bonded to the thiosemicarbazide group (r.m.s. deviation = 0.038 Å) adopts a sofa conformation, with one of the methylene-group C atoms as the flap. A short intramolecular N—H···N contact is observed. In the crystal, molecules are linked by weak N—H···S interactions to generate $C(4)$ chains propagating in the [010] direction, with adjacent molecules related by glide symmetry.

1. Related literature

For the antitumour activities of thiosemicarbazides, see: Vandresen et al. (2014); Xie et al. (2014); Gan et al. (2014). For the synthesis of the title compound, see: Mague et al. (2014)

2. Experimental

2.1. Crystal data

$C_{18}H_{19}N_3S$
$M_r = 309.42$
Orthorhombic, $Pbca$
$a = 11.9129$ (5) Å
$b = 9.6914$ (4) Å
$c = 27.8220$ (11) Å

$V = 3212.1$ (2) Å^3
$Z = 8$
Cu $K\alpha$ radiation
$\mu = 1.77$ mm^{-1}
$T = 150$ K
$0.22 \times 0.18 \times 0.05$ mm

2.2. Data collection

Bruker D8 VENTURE PHOTON 100 CMOS diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2014)
$T_{min} = 0.84$, $T_{max} = 0.91$

61936 measured reflections
3155 independent reflections
2795 reflections with $I > 2\sigma(I)$
$R_{int} = 0.050$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
$wR(F^2) = 0.098$
$S = 1.06$
3155 reflections

199 parameters
H-atom parameters constrained
$\Delta\rho_{max} = 0.56$ e Å^{-3}
$\Delta\rho_{min} = -0.22$ e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1A···N3	0.91	2.20	2.6219 (16)	108
N1—H1A···S1^i	0.91	2.85	3.5790 (13)	138

Symmetry code: (i) $-x + \frac{1}{2}, y - \frac{1}{2}, z$.

Data collection: APEX2 (Bruker, 2014); cell refinement: SAINT (Bruker, 2014); data reduction: SAINT; program(s) used to solve structure: SHELXT (Bruker, 2014); program(s) used to refine structure: SHELXL2014 (Sheldrick, 2015); molecular graphics: DIAMOND (Brandenburg & Putz, 2012); software used to prepare material for publication: SHELXTL (Bruker, 2014).

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supporting information

Crystal structure of 3-benzyl-1-[(1,2,3,4-tetrahydronaphthalen-1-ylidene)amino]thiourea

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S1. Comment

Recently, several kinds of thiosemicarbazone derivatives were synthesized and their antitumor activities were also reported (Vandresen *et al.*, 2014; Xie *et al.*, 2014; Gan *et al.*, 2014). As part of our studies in this area, we report here the synthesis and crystal structural determination of the title compound.

In the title molecule (Fig. 1), the dihedral angle between the phenyl ring (C1–C6) and the aromatic portion of the tetrahydronaphthylidene unit (C13–C18) is 58.64 (5)°. A Cremer-Pople analysis of the conformation of the C9–C13,C18 ring gave puckering parameters $Q = 0.434$ (2) Å, $\theta = 126.2$ (2)° and $\varphi = 305.1$ (2)°. The molecules pack in chains running parallel to the *b* axis assisted by weak N1—H1A···S1[i] (i: 1/2 - *x*, -1/2 + *y*, *z*) interactions (Table 1, Fig. 2).

S2. Experimental

The title compound was prepared according to our recently reported method (Mague *et al.*, 2014). The product was recrystallized from ethanol solution to afford colorless tablets (90% yield) *M*.p. 413 - 414 K.

S3. Refinement

H atoms attached to C atoms were placed in calculated positions (C—H = 0.95–0.99 Å) while those attached to N atoms were placed in locations derived from a difference map and their parameters adjusted to give N—H = 0.91 Å. All were included as riding contributions with isotropic displacement parameters 1.2–1.5 times those of the attached atoms.
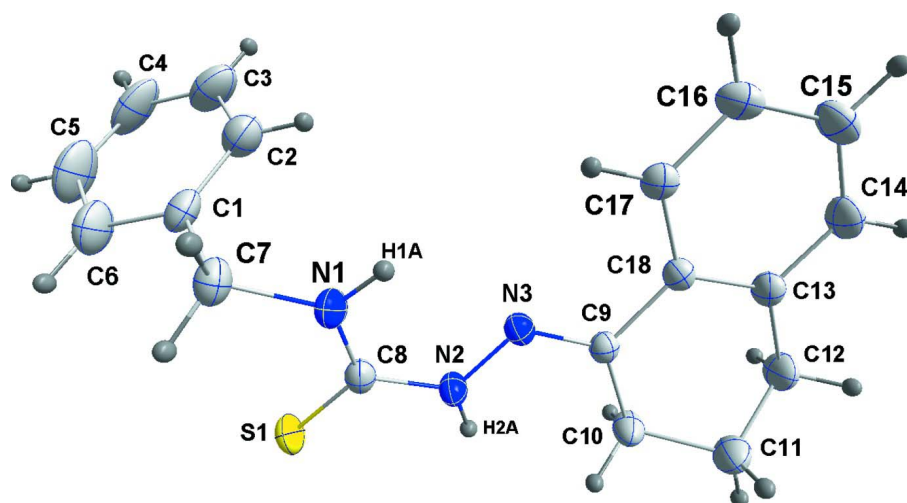
Figure 1

The title molecule with 50% probability displacement ellipsoids.
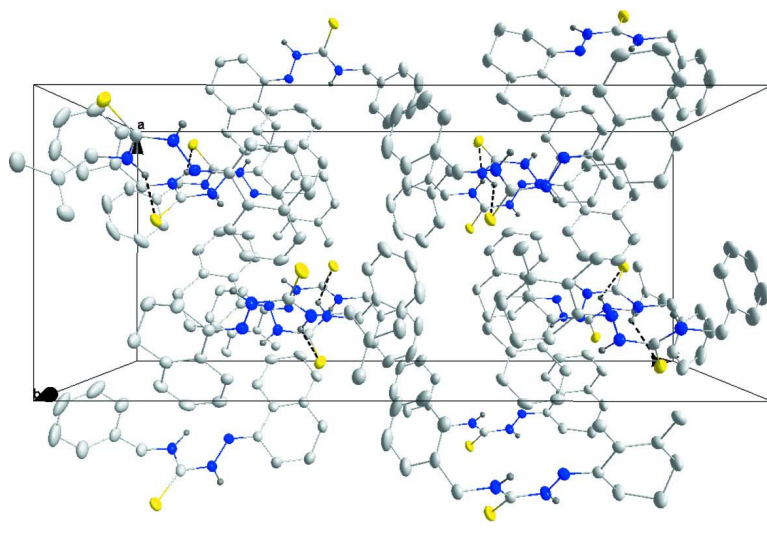
Figure 2

Packing viewed down the *b* axis. N—H···S interactions are shown by dotted lines.

3-Benzyl-1-[(1,2,3,4-tetrahydronaphthalen-1-ylidene)amino]thiourea

Crystal data

$C_{18}H_{19}N_3S$	$D_x = 1.280$ Mg m^{-3}
$M_r = 309.42$	Cu $K\alpha$ radiation, $\lambda = 1.54178$ Å
Orthorhombic, *Pbca*	Cell parameters from 9005 reflections
$a = 11.9129$ (5) Å	$\theta = 4.9$–72.0°
$b = 9.6914$ (4) Å	$\mu = 1.77$ mm^{-1}
$c = 27.8220$ (11) Å	$T = 150$ K
$V = 3212.1$ (2) Å^3	Tablet, colourless
$Z = 8$	$0.22 \times 0.18 \times 0.05$ mm
$F(000) = 1312$	

Data collection

Bruker D8 VENTURE PHOTON 100 CMOS diffractometer	$T_{min} = 0.84$, $T_{max} = 0.91$
Radiation source: INCOATEC IµS micro–focus source	61936 measured reflections
	3155 independent reflections
Mirror monochromator	2795 reflections with $I > 2\sigma(I)$
Detector resolution: 10.4167 pixels mm⁻¹	$R_{int} = 0.050$

Bruker D8 VENTURE PHOTON 100 CMOS
   diffractometer

Radiation source: INCOATEC IµS micro–focus
   source

Mirror monochromator

Detector resolution: 10.4167 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
   (*SADABS*; Bruker, 2014)

$T_{min} = 0.84$, $T_{max} = 0.91$

61936 measured reflections

3155 independent reflections

2795 reflections with $I > 2\sigma(I)$

$R_{int} = 0.050$

$\theta_{max} = 72.2°$, $\theta_{min} = 3.2°$

$h = -14 \rightarrow 14$

$k = -11 \rightarrow 11$

$l = -34 \rightarrow 34$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.035$

$wR(F^2) = 0.098$

$S = 1.06$

3155 reflections

199 parameters

0 restraints

Primary atom site location: structure-invariant
   direct methods

Secondary atom site location: difference Fourier
   map

Hydrogen site location: mixed

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0522P)^2 + 1.4441P]$
   where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.001$

$\Delta\rho_{max} = 0.56$ e Å⁻³

$\Delta\rho_{min} = -0.22$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger. H-atoms attached to carbon were placed in calculated positions (C—H = 0.95 - 0.99 Å) while those attached to nitrogen were placed in locations derived from a difference map and their parameters adjusted to give N—H = 0.91 Å. All were included as riding contributions with isotropic displacement parameters 1.2 - 1.5 times those of the attached atoms.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	*y*	*z*	U_{iso}*/U_{eq}
S1	0.08381 (3)	0.69807 (4)	0.37402 (2)	0.02936 (13)
N1	0.23130 (10)	0.48914 (12)	0.37501 (4)	0.0250 (3)
H1A	0.2789	0.4340	0.3581	0.030*
N2	0.18664 (10)	0.57805 (12)	0.30147 (4)	0.0243 (3)
H2A	0.1458	0.6361	0.2827	0.029*
N3	0.27037 (9)	0.49802 (12)	0.28232 (4)	0.0223 (3)
C1	0.29867 (14)	0.56296 (15)	0.45561 (5)	0.0287 (3)
C2	0.41047 (15)	0.57894 (19)	0.44281 (6)	0.0388 (4)
H2	0.4386	0.5343	0.4149	0.047*
C3	0.48124 (18)	0.6596 (2)	0.47050 (7)	0.0524 (5)
H3	0.5578	0.6695	0.4616	0.063*
C4	0.4413 (2)	0.7256 (2)	0.51086 (7)	0.0588 (6)
H4	0.4900	0.7817	0.5296	0.071*

C5	0.3306 (2)	0.7102 (2)	0.52401 (7)	0.0578 (6)
H5	0.3030	0.7551	0.5520	0.069*
C6	0.25919 (18)	0.62890 (19)	0.49641 (5)	0.0421 (4)
H6	0.1829	0.6185	0.5056	0.051*
C7	0.22186 (13)	0.47002 (15)	0.42688 (5)	0.0286 (3)
H7A	0.2394	0.3727	0.4347	0.034*
H7B	0.1433	0.4878	0.4367	0.034*
C8	0.17257 (12)	0.58125 (14)	0.35015 (5)	0.0231 (3)
C9	0.27768 (11)	0.49111 (13)	0.23608 (5)	0.0206 (3)
C10	0.19577 (13)	0.55643 (16)	0.20165 (5)	0.0298 (3)
H10A	0.2161	0.6546	0.1971	0.036*
H10B	0.1196	0.5531	0.2158	0.036*
C11	0.19402 (14)	0.48477 (18)	0.15289 (6)	0.0356 (4)
H11A	0.1620	0.3911	0.1564	0.043*
H11B	0.1459	0.5374	0.1304	0.043*
C12	0.31195 (14)	0.47477 (18)	0.13267 (5)	0.0339 (4)
H12A	0.3099	0.4231	0.1020	0.041*
H12B	0.3404	0.5687	0.1258	0.041*
C13	0.39089 (12)	0.40340 (15)	0.16706 (5)	0.0247 (3)
C14	0.48145 (13)	0.32664 (16)	0.14988 (6)	0.0305 (3)
H14	0.4926	0.3188	0.1162	0.037*
C15	0.55529 (13)	0.26181 (16)	0.18071 (6)	0.0328 (3)
H15	0.6154	0.2082	0.1683	0.039*
C16	0.54132 (12)	0.27535 (15)	0.22997 (6)	0.0299 (3)
H16	0.5928	0.2325	0.2514	0.036*
C17	0.45229 (12)	0.35136 (14)	0.24798 (5)	0.0242 (3)
H17	0.4436	0.3613	0.2817	0.029*
C18	0.37478 (11)	0.41390 (13)	0.21678 (5)	0.0206 (3)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0343 (2)	0.0285 (2)	0.0252 (2)	0.00308 (14)	0.00802 (14)	−0.00293 (13)
N1	0.0291 (6)	0.0258 (6)	0.0200 (6)	−0.0001 (5)	0.0013 (5)	−0.0034 (5)
N2	0.0271 (6)	0.0254 (6)	0.0204 (6)	0.0045 (5)	0.0021 (5)	−0.0016 (5)
N3	0.0219 (6)	0.0218 (6)	0.0232 (6)	0.0000 (4)	0.0021 (4)	−0.0022 (4)
C1	0.0406 (8)	0.0263 (7)	0.0192 (6)	−0.0016 (6)	−0.0040 (6)	0.0049 (6)
C2	0.0412 (9)	0.0445 (9)	0.0308 (8)	−0.0055 (8)	−0.0046 (7)	0.0048 (7)
C3	0.0518 (11)	0.0630 (12)	0.0425 (10)	−0.0206 (10)	−0.0188 (9)	0.0151 (9)
C4	0.0905 (16)	0.0536 (12)	0.0325 (10)	−0.0305 (12)	−0.0287 (10)	0.0096 (8)
C5	0.0977 (18)	0.0514 (12)	0.0244 (8)	−0.0166 (12)	−0.0052 (10)	−0.0066 (8)
C6	0.0599 (11)	0.0423 (9)	0.0242 (7)	−0.0045 (8)	0.0025 (8)	−0.0027 (7)
C7	0.0368 (8)	0.0273 (7)	0.0217 (7)	−0.0036 (6)	0.0024 (6)	0.0023 (6)
C8	0.0248 (7)	0.0235 (7)	0.0209 (6)	−0.0058 (5)	0.0023 (5)	−0.0023 (5)
C9	0.0220 (7)	0.0183 (6)	0.0214 (6)	−0.0013 (5)	0.0002 (5)	−0.0013 (5)
C10	0.0322 (8)	0.0321 (8)	0.0250 (7)	0.0105 (6)	−0.0022 (6)	−0.0015 (6)
C11	0.0356 (8)	0.0438 (9)	0.0275 (8)	0.0066 (7)	−0.0063 (6)	−0.0026 (7)
C12	0.0364 (9)	0.0453 (9)	0.0202 (7)	0.0052 (7)	0.0011 (6)	−0.0011 (6)

C13	0.0248 (7)	0.0248 (7)	0.0244 (7)	−0.0029 (6)	0.0024 (5)	−0.0022 (5)
C14	0.0297 (8)	0.0322 (8)	0.0297 (7)	−0.0028 (6)	0.0098 (6)	−0.0042 (6)
C15	0.0243 (7)	0.0279 (7)	0.0463 (9)	0.0019 (6)	0.0096 (7)	−0.0046 (7)
C16	0.0232 (7)	0.0248 (7)	0.0417 (8)	0.0014 (6)	−0.0010 (6)	0.0028 (6)
C17	0.0234 (7)	0.0220 (7)	0.0270 (7)	−0.0026 (6)	−0.0002 (5)	0.0007 (6)
C18	0.0208 (6)	0.0171 (6)	0.0239 (7)	−0.0028 (5)	0.0017 (5)	−0.0007 (5)

Geometric parameters (Å, º)

S1—C8	1.6855 (14)	C9—C18	1.4786 (18)
N1—C8	1.3285 (19)	C9—C10	1.5069 (19)
N1—C7	1.4592 (17)	C10—C11	1.524 (2)
N1—H1A	0.9098	C10—H10A	0.9900
N2—C8	1.3651 (18)	C10—H10B	0.9900
N2—N3	1.3713 (16)	C11—C12	1.516 (2)
N2—H2A	0.9099	C11—H11A	0.9900
N3—C9	1.2913 (18)	C11—H11B	0.9900
C1—C6	1.385 (2)	C12—C13	1.509 (2)
C1—C2	1.387 (2)	C12—H12A	0.9900
C1—C7	1.512 (2)	C12—H12B	0.9900
C2—C3	1.384 (3)	C13—C14	1.395 (2)
C2—H2	0.9500	C13—C18	1.4004 (19)
C3—C4	1.377 (3)	C14—C15	1.380 (2)
C3—H3	0.9500	C14—H14	0.9500
C4—C5	1.377 (3)	C15—C16	1.387 (2)
C4—H4	0.9500	C15—H15	0.9500
C5—C6	1.391 (3)	C16—C17	1.385 (2)
C5—H5	0.9500	C16—H16	0.9500
C6—H6	0.9500	C17—C18	1.405 (2)
C7—H7A	0.9900	C17—H17	0.9500
C7—H7B	0.9900		
C8—N1—C7	124.03 (12)	C9—C10—C11	112.53 (12)
C8—N1—H1A	116.9	C9—C10—H10A	109.1
C7—N1—H1A	119.0	C11—C10—H10A	109.1
C8—N2—N3	119.18 (12)	C9—C10—H10B	109.1
C8—N2—H2A	119.4	C11—C10—H10B	109.1
N3—N2—H2A	121.0	H10A—C10—H10B	107.8
C9—N3—N2	117.74 (12)	C12—C11—C10	110.29 (13)
C6—C1—C2	119.00 (16)	C12—C11—H11A	109.6
C6—C1—C7	120.17 (15)	C10—C11—H11A	109.6
C2—C1—C7	120.77 (14)	C12—C11—H11B	109.6
C3—C2—C1	120.34 (18)	C10—C11—H11B	109.6
C3—C2—H2	119.8	H11A—C11—H11B	108.1
C1—C2—H2	119.8	C13—C12—C11	111.79 (12)
C4—C3—C2	120.4 (2)	C13—C12—H12A	109.3
C4—C3—H3	119.8	C11—C12—H12A	109.3
C2—C3—H3	119.8	C13—C12—H12B	109.3

C5—C4—C3	119.80 (18)	C11—C12—H12B	109.3
C5—C4—H4	120.1	H12A—C12—H12B	107.9
C3—C4—H4	120.1	C14—C13—C18	118.90 (13)
C4—C5—C6	120.06 (19)	C14—C13—C12	120.60 (13)
C4—C5—H5	120.0	C18—C13—C12	120.50 (13)
C6—C5—H5	120.0	C15—C14—C13	121.51 (14)
C1—C6—C5	120.40 (19)	C15—C14—H14	119.2
C1—C6—H6	119.8	C13—C14—H14	119.2
C5—C6—H6	119.8	C14—C15—C16	119.66 (14)
N1—C7—C1	113.61 (12)	C14—C15—H15	120.2
N1—C7—H7A	108.8	C16—C15—H15	120.2
C1—C7—H7A	108.8	C17—C16—C15	119.98 (14)
N1—C7—H7B	108.8	C17—C16—H16	120.0
C1—C7—H7B	108.8	C15—C16—H16	120.0
H7A—C7—H7B	107.7	C16—C17—C18	120.62 (14)
N1—C8—N2	115.89 (12)	C16—C17—H17	119.7
N1—C8—S1	125.21 (10)	C18—C17—H17	119.7
N2—C8—S1	118.90 (11)	C13—C18—C17	119.28 (13)
N3—C9—C18	116.15 (12)	C13—C18—C9	120.18 (12)
N3—C9—C10	124.61 (12)	C17—C18—C9	120.54 (12)
C18—C9—C10	119.23 (12)		
C8—N2—N3—C9	176.26 (12)	C9—C10—C11—C12	−52.80 (18)
C6—C1—C2—C3	0.0 (3)	C10—C11—C12—C13	55.55 (19)
C7—C1—C2—C3	−177.27 (16)	C11—C12—C13—C14	149.52 (14)
C1—C2—C3—C4	−0.4 (3)	C11—C12—C13—C18	−30.9 (2)
C2—C3—C4—C5	0.6 (3)	C18—C13—C14—C15	−0.3 (2)
C3—C4—C5—C6	−0.4 (3)	C12—C13—C14—C15	179.30 (15)
C2—C1—C6—C5	0.2 (3)	C13—C14—C15—C16	−1.5 (2)
C7—C1—C6—C5	177.49 (16)	C14—C15—C16—C17	1.3 (2)
C4—C5—C6—C1	0.0 (3)	C15—C16—C17—C18	0.8 (2)
C8—N1—C7—C1	−87.70 (17)	C14—C13—C18—C17	2.3 (2)
C6—C1—C7—N1	136.73 (15)	C12—C13—C18—C17	−177.28 (13)
C2—C1—C7—N1	−46.0 (2)	C14—C13—C18—C9	−178.21 (13)
C7—N1—C8—N2	−175.61 (13)	C12—C13—C18—C9	2.2 (2)
C7—N1—C8—S1	4.1 (2)	C16—C17—C18—C13	−2.6 (2)
N3—N2—C8—N1	−9.87 (18)	C16—C17—C18—C9	177.96 (13)
N3—N2—C8—S1	170.43 (10)	N3—C9—C18—C13	−178.67 (13)
N2—N3—C9—C18	174.84 (11)	C10—C9—C18—C13	0.82 (19)
N2—N3—C9—C10	−4.6 (2)	N3—C9—C18—C17	0.80 (19)
N3—C9—C10—C11	−155.52 (14)	C10—C9—C18—C17	−179.72 (13)
C18—C9—C10—C11	25.04 (19)		

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1A···N3	0.91	2.20	2.6219 (16)	108

| N1—H1A···S1^{i} | 0.91 | 2.85 | 3.5790 (13) | 138 |

Symmetry code: (i) −x+1/2, y−1/2, z.